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(54)

Cable with adhesively bonded sheath.

(57)

An electrical cable comprising a core having at least one inner conductor surrounded by a dielectric material, a metallic sheath surrounding said core, and an adhesive bonding said metallic sheath to said core, said adhesive comprising a water dispersible copolymer of ethylene and an ethylenically unsaturated carboxylic acid forming a thin film between said metallic sheath and said core.

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CABLE WITH ADHESIVELY BONDED SHEATH

Background of the Invention

This invention relates to cables of the type which have a core formed of one or more conductors surrounded by a dielectric material, such as a foam plastic material, and with a metallic sheath surrounding the core and serving as the outer conductor of the cable. Cables of this type are used in the transmission of RF signals, such as television signals, for example.

It has been previously recognized that for cables of this type it is desirable for several reasons to bond the metallic sheath to the core. Bonding of the sheath to the core prevents the core from pulling back within the sheath, and bending and seals the cable against penetration of moisture. The foam dielectric used in the core of the cable is desirably formed of a nonpolar material, such as polyethylene foam for example, so as to avoid adversely affecting the electrical properties of the cable. However, the nonpolar nature of the dielectric makes it difficult to achieve a good bond between the dielectric and the metal sheath. It has also been previously recognized that a particularly suitable class of adhesive for accomplishing bonding of the metallic sheath to the foam dielectric of the core are polar thermoplastic copolymers of ethylene and an ethylenically unsaturated carboxylic acid, and in particular ethylene acrylic acid copolymers. For example, the use of such adhesives in cables is disclosed in the following United States patents:

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	<u>U.S. Patent No.</u>	<u>Inventor</u>	<u>Issue Date</u>
	3,315,025	Tomlinson	April 18, 1967
	3,459,877	Bullock et al	August 5, 1969
	3,662,090	Grey	May 9, 1972
5	3,681,515	Mildner	August 1, 1972
	3,795,540	Mildner	March 5, 1974
	4,010,315	Mildner	March 1, 1977

Because of the thermoplastic nature of the ethylene acrylic acid copolymer adhesive, it is typically applied to the substrate with heating, e.g. by extrusion coating or by application of the adhesive in powdered form accompanied by fusion. The adhesive layer is usually about one to five mils in thickness.

The present invention is based upon the recognition that the polar nature of the ethylene acrylic acid copolymer adhesive adversely affects the electrical characteristics of the cable, and for this reason, as well as for reduction in material cost, it is desirable to make the adhesive layer as thin as possible. However, with the conventional methods of application noted above there is a limitation to how thin the adhesive can be applied. Accordingly, one of the objectives of the present invention is to provide a method for producing cable with a metallic sheath bonded to the core and wherein the adhesive can be applied in a thinner layer than heretofore possible. Still another objective of the invention is to provide a suitable adhesive for bonding the core to the metallic sheath wherein the adhesive can be applied in a thinner layer than heretofore possible. A related objective is to provide an improved cable utilizing such an adhesive.

One of the problems which is encountered in applying thermoplastic ethylene acrylic acid copolymer adhesives by the conventional methods heretofore available is that the heat which is necessary for application of the adhesive adversely affects the thermoplastic foam dielectric of the core and thus makes it difficult or impractical to apply

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the adhesive directly to the core component. For example, extrusion coating of the adhesive directly onto the core component tends to collapse the foam dielectric and thus impair its electrical characteristics. Consequently, the usual practice is to apply the adhesive coating to the metal foil used for forming the metallic sheath, as will be seen from the patents cited above. However, this approach does not lend itself for application of the adhesive to cables in which the metallic sheath component is of an electrically and mechanically continuous construction and formed from extruded seamless tubing or from longitudinally welded tubing.

Summary of the Invention

In accordance with the present invention, these problems are overcome by providing a copolymer of ethylene and an ethylenically unsaturated carboxylic acid which can be applied as an aqueous dispersion to form a very thin film of adhesive for bonding the metallic sheath to the core component. The copolymer is rendered water dispersible by partially neutralizing the carboxylic acid, but while leaving sufficient residual carboxylic acid for imparting good adhesive properties for bonding the metallic sheath and the core.

The water dispersible copolymer adhesive may be used advantageously in the production of cables by any of several methods. A particularly suitable method for forming cables in accordance with the present invention involves forming the core by continuously extruding a foamed dielectric around one or more conductors, and directing the thus formed core through an adhesive bath containing an aqueous dispersion of a copolymer of ethylene and an ethylenically unsaturated carboxylic acid to form a thin adhesive coating on the core, and then surrounding the core with a metallic sheath and heating the adhesive coating to effect bonding of the core to the metallic sheath. The surrounding of the core with the metallic

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sheath may be accomplished by forming a metallic strip around the core and longitudinally welding to form a continuous tubular sheath. Alternatively, a seamless metallic tube of finite length can be formed by extrusion, and the
5 core may be drawn longitudinally into the seamless tube, followed by sinking of the tube onto the core. In another method of cable manufacture, the adhesive may be applied to a metal strip and the adhesive coated strip thereafter applied by wrapping about the core.

10 The adhesive polymer which is used in accordance with the present invention is a thermoplastic random copolymer of ethylene and one or more ethylenically unsaturated carboxylic acids, which polymer has been rendered water dispersible by partial neutralization of the carboxylic
15 acid as hereinafter more fully described.

The carboxylic acid monomer is preferably selected from ethylenically unsaturated mono- and poly-carboxylic acids and acid anhydrides having from 3 to 8 carbon atoms per molecule and partial esters of such polycarboxylic
20 acids where the acid moiety has at least one carboxylic acid group and the alcohol moiety has from one to 20 carbon atoms. The copolymer may consist essentially of ethylene and one or more of such ethylenically unsaturated carboxylic acids, or can also contain small amounts of other
25 monomers copolymerizable with ethylene. Specific examples of suitable ethylenically unsaturated carboxylic acids include acrylic acid, methacrylic acid, fumaric acid, maleic acid, crotonic acid, itaconic acid, maleic anhydride, monomethyl maleate, monoethyl maleate, monoethyl
30 fumarate, monoethyl fumarate, tripropylene glycol, monomethyl ether acid maleate, ethylene glycol, monophenyl ether acid maleate, etc. Among these carboxylic acids, acrylic acid itself is preferred.

The copolymers can be prepared by known methods, for
35 example, by subjecting a mixture of ethylene and an acrylic acid to high pressures and elevated temperature in the presence of a suitable catalyst.

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These copolymers are commercially available from various sources and contain varying amounts of acrylic acid ranging from about 3 percent to about 20 percent or higher. The copolymers useful in the present invention should have
5 an acrylic acid content of at least about 12 percent and most desirably at least about 15 percent, in order to be made water dispersible. The copolymer may be rendered water dispersible by reaction with an alkaline neutralizing agent, such as an alkali metal hydroxide or ammonium
10 hydroxide. Typically this is carried out by mixing the polymer and an aqueous solution of the alkaline neutralizing agent in a suitable reaction vessel for 2 to 3 hours with heating and stirring.

In order to maintain effective adhesive properties,
15 it is important that the copolymer be only partially neutralized. By "partially neutralized" it is meant that the copolymer is reacted with a stoichiometric quantity of neutralizing agent less than that required to react with all of the carboxyl groups of the carboxylic acid. Thus a
20 portion of the carboxyl groups of the copolymer remain unreacted and present in the copolymer. Preferably, the partially neutralized water dispersible copolymer should contain at least 2 percent by weight of unreacted ethylenically unsaturated carboxylic acid.

25 While various alkaline neutralizing agents can be used to react the copolymer and render it water dispersible, some agents have characteristics which render their use objectionable in certain applications as a cable sheath adhesive. For example, copolymers which have been solubi-
30 lized by reaction with ammonium hydroxide evolve ammonia gas when heated, which is highly corrosive to the copper center conductor. Trimethylamine as a neutralizing agent is not corrosive, but produces an undesirable fishy odor. For use in the present invention it is preferred that the
35 copolymer be reacted with monovalent metal hydroxides, e.g. hydroxides of sodium, lithium, potassium or rubidium. This

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class of alkaline neutralizing agent produces stable dispersions of the copolymer adhesive without undesirable effects such as those noted above.

EXAMPLE 1

5 Preparation of Partially Neutralized Adhesive

An aqueous dispersion of ethylene acrylic acid copolymer containing about 4 percent by weight of acrylic acid was prepared as follows:

To 132.5 liters of water was added 3.45 kg of sodium
10 hydroxide with stirring to dissolve the sodium hydroxide. The solution was heated to 180° F. (82° C) and held at this temperature while 41.7 kg of ethylene acrylic acid copolymer (500 melt index, 20% acrylic acid) (Dow Chemical Company EAA XO 2375.30) was added. Stirring was maintained
15 for about 3 hours until all of the EAA was dispersed in the water. The mix was then cooled. Five milliliters of an antifoam agent (BYK-W antifoam available from Mallinckrodt Chem was added with stirring and the mixture was filtered through fine cheesecloth.

20 EXAMPLE 2

Application of Adhesive

An arrangement of apparatus similar to that shown in the drawing is used for producing a coaxial cable with the sheath adhesively bonded to the core. The aqueous adhesive
25 dispersion prepared in Example 1 is placed in an open trough 20 at the discharge end of an extrusion apparatus 21.

A continuous copper center conductor is directed through the extrusion apparatus 21 and a foamable polyolefin polymer is coextruded around the center conductor 11
30 and allowed to expand to form a cable core comprised of the copper center conductor with a surrounding foam dielectric. This cable core is directed into the trough 20 where the cellular foam material is cooled and allowed to harden while
35 simultaneously a coating of the aqueous adhesive composition is applied to the cellular foam. Upon leaving the trough

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the heat of the cellular foam partially dries the adhesive. The adhesive coated core is directed through a heated tunnel 22 to complete the drying of the adhesive, leaving a thin adhesive coating (about .00254 to .0254 mm thickness) on the outer surface of the foam dielectric. The adhesive coated core is then directed through a cooling chamber 23. A narrow strip of aluminum is directed from a reel 24 and is formed into a tubular sheath surrounding the foam dielectric. The strip of aluminum then advances through a welding apparatus 25 and the opposing side edges of the strip are positioned into butting relationship and joined together by a continuous longitudinal weld. The assembly then passes through a reduction die 26 where the tubular aluminum sheath is reduced in diameter and brought into close snug relationship with the dielectric. The assembly then passes through an extrusion coating apparatus 27 where a heated fluent coating material is extruded therearound to form a protective outer jacket. The heat of the coating composition also serves to activate the thermoplastic EAA adhesive coating and to form a bond between the sheath and the outer surface of the dielectric.

In the foregoing specification and examples there has been set forth a preferred embodiment of the invention, but it is to be understood that the invention is not limited thereto and may be embodied and practiced in other ways within the scope of the following claims.

CLAIMS

1. An electrical cable comprising a core having at least one inner conductor surrounded by a dielectric material, a metallic sheath surrounding said core, and an adhesive bonding said metallic sheath to said core, characterized in that said adhesive comprises a water dispersible
5 copolymer of ethylene and an ethylenically unsaturated carboxylic acid forming a thin film between said metallic sheath and said core.
2. A cable according to Claim 1 wherein said thin film of adhesive has a thickness no greater than about .0254 mm.
3. A cable according to Claim 1 wherein said dielectric material comprises a foam plastic material and said metallic sheath comprises a longitudinally welded metal tube.
4. A cable according to Claim 1 wherein said dielectric material comprises a foam plastic material and said metallic sheath comprises an extruded seamless metal tube.
5. A cable according to Claim 1 wherein said dielectric material comprises a foam plastic material and said metallic sheath comprises a metal strip wrapped about said core.
6. A cable according to Claim 1 wherein said copolymer is rendered water dispersible by partial neutralization of the ethylenically unsaturated carboxylic acid, and wherein a portion of said carboxylic acid is unreacted to
5 provide carboxyl groups present in the adhesive for effecting bonding between said metallic outer conductor and said core.

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7. A cable according to Claim 6 wherein the ethylenically unsaturated carboxylic acid is partially neutralized by reaction with a monovalent metal hydroxide.

8. A cable according to Claim 6 wherein the copolymer contains at least 2 percent by weight of unreacted ethylenically unsaturated carboxylic acid.

9. A cable according to Claim 1 wherein the adhesive comprises an ethylene acrylic acid copolymer containing at least 15 percent by weight acrylic acid partially neutralized with a monovalent metal hydroxide.

10. An electrical cable comprising a core having at least one inner conductor and a foam dielectric surrounding said at least one inner conductor, a metallic sheath surrounding said core, and an adhesive bonding said
5 metallic sheath to said core, characterized in that said adhesive comprises a copolymer of ethylene and acrylic acid containing at least 15 percent by weight acrylic acid, with a portion of the acrylic acid being neutralized to render the copolymer water dispersible, and with at least 2 per-
10 cent of the acrylic acid being unreacted to provide carboxyl groups present in the adhesive for effecting bonding between said metallic sheath and said core.

11. A method of producing an electrical cable having a core with at least one inner conductor surrounded by a dielectric material and a metallic sheath surrounding the core and adhesively bonded thereto, characterized by
5 applying to at least one of the core or the metallic sheath an adhesive composition comprising an aqueous dispersion of a copolymer of ethylene and an ethylenically unsaturated carboxylic acid to form a thin adhesive coating thereon, assembling the core and the metallic sheath in contact with
10 one another with the thin adhesive coating at the interface therebetween, and heating the assembly to activate the adhesive to bond the core and the metallic sheath.

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12. A method according to Claim 11 wherein the step of assembling the core and the metallic sheath comprises forming a metal strip longitudinally around the core to produce a longitudinal seam and welding the longitudinal
5 seam, and wherein said aqueous dispersion is applied to the core prior to forming of the metallic strip therearound.

13. A method according to Claim 11 wherein the step of assembling the core and the metallic sheath comprises drawing the core longitudinally into a seamless metal tube, and wherein said aqueous dispersion is applied to the core
5 prior to its being positioned within the metal tube.

14. A method according to Claim 11 wherein the ethylenically unsaturated carboxylic acid of said copolymer is partially neutralized and a portion thereof is unreacted to provide carboxyl groups present in the copolymer for
5 effecting bonding between the metallic sheath and the dielectric.

15. A method according to Claim 14 wherein the copolymer contains at least 2 percent by weight of unreacted ethylenically unsaturated carboxylic acid.

16. A method of producing an electrical cable having a core with at least one inner conductor surrounded by a foam dielectric, and a metallic sheath surrounding said core and adhesively bonded thereto, characterized by
5 applying to at least one of the core or the metallic sheath an adhesive composition comprising an aqueous dispersion of an ethylene acrylic acid copolymer containing at least 15 percent by weight acrylic acid, with a portion of the acrylic acid being neutralized to render the copolymer
10 water dispersible and with at least 2 percent of the acrylic acid being unreacted to provide carboxyl groups present in the adhesive for effecting bonding, assembling the foam

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dielectric and the metallic sheath in contact with one another with the adhesive at the interface therebetween,
15 and heating the thus formed assembly to activate the adhesive to bond the foam dielectric and the metallic sheath.

17. A method of producing an electrical cable comprising forming a core by advancing at least one conductor through an extruder while extruding a foam plastic dielectric material therearound, directing the thus formed
5 core through an adhesive bath containing an aqueous dispersion of a copolymer of ethylene and an ethylenically unsaturated carboxylic acid to form a thin adhesive coating on the core, surrounding the core with a metallic sheath, and heating the adhesive coating to effect bonding of the core
10 to the metallic sheath.

18. An adhesive composition comprising a stable aqueous dispersion of a copolymer of ethylene and an ethylenically unsaturated carboxylic acid partially neutralized by reaction with a monovalent metal hydroxide.

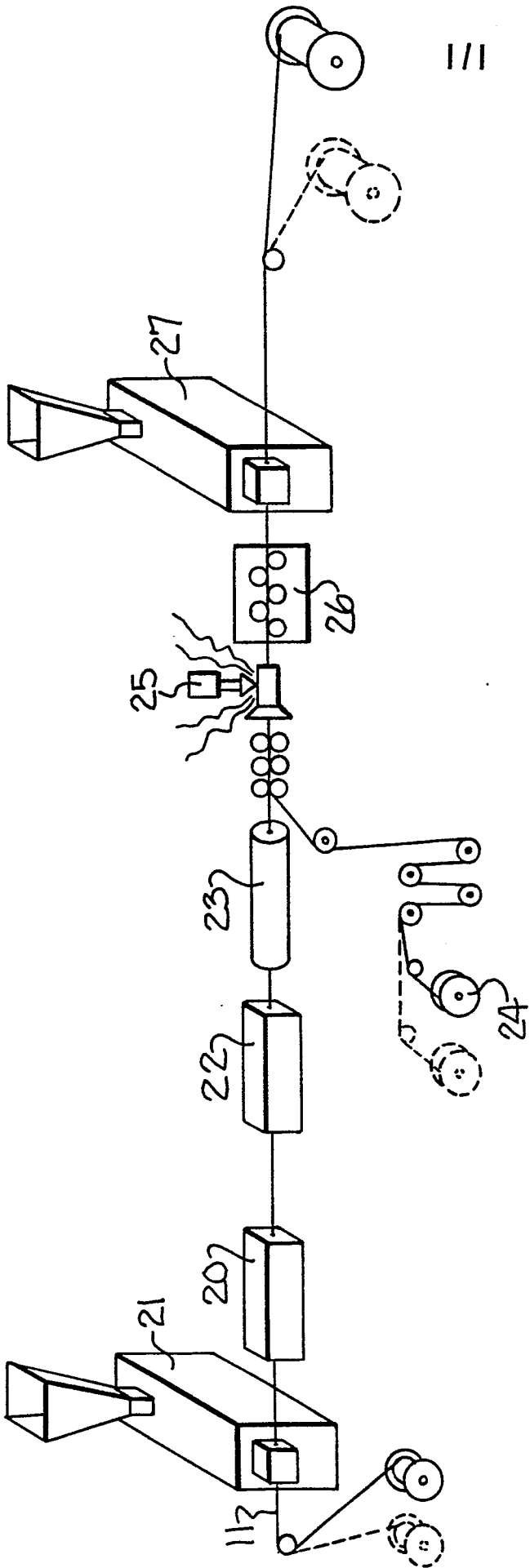
19. An adhesive composition according to Claim 18 wherein said copolymer contains at least 2 percent by weight of unreacted ethylenically unsaturated carboxylic acid.

20. An adhesive composition according to Claim 19 wherein the copolymer contains at least 12 percent by weight of said ethylenically unsaturated carboxylic acid.

21. An adhesive composition according to Claim 18 additionally comprising an antifoam agent.

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22. An adhesive composition comprising a stable aqueous dispersion of an ethylene acrylic acid copolymer containing at least 15 percent by weight acrylic acid partially neutralized with a monovalent metal hydroxide, with
5 at least 2 percent of the acrylic acid being unreacted to provide carboxyl groups present in the adhesive for effecting bonding.





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Application number

EP 83 30 4063

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 7)
A	US-A-4 104 481 (F.N. WILKENLOH et al.)		H 01 B 3/44 H 01 B 11/18 C 09 J 3/14 H 01 B 13/00
A	--- CHEMICAL ABSTRACTS, vol. 87, no. 4, 25th July 1977, page 84, no. 24880j, Columbus, Ohio, USA & JP - A - 77 40 583 (NIPPON PAINT CO., LTD.) 29-03-1977		
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The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 08-11-1983	Examiner STIENON P.M.E.
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			



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DOCUMENTS CONSIDERED TO BE RELEVANT			Page 2
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 3)
D, A	US-A-3 795 540 (R.C. MILDNER) -----		
			TECHNICAL FIELDS SEARCHED (Int. Cl. 3)
The present search report has been drawn up for all claims			
Place of search THE HAGUE		Date of completion of the search 08-11-1983	Examiner STIENON P.M.E.
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	