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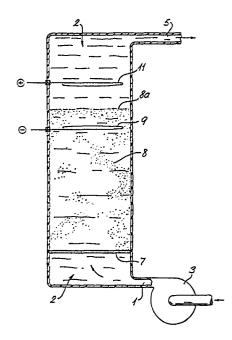
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54) Purifying mixed-cation electrolyte.

(9) An electrolyte containing 65 g/l zinc and 150 g/l Cu is purified in zinc, that is, the copper is removed, by causing the electrolyte to fluidise a bed (8) of ½ mm copper particles. The bed is fluidised by 25% to make it 42 cm deep. An anode (11) is disposed above the top of the bed. A cathode (9) is disposed part-way up the bed. Copper is electroplated onto the bed particles. Any zinc which may be electroplated onto the bed particles tends to redissolve with concomitant cementation, on the particles, of copper, which can be recovered. The elektrolyte is thus eventually completely stripped of copper and can be used for zinc electrowinning.

By controlling the pH of the electrolyte, substantially one metal, or one desired combination of metals, may be removed. In particular, pure copper deposition can be completed at low pH even in the presence of cadmium; upon a substantial increase in pH, cadmium deposition will occur.



197 769

Application number



EUROPEAN SEARCH REPORT

ΕP 86 30 2477

DOCUMENTS CONSIDERED TO BE RELEVANT Citation of document with indication, where appropriate. Relev			Relevant	vant CLASSIFICATION OF THE
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х	* Claims 3-18 *		3-18	
Y,A	US-A-4 240 886 al.) * Claim 4 *	 (D.R. HODGES et	1,2	
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	The present search report has t	been drawn up for all claims		
Place of search Date of completion of the search			Examiner	
	THE HAGUE	20-01-1987	DE	ANNA P.L.

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