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54 **Imaging media capable of displaying sharp indicia.**

57 Improved sheet material of the type wherein a base sheet is coated on at least one face with a layer comprising a film-forming polymer containing inter-connected microvoids, the layer being locally transparentizable to display contrasting indicia when there is applied to its exposed surface a transparent, colorless liquid which has a refractive index similar to that of the solid constituents of the layer. A substance which jellifies the liquid is incorporated in the layer, whereby the indicia maintain substantially the same dimensions throughout the time that the layer is locally transparentized.

DescriptionIMAGING MEDIA CAPABLE OF DISPLAYING SHARP INDICIABackground of the Invention

5 This invention relates to sheet material,
especially a base sheet obscured by an opaque but
transparentizable microporous, diffusely light-reflective
layer.

10 For centuries paper has been one of the most
versatile substances made by man. Formed from commonly
available cellulosic materials, it can be made stiff or
flexible, rough or smooth, thick or thin, and provided
with any desired color. After it has served its intended
purpose, it can often be repulped and used again. In
recent years, however, the demands for paper have
15 increased to the extent that it has finally been
recognized that the sources of cellulosic raw materials
are not inexhaustible. Further, the energy required to
manufacture paper is a significant consideration in a
world becoming increasingly aware that supplies of energy
20 are also finite. It has also become recognized that,
where paper is used as a carrier for indicia, it can
generally be used only once, it being impossible or
impractical to remove indicia which are no longer needed
or desired. There has thus arisen a desire for a
25 substitute for paper, especially one which can be
repeatedly and easily reused; even a substitute which was
more expensive to manufacture would be less expensive in
the long run if it could be reused a sufficient number of
times.

30 Several U.S. Patents (e.g., Nos. 2,299,991,
3,031,328 and 3,508,344) disclose composite sheet material
wherein a light-colored opaque blushed lacquer layer is
coated over a base sheet which is either dark-colored or
imprinted with dark-colored indicia. The opacity and
35 light color of the blushed lacquer coating are due to the

inclusion of numerous microvoids; the local application of (1) heat or pressure (either of which irreversibly collapses the microvoids) or (2) a non-solvent liquid having substantially the same refractive index as the lacquer (which fills the microvoids), causes the coating to become selectively transparent and the underlying dark backing to become visible. A liquid employed to impart transparency to the opaque microporous layer can subsequently be evaporated to restore the original appearance.

U.S. Patent No. 2,854,350 describes structures which are functionally similar to those just described, except that the blushed lacquer coatings are replaced by a microporous layer of finely divided calcium carbonate in an organic binder. Transparency is imparted by locally applying pressure or treating selected areas with a wax, oil or grease having a refractive index similar to that of the calcium carbonate; see U.S. Patent 2,854,350. Other pigments may be incorporated in a microporous highly plasticized resin binder; see U.S. Patent 3,247,006.

It is sometimes desirable to have microvoid-containing sheet material which can be transparentized by applying a liquid, but which cannot readily be transparentized by the application of heat or pressure. In such circumstances, a microvoid-containing layer of the type described in copending U.S. patent application Ser. No. 94,645 is preferred. This earlier U.S. application discloses a structure in which the microvoid-containing layer consists essentially of particles held in

pseudo-sintered juxtaposition by a thermoset binder and has a cohesion value of at least 400 grams force*.

To a greater or lesser extent, each of the products described in the preceding paragraphs suffers from the disadvantage that localized application of a transparentizing liquid results in an image which does not maintain its original sharp outlines with the passage of time. In other words, there is a tendency for the marking liquid not only to penetrate the microvoids perpendicularly to the surface but also to wick laterally. As a result, the longer the transparentizing liquid remains in contact with the microvoid-containing layer, the less distinct the original image. In some instances, it becomes difficult to distinguish similar numerals (e.g., 6, 8 and 9) or letters (e.g., l, i and t) from each other. Prior to the present invention, no way of combatting this problem was known.

Brief Description

The present invention provides an improved sheet material of the type wherein a base sheet is coated on at least one face with an opaque white or

* The cohesion value is determined by knife-coating a dispersion of a putative composition on a cleaned gray cold rolled steel panel, drying and curing as appropriate for the composition, to provide a coating 50-60 micrometers thick. Using a "Balance Beam Scrape-Adhesion and Mar Tester", sold by Gardner Laboratories, Inc., Bethesda, Maryland, a sapphire-tipped stylus is lowered into contact with the test panel and held in fixed position while a ball bearing-supported platform moves the panel. The minimum grams-force required to form a 50-micrometer deep scratch in the coating in a single pass is determined at a magnification of 40X and reported as cohesive value.

pastel layer comprising a film-forming polymer containing interconnected microvoids. The sheet material can be made to display contrasting indicia by applying to the exposed surface a desired pattern of a colorless, transparent marking liquid which has a refractive index approximating that of the constituents of the layer, thereby rendering the marked portions transparent. The marking liquid for these conventional products is selected on the basis that it is unreactive with the constituents of the layer and has a volatility suited to the desired end use.

The improvement provided by the present invention comprises, incorporated into the microvoid-containing layer, a substance (typically an organic polymer) which significantly increases the viscosity of the marking liquid. (For convenience, this thickening action is referred to herein as "jellification.") When applied to the surface of the microvoid layer the marking liquid penetrates into the layer and then thickens, or gels, when it contacts the jellifying substance, its viscosity becoming so high that it retards passage of the marking liquid through the microvoids, i.e., it inhibits lateral wicking. While it might be anticipated that vertical penetration of the microvoids would likewise be inhibited, so that transparentization would not occur, such is surprisingly not the case. As a result, indicia can be readily generated but nevertheless maintain substantially the same dimensions throughout the time that the layer is locally transparentized.

The jellifying substance employed in the practice of the invention can be either natural or synthetic but is characterized by the property of jellifying the marking liquid employed. A simple test for determining whether a given substance is suitable for use with a specific marking liquid involves placing 10 grams of the putative jellifying substance and 90 grams of marking liquid in a 500-cc glass jar, tightly capping the

jar, and tumbling it for 24 hours. If the substance and marking liquid have formed a homogeneous gelatinous ball with no liquid remaining, the combination is deemed suitable for use in practice of the invention.

5 To determine whether a given jellifying substance-marking liquid combination will probably be effective in practicing the invention, it has been found useful to consider their respective solubility parameters, δ (measured in hildebrands).^{*} Generally speaking, if the
10 solubility parameters of the jellifying substance and the marking liquid differ by approximately 2 hildebrands, the combination is likely to be effective in practice of the present invention; smaller differences tend to result in lower solution viscosities, and greater differences tend
15 to result in insufficient gelling to inhibit lateral wicking.

Presently Preferred Embodiments

Understanding of the invention will be further enhanced by referring to the following illustrative but
20 non-limitative examples, in which all parts, ratios and percentages are by weight unless otherwise noted.

EXAMPLE 1

Following the general procedure described in U.S. Patent No. 2,854,350, a control was prepared by
25 placing 39.6 grams of water, 0.4 gram sodium alginate and

* Detailed discussions of solubility parameters, their measurement and calculation are found in (1) Encyclopedia of Polymer Science and Technology, Interscience, New York (1965), Vol. 3, page 833 et seq., and (2) Encyclopedia of
30 Chemical Technology, Interscience, New York, (1971), Supplement Vol., page 889 et seq.)

5 grams precipitated calcium carbonate in a 4-ounce (approximately 125-cc) jar and ball milling for several days. The resulting composition was knife-coated, at a thickness of approximately 200 micrometers, onto the surface of black 60-micrometer greaseproof paper and allowed to dry at room temperature overnight; the dried coating was approximately 25 micrometers thick.

In accordance with the invention, a sample was prepared which was identical to the control except that 0.2 gram cellulose acetate butyrate (CB 500-1 available from Eastman Chemical Products, Inc.) was included in the composition. Using a felt-tipped pen filled with diethylphthalate, a 1.5-mm line was stroked on the surface of each of the two products. The table below shows the width of the stroked lines, measured after various time intervals.

		<u>Image width, mm</u>	
		<u>Control</u>	<u>Example 1</u>
20	Time, min.		
	0	1.5	1.5
	2	1.9	1.5
	4	2.2	1.5
	8	3.2	1.5
	12	3.2	1.5
	16	3.2	1.5

It will be observed that the control product suffered from lateral wicking which more than doubled the width of the initial line; in contrast, the EXAMPLE 1 product maintained a substantially constant line width, the diethylphthalate gelling when it contacted the cellulose acetate butyrate, and thus preventing lateral wicking.

EXAMPLE 2

Following the general procedure described in U.S. Patent No. 3,508,344, a control was prepared by placing 75.1 grams acetone, 6.6 grams polymethyl

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methacrylate ("Elvacite" 2041 available from E. I. duPont de Nemours & Company), 1.6 grams diethylphthalate, 1.1 grams polyacrylate ("Rhoplex" B-15, available from Rohm & Haas Company) and 11.0 grams of water in an 8-ounce (approximately 250-cc) jar and ball milling overnight. The resulting composition was knife-coated onto the black greaseproof paper used in EXAMPLE 1 and allowed to dry at room temperature to leave a coating approximately 50 micrometers thick.

In accordance with the present invention, a sample was prepared which was identical to the control except that 1.0 gram methyl cellulose ("Methocel" MC 4000 cp, available from the Dow Chemical Company) was incorporated in the composition prior to ball milling. Using a felt-tipped pen filled with 1,2-propanediol, 0.9-mm lines were drawn on the surface of each product and measured after various time intervals. Results are shown below:

Time, min.	Image width, mm	
	Control	Example 2
0	0.9	0.9
1	1.2	1.0
5	1.2	1.0
10	1.2	0.9

While not so striking as the results shown in EXAMPLE 1, the results of this example nevertheless show that the construction of the present invention, where the 1,2-propanediol was jellified by the methyl cellulose and prevented lateral wicking, maintained a significantly sharper image than did the control.

EXAMPLE 3

Following the procedure described in more detail in the aforementioned U.S. Patent Application Ser. No. 94,645, a control coating composition was prepared by

mixing 16 parts xylene, 16 parts diisobutyl ketone, 8 parts heptane, 11.39 parts thermosetting acrylic resin (G-CURE 868 RX-60, available from Henkel Corporation), 0.2 part di(dioctyl pyrophosphato) ethylene titanate (KR-2385, available from Kenrich Petrochemicals Inc.), 100 parts calcium carbonate having a 0.5-15 micrometer particle size, and 2.17 parts "Desmodur" N-75 (75% solid solution in 1:1 xylene:2-ethoxyethyl acetate of the high molecular weight biuret of 1,6-hexamethylene diisocyanate, having an equivalent weight of 195, available from Mobay Chemical Corporation). The composition was coated on the black greaseproof paper of EXAMPLE 1 and cured 30 minutes at approximately 90°C to leave a dried coating approximately 25 micrometers thick.

In accordance with the present invention a product was prepared which was identical to that of the control except that 4 parts of carboxypolymethylene ("Carbopol" 941, available from B. F. Goodrich Chemical Co.) was included in the coating composition. A felt-tipped pen filled with tetraethylene glycol was then used to mark a 1.0-mm line on the surface of each sheet. After 30 minutes, the line width on the control product was 1.9 mm, while the line width on the product of this EXAMPLE 3 was only 1.1 mm.

25 EXAMPLE 4

A microvoid-forming control coating composition was prepared by mixing together 20 grams xylene, 7.6 grams methylisobutyl ketone, 7.6 grams ethylene glycol monoethylether acetate, 0.2 gram di(dioctylpyrophosphato) ethylene titanate, 13.0 grams thermosetting acrylic resin, 100.0 grams 0.5-15 micrometer calcium carbonate, and 2.5 grams "Desmodur" N-75. The composition was coated on the black greaseproof paper of EXAMPLE 1 and cured by heating 45 minutes at approximately 90°C.

35 In accordance with the present invention, a coating composition was prepared which was identical to

that of the control except that 2 grams of "Kraton" 1107 (isoprene:styrene:isoprene block copolymer, having a 25% toluene solution viscosity at 23°C of 1.6 Pa.s) was included in the coating composition. A felt-tipped pen
5 containing a mixture of saturated branched chain hydrocarbons, having a distillation range of 244°-286°C, was then used to mark lines on both the control and EXAMPLE 4 products, the width being measured after various time intervals. Results are tabulated below:

10	Time, min.	Image width, mm	
		Control	Example 4
	0	1.0	1.0
	3	1.2	1.0
	5	1.2	1.0
15	11	1.2	1.0
	30	1.3	1.0
	50	1.2	0.9

EXAMPLE 5

A microvoid-forming control composition was
20 prepared by mixing together 8.0 parts of an oil-free thermosetting alkyd resin having an equivalent weight of 400 ("Aroplaz" 6022, available from Ashland Chemical Company), 56.0 parts ethylene glycol monoethylether acetate, 100.0 parts isopropyl triisostearoyl titanate
25 pigment, and 3.8 parts "Desmodur"N-75. The composition was coated on 64-micrometer dark brown greaseproof paper and cured 6 minutes at 230°C to leave a dry-coating approximately 25 micrometers thick.

In accordance with the present invention, a
30 product was prepared which was identical to the control except that the coating composition also contained 2.0 parts cellulose acetate butyrate and 2.0 parts carboxypolymethylene. A felt-tipped pen containing dibutyl phthalate was then used to mark a 1.5-mm line on
35 each coated product and the width measured after various

time intervals. Results are tabulated below:

	<u>Time, min.</u>	<u>Image width, mm</u>	
		<u>Control</u>	<u>Example 5</u>
5	0	1.5	1.5
	0.5	1.8	1.5
	1.0	2.0	1.5
	2.0	2.4	1.6
	1440	4.1	1.6

Similar results were obtained using propylene glycol as a marking liquid, except that the marks on both products disappeared by evaporation of the liquid before 1,440 minutes had elapsed.

The preceding examples illustrate the incorporation of a jellifying substance directly into a microvoid-forming coating composition. It is also feasible to post-treat a microvoid coating to introduce a jellifying material into the microvoids, as the following examples illustrate.

EXAMPLE 6

A sample was prepared in substantially the same manner as the control sheet of EXAMPLE 1. In accordance with the present invention, a portion of this control was coated with a 5% solution of cellulose acetate butyrate in methylisobutyl ketone and all excess wiped from the surface, after which the solvent was allowed to evaporate at room temperature. A felt-tipped pen containing diethylphthalate was stroked across the treated and untreated portions to produce a line which was initially 2 mm wide. After 30 minutes, the line on the untreated portion was 5 mm wide, while the line on the treated portion was still only 2 mm wide. In a similar manner, a felt-tipped pen containing dioctylphthalate was stroked across the treated and untreated portions to produce a line which was initially 1 mm wide. After 13 days, the

line on the untreated portion had become so wide that it could no longer be distinguished, while the line on the treated portion was still only 1 mm wide and easily recognized.

5 EXAMPLE 7

A control sheet substantially identical to the control sheet of EXAMPLE 2, was prepared. In accordance with the present invention, a portion of this control was then treated with a 2% aqueous solution of methyl
10 cellulose ("Methocel" MC Standard 4000 cps), the excess solution wiped from the surface, and the sample allowed to dry at room temperature. A felt-tipped pen containing 1,2-propanediol was then used to mark a 0.7-mm line on the surface of both the control and the treated material of
15 this EXAMPLE 7, line width on each being measured after various time intervals. Results are tabulated below:

	<u>Time, min.</u>	<u>Image width, mm</u>	
		<u>Control</u>	<u>Example 7</u>
	0	0.7	0.7
20	1	1.4	0.7
	2	1.6	0.7
	10	1.9	0.7
	20	2.0	0.8

EXAMPLE 8

25 Following the procedure described in more detail in copending U.S. Patent Application Ser. No. 94,465, a control sheet was prepared by mixing 17.5 parts xylene, 17.5 parts methyl ethyl ketone, 100 parts dry ground calcium carbonate having a 0.5-15 micrometer particle
30 size, and 13 parts thermosetting acrylic resin and ball milling the mixture for 2 days. Then 2.5 parts of the high molecular weight biuret of 1,6-hexamethylene diisocyanate was added and ball milling continued for an additional 4 hours. The dispersion was knife-coated onto

the black greaseproof paper of EXAMPLE 1, dried 1 minute at 90°C and cured 45 minutes at the same temperature to leave a dried coating about 50 micrometers thick. To a portion of the control sheet coated surface there was
5 applied a 5% methyl ethyl ketone solution of cellulose acetate butyrate (Eastman CAB 500-1) and, after the solution had penetrated the surface, the excess was wiped off and the sample allowed to dry at room temperature. A felt-tipped pen containing diethylphthalate was then used
10 to mark a 1.2-mm line on the surface of both the control and the treated material of this EXAMPLE 8, line width on each being measured after various time intervals. Results are tabulated below:

15	Time, min.	Image width, mm	
		Control	Example 8
	0	1.2	1.2
	3	1.3	1.2
	36	1.5	1.2

As previously indicated, maintaining a sharp,
20 distinct image is highly desirable when the applied indicia are to be read visually. It is perhaps even more critical, however, that the indicia maintain their original dimensions when they are to be scanned by optical character recognition (OCR) or optical mark reading (OMR)
25 devices.

Whether incorporated in the microvoid-forming coating composition or subsequently applied to a microvoid-containing coating, it is preferred that the jellifying substance occupy a minimum of the void-volume.
30 The incorporation of an excessive amount of jellifying substance will reduce the size of the voids to the point where they are unable to scatter white light effectively. While the jellifying substance can occupy up to 50% of the void volume, it is preferred that it occupy 1-10%.

As a general guide to the selection of jellifying agents and marking liquids, which work effectively together, attention is directed to the following illustrative table:

5	<u>Marking Liquid</u>	<u>Jellifying Agent</u>
	Diethyl phthalate	Cellulose acetate butyrate
	Dibutyl phthalate	DO
	Dioctyl phthalate	DO
	Triacetin	DO
10	Esters in general	DO
	Isopropanol	Carboxypolymethylene
	Water	DO
	Ethyleneglycol	DO
	Tetraethylene glycol	DO
15	Glycols and alcohols in general	DO
	Dodecane	Styrene:isoprene:styrene block copolymer
	Undecane	DO
20	Octane	DO
	Branched and straight chain saturated hydrocarbons in general	DO

In most instances, a marking liquid having a desired
25 volatility will be chosen, after which an appropriate
jellifying agent will be selected.

The invention having been described and
exemplified as above, it will be recognized that numerous
variations of coating compositions, marking liquids, etc.
30 are within the ordinary skill of the art.

The trade names "Aroplaz", "Carbopol",
"Desmodur", "Elvacite", "Methocel", and "Rhoplex" are believed
to be registered trademarks.

CLAIMS:

1. Sheet material of the type wherein a base sheet is coated on at least one face with a layer comprising a film-forming polymer containing interconnected microvoids, said layer being locally transparentizable to display contrasting indicia when there is applied to its exposed surface a transparent, colorless liquid which has a refractive index similar to that of the solid constituents of the layer, characterized in that
- there is incorporated in said layer a substance which jellifies the liquid,
- whereby the indicia maintain substantially the same dimensions throughout the time that the layer is locally transparentized.
2. The invention of claim 1 further characterized in that the layer consists essentially of particles held in pseudo-sintered juxtaposition by a thermoset binder.
3. The invention of claim 1 or 2 further characterized in that the jellifying substance comprises cellulose acetate butyrate.
4. The invention of claim 1 or 2 further characterized in that the jellifying substance comprises methyl cellulose.
5. The invention of claim 1 or 2 further characterized in that the jellifying substance comprises carboxypolymethylene.
6. The invention of claim 1 or 2 further characterized in that the jellifying substance comprises a styrene:isoprene:styrene block copolymer.

7. Sheet material according to any preceding claim further characterized in that the sheet material comprises in combination:

- a. a self-supporting base sheet,
- 5 b. bonded over at least one side of the base sheet, a reflective opaque white to pastel layer having a cohesive value of at least 400 grams and consisting essentially of particles held in pseudo-sintered
- 10 juxtaposition by a thermoset binder so that interconnected microvoids are present throughout the layer, and
- c. a jellifying substance other than the binder partially filling said microvoids,
- 15 whereby, when there are applied to the exposed surface of said layer desired indicia of a marking liquid which has a refractive index approximately that of the particles and which is jellified by the jellifying substance, the liquid not only penetrates the microvoids and is jellified by the
- 20 jellifying substance, thereby reducing the reflectivity of the layer in the vicinity of the liquid-penetrated microvoids to impart transparency and maintaining substantially constant dimensions of the indicia throughout the time that the layer is locally
- 25 transparentized, but also is inhibited from lateral wicking.

8. A method of making the sheet material of any preceding claim characterized by incorporating, in an opaque microvoid-containing layer which is locally

30 transparentizable when contacted with a liquid having a refractive index similar to that of the solid constituents of said layer, a substance which jellifies said liquid.

9. The method of claim 8 further characterized in that, prior to formation of said layer, the jellifying substance is incorporated in the composition which is used to form the layer.

- 5 10. The method of claim 8 further characterized in that, after said layer has been formed, the jellifying substance is incorporated in said layer.