

(19)



Europäisches Patentamt  
European Patent Office  
Office européen des brevets

(11) Publication number:

**0 073 615**  
**A2**

(12)

# EUROPEAN PATENT APPLICATION

(21) Application number: **82304401.1**(51) Int. Cl.<sup>3</sup>: **C 10 L 1/18, C 10 L 1/24**(22) Date of filing: **20.08.82**(30) Priority: **25.08.81 US 296016**(71) Applicant: **Exxon Research and Engineering Company,  
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Tower High Street, New Malden Surrey KT3 4DJ (GB)**(54) **Residual fuel oil compositions and the preparation and combustion thereof.**

(57) A residual fuel oil composition comprises a residual fuel oil, preferably at No. 6 fuel oil, having dissolved therein an effective total amount of at least one zirconium salt of (i) an alcohol, or (ii) a phenol, each having a C<sub>2</sub> to C<sub>24</sub> hydrocarbyl group, or (iii) of a hydrocarbyl sulfonic acid such that the salt has a mol. wt of 100 to 2500, or of any mixture thereof; said effective total amount of the salt(s) being sufficient to reduce the amount of particulate matter which will be formed when, in use, the residual fuel oil is combusted. Suitably 1 to 1000 ppm of zirconium salt(s), calculated as metallic zirconium, are present. Preferred zirconium salts are zirconium propoxide or dodecyl benzene sulfonate. The composition is suitable for combustion in industrial boilers and refinery furnaces, preferably being fed in air or steam-atomized form.

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1           This invention relates to residual fuel oil compositions and the preparation and combustion thereof, which fuel oil compositions contain certain zirconium salts to reduce the amount of particulate matter formed during combustion.

5           Residual fuel oils, including Grades Nos. 4, 5 and 6 (ASTM D-396), are widely used in a variety of industrial heating and steam boiler applications. A particularly desired fuel oil is No. 6, which is extensively used by utility and power companies.

10           State, federal, EPA, and other, emission standards currently limit the use of residual fuels which produce excessive amounts of particulate emission during combustion and thus are not in compliance with standards.

          However, the situation is relatively complicated, since country-to-country, or state-to-state, emission standards tend to be different and compliance by a residual fuel oil in one location may not necessarily be achieved in another. Further, since standards are continuously subject to change, a residual fuel oil currently in compliance with emission standards may not be in compliance in the near future in the same location and under the same end-use conditions.

20           Residual fuel oils which tend to produce excessive amounts of particulate emission generally have one or more characteristics associated with them: a sulfur content above about 1 percent; a Conradson Carbon Residue (ASTM D-189, also termed "Con Carbon" in the art) above about 7 percent; or a high asphaltene content. Residual fuel oils yielding particulate emissions that surpass the existing standards can't be directly used, but in some cases can be blended in admixture with fuels that do meet existing standards, which are generally low in sulfur and/or low in "Con Carbon" and asphaltene content. This situation has resulted in an overall increased demand for fuel oils which meet emission standards despite their diminishing supply and attendant increase in cost.

          What is desired is a process for increasing the

1 utility of these high emission yielding residual fuel oils  
2 for industrial heating purposes in a manner that results  
3 in acceptable particulate emissions, despite a high sulfur  
4 content, a high Con Carbon residue and/or high asphaltene  
5 content.

6 - In the area of related problems, it is known in  
7 the art that the use of specific additives in certain  
8 hydrocarbon fuels, can reduce smoke or soot upon combus-  
9 tion, in certain instances. It is also known to use  
10 specific additives in fuels to inhibit corrosion, inhibit  
11 slag\_formation in boilers and to reduce the deleterious  
12 effect of vanadium present in such fuels.

13 However, it has not been shown to use selected  
14 additives to reduce particulate emission during combustion  
15 of residual fuel oil, and particularly No. 6 fuel oil.

16 Summary of Invention

17 It has unexpectedly been found that by adding a  
18 selected oil soluble zirconium salt of an alcohol/phenol  
19 or sulfonate to a residual fuel oil, and particularly No.  
20 6 fuel oil the amount of particulate matter formed during  
21 combustion can be significantly reduced in amounts of 10  
22 to 25 percent or greater.

23 In accordance with this invention, there is  
24 provided a process for reducing the amount of particulate  
25 matter formed during the combustion of a residual fuel oil  
26 comprising the steps of:

27 (a) dissolving in said fuel oil an effective  
28 trace amount of an additive selected from the group  
29 consisting of:

30 i) an oil soluble zirconium salt of an  
31 alcohol or phenol having the formula:

32 
$$\text{ROH}$$

33 where R is a hydrocarbyl group of 2-24 carbon atoms; or

34 ii) an oil soluble zirconium salt of  
35 sulfonic acids having the formula:

36 
$$\text{RSO}_3\text{H}$$

1 where R is an alkyl cycloalkyl, aryl, alkaryl or aralkyl  
2 group and said salt has a molecular weight of about 100 to  
3 about 2500;

4 said amount being effective in reducing the amount of  
5 particulate matter formed during combustion as compared to  
6 said process conducted in the absence of said zirconium  
7 salt; and

8 (b) combusting said residual fuel oil.

9 DESCRIPTION OF THE INVENTION AND PREFERRED EMBODIMENTS

10 The novelty of the present invention resides in  
11 the discovery that zirconium salts of certain alcohols/  
12 phenols or sulfonic acids exert a beneficial effect on  
13 residual fuel oil, particularly No. 6 fuel oil, in  
14 reducing the amount of particulate matter formed during  
15 combustion. The term "residual fuel oil" as used herein,  
16 is well-known and as described hereinabove, and includes  
17 Grades No. 4, No. 5 and No. 6 residual fuel oils, meeting  
18 the specifications of ASTM D-396. Particularly preferred  
19 is No. 6 fuel oil.

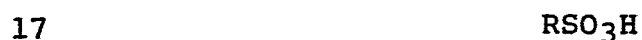
20 The reason these particular zirconium additives  
21 exhibit this surprising effect is not clearly understood.  
22 It may be that the subject compounds promote and acti-  
23 vate the complete oxidation of hydrocarbon and sulfur-  
24 containing constituents in the fuel to volatile or gaseous  
25 compounds during combustion, in a highly specific manner.

26 The subject zirconium salts or compounds, also  
27 termed "additives" herein, operative in the instant in-  
28 vention, comprise oil soluble zirconium salts of an  
29 alcohol/phenol or sulfonate. The zirconium salt of  
30 selected alcohols or phenols will be a zirconium salt of  
31 an alcohol or phenol having the formula:

32 ROH

33 where R is a hydrocarbyl group of 2 to 24 carbon atoms.  
34 More particularly R is a branched or unbranched, hydro-  
35 carbyl group preferably having 2 to 13 carbon atoms.  
36 Preferred compounds are those where R is a saturated or

14           The zirconium salt of sulfonic acids useful in  
15 this invention are the zirconium salts of sulfonic acids  
16 having the formula:



24 The sulfonic acids are characterized by the  
25 presence of the sulfo group  $\text{-SO}_3\text{H}$  (or  $\text{-SO}_2\text{OH}$ ) and can  
26 be considered derivatives of sulfuric acid with one of the  
27 hydroxyl groups replaced by an organic radical. Compounds  
28 of this type are generally obtained by the treatment of  
29 petroleum fractions (petroleum sulfonates). Because of  
30 the varying natures of crude oils and the particular oil  
31 fraction used, sulfonates generally constitute a complex  
32 mixture and it is best to define them in a general manner  
33 giving the molecular weight as defined above. Partic-  
34 ularly preferred sulfonates are those having an alkaryl  
35 group, i.g. alkylated benzene or alkylated naphtalene.  
36 Illustrative examples of sulfonic acids useful

1 in this invention are: dioctyl benzene sulfonic acid,  
2 dodecyl benzene sulfonic acid, didodecyl benzene sulfonic  
3 acid, dinonyl naphthalene sulfonic acid, dilauryl benzene  
4 sulfonic acid, lauryl cetyl benzene sulfonic acid, poly-  
5 olefin alkylated benzene sulfonic acid such as poly-  
6 butylene and polypropylene, etc. Further details regard-  
7 ing sulfonic acids may be found in Kirk-Othmer, "Encyclo-  
8 pedia of Chemical Technology", Second Edition, 1969,  
9 Vol. 19, pp 311 to 319 and in "Petroleum Sulphonates" by  
10 R. Leslie in Manufacturing Chemist, October 1950 (XX1, 10)  
11 pp 417 to 422.

12           Methods of preparing the subject zirconium salts  
13 described above are well known in the art and generally  
14 said salts are commercially available.

15           The zirconium additive is incorporated into the  
16 residual fuel oil by dissolving therein. This is accom-  
17 plished by conventional methods as by heating, stirring  
18 and the like.

19           The amount of zirconium additive to be used is  
20 an "effective trace amount" that will reduce the amount of  
21 particulate matter formed during combustion of the resi-  
22 dual fuel oil as compared to the combustion of said fuel  
23 oil in the absence of said additive. By the term "effec-  
24 tive trace amount" is quantitatively <sup>generally</sup> meant an amount of  
25 about 1 to 1000 ppm by weight and preferably 10-500 ppm  
26 by weight, zirconium additive, taken as metallic zirconium,  
27 in said fuel oil, and particularly preferred about 50 to  
28 150 ppm by weight zirconium additive, taken as metallic  
29 zirconium, in said fuel oil. However, lower and higher  
30 amounts than the 1-1000 ppm range can also be present  
31 provided an effective trace amount, as defined herein, is  
32 present in the residual fuel oil.

33           By the term "reduce the amount of particulate  
34 matter formed during combustion", as used herein, is <sup>normally</sup> meant  
35 that at least about a five percent reduction in formed  
36 particulate matter, and preferably from about 10 to 25  
37 percent and greater, reduction in formed particulate  
38 matter is achieved as compared to the combustion of the

1 residual fuel oil in the absence of the subject zirconium  
2 additive.

3           In the process, the fuel oil containing said  
4 additive is generally mixed with oxygen, usually in the  
5 form of air, to form a fuel/air mixture prior to combus-  
6 tion. Generally, the amount of air utilized is an excess  
7 over the stoichiometric amount to completely combust the  
8 fuel oil to carbon dioxide and water. The reason for  
9 utilizing this excess is that complete mixing does not  
10 always occur between the fuel oil and the air, and that  
11 also a slight excess of air is desirable since it serves  
12 to reduce the tendency of soot and smoke formation during  
13 combustion. Generally, the excess of air used is about 2  
14 to 35 percent (0.4 to 7 percent based on oxygen) over  
15 the stoichiometric amount depending upon the actual  
16 end-use conditions which may vary considerably from one  
17 type of industrial boiler to the next. One disadvantage  
18 in using a large excess of air is that a greater amount of  
19 heat is lost through entrainment that would otherwise be  
20 utilized for direct heating purposes. We have found  
21 that by use of the subject zirconium additives, less  
22 excess air is required to reduce smoke and soot formation  
23 and thus the heating efficiency of the residual fuel oil  
24 is greater, as well as resulting in a reduction of par-  
25 ticulate emission.

26           The above-described step of mixing fuel oil and  
27 air is conventional and is usually accomplished for  
28 example, by steam or air atomization to produce a fine  
29 spray which is then combusted to maintain and support a  
30 flame. The combustion is controlled and conducted  
31 at a particular "firing rate" which is usually expressed  
32 as lbs/minute of fuel oil combusted.

33           The combustion of residual fuel oil is usually  
34 carried out in conventional industrial boilers, utility  
35 boilers, refinery furnaces and the like.

36           The amount of particulate matter formed during  
37 combustion of residual fuel oil will vary over a broad  
38 range and is dependent upon a number of factors such as

1 type of boiler, boiler size, number and type of burners,  
2 source of the residual fuel oil used, amount of excess air  
3 or oxygen, firing rate and the like. Generally, the  
4 amount of particulate matter formed will be in the range  
5 of about 0.01 to 1.0 weight percent of the fuel oil used  
6 and higher. One weight percent corresponds to one gram  
7 particulate matter formed from the combustion of 100 grams  
8 of fuel oil. The amount of particulate matter formed,  
9 herein termed "total particulate matter," is actually  
10 the sum of two separate measurements; "tube-deposits," the  
11 amount of particulate matter deposited inside of the  
12 boiler, and two, "filtered stack particulate," which is  
13 the amount of particulate matter formed which escapes the  
14 boiler and is actually emitted out of the stack into  
15 the air. EPA measurements are generally only concerned  
16 with filtered stack particulate which is directly released  
17 into the air environment and contributes to a decrease in  
18 air quality. However, "tube deposits" lead to corrosion  
19 of the equipment, frequent "clean-outs" and add to the  
20 total operating costs. Furthermore, as tube deposits  
21 collect on the inside of the apparatus, a critical crust  
22 thickness is reached and further tube deposits are then  
23 entrained in stack particulate, which significantly  
24 increases the amount of particulate emission. Thus, in  
25 order to fully assess the overall operating advantages of  
26 a particular residual fuel oil in a boiler operation, the  
27 amount of tube deposits should also be considered, as well  
28 as total stack particulate for compliance with emission  
29 standards.

30           The amount of allowed stack particulate will  
31 vary from state to state and is also subject to a minimum  
32 amount allowed under Federal EPA standards. For example,  
33 in Florida, the currently allowable limit for existing  
34 power plants is 0.10 lbs. particulate emission per million  
35 BTU, which is equivalent to about 0.185 weight percent of  
36 particulate stack emission per weight of combusted fuel  
37 oil. Since the allowable emission standards will vary  
38 from jurisdiction to jurisdiction, differing amounts of



1 the subject zirconium additive will be necessary to  
2 produce a residual fuel oil composition in compliance with  
3 those standards.

4 Measurement of the amount of "stack particulate  
5 matter" is conducted by EPA Method #5 Stack Sampling  
6 System, "Determination of Particulate Emissions from  
7 Stationary Sources" and is described in the Federal  
8 Register.

9 The particulate stack emissions are generally  
10 comprised of particulate carbon, sulfur-containing hydro-  
11 carbons, inorganic sulfates and the like.

12 The following examples are illustrative of this  
13 invention and should not be construed as being limitations  
14 on the scope and spirit of such invention.

15 Example 1

16 Combustion runs were carried out in a 50 horse-  
17 power ABCO, 2-pass, water jacketed forced draft boiler  
18 with an air-atomizing burner and a nominal firing rate of  
19 1.2 lbs/min. of residual fuel oil. The boiler was modi-  
20 fied so that closure on each end could be opened easily  
21 for recovery of deposits laid down in the boiler. Two  
22 other modifications included installation of a second fuel  
23 system so the boiler could be heated to operating tempera-  
24 tures on No. 2 oil and then switched over to the test fuel  
25 without shutting down or upsetting the boiler operation  
26 unduly and installation of a two foot length of firebrick  
27 lining at the burner end of the firetube and a Cleaver-  
28 Brooks nozzle assembly in place of the Monarch nozzle.  
29 These modifications eliminated oil pooling and rapid  
30 carbon deposits on the firetube walls when residual fuel  
31 was fired. The first pass is a 49 cm (18.375 in.) diameter  
32 x 178 cm (5 ft. 10 in.) long fire tube; the second pass  
33 consists of 52 tubes each 6 cm (2.375 in.) diameter x 188  
34 cm (6 ft. 2 in.) long.

35 Atomization of the fuel was accomplished using a  
36 low pressure air-atomizing nozzle. Viscosity of the fuel  
37 oil at the nozzle was maintained at 30 centistokes by  
38 heating the oil to a predetermined temperature (about

1 105°C). Prior to contacting the burner gun, the atom-  
2 ized fuel oil was mixed with a measured amount of excess  
3 "secondary" air which was forced through a diffuser  
4 plate to insure efficient combustion. The secondary air  
5 was provided by a centrifugal blower mounted in the boiler  
6 head. The amount of secondary air was controlled by means  
7 of a damper which was regulated to keep the oxygen level  
8 in the atomized fuel at about 1.5% in excess (over that  
9 needed stoichiometrically to completely combust the fuel).

10 A run was started by firing the boiler and  
11 heating it to operating temperature for 55 minutes using  
12 No. 2 oil. The feed was then switched to test fuel and  
13 after allowing sufficient time for conditions to stabilize  
14 (about 25 minutes) samples of about 10 minutes duration  
15 were collected isokinetically from the stack on tared,  
16 Gelman, Type A (20.3 x 25.4 cm) fiber glass filters. The  
17 test fuel was a No. 6 fuel oil.

18 Total particulate matter formed was determined  
19 by adding the amount of Stack particulate measured iso-  
20 kinetically (EPA Method 5 Stack Sampling System) to the  
21 amount deposited in the tubes of the boiler i.e. "tube  
22 deposits".

23 The EPA Method 5 Stack Sampling System was  
24 conducted with a commercially available system for this  
25 purpose. This unit consisted of an 18-inch glass lined  
26 probe, a cyclone, a 125 mm glass fiber filter and four  
27 impingers. The first two impingers contained water, the  
28 third was empty and the last one contained silica gel.  
29 With the exception of the impingers, the entire sampling  
30 train was maintained at about 175°C to insure that the  
31 stack gases entering the sampling system were above  
32 the H<sub>2</sub>SO<sub>4</sub> dew point.

33 The deposits laid down in each of the 52 tubes  
34 is collected on a separate, tared 20.3 x 25.4 cm fiber-  
35 glass filter. Deposits are collected by positioning a  
36 specially-designed filter holder against the end of each  
37 tube in turn, pulling air through the tube and the filter  
38 using a high-volume vacuum pump and manually brushing the

1 tube from end-to-end ten times with a 2.50 inch diameter  
2 wire shank brush. The brush is mounted on a 8 ft. long,  
3 0.25 in. diam. SS rod driven by an electric drill. This  
4 method gives almost 100% recovery of the deposits laid  
5 down in the tubes. All the tubes are sampled because for  
6 a given run there are large differences in deposit weight  
7 from tube-to-tube in each row of tubes across the boiler  
8 and from top row to bottom row and there is no consistent  
9 ratio of the weight of deposit collected from a given tube  
10 from run-to-run.

11 The fuel oil used (Test Fuel) in the runs  
12 analyzed for the following constituents:

13	<u>Analysis of Test Fuel</u>	
14	Sulfur	2.0 wt%
15	ConCarbon <sup>a</sup>	14.8 wt%
16	Ash <sup>b</sup>	0.1 wt%
17	Vanadium	469 ppm
18	Nickel	70 ppm
19	Iron	2.9 ppm

20 <sup>a</sup>ASTM-D-189

21 <sup>b</sup>ASTM-D-482

22 The zirconium additive used in the run was  
23 zirconium propoxide an alcohol salt and was present in a  
24 concentration of 100 ppm taken as metallic zirconium.

25 For test fuel alone, the stack particulate was  
26 0.34 wt% on fuel, while the tube deposits was 0.20 wt% on  
27 fuel for a total test particulate wt% of 0.54. The sample  
28 of fuel containing the zirconium propoxide measured a  
29 stack particulate of 0.24 wt% on fuel and tube deposits of  
30 0.16 wt% on fuel for a total particulate wt% of 0.40. The  
31 improvement in using the zirconium additives was a reduc-  
32 tion in total particulates of 25.9%.

33 Example 2

34 Following the same general procedure and using  
35 the ABCO boiler described in Example 1, a sample run using  
36 100 ppm of a zirconium sulfonate additive, i.e. zirconium

1 salt of dodecyl benzene sulfonic acid, was made with the  
2 same No. 6 fuel oil as in said Example 1.

3           The results for the sample containing zirconium  
4 sulfonate were a stack particulate of 0.29 wt% on fuel and  
5 tube deposits of 0.18 wt% on fuel for a total particulate  
6 of 0.47 wt% on fuel. The improvement in using the zir-  
7 conium additive was a reduction in total particulate of  
8 13.0 %.

CLAIMS:

1. A residual fuel oil composition, characterised by a residual fuel, preferably a No.6 fuel oil, having *present* therein an effective total amount of at least one additive selected from (i) an oil soluble zirconium salt of an alcohol or phenol having the formula ROH, where R is a hydrocarbyl group of 2 to 24 carbon atoms, or (ii) an oil soluble zirconium salt of a sulfonic acid having the formula  $\text{RSO}_3\text{H}$ , where R is an alkyl, cycloalkyl, aryl, alkaryl or aralkyl group and said salt has a molecular weight of 100 to 2500; said effective total amount of the salt(s) being sufficient to reduce the amount of particulate matter which will be formed when, in use, the residual fuel oil is combusted.
2. A residual fuel oil composition as claimed in claim 1, characterised in that the zirconium salt or salts is or are present in a total amount of from 1 to 1000 ppm, preferably 50 to 150 ppm, by weight, calculated as metallic zirconium.
3. A composition as claimed in claim 1 or claim 2, wherein the R group in said alcohol salt is a saturated or unsaturated aliphatic group having 2 to 8 carbons.
4. A composition as claimed in claim 1 or claim 2, wherein the R group in said sulfonic acid is an alkaryl group and the salt has a molecular weight of 200 to 700.
5. A residual fuel oil composition as claimed in claim 1 or claim 2, characterised in that the zirconium salt or salts is or are selected from zirconium propoxide, zirconium dodecyl benzene sulfonate, or mixtures thereof.
6. A process of preparing a residual fuel oil, preferably a No.6 fuel oil, characterised by dissolving in said fuel oil an effective total amount of at least one additive selected from (i) an oil soluble zirconium salt of an alcohol or phenol having the formula ROH, where R is a hydrocarbyl group of 2 to 24 carbon atoms, or (ii) an oil soluble zirconium salt of a sulfonic acid having the formula  $\text{RSO}_3\text{H}$  where R is an alkyl, cycloalkyl, aryl, alkaryl or aralkyl group and said salt has a molecular weight of 100 to 2500, said effective total amount

of the salt(s) being sufficient to reduce the amount of particulate matter which will be formed when, in use, the residual fuel oil is combusted and said effective amount being preferably 1 to 1000 ppm, especially 50 to 150 ppm, by weight, calculated as metallic zirconium.

7. A composition as claimed in claim 6, wherein said fuel oil contains above 1 weight percent sulfur.

8. A process in which a residual fuel oil, preferably a No.6 fuel oil is combusted; characterised by

(a) dissolving in the fuel oil an amount defined in claim 1 or claim 2 of at least one zirconium salt defined in any one of claims 1 and 3 to 5, and

(b) combusting the resultant residual fuel oil composition.