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(54) CORROSION-RESISTANT, RARE EARTH-TRANSITION METAL MAGNET AND METHOD OF PRODUCTION THEREOF.

(57) A two-phase permanent magnet which has not only excellent magnetic characteristics but also significantly high corrosion resistance. It is manufactured by use of two magnetically useful phases as the starting materials. One is an $RE_2TM_{14}B$ phase having a high residual flux density; the other is a low-melting RE-TM phase and/or RE-TM-B phase which contributes to the improvement of sinterability and has a function of cleaning the main phase grain boundary and a property of causing the resultant material to be an electrochemically noble composition.

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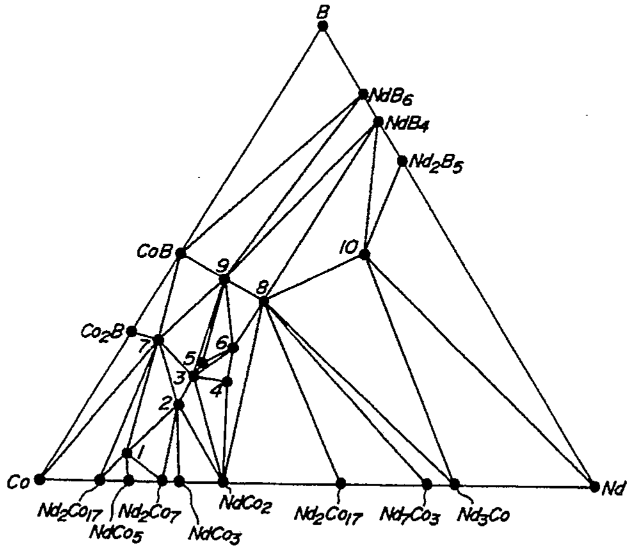
FIG_2



1 --- $Nd_2Co_{14}B$
2 --- $NdCo_4B$
3 --- $Nd_3Co_{11}B_4$
4 --- $Nd_2Co_5B_2$
5 --- $Nd_2Co_7B_3$
6 --- $\sim Nd_2Co_5B_3$
7 --- $NdCo_{12}B_6$
8 --- $NdCo_2B_2$
9 --- $NdCo_4B_4$
10 --- $\sim Nd_2CoB_3$

TECHNICAL FIELD

This invention relates to rare earth metal-transition metal series magnets having not only excellent magnetic properties but also improved corrosion resistance and temperature-dependent properties and a method of producing the same.

BACKGROUND ART

As a typical permanent magnet manufactured at the present, there are mentioned Alnico magnets, ferrite magnets, rare earth metal magnets and the like. The Alnico magnets are manufactured from the old time, but their demand is lowering in accordance with the development of cheap ferrite magnets and rare earth metal magnets having higher magnetic properties. On the other hand, the ferrite magnets are chemically stable and low in the cost because oxides are used as a main starting material, so that they are the main current as a magnet material even at the present, but they have a drawback that maximum energy product is small.

Recently, Sm-Co series magnets having a combination of magnetic isotropy inherent to rare earth metal ion and magnetic moment inherent to transition metal element have been developed, whereby the conventional value of maximum energy product is largely increased. However, the Sm-Co series magnet is mainly composed of resourceless Sm and Co, so that it is obliged to become expensive.

Now, it has been attempted to develop cheap magnet alloys containing no expensive Sm and Co and having high magnetic properties, and consequently Egawa et al developed stable ternary alloys by sintering process (Japanese Patent Application Publication No. 61-34242 and Japanese Patent laid open No. 59-132104) and J. J. Groat et al developed alloys having a high coercive force by liquid quenching process (Japanese Patent laid open No. 59-64739). These magnets are composed of Nd, Fe and B, and their maximum energy product exceeds that of Sm-Co series magnet.

However, Nd-Fe-B series magnets contain greater amounts of a light rare earth element such as ND having very high activity or the like and a corrosive Fe as a main component, so that the corrosion resistance is poor and hence the magnetic properties are degraded to damage the reliability as an industrial material.

Therefore, in order to improve the corrosion resistance, there are taken countermeasures such as surface plating (Japanese Patent laid open No. 63-77103), coating treatment (Japanese Patent laid open No. 60-63901) and the like on the sintered magnets, and surface treatment on resin bonded type magnets before kneading magnet powder with a resin and the like, but they can not be said to be an effective rustproof treatment over a long period of time, and the cost becomes higher due to such a treatment and further there are caused problems such as magnetic flux loss due to the presence of protective film and the like.

As a solution on the above problems, the inventors have previously proposed rare earth metal-transition metal-boron series magnet alloys in which Fe in the Nd-Fe-B series magnet is replaced with high concentrations of Co and Ni (Japanese Patent laid open No. 2-4939).

Such magnets are excellent in the corrosion resistance and high in the Curie point, so that the reliability as a magnet material is largely increased.

The invention is concerned with rare earth metal-transition metal series magnets of two phase structure further developed from the above magnet.

Moreover, magnets having excellent magnetic properties through two alloying process in which rare earth rich phase and rare earth poor phase are mixed and sintered at liquid phase state have previously been proposed as Nd series magnet of two phase structure (Japanese Patent laid open No. 63-93841 and No. 63-164403). In this case, the magnetic properties are improved, but there is still remained a problem on the corrosion resistance.

DISCLOSURE OF INVENTION

The invention is to advantageously solve the aforementioned problems and to propose rare earth metal-transition metal series magnets of two phase structure being excellent in not only the magnetic properties but also the corrosion resistance and a method of advantageously producing the same.

At first, details of elucidating the invention will be described.

The inventors have made various metallographical studies on the above magnet using high resolution electron microscope or the like, and confirmed that this magnet contains $Nd_2(Fe, Co, Ni)_{14}B$ phase having a large saturated magnetic flux density, and intergranular phases surrounding crystal grains of the above phase and developing a strong coercive force such as $Nd_2(Fe, Co, Ni)_{17}$, $Nd(Fe, Co, Ni)_5$, $Nd_2(Fe, Co, Ni)_7$, $Nd(Fe, Co, Ni)_4B$ and $Nd(Fe, Co, Ni)_{12}B_6$ and further $Nd_{1-x}TM_x$ of CrB structure (TM is mainly Ni) and the like.

Furthermore, it has been found that better corrosion resistance is exhibited as the amount of Nd phase being a point of causing corrosion is less and the concentration of Ni or Co in the above intergranular phase becomes high.

Now, the inventors have made further studies with respect to this point and found that the above intergranular phase hardly appears in a range of Nd-Fe-B ternary phase diagram other than $Nd_2(Fe, Co, Ni)_{17}$ andis rather a phase appearing only in the range of Nd-Co-B system.

For the reference, Nd-Fe-B ternary phase diagram is shown in Fig. 1 (N. F. Chaban, Yu. B. Kuzma, N. S. Bilonizhko, O. O. Kachmar and N. U. Petrov, Akad Nauk, SSSR, SetA, Fiz.-Mat. Tekh, Nauki No. 10 (1979) 873), and Nd-Co-B ternary phase diagram is shown in Fig. 2 (N. S. Bilonizhko and Yu. B. Kuzma, Izv. Akad. Nauk SSSR Neorg. Mater, 19 (1983) 487) (In the original report, $Nd_2Fe_{14}B$ phase and $Nd_2Co_{14}B$ phase are misinterpreted as Nd_2Fe_9B phase and Nd_2Co_9B phase, so that they are corrected in Figs. 1 and 2).

In Fig. 1, a phase of number 1 is $Nd_2Fe_{14}B$ phase, and $NdFe_4B_4$ phase (phase of number 2), Nd phase, Nd_2Fe_{17} phase and Fe phase appear as a composition near thereto. In Fig. 2, however, Nd_2Co_{17} phase, $NdCo_5$ phase, Nd_2Co_7 phase, $NdCo_4B$ phase (phase of number 2) and $NdCo_{12}B_6$ phase (phase of number 7) appear in a magnet prepare from a composition near to $Nd_2Co_{14}B$ phase of number 1, and Nd phase does not naturally appear at an equilibrium state.

As previously mentioned, Nd phase is not only a point of causing rust but also a magnetically useless phase, so that it should be eliminated.

It is, therefore, an object of the invention to provide permanent magnets having excellent magnetic properties and corrosion resistance by using magnetically useful two phases, i.e. $RE_2TM_{14}B$ phase having a high residual magnetic flux density and a low melting point RE-TM phase or RE-TM-B phase enhancing the sinterability and possessing a cleaning action against grain boundary of main phase and further forming an electrochemically noble composition as a starting material to prepare a two phase magnet.

That is, the invention lies in a corrosion-resistant rare earth metal-transition metal series permanent magnet consisting essentially of RE: not less than 10 at% but not more than 25 at% (where RE: one or more of Y, Sc and lanthanoid, B: not less than 2 at% but not more than 20 at% and the reminder being substantially TM (TM is one or more of Fe, Co and Ni), whose texture being comprised of a phase of $RE_2TM_{14}B$ (TM is the same as mentioned above) having $Nd_2Fe_{14}B$ structure, RE-TM series intermetallic compound phase (TM is Ni or a mixture of Ni and Fe or Co) or RE-TM series eutectic structure (TM is the same as mentioned above) and/or RE-TM-B series intermetallic compound phase (TM is the same as mentioned above) having a melting point lower than that of the above phase.

Furthermore, the invention lies in a method of producing a corrosion-resistant rare earth metal-transition metal series magnet, which comprises subjecting a mixture of powder composed mainly of $RE_2TM_{14}B$ series intermetallic compound phase (TM is one or more of Fe, Co and Ni) and powder having a melting point lower than that of the above powder and composed of mainly of RE-TM series intermetallic compound phase (TM is Ni or a mixture of Ni and Fe or Co) or RE-TM series eutectic structure (TM is the same as mentioned above) and/or RE-TM-B series intermetallic compound phase (TM is the same as mentioned above) to a compression molding and then sintering it.

In the invention, in order to more improve the corrosion resistance, it is effective to make the intergranular phase electrochemically more noble than the main phase, so that it is preferable that a ratio of Ni and/or Co in TM of the low melting point RE-TM and RE-TM-B series phases is higher than that in $RE_2TM_{14}B$ phase. Particularly, the increase of Ni ratio is effective to the improvement of corrosion resistance and the reduction of cost.

In the invention, it favorable that a ratio of $RE_2TM_{14}B$ intermetallic compound phase to RE-TM, RE-TM-B series intermetallic compound phase is about 95:5 to 40:60 as a formula unit. Because, when this ratio is outside the above range, there is caused a disadvantage of bringing about considerable degradation of coercive force and saturated magnetic flux density. The term "formula unit" used herein corresponds to a case that $Nd_2Fe_{14}B$ is considered as one molecule (this is called as formula in case of solid). The particle size of each of the above powders to be mixed is desirable to be about 0.5-5 μm for handling easiness and homogeneous mixing.

A typical composition of RE-TM series intermetallic compound phase (inclusive of eutectic structure, same as above) and RE-TM-B series intermetallic compound phase having a melting point lower than that of $RE_2TM_{14}B$ intermetallic compound phase is as follows.

- RE-TM series

RE_2TM_{17}, $RETM_5$, RE_2TM_7, $RETM_3$, $RETM_2$, RE_1TM_{1-x}, RE_7TM_3, RE_3TM and RE-TM eutectic structure
•RE-TM-B series
$RETM_4B$, $RE_3TM_{11}B_4$, $RE_2TM_5B_2$, $RE_2TM_7B_3$, $RE_2TM_5B_3$, $RETM_{12}B_6$, $RETM_2B_2$, $RETM_9B_4$, RE_2TMB_3

Moreover, powder composed mainly of the above $RE_2TM_{14}B$, RE-TM series and RE-TM-B series intermetallic compound phases can be obtained as follcws.

That is, constitutional elements are weighed so as to have a given composition and shaped into an ingot by arc melting or high frequency melting under vacuum or in an inert gas atmosphere. Then, the ingot is held at a temperature of 600-1000°C under vacuum or in an inert gas atmosphere for 1-30 days to form a single phase of intermetallic compound. In general, the intermetallic compound phase has frequently a solid solution range to a certain extent (~20%), so that the starting composition is allowed to have a composition width in accordance therewith.

The single phase of the intermetallic compound is roughly ground by means of a hammer mill and then finely divided into a particle size of 0.5-5 µm by using a jet mill or an attritor . Moreover, when the hardness is low and the pulverization is difficult in the low melting point RE-TM and RE-TM-B phases, hydrogen brittleness is previously carried out within a temperature range of room temperature to about 350°C for several hours before the grinding with a hammer mill, whereby the subsequent pulverization is made easy.

According to the invention, powder composed mainly of the previously prepared intermetallic compound having a composition of $RE_2TM_{14}B$ is mixed with at least one powder composed mainly of the previously prepared RE-TM series intermetallic compound and RE-TM-B series intermetallic compound phases having a melting point lower than that of the above powder, pressed and sintered, whereby high magnetic properties and high corrosion resistance can simultaneously be provided.

This is considered to be due to the fact that the powder having a melting point lower than that of the powder composed mainly of $RE_2TM_{14}B$ intermetallic compound phase promotes the sintering and forms an intergranular phase between crystal grains of $RE_2TM_{14}B$ to improve coercive force.

In $RE_2TM_{14}B$ phase, Nd and Pr are desirable as RE from viewpoints of magnitude of magnetic moment and magnetic coupling with TM atm as well as the cost, but it is needless to say that the other RE or a combination of Nd, Pr therewith may be used.

As to TM, one or more of Fe, Co and Ni is sufficient, and particularly it is preferable to increase the ratio of Ni from a viewpoint of high corrosion resistance of the magnet. Further, $RE_2TM_{14}B$ phase bears the saturated magnetic flux density of the magnet, so that the ratios of Fe, Co and Ni in TM are desirable to be not less than 10 at% but less than 73 at% in Fe, not less than 7 at% but not more than 50 at% in Co and not less than 5 at% but not more than 30 at% in Ni. Even when the main phase is $RE_2TM_{14}B$ phase in which Fe as TM is 100%, the corrosion resistance of the permanent magnet according to the invention is superior to that of the conventional RE-TM-B magnet, so that the above phase can naturally be used as a main phase in accordance with the use purpose of the magnet.

As RE in the low melting point phase of RE-TM system and RE-TM-B system, light rare earth element such as La, Ce, Pr, Nd or the like is advantageously adaptable importantly considering the cost, and middle to heavy rare earth elements from Sm to Lu and Y, Sc and the like are adaptable for more enhancing the corrosion resistance.

As to TM, the presence of Ni and/or Co, particularly Ni is effective to improve the corrosion resistance, so that according to the invention Ni is necessarily contained as TM. In this case, the content in TM is preferable to be not less than about 8%.

The addition effect of Ni is as follows.

i) The melting point of RE-TM system and RE-TM-B system is lowered, and the wetting of liquid phase in the liquid phase sintering is promoted to increase the sintering density and enhance the residual magnetic flux density.

ii) The effect of cleaning grain boundary in liquid phase is enhanced in the liquid phase sintering to more increase the coercive force by the same reason as in the above item i).

iii) It is effective to the improvement of corrosion resistance and cheap as compared with Co.

Furthermore, when the ratio of Ni and/or Co in the low melting point phase is made higher than that of $RE_2TM_{14}B$ phase, the corrosion resistance can be more improved because the phases of these powders tend to preferentially corrode in the grain boundary as compared with $RE_2TM_{14}B$ phase in the sintered body if the structure of TM is same and is advantageously acted by previously making electrochemically noble. Furthermore, the magnetically useless Nd phase can be eliminated, so that the residual magnetic flux density increases and hence the maximum energy product $(BH)_{max}$ also increases.

In this connection, even when an alloy having an average composition as a whole magnet is melted from the first as in the conventional technique, pulverized, pressed and sintered so as to approach to an equilibrium state, the Nd phase is not obtained. For this purpose, it is necessary to conduct the heating at a high temperature for long time, during which abnormal growth of crystal grain is undesirably caused to considerably degrade the coercive force.

Moreover, it is not necessary that the same element is used in RE of the main phase and RE of the low melting point phase. And also, in the magnet consisting of the above two phases, the effect of the invention is not lost even when a part of RE and TM is replaced with at least one of Mg, Al, Si, Ti, V, Cr, Mn, Cu, Ag, Au, Cd, Rh, Pd, Ir, Pt, Zn, Ga, Ge, Zr, Nb, Mo, In, Sn, Hf, Ta and W in an amount up to 8 at% of a full magnet.

As to the production method, there may be carried out a method wherein a mixture of powder of $RE_2TM_{14}B$ composition and powder composed mainly of low melting point RE-TM series and/or RE-TM-B series intermetallic compound phases is placed in an iron pipe under vacuum and then sintered while hot rolling as a method of producing large size magnets in addition to the method in which the above powder mixture is subjected to compression molding and then sintered.

BRIEF DISCLOSURE OF DRAWINGS

Fig. 1 is a Nd-Fe-B three component phase diagram; and

Fig. 2 is a Nd-Co-B three component phase diagram.

BEST MODE FOR CARRYING OUT THE INVENTION

Example 1

An alloy button was prepared by arc melting neodymium, transition metal and boron at an atomic ratio of 2:14:1, which was subjected to a normalizing treatment in a vacuum furnace at 950°C for 7 days and further to rough grinding and fine pulverization, whereby fine powder having a particle size of few microns was obtained. In this case, the ratios of Fe, Co, Ni in the transition metal were varied to produce a plurality of alloy powders.

Similarly, powder having a ratio of neodymium or (neodymium + dysprosium) to nickel of 1:1 was prepared. In this case, the normalizing treatment conditions were 680°C and 5 days.

Then, powders selected from the above two groups were mixed at a mixing ratio shown in Table 1, pressed while applying a magnetic field of 15 kOe, sintered at 1000°C under vacuum for 2 hours and then quenched to room temperature.

The magnetic properties and corrosion property of the thus obtained samples were measured to obtain results shown in Table 1. Moreover, the corrosion property was evaluated by exposing the sample to an environment at a temperature of 70°C and a humidity of 95% for 48 hours and measuring a rusted area ratio on the surface of the sample.

For the comparison, the measured results of a sample produced by the conventional method in which a full composition for the sintered magnet was melted at once and subjected to rough grinding - fine pulverization - pressing in magnetic field - sintering steps are also shown in Table 1.

Table 1

No.	Powder mixing ratio (formula unit.ratio %)		Ratio of transition metals (atomic ratio %)	Composition of magnet (at%)						Br (kG)	iHc (kOe)	$(BH)_{max}$ (MGOe)	Rusted area (%)	Remarks
				Nd	Dy	Fe	Co	Ni	B					
1	$Nd_2TM_{14}B_1$ Nd_1Ni_1	50 50	Fe : Co : Ni 100 : 0 : 0	15.79	-	73.69	-	5.26	5.26	12.0	7.2	32.5	11	Acceptable Example 1
2	Arc melting of the above whole composition			"	-	"	-	"	"	13.3	5.7	35.2	30	Comparative Example 1
3	$Nd_2TM_{14}B_1$ Nd_1Ni_1	55 45	Fe : Co : Ni 75 : 25 : 0	15.12	-	56.34	18.78	4.39	5.37	11.8	5.8	29.8	2	Acceptable Example 2
4	Arc melting of the above whole composition			"	-	"	"	"	"	12.5	4.0	28.0	9	Comparative Example 2
5	$Nd_2TM_{14}B_1$ Nd_1Ni_1	50 50	Fe : Co : Ni 65 : 30 : 5	15.8	-	47.9	22.1	8.9	5.3	11.5	13.2	32.0	0	Acceptable Example 3
6	Arc melting of the above whole composition			"	-	"	"	"	"	12.0	4.0	26.4	6.5	Comparative Example 3
7	$Nd_2TM_{14}B_1$ Nd_1Ni_1	45 55	Fe : Co : Ni 65 : 30 : 5	16.6	-	46.8	21.6	9.9	5.1	11.6	8.0	29.6	0	Acceptable Example 4
8	Arc melting of the above whole composition			"	-	"	"	"	"	12.0	3.7	22.0	2	Comparative Example 4
9	$Nd_2TM_{14}B_1$ $(Nd_{0.99}Dy_{0.01})_1Ni_1$	50 50	Fe : Co : Ni 100 : 0 : 0	15.74	0.05	73.69	-	5.26	5.26	11.4	9.7	31.5	1	Acceptable Example 5
10	Arc melting of the above whole composition			"	"	"	-	"	"	12.8	8.5	35.0	28	Comparative Example 5
11	$Nd_2TM_{14}B_1$ $(Nd_{0.99}Dy_{0.01})_1Ni_1$	55 45	Fe : Co : Ni 80 : 20 : 0	15.08	0.04	56.34	18.78	4.39	5.37	11.3	8.1	30.0	3	Acceptable Example 6
12	Arc melting of the above whole composition			"	"	"	"	"	"	12.6	5.0	31.7	7	Comparative Example 6

As seen from the above table, the rare earth metal-transition metal series magnets of two phase structure according to the invention considerably improve not only the magnetic properties but also corrosion resistance as compared with those obtained by melting the full composition from the first as in the conventional technique.

Example 2

An alloy button was prepared by arc melting neodymium, transition metal and boron at an atomic ratio of 2:14:1, which was subjected to a normalizing treatment in a vacuum furnace at 950° C for 7 days and further to rough grinding and fine pulverization, whereby fine powder having a particle size of few microns was obtained. In this case, the ratios of Fe, Co, Ni in the transition metal were varied to produce a plurality of alloy powders.

Similarly, powder having a ratio of neodymium and/or dysprosium or praseodymium to nickel or (nickel + cobalt) of 3:1 was prepared. In this case, the normalizing treatment conditions were 485° C and 5 days.

The magnetic properties and corrosion property of the thus obtained samples were measured to obtain results shown in Table 2.

For the comparison, the measured results on the properties of a magnet produced by the technique disclosed in Japanese Patent laid open NO. 63-164403 are also shown in Table 2.

Table 2

No.	Powder mixing ratio (formula unit ratio %)	Ratio of transition metals (atomic ratio %)	Composition of magnet (at%)							Br (kG)	iHc (kOe)	$(BH)_{max}$ (MGOe)	Rusted area (%)	Remarks
			Nd	Dy	Pr	Fe	Co	Ni	B					
13	$Nd_2TM_{14}B_1$ 65 Nd_3Ni_1 35	Fe : Co : Ni 100 : 0 : 0	18.88	-	-	73.09	-	2.81	5.22	12.6	12.0	34.0	5	Acceptable Example 7
14	Arc melting of the above whole composition		"	-	-	"	-	"	"	13.1	11.2	36.2	50	Comparative Example 7
15	$Nd_2TM_{14}B_1$ 65 Nd_3Ni_1 35	Fe : Co : Ni 70 : 30 : 0	18.88	-	-	51.16	21.92	2.81	5.22	12.0	15.0	35.2	0	Acceptable Example 8
16	Arc melting of the above whole composition		"	-	-	"	"	"	"	12.8	7.9	34.8	10	Comparative Example 8
17	$Nd_2TM_{14}B_1$ 65 Nd_3Ni_1 35	Fe : Co : Ni 65 : 30 : 5	18.88	-	-	47.51	21.92	6.47	5.22	11.5	10.5	32.0	0	Acceptable Example 9
18	Arc melting of the above whole composition		"	-	-	"	"	"	"	12.3	6.8	32.2	5	Comparative Example 9
19	$Nd_2TM_{14}B_1$ 65 $(Nd_{0.97}Dy_{0.03})_3Ni_1$ 35	Fe : Co : Ni 80 : 20 : 0	18.63	0.25	-	58.48	14.61	2.81	5.22	11.6	15.5	34.0	3	Acceptable Example 10
20	Arc melting of the above whole composition		"	"	-	"	"	"	"	12.3	8.1	32.6	20	Comparative Example 10
21	$Pr_2Tm_{14}B_1$ 70 $Pr_{2.5}Ni_1$ 30	Fe : Co : Ni 85 : 15 : 0	-	-	16.60	64.32	11.35	2.32	5.41	11.8	9.5	33.0	2	Acceptable Example 11
22	Arc melting of the above whole composition		-	-	"	"	"	"	"	12.5	6.2	32.3	25	Comparative Example 11
23	$Nd_2TM_{14}B_1$ 70 $Nd_3(Ni_{0.8}Co_{0.2})_1$ 30	Fe : Co : Ni 80 : 20 : 0	17.56	-	-	59.85	15.42	1.83	5.34	12.2	10.5	35.1	3	Acceptable Example 12
24	Arc melting of the above whole composition		"	-	-	"	"	"	"	12.8	6.3	34.6	25	Comparative Example 12
25	$Nd_2TM_{14}B_1$ 65 $Nd_3(Ni_{0.1}Co_{0.9})_1$ 35	Fe : Co : Ni 70 : 30 : 0	18.88	-	-	51.16	24.46	0.28	5.22	12.1	12.3	36.8	3	Acceptable Example 13
26	Arc melting of the above whole composition		"	-	-	"	"	"	"	13.3	10.1	38.0	20	Comparative Example 13
27	$Nd_2TM_{14}B_1$ 55 Dy_3Ni_1 45	Fe : Co : Ni 80 : 20 : 0	9.87	12.11	-	55.25	13.80	4.04	4.93	7.0	28.5	17.2	3	Acceptable Example 14
28	Arc melting of the above whole composition		"	"	-	"	"	"	"	8.8	15.3	16.5	8	Comparative Example 14
29	$Nd_2TM_{14}B_1$ 65 $Nd_3(Co_{0.5}Fe_{0.5})_1$ 35	Fe : Co : Ni 70 : 30 : 0	18.88	-	-	52.57	23.33	-	5.22	12.3	16.0	35.5	14	Conventional Example

As seen from the above table, the rare earth metal-transition metal series magnets of two phase structure according to the invention are excellent in the magnetic properties and corrosion resistance. Furthermore, when Acceptable Example 8 is compared with Acceptable Example 13, it is apparent that the corrosion resistance is improved as the Ni ratio in $RE_3(Ni, Co)_1$ becomes particularly higher. Moreover, in the conventional example, the magnetic properties are good, but the corrosion resistance is poor because

Ni is not contained.

Example 3

A fine alloy powder of $RE_2TM_{14}B$ composition was prepared by the same manner as in Example 1, while a fine alloy powder in which ratios of Ni and Co in TM were made higher than those of $RE_2TM_{14}B$ powder was prepared as a starting powder. After these powders were mixed, a sintered magnet was produced by the same manner as in Example 1.

The properties of the thus obtained sintered magnet are shown in Table 3 together with those of the sintered magnet produced by the conventional method.

Table 3

No.	Powder mixing ratio (formula unit ratio %)	Ratio of transition metals (atomic ratio %)	Composition of magnet (at%)						Br (kG)	iHc (kOe)	(BH)max (MGOe)	Rusted area (%)	Remarks
			Nd	Dy	Fe	Co	Ni	B					
30	$Nd_2TM_{14}B_1$ 60 $NdTM_4B_1$ 40	Fe : Co : Ni 65 : 30 : 5 2.25 : 61.5 : 36.25	12.69	-	43.65	27.78	7.94	7.94	11.9	7.2	31.7	0	Acceptable Example 15
31	Arc melting of the above whole composition	55 : 35 : 10	"	-	"	-	"	"	12.7	5.1	31.5	3	Comparative Example 15
32	$Nd_2TM_{14}B_1$ 55 $NdTM_4B_1$ 45	70 : 30 : 0 17.2 : 30 : 52.8	12.86	-	47.30	23.65	7.89	8.30	11.5	6.5	29.8	0	Acceptable Example 16
33	Arc melting of the above whole composition	60 : 30 : 10	"	-	"	"	"	"	12.3	4.8	30.5	3	Comparative Example 16
34	$Nd_2TM_{14}B_1$ 45 $NdTM_4B_1$ 55	100 : 0 : 0 3.4 : 58.0 : 38.6	13.24	-	47.30	23.65	7.89	8.30	11.8	8.0	32.1	0	Acceptable Example 17
35	Arc melting of the above whole composition	60 : 30 : 10	"	-	"	"	"	"	12.6	5.2	31.7	3	Comparative Example 17

As seen from the above table, when using the fine alloy powder in which the ratios of Ni and Co in TM are higher than those of $RE_2TM_{14}B$ powder as a powder to be mixed, the more improvement of the corrosion resistance is attained.

INDUSTRIAL APPLICABILITY

According to the invention, the rare earth metal-transition metal series magnets having improved corrosion resistance and magnetic properties can be produced as compared with the conventional production method. Particularly, the corrosion resistance is improved, so that the considerable improvement of reliability as an industrial material is realized.

Claims

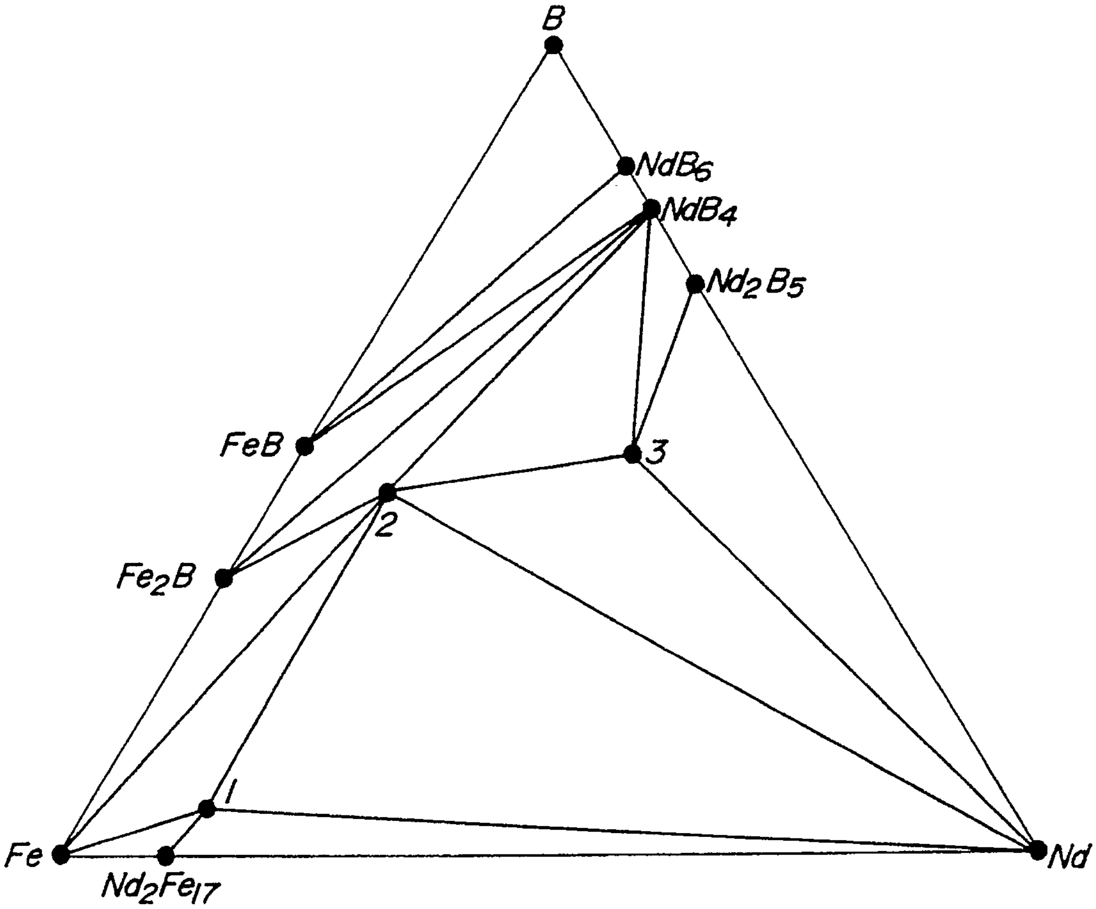
1. A corrosion-resistant rare earth metal-transition metal series permanent magnet consisting essentially of RE: not less than 10 at% but not more than 25 at% (where RE: one or more of Y, Sc and lanthanoid, B: not less than 2 at% but not more than 20 at% and the remainder being substantially TM (TM is one or more of Fe, Co and Ni), whose texture being comprised of a phase of $RE_2TM_{14}B$ (TM is the same as mentioned above) having $Nd_2Fe_{14}B$ structure, RE-TM series intermetallic compound phase (TM is Ni or a mixture of Ni and Fe or Co) or RE-TM series eutectic structure (TM is the same as mentioned above) and/or RE-TM-B series intermetallic compound phase (TM is the same as mentioned above) having a melting point lower than that of the above phase.

2. The corrosion-resistant rare earth metal-transition metal series permanent magnet according to claim 1, wherein a ratio of Ni and/or Co in TM in said low melting point RE-TM series and RE-TM-B series phases is made higher than that of said $RE_2TM_{14}B$ phase.

3. A method of producing a corrosion-resistant rare earth metal-transition metal series magnet as claimed in claim 1, which comprises subjecting a mixture of powder composed mainly of $RE_2TM_{14}B$ series intermetallic compound phase (TM is one or more of Fe, Co and Ni) and powder having a melting point lower than that of the above powder and composed of mainly of RE-TM series intermetallic compound phase (TM is Ni or a mixture of Ni and Fe or Co) or RE-TM series eutectic structure (TM is the same as mentioned above) and/or RE-TM-B series intermetallic compound phase (TM is the same as mentioned above) to a compression molding and then sintering it.

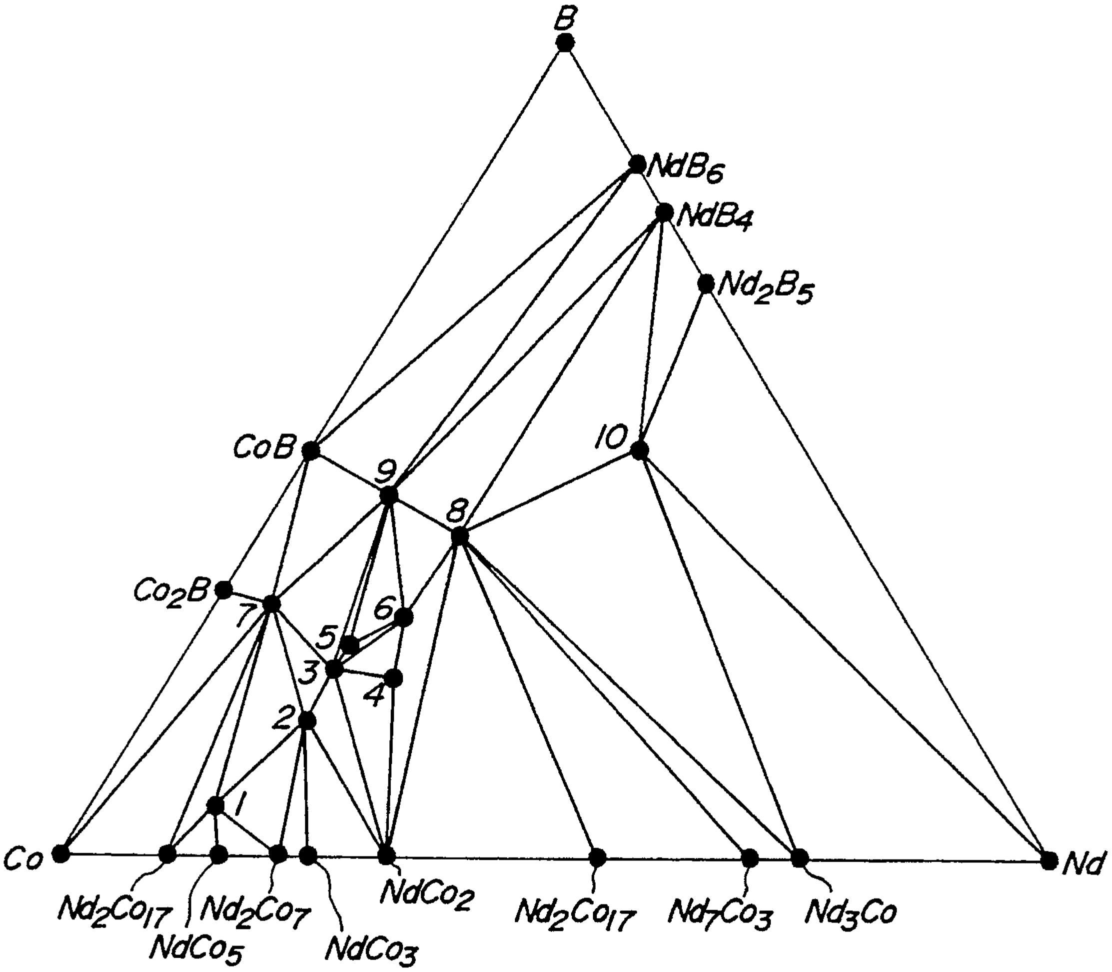
4. A method of producing a corrosion-resistant rare earth metal-transition metal series magnet as claimed in claim 2, which comprises subjecting a mixture of powder composed mainly of $RE_2TM_{14}B$ series intermetallic compound phase (TM is one or more of Fe, Co and Ni) and powder having a melting point lower than that of the above powder and composed of mainly of RE-TM series intermetallic compound phase (TM is Ni or a mixture of Ni and Fe or Co) or RE-TM series eutectic structure (TM is the same as mentioned above) and/or RE-TM-B series intermetallic compound phase (TM is the same as mentioned above) to a compression molding and then sintering it.

FIG_1



1 --- $Nd_2Fe_{14}B$
2 --- $NdFe_4B_4$
3 --- Nd_2FeB_3

FIG_2



1 --- $Nd_2Co_{14}B$
2 --- $NdCo_4B$
3 --- $Nd_3Co_{11}B_4$
4 --- $Nd_2Co_5B_2$
5 --- $Nd_2Co_7B_3$
6 --- $\sim Nd_2Co_5B_3$
7 --- $NdCo_{12}B_6$
8 --- $NdCo_2B_2$
9 --- $NdCo_4B_4$
10 --- $\sim Nd_2CoB_3$

INTERNATIONAL SEARCH REPORT

International Application No PCT/JP90/01315

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) [6]

According to International Patent Classification (IPC) or to both National Classification and IPC

Int. Cl^5 H01F1/053, 1/08, C22C1/04, 19/00, 33/02, 38/00

II. FIELDS SEARCHED

Minimum Documentation Searched [7]

Classification System	Classification Symbols
IPC	H01F1/053-1/08, C22C33/02, 38/00, 19/00

Documentation Searched other than Minimum Documentation
to the Extent that such Documents are Included in the Fields Searched [8]

Jitsuyo Shinan Koho	1926 - 1990
Kokai Jitsuyo Shinan Koho	1971 - 1990

III. DOCUMENTS CONSIDERED TO BE RELEVANT [9]

Category *	Citation of Document, [11] with indication, where appropriate, of the relevant passages [12]	Relevant to Claim No. [13]
X	JP, A, 63-254703 (Tohoku Kinzoku Kogyo Co., Ltd.), 21 October 1988 (21. 10. 88) & EP, A1, 261579	1 - 4
Y	JP, A, 63-93841 (Shin-Etsu Chemical Co., Ltd.), 25 April 1988 (25. 04. 88), (Family: none)	1 - 4
Y	JP, A, 63-164403 (Shin-Etsu Chemical Co., Ltd.), 7 July 1988 (07. 07. 88), (Family: none)	1 - 4
Y	JP, A, 63-313807 (Sumitomo Special Metals Co., Ltd.), 21 December 1988 (21. 12. 88), (Family: none)	1 - 2

* Special categories of cited documents: [10]

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

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"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

IV. CERTIFICATION

Date of the Actual Completion of the International Search	Date of Mailing of this International Search Report
November 14, 1990 (14. 12. 90)	December 3, 1990 (03. 12. 90)
International Searching Authority	**Signature of Authorized Officer**
Japanese Patent Office	