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(54) High-contrast silver halide photographic material.

The improve silver halide photographic material contains a specified hydrazide compound and a specified polyoxyethylene based compoundand and is capable of providing high-contrast image with high sensitivity in a rapid and consistent way.

EP 0 382 455 /

HIGH-CONTRAST SILVER HALIDE PHOTOGRAPHIC MATERIAL

BACKGROUND OF THE INVENTION

The present invention relates to a silver halide photographic material. More particularly, the present invention relates to a silver halide photographic material capable of providing a high-contrast and sensitivity photographic image in a rapid and consistent way.

High-contrast photographic image is usually employed in forming characters or halftone image in photomechanical processes, or forming fine-line image in ultra-high-precision photomechanical processes. Certain kinds of silver halide photographic materials are known to be capable of forming photographic image with very high contrast. In one conventional method, a light-sensitive material composed of a silver chlorobromide emulsion that has an average grain size of 0.2 µm, that has a narrow grain size distribution, that comprises uniformly shaped grains and that has a high silver chloride content (at least 50 mol%) is processed with an alkaline hydroquinone developer of low sulfite ion concentration to form an image of high sharpness and high resolution, for example, a halftone image or a fine-line image. Silver halide light-sensitive materials

The photomechanical process involves a step of converting the original of continuous tone to be reproduced into a halftone image, or a step in which the continuous change in the density of the original is converted to a set of dots having areas proportional to the varying densities. To form a halftone image, the original is imaged onto a "lithographic" light-sensitive material via a crossline screen or a contact screen and subsequently developed. However, development of a lith-type light-sensitive material (i.e., a silver halide photographic material containing a silver halide emulsion consisting of fine grains of uniform size and shape) with an ordinary black-and-white developer results in the formation of a lower quality image than when it is developed with a lith-type developer. Thus, the lith-type light-sensitive material is conventionally processed with a lith-type developer that has a very low concentration of sulfite ions and which contains hydroquinone as the sole developing agent. However, developers of the lith type are very low in storage stability because they are highly sensitive to autooxidation. Hence, there is a strong need for controlling them to provide consistent quality of development in their continuous use and substantial efforts have been made to improve the storage stability of lith-type developers.

One of the methods proposed so far to attain object is replenishment with two separate liquids, one being a replenisher intended to compensate for the deterioration in developer's activity during development (compensation for process fatigue) and the other being a replenisher intended to compensate for the oxidative deterioration over time (compensation for aging fatigue). This method of using two types of replenishers is commonly adopted with automatic processors in photomechanical processes. However, because of the need to control the balance in replenishment with two liquids, this method suffers the problem of complexity in both equipment and operations.

Another problem with the use of lith-type developers is that they have a long induction period (i.e., the time required for an image to appear upon development) and hence are incapable of providing rapid access to image.

Methods are known that are capable of rapid formation of high-contrast image without using developers of the lith type. They are characterized by containing hydrazine compounds in silver halide light-sensitive materials [see U.S. Patent 2,419,975 and JP-A-51-16623 (the term "JP-A" as used herein means an "unexamined published Japanese patent application"), JP-A-51 20921, etc.]. According to these methods, the concentration of sulfite ions in developers can be held high enough to accomplish development with their storage stability being maintained at enhanced levels. However, in order to obtain a contrasty image by these methods, a developer with fairly high pH is necessary, which presents a problem in keeping the developer stable to produce a photographic image rapidly and with high sensitivity. The high pH of the developer tends to cause fogging and to prevent it, various organic restrainers must be contained in high concentrations but then sensitivity is compromised. A further problem is that a contrasty image is susceptible to defects such as pinholes if dust particles or other foreign substances are deposited on the light-sensitive material.

SUMMARY OF THE INVENTION

A first object, therefore, of the present invention is to provide a silver halide photographic material that is capable of producing a contrasty image in a rapid and consistent way.

A second object of the present invention is to provide a silver halide photographic material that is capable of consistent production of a contrasty image with minimum loss in sensitivity.

A third object of the present invention is to provide a silver halide photographic image capable of producing a contrasty and low-fog image.

A fourth object of the present invention is to provide a silver halide photographic image capable of producing a contrasty halftone image of high quality.

These and other objects of the present invention will become apparent by reading the following description.

The above-described objects of the present invention can be attained by a silver halide photographic material having at least one light-sensitive silver halide emulsion layer and which contains at least one of the compounds represented by the general formulas (I), (II) and (III) shown below and at least one compound represented by the general formula (IV) also shown below:

$$(R)_{n} = \begin{pmatrix} CONHNH-R_{1} \\ (CO)_{m}NHNH-R_{2} \end{pmatrix}$$

(where R_1 and R_2 each represents an aryl group or a heterocyclic group; R is an organic linkage; n is 0 - 6; and m is 0 or 1, provided that when n is 2 or more, R may be the same or different);

$$\begin{array}{c|cccc}
P_1 & P_2 & O & O \\
& & & & & & & & & \\
R_1 - N - N - C - C - C - R_2
\end{array}$$
(II)

(where R₁ is an aliphatic, aromatic or heterocyclic group; R₂ is a hydrogen atom or an optionally substituted alkoxy, heterocyclicoxy, amino or aryloxy group; P₁ and P₂ each represents a hydrogen atom, an acyl group or a sulfinic acid group);

$$\begin{array}{ccc}
R' & O \\
I & II \\
Ar-NNH-C-R_1
\end{array}$$

(where Ar is an aryl group containing at least one of a non-diffusible group or a silver halide adsorption accelerating group; R₁ is an optionally substituted alkyl, alkoxy or amino group; R' is a hydrogen atom or a substituent);

 $R_1 - O(CH_2CH_2O)nH$ (IV

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(where R₁ is a hydrogen atom or a substituted or unsubstituted aromatic group; n is an integer of 10 - 200).

DETAILED DESCRIPTION OF THE INVENTION

The specific construction of the present invention is described hereinafter in detail.

First, the compounds represented by the general formulas (I), (II) and (III) are specifically described below.

General formula (I) is as follows:

(R)n
$$CONHNH-R_1$$
(CO)mNHNH-R₂

where R_1 and R_2 each represents an aryl group or a heterocyclic group; R is a divalent organic group; R is a divalent org

Examples of the aryl group represented by R_1 and R_2 include phenyl, naphthyl, etc. Examples of the heterocyclic group represented by R_1 and R_2 include pyridyl, benzothiazolyl, quinolyl, thienyl, etc. A preferred example of R_1 and R_2 is an aryl group. Various substituents can be introduced into the aryl or heterocyclic group represented by R_1 and R_2 . Exemplary substituents include a halogen atom (e.g. chlorine or fluorine), an alkyl group (e.g. methyl, ethyl or dodecyl), an alkoxy group (e.g. methoxy, ethoxy, isopropoxy, butoxy, octyloxy or dodecyloxy), an acylamino group [e.g. acetylamino, pivalylamino, benzoylamino, tetradecanoylamino or α -(2,4-di-t-amylphenoxy) butyrylamino], a sulfonylamino group (e.g. methanesulfonylamino, butanesulfonylamino, dodecanesulfonylamino or benzenesulfonylamino), a urea group (e.g. phenylurea or ethylurea), a thiourea group (e.g. phenylthiourea or ethylthiourea), a hydroxy group, an amino group an alkylamino group (e.g. methylamino or dimethylamino), a carboxyl group, an alkoxycarbonyl group (e.g. ethoxycarbonyl), a carbamoyl group and a sulfo group. Examples of the divalent organic group represented by R include an alkylene group (e.g. methylene, ethylene, trimethylene and tetramethylene), an arylene group (e.g. phenylene and naphthylene), and an aralkylene group. The alkylene group may contain in the bond an oxy group, a thio group, a seleno group, a carbonyl group, group



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(where R_3 is a hydrogen atom, an alkyl group or an aryl group), a sulfonyl group, etc. Various substituents may be introduced into the group represented by R. Exemplary substituents include -CONHNHR₄ (R₄ has the same meaning as R₁ and R₂ described above), an alkyl group, an alkoxy group, a halogen atom, a hydroxy group, a carboxyl group, an acyl group, an aryl group, etc. A preferred example of R is an alkylene group.

Preferred examples of the compound represented by the general formula (I) are those wherein R_1 and R_2 are each a substituted or unsubstituted phenyl group, n = m = 1 and R is an alkylene group.

The following are typical examples of the compound represented by the general formula (I):

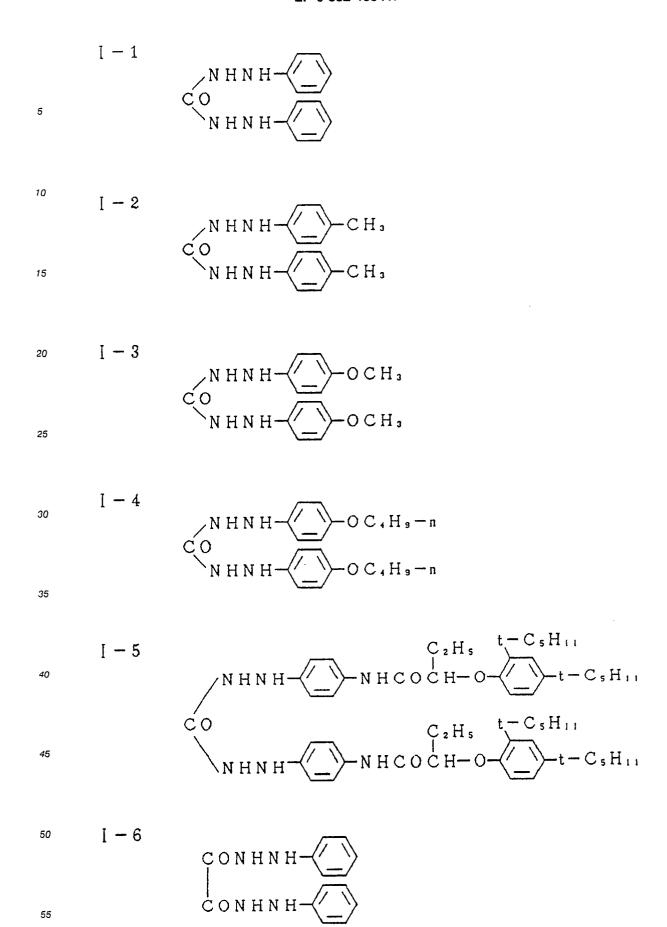
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The general formula (II) is as follows:

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where R_1 is an aliphatic, aromatic or heterocyclic group; R_2 is a hydrogen atom or an optionally substituted alkoxy, heterocyclicoxy, amino or aryloxy group; P_1 and P_2 each represents a hydrogen atom, an acyl group or a sulfinic acid group.

The aliphatic group represented by R_1 is preferably one having at least 6 carbon atoms, with a straight-chained, branched or cyclic alkyl group having 8 - 50 carbon atoms being particularly preferred. The branched alkyl group may be cyclized to form a saturated hetero ring containing one or more hetero atoms. the alkyl group represented by R_1 may contain a substituent such as an aryl, alkoxy or sulfoxy group.

The aromatic group represented by R_1 is a monocyclic or bicycyclic aryl group or an unsaturated heterocyclic group, which may be condensed with a mono- or bicyclic aryl group to form a hetero aryl group. Examples of the aromatic group represented by R_1 included those which contain a benzene ring, a naphthalene ring, a pyridine ring, a pyrimidine ring, an imidazole ring, a pyrazole ring, a quinoline ring, an isoquinoline ring, a benzene ring is particularly preferred. A particularly preferred example of R_1 is an aryl group.

The aryl group or unsaturated heterocyclic group represented by R_1 may be substituted and typical substituents include, for example, a straight-chained, branched or cyclic alkyl group (preferably one in which the alkyl portion is mono- or bicyclic with 1 - 20 carbon atoms), an alkoxy group (preferably one having 1 - 20 carbon atoms), a substituted amino group (preferably one which is substituted with an alkyl group having 1 - 20 carbon atoms), an acylamino group (preferably one having 2 - 30 carbon atoms), a sulfonamido group (preferably one having 1 - 30 carbon atoms) and a ureido group (preferably one having 1 - 30 carbon atoms).

The optionally substituted alkoxy group represented by R_2 in the general formula (II) is one having 1 - 20 carbon atoms which may have a substituent such as a halogen atom or an aryl group.

The optionally substituted aryloxy or heterocyclic oxy group which is represented by R_2 in the general formula (II) is preferably monocyclic and exemplary substituents include a halogen atom, an alkyl group, an alkoxy group, a cyano group, etc.

A preferred example of the group represented by R_2 is an optionally substituted alkoxy or amino group. In the case of an amino group, it is group

where A_1 and A_2 are each an optionally substituted alkyl or alkoxy group, or they may combine to form a cyclic structure containing an -O-, -S- or -N- bond. It should however be noted that R_2 is in no case a hydrazino group.

A ballast group commonly employed in couplers and other immobilized photographic additives may be incorporated into R_1 or R_2 in the general formula (II). A ballast group is a group having at least 8 carbon atoms which is fairly innert to photographic properties and may be selected from among, for example, an alkyl group, an alkoxy group, a phenyl group, an alkylphenyl group, a phenoxy group, an amino group, an amido group and an ureido group.

A group that enhances adsorption onto the surface of silver halide grains may be incorporated into R_1 or R_2 in the general formula (II). Examples of such adsorption enhancing group include those which are described in U.S. Patent 4,355,105 such as a thiourea group, a heterocyclic thioamido group, a mercapto heterocyclic group and a triazole group. Among the compounds represented by the general formula (II), those which are represented by the following general formula (IIa) are particularly preferred:

where R_3 and R_4 each represents a hydrogen atom, an optionally substituted alkyl group (e.g. methyl, ethyl, butyl, dodecyl, 2-hydroxypropyl, 2-cyanoethyl or 2-chloroethyl), or an optionally substituted phenyl, naphthyl, cyclohexyl, pyridyl or pyrrolidyl group (e.g. phenyl, p-methylphenyl, naphthyl, α -hydroxynaphthyl, cyclohexyl, p-methylcyclohexyl, pyridyl, 4-propyl-2-pyridyl, pyrrolidyl or 4-methyl-2-pyrrolidyl); R_5 represents a hydrogen atom or an optionally substituted benzyl, alkoxy or alkyl group (e.g. benzyl, p-methylbenzyl, methoxy, ethoxy, ethyl or butyl); R_6 and R_7 each represents a divalent aromatic group (e.g. phenylene or naphthylene); Y represents a sulfur or oxygen atom; L represents a divalent linkage (e.g. $-SO_2CH_2CH_2NH-SO_2NH$, $-OCH_2SO_2NH$, -O-, or -CH=N-); R_8 represents -NR'R'' or $-OR_9$ [where R', R'' and R_9 each represents a hydrogen atom or an optionally substituted alkyl (methyl, ethyl or dodecyl), phenyl (e.g. phenyl, p-methylphenyl or p-methoxyphenyl) or naphthyl (e.g. α -naphthyl or β -naphthyl) group; and m and a n each represents 0 or 1, provided that when R_8 is OR_9 , Y is preferably a sulfur atom.

The following are typical examples of the compounds represented by the general formulas (II) and (IIa):

II - 37

$$(CH_3)_3CCCH_2CNH \longrightarrow CONH-(CH_2)_2CNH \longrightarrow NHNHCC-O-C_2H_5$$

$$[I-43]$$

$$CH_3 COOC_4H_{9}N OO$$

$$H_7C_3-C \longrightarrow OCHCONH \longrightarrow NHNHCCOCH_2CH_2-N NH$$

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II - 46

11 - 47

II - 45C2H5NHCSNH-OCH2CONH-NHNHCOCONHCH3 25

C₂H₅N H C N H — O C H₂S O₂N H — NHN H C C N H C H₃

CH3NHCNH-_SO2NH-_NHNHCCNHCH3 40

The method of synthesizing the compounds listed above is described below with compounds II-45 and II-47 taken as examples.

Synthesis of compound II-45 Scheme of synthesis:

$$\frac{H_2}{P d/C} \longrightarrow N H_2 \longrightarrow NHNHC - COC_2H_5$$
(B)

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$$\begin{array}{c}
 & \text{OO} \\
 & \text{H}_2 \\
\hline
 & \text{Pd/C}
\end{array}$$

$$\begin{array}{c}
 & \text{NH}_2 \\
\hline
 & \text{C}_2 \text{H}_5
\end{array}$$

$$\begin{array}{c}
 & \text{OO} \\
 & \text{NHNHC-COC}_2 \text{H}_5 \\
\hline
 & \text{(E)}
\end{array}$$

Compound II-45

Two compounds, 4-nitrophenylhydrazine (153 g) and diethyl oxalate (500m ml), are mixed and refluxed for 1 h. As the reaction proceeds, ethanol is removed and the reation mixture is finally cooled to cause crystal precipitation. After filtration, the crystal is washed with petroleum several times and recrystallized. Fifty grams of the resulting crystal (A) is dissolved in 1000 ml of methanol by heating and reduced in a hydrogen atmosphere under pressure (50 psi) in the presence of a Pd/C (palladium on carbon) catalyst to obtain compound (B).

Twenty-two grams of the compound (B) is dissolved in a mixture of acetonitrile (200 ml) and pyridine (16 g). To the resulting solution, 24 g of compound (C) in acetonitrile is added dropwise at room temperature. The insoluble matter is removed by filtration and the filtrate is concentrated and purified by

recrystallization to obtain 31 g of compound (D).

A portion (30 g) of the compound (D) is hydrogenated by the same procedure as described above to obtain 20 g of compound (E).

Ten grams of compound (E) is dissolved in 100 ml of acetonitrile and after adding 3.0 g of ethyl isothiocyanate, the solution is refluxed for 1 h. After distilling off the solvent, the residue is purified by recrystallization to obtain 7.0 g of compound (F). Five grams of compounds (F) is dissolved in 50 ml methanol and the solution is stirred after adding methylamine (8 ml of a 40% aqueous solution). After concentrating methanol to some extent, the precipitating solids are recovered and purified by recrystallization to obtain compound II-45.

Synthesis of compound II-47 Scheme of synthesis:

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$$\begin{array}{c|c}
& OO \\
& || || \\
& || || \\
& Pd/C
\end{array}$$

$$C H_{3}N H C S N H \longrightarrow S O_{2}N H \longrightarrow NHNH C COC_{2}H_{5} \xrightarrow{C H_{3}N H_{2}}$$

$$(E)$$

Compound II -47

Compound (B) (22 g) is dessolved in 200 ml of pyridine and to the stirred solution, 22 g of pnitrobenzenesulfonyl chloride is added. After pouring the reaction mixture into water, the precipitating solids are recovered to obtain compound (C). In accordance with the scheme described above, compound (C) is subjected to the same reactions as performed in the synthesis of compound II-45 to obtain compound II-47.

The general formula (III) is as follows:

$$\begin{array}{ccc}
R' & O \\
& \parallel & \parallel \\
Ar-NNH-C-R_1
\end{array}$$
(III)

where Ar is an aryl group containing at least one of a non-diffusible group or a silver halide adsorption accelerating group; R₁ is an optionally substituted alkyl, alkoxy or amino group. A preferred non-diffusible group is a ballast group which is commonly employed in couplers and other immobilized photographic additives. A ballast group is a group having at least 8 carbon atoms which is fairly inert to photographic properties and may be selected from among, for example, an alkyl group, an alkoxy group, a phenoxy group, an alkylphenoxy group, an amino group, an amido group and an ureido group. Examples of the silver halide adsorption accelerating group include those which are described in U.S. Patent 4,385,108 such as a thiourea group, a thiourethane group, a heterocyclic thioamido group, a mercapto heterocyclic group and a triazole group.

An alkyl group is represented by R₁ in the general formula (III) and it is a straight-chained, branched or cyclic alkyl group as exemplified by methyl, ethyl, propyl, butyl, isopropyl, pentyl, cyclohexyl, etc. Examples of the substituent that is introduced into these alkyl groups include: an alkoxy group e.g. methoxy or ethoxy; an aryloxy group e.g. phenoxy or or p-chlorophenoxy; a heterocyclic oxy group e.g. pyridyloxy; a mercapto group; an alkylthio group e.g. methylthio or ethylthio; an arylthio group e.g. phenylthio or pchlorophenylthio; a heterocyclic thio group e.g. pyridylthio, pyrimidylthio or thiazolylthio; an alkylsulfonyl e.g. methanesulfonyl or butanesulfonyl; an arylsulfonyl group e.g. benzenesulfonyl; a heterocyclic sulfonyl group e.g. pyridylsulfonyl or morpholinosulfonyl; an acyl group e.g. acetyl or benzoyl; a cyano group; a chlorine atom; a bromine atom; an alkoxycarbonyl group e.g. ethoxycarbonyl or methoxycarbonyl; an aryloxycarbonyl group e.g. phenoxycarbonyl; a carboxy group; a carbamoyl group; an alkylcarbamoyl group e.g. N-methylcarbamoyl or N,N-dimethylcarbamoyl; an arylcarbamoyl group e.g. N- phenylcarbamoyl; an amino group; an alkylamino group e.g. methylamino or N,N-dimethylamino; an arylamino group e.g. phenylamino or naphthylamino; an acylamino group e.g. acetylamino or benzoylamino; an alkoxycarbonylamino group e.g. ethyoxycarbonylamino; an aryloxycarbonylamino group e.g. phenoxycarbonylamino; an acyloxy group e.g. acetyloxy or benzoyloxy; an alkylaminocarbonyloxy group e.g. methylaminocarbonyloxy; an arylaminocarbonyloxy group e.g. phenyl-aminocarbonyloxy; a sulfo group; a sulfamoyl group; an alkylsulfamoyi group e.g. methylsulfamoyi; and an arylsulfamoyi group e.g. phenylsulfamoyi.

Examples of the substituent represented by R include sulfonyl group (e.g. methanesulfonyl or toluenesulfonyl), an acyl group (e.g. acetyl or trifluoroacetyl) or an oxalyl group (e.g. ethoxalyl).

Of the compounds of the general formula (III), those represented by the following general formula (IIIa) are particularly preferable:

$$\begin{array}{c|c}
R_2 & Y & O \\
N-N-C-NH-R_5-NHNHC-R_6 \\
R_3 & R_4
\end{array}$$
(IIIa)

where R₂, R₃ and R₄ each represents a hydrogen atom, an optionally substituted aliphatic group, e.g. an alkyl group such as methyl, ethyl, buthyl, 3-aryloxypropyl or cyclohexyl; an optionally substituted aromatic group, e.g. a phenyl or naphthyl group; an optionally substituted heterocyclic group, e.g. a pyridyl or pyrrolidyl group; an optionally substituted alkoxy group, e.g. a methoxy, ethoxy or buthoxy group; or an optionally substituted aryloxy group, e.g. a phenoxy or 4pmethylphenoxy group.

Preferable as R_2 and R_3 are an alkoxy group and a substituted alkyl group, the substituent being an alkoxy or aryl group or the like.

Preferable as R₄ is a hydrogen atom or an alkyl group.

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R₅ represents a divalent aromatic group, e.g. a phenylene or naphthylene group, and Y represents a sulfur atom or an oxygen atom.

 R_6 represents an optionally substituted alkyl, alkoxy or amino group, the substituent being an alkoxy, cyano or aryl group or the like.

The following are typical examples of the compound represented by the general formula (III):

$$t - C_{5}H_{1T}$$

$$t - C_{5}H_{1T}$$

$$O(CH_{2})_{3}NHCONH$$

$$NHNHCO(CH_{2})_{3}O$$

$$OCH_{3}$$

$$t - C_{5}H_{11}$$

$$t - C_{5}H_{11}$$

$$0(CH_{2})_{4}NHCONH - NHNHCOCH_{2}OCH_{3}$$

$$\begin{array}{c} II - 6 \\ t - C_5 H_1, \\ \downarrow \\ C_2 H_5 \end{array}$$

II - 7

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 $C_{16}H_{33}SO_{2}NH$ — NHNHCOCH₂CH₂NHCOOCH₃

$$III - 17$$
 $C_2 H_5 N H C S N H \longrightarrow NHNHCOCH_2 S \longrightarrow$

$$\begin{array}{c} \mathbb{I} - 18 \\ \text{C}_{12} \text{H}_{25} \text{N} \text{H} \text{COO} & \\ & \\ \end{array} \\ \begin{array}{c} \text{N} - \text{NHNHCOCH}_2 - \text{S} & \\ \text{N} - \\ \end{array}$$

$$III - 22$$
 S O $IIII - 22$ S O $IIIII - 20$ C $IIII - 20$

-33

²⁵ **∏** −34

$$\begin{array}{c|c}
C H_{2} & \longrightarrow & C H_{2} \\
\hline
C H_{2} & \longrightarrow & C H_{2}
\end{array}$$

$$\begin{array}{c|c}
O & O \\
\parallel & \parallel \\
N-N H C N H
\end{array}$$

$$\begin{array}{c|c}
N+N H C H_{2} C H_{$$

35 Ⅲ −35

I −37

II −38

− 39

Ⅲ - 40

II −48

O

N−N H C N H

N+N+COCH

1 − 49

O

N−N H C N H − NHNHCCH₂OCH

C H 3

Compound III-5 may be synthesized by the following procedure.

Synthesis of compound III-5

Scheme of synthesis:

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$$NO_2 \xrightarrow{\text{CH}_3 \text{OCH}_2 \text{COOC}_2 \text{H}_5} NO_2 \xrightarrow{\text{NHNH}_2 \text{CH}_3 \text{OCH}_2 \text{OCH}_3} NO_2 \xrightarrow{\text{NHNHCOCH}_2 \text{OCH}_3} NO_2$$

$$\xrightarrow{C_2H_5NCS} C_2H_5NHCSNH- \longrightarrow NHNHCOCH_2OCH_3$$

Compound III-5 can be synthesized as in the case of synthesis of compound II-45.

One or more of the compounds (I), (II) and (III) are contained in the silver halide photographic material of the present invention in a total amount which preferably ranges from 5×10^{-7} to 5×10^{-1} moles, more preferably from 5×10^{-5} to 1×10^{-2} mole, per mole of silver halide contained in said photographic material

Hydrazide compounds may be incorporated in either the light-sensitive material or a developer or both in accordance with the present invention. If they are to be incorporated in the light-sensitive material, it may

be contained in a light-sensitive silver halide emulsion layer containing a 3-pyrazolidone compound and/or a trihydroxybenzene compound and/or in at least one of other layers coated on the support in such a way that the incorporated hydrazide compounds will have diffused into said emulsion layer by the time development starts. If hydrazide compounds are to be incorporated in the light-sensitive material, they are suitably used in amounts ranging from 10^{-6} to 10^{-1} mole per mole of silver halide, preferably from 10^{-4} to 10^{-2} mole per mole of silver halide. A suitable amount can be determined in consideration of the balance with other factors such as the silver halide composition, the size of silver halide grains, the degree of their chemical ripening, the amount of a hydrophilic colloid used as a binder, and the amounts of other additives such as a stabilizer, a restrainer and an accelerator.

If hydrazide compounds are to be incorporated in a developer, they are usually added in amounts ranging from 10^{-5} to 10^{-2} mole, preferably from 10^{-4} to 10^{-3} mole, per liter of the developer. A suitable amount is determined in consideration of the balance with other factors such as the amino compound used, pH and fog restrainer used.

The general formula (IV) is as follows:

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15 R_1 - 0 (CH_2CH_2O)_0H (IV)
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Where R₁ is a hydrogen atom or a substituted or unsubstituted aromatic ring; and n is an integer of 10 -200.

The following are non-limiting preferred examples of the compound represented by the general formula (IV):

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20 IV - 1 HO(CH_2CH_2O)nH n = 10

IV - 2 HO(CH_2CH_2O)nH n = 30

IV - 3 HO(CH_2CH_2O)nH n = 50

IV - 4 HO(CH_2CH_2O)nH n = 70

IV - 5 HO(CH_2CH_2O)nH n = 150

25 IV - 6 HO(CH_2CH_2O)nH n = 200
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$$V-7$$

$$t - C_8 H_{17} - O - (C H_2 C H_2 O) n H$$
 $(n = 20)$

V - 8

$$C_9 H_{19} - (C H_2 C H_2 O)_n H$$
 (n=13)

N-9

$$C_9 H_{19} - (C H_2 C H_2 O) n H$$
 $(n = 20)$

[V - 10]

$$C_9 H_{19} - O - (C H_2 C H_2 O) n H$$
 $(n=35)$

N - 11

$$C_{12}H_{25}$$
 \longrightarrow $O-(CH_{2}CH_{2}O)nH$ $(n=20)$

V - 12

$$C_{20}H_{4T}$$
 $O-(CH_{2}CH_{2}O)nH$ $(n=50)$

$$V - 13$$
 $C_9 H_{19} \longrightarrow O - (C H_2 C H_2 O)_n H \qquad (n = 30)$

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$$V = 14$$

$$C_{6}H_{19} \longrightarrow O - (CH_{2}CH_{2}O)nH \qquad (n = 10)$$

$$CH_{3}$$

$$V = 15$$

$$C_{9}H_{19} \longrightarrow O - (CH_{2}CH_{2}O)nH \qquad (n = 30)$$

$$V = 16$$

$$C_{6}H_{13} \longrightarrow O - (CH_{2}CH_{2}O)nH \qquad (n = 25)$$

$$V = 17$$

$$C_{9}H_{19} \longrightarrow O - (CH_{2}CH_{2}O)nH \qquad (n = 50)$$

Of the above compounds of the general formula (IV), those of IV-1 to IV-6, IV-9 and IV-10 are more preferable, particularly IV-3, IV-9 and IV-10.

These compounds are readily available from the commercial market. They are preferably added in an amount of 0.01 - 4.0 moles, more preferably from 0.02 to 2 moles, per mole of silver halide. Two or more compounds having different values of n may be used in admixture.

The silver halide emulsion to be used in the light-sensitive material of the present invention may employ various silver halides such as silver chloride, silver bromide, silver chlorobromide and silver chloroiodobromide. The present invention proves effective with silver bromide and silver iodobromide, with particular advantage being exhibited with a high-sensitivity light-sensitive material having a small content (≤5 mol%, particularly 1.5 to 3 mol%) of silver iodide. The present invention proves particularly effective with an emulsion obtained by adding a water-soluble iodide at the stage of adjustment.

The silver halide emulsion to be used in the present invention may be such that silver halide grains are suspended in a hydrophilic colloid by a known method such as the neutral method, ammoniacal method, single-jet method or double-jet method.

The silver halide emulsion layer in the light-sensitive material of the present invention preferably contains a silver halide composed of grains having an average size of $0.1 - 1.0 \mu m$, more preferably $0.1 - 0.7 \mu m$, with at least 75%, preferably at least 80%, of the total grains having a grain size ranging from 0.7 - 1.3 times the average grain size.

A silver halide emulsion having polyvalent metallic ions (e.g. iridium or rhodium) occluded therein may also be used and examples of such emulsion are described in U.S. Patent 3,271,157, 3,447,927 and 3,531,291. Silver halide emulsions used in the present invention may be chemically sensitized by a common method such as the use of sulfur compounds or gold compounds exemplified by chloroaurates and gold trichloride.

The silver halide emulsion to be used in the present invention may be treated with spectral sensitizers to have sensitivity to colors in desired spectral ranges of wavelength. Common spectral sensitizers may be used for this purpose and they include methine and styryl dyes such as cyanine, hemicyanine, rhodacyanine, merocyanine, exanole and hemioxonole. Regarding spectral sensitizers, reference may be made to various prior patents such as U.S. Patent Nos. 2,742,833, 2,756,148, 3,567,458, 3,615,517, 3,615,519, 3,632,340, 3,155,519, 3,384,485, 4,232,115, 2,796,580, 4,028,110, 3,752,673 and JP-A-55-45015.

The silver halide photographic emulsion in the light-sensitive material of the present invention may be used with common hardeners such as aldehydes (e.g. formaldehyde, glyoxal, glutaraldehyde and mucochloric acid), N-methylol compounds (e.g. dimethylolurea and methylol dimethyl hydantoin), dioxane derivatives (e.g. 2,3-dihydroxydioxane), active vinyl compounds (e.g. 1,3,5-triacryloylhexahydro-s-triazine and bis (vinylsulfonyl) methyl ether), and active halide (e.g. 2,4-dichloro-6-hydroxy-s-triazine), which may be used

either on their own or as admixtures. The emulsion may also be used with common thickeners, matting agents and coating aids. Suitable binders are hydrophilic binders that work as protective colloids and which are commonly used in the art.

The light-sensitive material of the present invention may further contain various other additives such as couplers, ultraviolet absorbers, brighteners, image stabilizers, antioxidants, lubricants, metallic ion sequestering agents and emulsifying/dispersing agents, which are selected in accordance with the specific object of interest.

Besides the silver halide emulsion layer, the light-sensitive material of the present invention may have other photographic layers such as a protective layer, an intermediate layer, a filter layer, an anti-halation layer, a subbing layer, an auxiliary layer, an anti-irradiation layer and a backing layer. Various supports may be employed as selected from among, for example, baryta paper, polyethylene-coated paper, cellulose acetate, cellulose nitrate and polyethylene terephthalate, in accordance with the specific use of the light-sensitive material.

Polymer latices made of homo- or copolymers of monomeric compounds such as alkyl acrylates, alkyl methacrylates, acrylic acid or glycidyl acrylate may be contained in the light-sensitive silver halide emulsion layer or some other layers coated on the support in order to attain certain purposes such as improvements in the dimensional stability of the photographic material and in its film properties.

A stabilizer or an anti-foggant may be contained in the light-sensitive silver halide emulsion, with a suitable one being selected from among many compounds including 4-hydroxy-6-methyl-1,3,3a, 7-tetraazaindene, 1-phenyl-5-mercaptotetrazole, resorcinol oxime and hydroquinone aldoxime. These compounds can be incorporated in amounts ranging from 10^{-4} to 10^{-1} mole per mole of silver halide.

The following examples are provided for the purpose of further illustrating the present invention but are in no way to be taken as limiting.

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Example 1

Preparation of silver halide photographic materials:

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A polyethylene terephthalate film with a thickness of 100 µm that had a subbing layer (see Example 1 of JP-A-59-19941) on each sides in a thickness of 0.1 µm was used as a support. The subbing layer on one side was coated with a silver halide emulsion layer (for its composition, see recipe (1) below) to give a gelatin and a silver deposit of 2.0 g/m² and 3.5 g/m², respectively. The resulting emulsion layer was coated with a protective layer according to recipe (2) shown below, to give a gelatin deposit of 1.5 g/m². The subbing layer on the other side was coated with a backing layer according to recipe (3) shown below, to give a gelatin deposit of 2.7 g/m². The backing layer was coated with a protective layer according to recipe (4) shown below, to give a gelatin deposit of 1 g/m². By these procedures, sample Nos. 1 - 17 of photographic material were prepared.

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Recipe (1) (formulation of silver halide emulsion layer) $2.0 \, \text{g/m}^2$ Gelatin $3.5 \, g/m^2$ Silver iodobromide emulsion (a) (Silver iodobromide emulsion (a) was prepared by the double-jet method and it comprised AgIBr grains having an average size of 0.20 μm and consisting of 98 mol% AgBr and 2 mol% AgI. In the mixing step, K2IrCl6 was added in an amount of 6 \times 10-7 moles per mole of Ag. This emulsion was washed and desalted in the usual manner and subsequently treated with an aqueous solution of KI to adjust its pAg to 8.80 at 40°C.) 8 mg/m^2

(CH₂)₃SO₃⊖

Spectral sensitizer

 25 mg/m^2 Hydroquinone 30 mg/m^2 4-Methyl-6-hydroxy-1,3,3a,7-tetrazaindene Compound within the scope of the present invention or comparative compound (see Table 1) $0.1 \, g/m^2$ Saponin

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	Recipe (2) (formulation of emulsion protective lay	ver)
	Gelatin	1.5 g/m^2
5	Matting agent (particulate polymethyl	
	methacrylate with an average grain	
40	size of $3.0 - 5.0 \mu m$)	0.05 g/m^2
10	Colloidal silica	0.02 g/m^2
	Surfactant A CH2COOCH2CH(C2H5)C4H9	
15	CHCOOCH2CH(C2H5)C4H9	
	Surfactant A CH2COOCH2CH(C2H5)C,H9 CHCOOCH2CH(C2H5)C,H9 SO3Na	0.01 g/m^2
22	Hardener: (CH ₂ =CHSO ₂ CH ₂) ₂ O	0.10 g/m^2
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Recipe (3) (formulation of backing layer) 2.7 g/m^2 Gelatin 5 Dye (d) 10 0.12 g/m^2 SOaNa 15 Dye (e) CH = CH - CH =20 0.03 g/m^2 25 SO₃Na Dye (f) 30 0.04 g/m^2 35 CH2SO3Na 40 0.1 g/m^2 Surfactant (saponin) 0.01 g/m^2 Sodium dodecylbenzenesulfonate 0.01 g/m^2 45 Hardener (glyoxal)

501	Recipe (4) (formulation of backing protective layer)										
Gelat Matti Surfa	itin ing agent (particulate polymethyl methacrylate with an average grain size of 3.0 - 5.0 μm) actant A dener (glyoxal)	1 g/m ² 0.05 g/m ² 0.01 g/m ² 0.01 g/m ²									

The samples prepared were subjected to a halftone dot quality test by the following method.

Halftone dot quality test

A contact screen (50% dot area; 150 lines per inch) was partly attached to a step wedge and a sample placed in close proximity with the contact screen was exposed to light from a xenon lamp for 5 seconds. The exposed sample was then developed with a rapid access automatic processor supplied with a developer having the formulation shown below. The quality of halftone dots in the processed samples was observed with a magnifier (X 100) and rated by a five-score method, with the best sample (having the highest quality of halftone dots) being given score 5 and followed by 4, 3, 2 and 1 in the decreasing order of quality. Samples rated "1" and "2" were unacceptable for practical purposes.

The samples were also evaluated for the presence of black peppers in halftone dots; the sample which was entirely free from black peppers in halftone dots was given a maximum score 5 and followed by 4, 3, 2 and 1 in the increasing order of the frequency of black peppers that developed in halftone dots. Samples rated "1" and "2" were unacceptable for practical purposes because of the great size of black peppers that developed.

Density measurements were conducted on the samples with a digital densitometer (Model PDP-65 of Konica Corp.) and the results were evaluated in terms of relative values, with the value for sample No. 1 at density 3.0 being taken as 100. Gamma values were also calculated (γ = the tangent of angle α formed by the horizontal axis and the straight-line portion of the characteristic curve connecting two points at densities of 0.3 and 3.0).

Formulation of developer	
Ethylenediaminetetraacetic acid disodium salt	1 g
Sodium sulfite	60 g
Hydroquinone	35 g
5-Amino-1-pentanol	50 g
Sodium bromide	2.5 g
5-Methylbenzotriazole	0.3 g
1-Phenyl-3-pyrazolidone	0.2 g
Water	to make 1,000 ml
pH adjusted to 11.5 with sodium hydroxide	

Formulation of fixing solution	
Recipe A:	
Ammonium thiosulfate (72.5% W/V aq. sol.) Sodium sulfite Sodium acetate (3H ₂ O) Boric acid Sodium citrate (2H ₂ O)	240 ml 17 g 6.5 g 6 g 2 g
Recipe B:	
Pure water (ion-exchanged water) Sulfuric acid (50% W/V aq. sol.) Aluminum sulfate (8.1% W/V aq. sol. in terms of Al ₂ O ₃ content)	17 ml 4.7 g 26.5 g

Recipes A and B were successively dissolved in 500 ml of water and the solution was worked up to 1,000 ml to be used as a fixing solution. The fixing solution as prepared was adjusted to a pH of 6 with acetic acid.

Processing scheme											
Steps	Temperature	Time									
Development Fixing Rinsing Drying	40 ° C 35 ° C 30 ° C 50 ° C	15 sec 10 sec 10 sec 10 sec									

The following compounds (a) to (c) were added as comparative compounds to the silver halide emulsion layer in Recipe (1).

(e)
$$C_{2}H_{5}$$

$$tC_{5}H_{1} \longrightarrow OCHCONH \longrightarrow NHNHCHO$$

	Remarks			comparison	do.	do.	do.	do.	the invention	do.	do.	do.	do.	og Og	<u>.</u>		j <u>c</u>		oomoonioon	the invention
	Black			3	က	က	က	4	5	5	5	4	2	4	ۍ.	- 52	יט נ	ຸນເ) c	v ro
	Quality of			6	က	က	က	4	5	5	4	4	ಬ	2	5	2	2	. rc	- er	ch c
	Gamma			10.0	10.9	10.1	10.2	15.3	16.7	16.5	15.5	15.5	15.7	15.8	15.3	15.8	15.7	15.8	40	16.7
le 1	Relative sensitivity	`		100	101	105	105	110	137	130	121	122	125	128	121	131	130	128	30	138
Table	rmula (IV)	Amount,	mg/m ²	•	,	1	100	1	100	100	100	100	100	100	100	100	100	100	1	100
	Compound formula (IV)	Identification		1	1	1	IV-4	ı	IV-4	IV-3	1V-8	6-71	IV-4	IV-4	IV-4	IV-4	IV-4	1V-4	ı	IV-4
	punodwo	Amount,	mg/m²	35	35	35	35	35	35	35	35	35	35	35	35	35	35	35	1	35
	Hydrazide co	Identification		a	a	ပ	ပ	E-13	8-111	E-13	£-11	E-3	1-10	1-5	11-13	111-16	III-20	111-22	1	111-35
	Sample No.			_	7	က	4	S.	9	_	89	<u>ග</u>	10	7	12	13	14	15	16	17

The results of Example 1 are shown in Table 1.

As is clear from Table 1, sample Nos. 6 - 15 and 17 according to the present invention had higher sensitivity and formed more contrasty image with better quality of halftone dots and fewer black peppers than the comparative samples.

Example 2

The procedure of Example 1 was repeated except that the samples prepared in it were processed with a developer having the formulation described below. The results are shown in Table 2.

Formulation of developer	
Hydroquinone	45.0 g
N-Methyl-p-aminophenol hemisulfate	0.8 g
Sodium hydroxide	18.0 g
Potassium hydroxide	55.0 g
5-Sulfosalicylic acid	45.0 g
Boric acid	25.0 g
Potassium sulfite	110.g g
Ethylenediaminetetraacetic acid disodium salt	1.0 g
Potassium bromide	6.0 g
5-Methylbenzotriazole	15.0 g
n-Butyl diethanolamine	15. g
Water	to make 1,000 ml
pH adjusted to 11.6	

	Remarks				comparison	do.	do.	do.	the invention	do.	do.	do.	do,	do.	i q	do .	; c	<u> </u>	900000000000000000000000000000000000000	the invention
	Black	pepper	•		0	۱ ۵	7	က	5	2	4	5	4	4	4	ſ.		, 4	. بر) (°	
	Quality of	halftone dots			2	2	2	2	က	S	S	S	4	4	4	2	4	4	rc:) 	· 2
	Gamma				9.0	9.5	8.8	9.6	12.1	16.1	15.5	15.3	16.2	16.3	16.2	15.9	15.8	15.9	15.7	4.0	16.3
Table 2	Relative	sensitivity			100	102	103	103	103	135	133	130	138	120	125	125	122	123	121	40	136
Tab	rmula (IV)		Amount,	mg/m ²	,	1	1	100	1	100	100	100	100	100	100	100	100	100	100	1	100
	Compound formula (IV)		Identification		ı	ı	,	1V-4	•	IV-4	IV-3	1V-8	1V-9	IV-4	1V-4	IV-4	IV-4	IV-4	IV-4	•	IV-4
	punodwo		Amount,	mg/m²	35	35	35	35	35	35	35	35	35	35	35	35	35	35	35	1	35
	Hydrazide compound		Identification		В	q	ပ	ပ	II-3	E-III-3	£-3		£- <u>⊞</u>	l-10	9-11	II-13	111-16	111-20	111-22		111-35
	Sample	S			-	2	က	4	5	9	7	8	6	10	=	12	13	14	15	16	17

Like Table 1, Table 2 shows that sample Nos. 6 - 15 and 17 according to the present invention had higher sensitivity and formed more contrasty image with better quality of halftone dots and fewer black peppers than the comparative samples.

Example 3

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Silver halide (90 mol% AgCl and 10 mol% AgBr) grains having an average size of 0.08 μ m and containing rhodium in an amount of 2 \times 10⁻⁶ mole per mole of silver halide were sulfur-sensitized in the usual manner. Thereafter, a stabilizer or 6-methyl-4-hydroyx-1,3,31,7-tetraazaindene (570 mg per mole of silver halide) and gelatin (25 g) were added. The coating solutions for the necessary layers had the following formulations. The emulsion layer was applied to give a silver deposit of 3.5 g/m².

Recipe (1) (formulation of silver halide emulsion layer)

Saponin

 100 mg/m^2

Potassium bromide

 3 mg/m^2

Desensitizing dye

 $CH_3 CH_3$ $SO_2 CH_3$ 1 mg/m^2

N (CH₂),50₃Θ

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	Sodium hydroxide	10 mg/m^2
	Compound according to the present	
5	invention [as represented by the	
	general formula (I), (II), (III) or	
10	(IV) or comparative compound	(see Table 3)
70	Sodium dodecylbenzenesulfonate	21 mg/m^2
	Butyl acrylate/styrene/acrylic	
15	acid copolymel latex	1 g/m^2
	5-Methyl benzotriazole	10 mg/m^2
	5-Phenyl-1-mercaptotetrazole	11.5 mg/m^2
20	2-Mercaptobenzimidazolw-5-sulfonic acid	1 mg/m^2
	Benzyl-triphenyl phosphonium chloride	5 mg/m^2
25	Compound (Z)	5.8 mg/m ²
	Company (7)	

Compound (Z)

30 C H 3 C H 3 O H
H O C H 3 C H 3

Formulation of emulsion protective layer

Gelatin 1.14 g/m^2 5 Compound (M) (see below) 32.7 mg/m^2 20 mg/m^2 Amorphous silica Compound (S) (see below) $0.5 \, g/m^2$ 10 Citric acid $5.4 \, \text{g/m}^2$ 71.5 mg/m^2 Formaldehyde 100 mg/m^2 Compound (K) (see below) 15

Compound (M)

Compound (S)

$$CH_2 O-CH_2(CH_2)_8CH_3$$
 $N_8O_3S-CH OCH_2CH_2CH(CH_3)_2$
 $C\ell$
 S
 N_7H

Compound (K)

CH₃ CH C_2H C_2H

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Backing layer

Latex polymer (butyl acrylate/styrene

copolymer)

 $0.5 \, g/m^2$

Sponin

 200 mg/m^2

Backing dye (g)

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CH, CH CH CH S

200 mg/m²

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Dye (e) (as used in Example 1)

 70 mg/m^2

Ossein gelatin

 2.0 g/m^2

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Backing protective layer	
Dioctyl sulfosuccinate ester	300 mg/m ²
Matting agent (particulate methyl methacrylate with an average grain size	e of 4.0 μm) 100 mg/m²
Ossein gelatin (isoelectric point, 4.9)	1.1 g/m ²
Fluorinated sodium dodecylbenzenesulfonate	50 mg/m ²

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The samples prepared in accordance with the recipes shown above were exposed under a quartz iodine light source (halogen lamp) at 1.5 kW for 10 seconds. The discharge tube was a light source having a substantial energy intensity at 400 nm and longer wavelengths. The samples as prepared were such that they could be handled under illumination with an ordinary fluorescent lamp at 300 lux. The exposed samples were developed and subsequently processed as in Example 1 and subjected to similar tests. The test results are shown in Table 3, from which one can see that sample Nos. 6 - 15 and 17 according to the present invention had higher sensitivity and formed more contrasty image with better quality of halftone dots and fewer black peppers than the comparative samples.

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•	Remarks	- <u>-</u>	comparison	do.	do.	do.	do.	the invention	do.	do.	g G	do.	do.	do.	do.	go.	do.	comparison	the invention
	Black	•	6	1 01	2	က	2	9	4	2	4	4	4	2	2	5	5	6	2
	Quality of halftone dots		2	8	8	7	ෆ	2	5	2	4	4	4	ည	4	4	5	,	လ
	Gamma		7.0	7.1	7.8	7.9	8.0	16.1	16.2	17.3	17.1	16.0	15.9	18.1	17.1	16.1	15.8	3.9	17.4
le 3	Relative sensitivity		100	101	101	102	103	131	130	132	129	128	130	125	121	120	125	40	135
Table 3	rmula (IV)	Amount,	1	•	•	100	•	100	100	100	100	100	100	100	100	100	100	•	100
	Compound formula (IV)	Identification	-	ı	,	IV-4	t	1V-4	14-3	10-8	6-71	IV-4	1V-4	IV-4	IV-4	1V-4	1V-4	1	IV-4
	punodu	Amount,	35	35	35	35	35	35	35	35	35	35	35	35	35	35	35	1	35
	Hydrazide compound	Identification	а	q	O	ပ	III-3	E-III	E-⊞-3	E-III	E-⊞	1-10	1-5	1-13	III-16	111-20	111-22	•	111-35
	Sample No.		-	2	ო	4	5	9	7	8	6	01	=	12	13	14	15	16	17

Example 4

Silver halide (98 mole% AgCl and 2 mol% AgBr) grains having an average size of 0.17 pm and containing rhodium in an amount of 1.0×10^{-4} mole per mole of silver halide were sulfur-sensitized in the usual manner. Thereafter, 6-methyl-4-hydroxyl,3,3a, 7-tetraazaindene was added as a stabilizer. The coating solutions for the necessary layers had the following formulations. The emulsion layer was applied to give a silver deposit of 3.0 g/mz.

Recipe (1) (formulation of silver halide emulsion layer)	
Compound (S) (as used in Example 3)	1 mg/m²
Sodium hydroxide	14
	mg/m²
Compound according to the present invention [as represented by	(see
the general formula (I), (II), (III) or (IV)] or comparative compound	Table 4)
Saponin	120
	mg/m²
Sodium dodecylbenzenesulfonate	80
	mg/m²
Butyl acrylate/styrene/acrylic acid copolymer latex	1100
	mg/m²
5-Methylbenzotriazole	10
	mg/m²
Compound (Z) (as used in Example 3)	10
	mg/m²
Aqueous styrene/maleic acid copolymer	50
	mg/m²

Formulation of emulsion protective layer	
Gelatin	1.1 g/m ²
Compound (M) (as used in Example 3)	30 mg/m ²
Amorphous silica	20 mg/m ²
Compound (K) (as used in Example 3)	120 mg/m ²
Gallic acid	70 mg/m ²
Citric acid	5 mg/m ²
Formaldelyde	75 mg/m ²

	Backing layer	
	Gelatin	2.0 g/m ²
50	Dye (e)	20 mg/m ²
	Dye (d)	160 mg/m ²
	Dye (f)	90 mg/m ²
	Dye (g)	100 mg/m ²
	Butyl acrylate/styrene copolymer	300 mg/m ²
55	Saponin	120 mg/m ²

Backing protective layer

Same as in Example 3.

The samples prepared in accordance with the recipes shown above were exposed with a room-light printer (Model P-627-FM of Danippon Screen Mfg. Co., Ltd.) through an ordinary wedge. The exposed damples were developed and subsequently processed as in Example 1 and subjected to similar tests. The test results are shown in Table 4, from which one can see that sample Nos. 6 - 15 and 17 according to the present invention had higher sensitivity and formed more contrasty image with better quality of halftone dots and fewer black peppers than the comparative samples.

	Remarks				comparison	do	. c	. .	j e	the invention	do.	do.	do.	do.	do.	d o.	do.	do.	qo	Comparison	the invention
	Black	pepper			0	۰ م	2	ا م	m	2	2	5	2	5	2	5	4	. 73	, ro	· er	2
	Quality of	halftone dots			2	1 (1)	က	2	က	ಬ	ည	2	S.	5	4	4	4	2	2	-	5
	Gamma				8.1	8.3	7.9	8.8	8.5	16.3	15.9	16.2	16.1	16.2	16.3	15.5	15.3	15.9	15.7	5.0	16.5
le 4	Relative	sensitivity			100	66	86	102	102	131	132	135	129	128	127	123	128	131	127	30	135
Table 4	rmula (IV)		Amount,	mg/m ²		ı	ı	100	ı	100	100	100	100	100	100	100	100	100	100		100
	Compound formula (IV)		Identification		ı	ı	ı	1V-4	ı	IV-4	17-3	1.7-8	6-/1	1V-4	IV-4	1V-4	1V-4	1V-4	1V-4	ı	IV-4
	punodwo		Amount,	mg/m²	35	35	35	35	35	35	35	35	35	35	35	35	35	35	35	1	35
	Hydrazide cor		Identification		a	q	ပ	ပ	E-3	€-=	E-3	E-∃	e-3 ==	F-10	1-5	II-13	111-16	111-20	111-22		111-35
	Sample	S			,	2	ო	4	9	9		80	ත 	9	=	12	13	14	15	91	17

Example 5

Using the silver halide grains described below, emulsions were prepared as in Example 1 except that compounds within the scope of the present invention or comparative compounds were added as shown in Table 5. The silver halide grains used were cubic, monodispersed silver iodobromide grains (grains A) having an average size of 0.3 µm and containing 1 mol% Agl and 10⁻⁷ mole of iridium per mole of Agl, 10 and sulfur-sensitized cubic, monodispersed silver iodobromide grains (grains B) having an average size of 0.15 µm and containing 0.1 mol% Agl. The two types of grains were mixed at a ratio of 1:1 to prepare emulsions.

Using these emulsions, light-sensitive materials were fabricated and exposed under an iodine-filled light source (i.e., an iodine light source) having a tungsten filament. The exposed samples were developed and 15 subsequently processed as in Example 1. The samples were then subjected to similar tests, the results of which are shown in Table 5.

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	Remarks			comparison	do.	do.	do.	do.	the invention	do.	do.	do.	do.	do.	do.	do.	do.	do.	comparison	the invention
	Black			2	2	က	2	က	വ	2	ည	വ	2	S	S	2	5	5	e	2
	Quality of halftone dots			2	2	2	2	က	2	S	2	2	4	4	4	വ	4	S	,	5
	Gamma			8.7	9.0	9.1	9.2	8.8	15.9	16.3	16.1	16.0	15.9	15.3	16.1	16.2	16.1	15.8	4.8	16.8
Table 5	Relative sensitivity			100	66	102	103	101	129	128	130	131	132	131	129	125	133	129	46	132
Tab	rmula (IV)	Amount,	mg/m²	•	,	1	100	,	100	100	100	100	100	100	100	100	100	100	ı	100
	Compound formula (IV)	Identification		•	ı	1	IV-4	ı	IV-4	1/-3	IV-8	6-VI	IV-4	1V-4	IV-4	IV-4	1V-4	1V-4		IV-4
	punoduo	Amount,	mg/m²	35	35	35	35	35	35	35	35	35	35	35	35	35	35	35	1	35
	Hydrazide cor	Identification		ď	q	ပ	ပ	II-3	II-3	£-3	E-3	E-H-3	1-10	<u>-</u> -2	11-13	111-116	111-20	111-22	1	111-35
	Sample No.			-	2	က	4	2	9		89	6	10	=	12	13	14	15	16	17

As one can see from Table 5, sample Nos. 6 - 15 and 17 according to the present invention had higher sensitivity and formed more contrasty image with better quality of halftone dots and fewer black peppers than the comparative samples.

Example 6

Emulsions were prepared as in Example 5 except that grains A were replaced by tabular grains (grains C), which were made from Emulsion 1 described on page 13 of JP-A-58-113926. This emulsion was a silver iodobromide emulsion having an average diameter of 2.8 μm and an average thickness of 0.095 μm, with an average aspect ratio being 29.5:1. Grains B used in Example 5 and grains C were mixed at a ratio of 5:1. Using the resulting mixture, light-sensitive materials were fabricated as in Example 4 except that compounds within the scope of the present invention or comparative compounds were used as shown in Table 6. The samples were exposed, developed and subsequently processed. Thereafter, the samples were subjected to a halftone dot quality test and the test results are shown in Table 6, from which one can see that sample Nos. 6 - 15 and 17 according to the present invention and higher sensitivity and formed more contrasty image with better quality of halftone dots and fewer black peppers. It was therefore clear that the compounds within the scope of the present invention produced satisfactory results even when tabular silver halide grains were used instead of cubic grains.

	Remarks				comparison	do	do.	do.	do.	the invention	do.	do.	do.	do.	do.	do.	do.	do.	do	comparison	the invention
	Black	pepper			3	5	2	2	3	4	2	5	2	2	2	4	4	4	S.	· m	വ
	Quality of	halftone dots			2	က	2	က	က	5	2	2	ა	S	4	2	4	4	2	-	5
	Gamma				9.3	8.9	8.5	9.1	9.2	15.9	15.3	13.3	14.8	16.1	16.2	15.5	15.3	14.9	14.8	4.3	16.3
le 6	Relative	sensitivity			100	102	86	66	101	133	132	135	131	129	128	130	131	132	131	45	136
Table 6	rmula (IV)		Amount,	mg/m ²	•	ı		100	ı	100	100	100	100	100	100	100	100	100	100		100
	Compound formula (IV)		Identification		•	ı		IV-4	•	IV-4	1V-3	8-/1	6-/1	IV-4	IV-4	IV-4	IV-4	IV-4	IV-4		IV-4
	punodwo		Amount,	mg/m²	35	35	35	35	35	32	35	35	35	35	35	35	35	35	35	1	35
	Hydrazide compound		Identification		æ	q	ပ	၁	€-=	E-3	II-3	E-III-3	⊪ -3	1-10	11-5	11-13	111-16	III-20	111-22	ı	111-35
	Sample	<u>;</u>			_	2	က	4	2	9	_	80	6	10	=	15	13	14	15	91	17

Example 7

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Emulsions were prepared as in Example 1. The composition of the silver halide emulsion layer used in this example was the same as what was employed in Example 1 except that the cyanine dye and spectral sensitizer were replaced by the following dye and compound, respectively, with each being added in an amount of 1 mg/m²;

Dye:

$$CH_3$$

Compound:

H CH_2 H

³⁵ Formulation of emulsion protective layer

Same as in Example 1.

 150 mg/m^2

 10 mg/m^2

Formulation of backing layer

	Gelatin	3	g/m^2
5	Dye (d)	80	mg/m^2
	Dye (e)	40	mg/m²
10	Dye		
15	$ \begin{array}{c c} H \\ N \\ N \\ SO_3K \end{array} $ Fe	100	mg/m²
20		120	/ 7
	Saponin	130	mg/m ²
	Latex polymer (butyl acrylate/styrene		

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Formulation of backing protective layer

Glyoxal

copolymer)

Same as in Example 1.

The samples of light-sensitive material prepared were exposed for 10⁻⁵ second with an apparatus using as a light source an infrared semiconductor laser emitting light having a main peak at 780 nm. The exposed samples were developed and subsequently processed as in Example 1. The processed samples were subjected to similar tests and the test results are shown in Table 7.

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	Remarks			comparison					the invention										comparison	the invention
	Re			comp	op	do.	do.	do.	the in	do.	do.	do.	do.	do.	go.	do.	9	do.	Comp	the in
	Black pepper			2	က	2	5	က	5	2	2	5	5	5	2	4	. ro	့	· m	2
	Quality of halftone dots			က	2	7	2	3	5	2	S	2	ß	4	5	2	2	2	_	ಎ
	Gamma			7.7	7.3	8.1	7.7	8.9	16.6	16.3	16.1	16.4	16.3	15.8	16.1	15.9	16.3	16.2	4.9	16.9
able 7	Relative sensitivity			100	101	88	66	86	136	132	131	133	130	133	129	128	133	130	40	135
lab	rmula (IV)	Amount,	mg/m-	•	•	•	100	1	100	100	100	100	100	100	100	100	100	100	•	100
	Compound formula (IV)	Identification		ı	1	1	IV-4	ı	IV-4	1٧-3	IV-8	6-71	17-4	1V-4	IV-4	10-4	IV-4	IV-4	ı	IV-4
	ompound	Amount,	-mg/m	35	35	35	32	35	35	35	35	35	35	35	35	35	35	35	ı	35
	Hydrazide compound	Identification		a	q	ပ	ပ	III-3	III-3	E-III	E-III-3	E-III	l-10	1-5	11-13	111-16	111-20	III-22		111-35
	Sample No.	-		-	2	က	4	2	9	7	80	တ	9	=	12	13	4.	15	16	17

As one can see from Table 7, sample Nos. 6 - 15 and 17 according to the present invention had higher sensitivity and formed image with better quality of dots and fewer black peppers.

Example 8

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A total of 17 samples were prepared in the same manner as in Example 1 except that the hardener in Recipe (2) was replaced by a hardener which is a combination of compound and formalin as a mentioned below:

ONa

N N

Cl
N Cl

Formalin 30 mg/m²

With the samples thus prepared, the same results as in Example 1 were obtained.

As will be understood from the foregoing description and data, the present invention provides a silver halide photographic material that has high sensitivity and forms a contrasty image with good quality of halftone dots and few black peppers.

Claims

1. In a silver halide photographic material having at least one light-sensitive silver halide emulsion layer, the improvement wherein said photographic material contains at least one of the compounds represented by the general formulas (I), (II) and (III) shown below and at least one compound represented by the general formula (IV) also shown below:

$$(R)n = \begin{pmatrix} CONHNH-R_1 \\ (CO)mNHNH-R_2 \end{pmatrix}$$
 (I)

(where R_1 and R_2 each represents an aryl group or a heterocyclic group; R is an organic linkage; n is 0 - 6; and m is 0 or 1, provided that when n is 2 or more, R may be the same or different);

(where R₁ is an aliphatic, aromatic or heterocyclic group; R₂ is a hydrogen atom or an optionally substituted alkoxy, heterocyclicoxy, amino or aryloxy group; P₁ and P₂ each represents a hydrogen atom, an acyl group or a sulfinic acid group);

(where Ar is an aryl group containing at least one of a non-diffusible group or a silver halide adsorption accelerating group; R₁ is an optionally substituted alkyl, alkoxy or amino group; R' is a hydrogen atom or a substituent);

 $H_1-O(CH_2CH_2O)nH$ (IV)

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(where R₁ is a hydrogen atom or a substituted or unsubstituted aromatic group; n is an integer of 10 - 200).

- 2. A silver halide photographic material according to claim 1 wherein R_1 and R_2 in the general formula (I) each represents a substituted or unsubstituted phenyl group, n=m=1 and R represents an alkylene group.
- 3. A silver halide photographic material according to claim 1 wherein the compound represented by the general formula (II) is a compound represented by the following general formula (IIa):

$$R_{3}(NR_{4})_{n}CN \xrightarrow{(} R_{6}-L)_{m}R_{7}-NHNHC-C-R_{8} (IIa)$$

$$R_{5}$$

where R_3 and R_4 each represents a hydrogen atom, an optionally substituted alkyl group, or an optionally substituted phenyl, naphthyl, cyclohexyl, pyridyl or pyrrolidyl group; R_5 represents a hydrogen atom or an optionally substituted benzyl, alkoxy or alkyl group; R_6 and R_7 each represents an aromatic group; Y represents a sulfur or oxygen atom; L represents a divalent linkage; R_8 represents -NR $^{'}$ R $^{''}$ or -OR $_9$ (where R $^{''}$, R $^{''}$ and R_9 each represents a hydrogen atom or an optionally substituted alkyl, phenyl or naphthyl group); and m and n each represents 0 or 1.

4. A silver halide photographic material according to claim 1 wherein the compound represented by the general formula (III) is a compound represented by the following general formula (IIIa):

$$\begin{array}{c|c}
R_2 & Y & O \\
N-N-C-NH-R_5-NHNHC-R_6 & (IIIa) \\
R_3 & R_4
\end{array}$$

where R_2 , R_3 and R_4 each represents a hydrogen atom, an optionally substituted aliphatic, aromatic, heterocyclic, alkoxy or aryloxy group; R_5 represents a divalent aromatic group; R_6 represents an optionally substituted alkyl, alkoxy or amino group; Y represents a sulfur or oxygen atom.

5. A silver halide photographic material according to claim 1 wherein the compound represented by the general formula (IV) is at least one compound selected from among the following compounds IV-3, IV-9 and IV-10.

IV-3
$$HO(CH_2CH_2O)nH$$
 $n = 50$

IV-9
$$C_9H_{19}$$
 $O-(CH_2CH_2O)nH$ $(n = 20)$

$$C_{9}H_{19} - (CH_{2}CH_{2}O)_{n}H \qquad (n = 35)$$

- 6. A silver halide photographic material according to claim 1 wherein the silver halide contained in the silver halide emulsion layer is silver bromide or silver iodobromide.
 - 7. A silver halide photographic material according to claim 1 wherein the silver halide contained in the silver halide emulsion layer is composed of grains having an average size of 0.1 1.0 μ m, with at least 75% of the total grains having a grain size ranging from 0.7 to 1.3 times the average grain size.



EUROPEAN SEARCH REPORT

	DOCUMENTS CONS	SIDERED TO BE RE	LEVANT		EP 90301187.2
Category		th indication, where appropri vant passages	ate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CI.')
X	22,25; pa	41 428 -3,5,7-10,12-1 age 37, lines lines 9-22 *	.4,16,	1-7	G 03 C 1/10 G 03 C 1/043 G 03 F 7/06
A	EP - A2 - 0 21 (ANITEC IMAGE * Claims 1, lines 34-	CORPORATION) 29-41; column	}	1,3,6	
					TECHNICAL FIELDS SEARCHED (Int. CI.')
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	The present search report has b	peen drawn up for all claims			
	Place of search	· · · · · · · · · · · · · · · · · · ·	The search	i	Examiner -
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Y: parti docu A: tech O: non-	CATEGORY OF CITED DOCL cularly relevant if taken alone cularly relevant if combined w ment of the same category nological background written disclosure mediate document	JMENTS T: E:	earlier patent of after the filing document cite document cite	ciple underl document, i date d in the app d for other	ying the invention but published on, or plication