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(54) Hard magnetic material.

 $\bigcirc$  A hard magnetic material having the composition RE<sub>2</sub>Fe<sub>14-x</sub>Mn<sub>x</sub>C, wherein RE is at least one element selected from the group formed by Nd, Pr, Ce and La, and wherein  $0.2 \le x \le 2$ .

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# Hard magnetic material

The invention relates to a hard magnetic material comprising a rare earth metal, iron and carbon. The invention also relates to a magnet on the basis of hard magnetic material.

A hard magnetic material of the type described in the opening paragraph is known from, inter alia, J. Appl. Phys. 61 3574-3576 (1987). The authors of the article classify the compound Nd<sub>2</sub>Fe<sub>14</sub>C described therein as not stable. Consequently, this material cannot be used as a starting material for a permanent magnet.

It is an object of the invention to provide a hard magnetic material which comprises a rare earth metal (RE), iron and carbon, and which is stable. This object is achieved in accordance with the invention by a hard magnetic material which is further characterized in that the material contains a hard magnetic phase the composition of which corresponds to the formula  $E_2Fe_{14-x}Mn_xC,$ 

wherein RE is at least one element selected from the group formed by Nd (Neodymium), Pr (Praseodymium), Ce (Cerium) and La (Lanthanum), and wherein  $0.2 \le x \le 2$ . It has been found that the value of x cannto be selected randomly. The materials in which x is smaller than 0.2 were found to be insufficiently stable. The Curie temperature of the materials in which x is larger than 2 is about room temperature or below room temperature, which is problematic for the magnetic application of these materials. In principle, the Curie temperature can be raised by substituting a part of the iron present in the material with cobalt. The absence of B in the compounds according to the invention is an advantage because as a consequence thereof no toxic, volatile boroncompounds can be formed during the preparations of said compounds.

In experiments it has further been established that the hard magnetic phase of the compounds according to the invention has a tetragonal crystal structure of the ThMn<sub>12</sub> type, which is also the case with the known Nd<sub>2</sub>Fe<sub>14</sub>B. As is shown in the Table below, the Curie temperature (T<sub>c</sub>) of the compounds according to the invention is between 300 and 550 K. The Curie temperature of compounds according to the invention having an identical rare earth metal is lower as the Mn content is higher.

Table

Curie temperature (K)	Compound
C 505	Nd <sub>2</sub> Fe <sub>13.7</sub> Mn <sub>0.3</sub> C
C 492	Nd <sub>2</sub> Fe <sub>13.5</sub> Mn <sub>0.5</sub> C
452	Nd <sub>2</sub> Fe <sub>13</sub> MnC
C 400	Nd <sub>2</sub> Fe <sub>12.5</sub> Mn <sub>1.5</sub> C
355	Nd <sub>2</sub> Fe <sub>12</sub> Mn <sub>2</sub> C
481	Pr <sub>2</sub> Fe <sub>13.7</sub> Mn <sub>0.3</sub> C
460	Pr <sub>2</sub> Fe <sub>13.5</sub> Mn <sub>0.5</sub> C
412	Pr <sub>2</sub> Fe <sub>13</sub> MnC
306	Pr <sub>2</sub> Fe <sub>12</sub> Mn <sub>2</sub> C

The hard magnetic materials according to the invention can be obtained in a customary way by melting together suitable starting materials in ratios corresponding to the compositional formula, after which they are subjected to an annealing treatment in a temperature range from 800° C - 950° C in a protective gas or a vacuum for two days; during which treatment recrystallization occurs.

It is noted that in the experiments leading to the present invention it has become clear to Applicants that the manufacture of a stable material having a hard magnetic phase, the composition of which corresponds to the formula Nd<sub>2</sub>Fe<sub>14</sub>C, can be achieved in principle. This was attained by subjecting a casting of the compositional formula Nd<sub>2</sub>Fe<sub>14</sub>C to a prolonged annealing treatment within a narrow temperature range, preferably between 850 and 880° C, during which treatment recrystallization occurs. Applicants have reason to believe that in comparison with the known method the preparation of the material according to the invention requires a shorter annealing treatment for recrystallization in order to obtain the desired tetragonal structure. Moreover, in the case of the materials according to the invention, the annealing

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treatment can take place in a wider temperature range. On the other hand, this stable material can also be prepared by melt spinning a composition with formula Nd<sub>2</sub>Fe<sub>14</sub>C, followed by an annealing treatment of the ribbon formed, during which treatment recrystallization occurs. Both known methods have the disadvantage that they are less suitable from an economical point of view.

The hard magnetic properties of the compounds in accordance with the invention are comparable with those of materials on the basis of  $Nd_2Fe_{14}B$ . For example, the magnetization measured at  $Nd_2Fe_{13.7}Mn_{0.3}C$  powder at room temperature is approximately 100 Am²/kg. This value is comparable with the saturation magnetization which is measured at  $Nd_2Fe_{14}B$  powder at room temperature. It is noted that this value cannot be regarded as the saturation magnetization, since the material under consideration is magnetically very anisotropic.

After the materials obtained have been pulverized, magnets in the form of shaped bodies can be manufactured from said materials in the usual manner.

#### 5 Claims

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- 1. A hard magnetic material comprising a rare earth metal, iron and carbon, characterized in that the material contains a hard magnetic phase the composition of which corresponds to the formula  $RE_2Fe_{14-x}Mn_xC$
- wherein RE is at least one element selected from the group formed by Nd (Neodymium), Pr (Praseodymium), Ce (Cerium) and La (Lanthanum), and wherein  $0.2 \le x \le 2$ .
  - 2. A hard magnetic material as claimed in Claim 1, characterized in that a part of the iron present in the material is replaced by cobalt.

3. A permanent magnet on the basis of a hard magnetic material as claimed in Claim 1.

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# **EUROPEAN SEARCH REPORT**

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The present search report has been d	lrawn up for all claims		
Place of search E HAGUE	Date of completion of the search 15–06–1989	DECA	Examiner ANNIERE L.J.
CATEGORY OF CITED DOCUMENTS ticularly relevant if taken alone ticularly relevant if combined with another ument of the same category honlogical background	E : earlier patent do after the filing o D : document cited L : document cited	T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons	
	intrinsic coercivities earth-carbon-boron all carbide or boro-carbid Fe14R2X(X=BxC1-x)"  ———————————————————————————————————	intrinsic coercivities in iron-rare earth-carbon-boron alloys through the carbide or boro-carbide Fe14R2X(X=BxC1-x)"  ———————————————————————————————————	intrinsic coercivities in iron-rare earth-carbon-boron alloys through the carbide or boro-carbide Fe14R2X(X=BxCl-x)"  ———————————————————————————————————