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(54) **Bleaching compositions**

(57) The present invention relates to bleaching compositions comprising a bleach, a borate or boric acid or a mixture thereof and less than 5% by weight of the total composition of a metal oxide selected from the group consisting of Sn, Ti, Zn, Al, Bi, Ga, Ge, Sb and Mg or a mixture thereof. These bleaching compositions exhibit excellent fabric yellowing-prevention performance and/or fabric safety.

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DescriptionField of the invention

- 5 **[0001]** The present invention relates to bleaching compositions, which can be used to bleach various surfaces including but not limited to, fabrics, clothes, carpets and the like as well as hard-surfaces like walls, tiles, floors, glass, bathrooms surfaces, kitchen surfaces, toilet bowls and dishes.

Background of the invention

- 10 **[0002]** Bleach-containing compositions for bleaching various surfaces, e.g., fabrics, are well known in the art.
- [0003]** Amongst the different bleaching compositions available, those relying on bleaching by hypohalite bleach, such as hypochlorite, and those relying on peroxygen bleach, such as hydrogen peroxide, are often preferred, mainly for bleaching performance reasons.
- 15 **[0004]** However, a drawback associated with the use of bleach-containing compositions, e.g., hypochlorite-based compositions, is that said compositions may damage and/or cause yellowing of the fabrics being bleached. During the bleaching operation, a liquid hypochlorite-containing composition for example, can be applied onto the fabric in neat or diluted form. The hypochlorite bleach can be present in various forms, including hypochlorous acid and in the form of various molecule fragments resulting from its decomposition pathways. Hypochlorous acid and these molecule frag-
- 20 ments are reactive with fabrics and may cause damage and/or yellowing of fabrics being bleached.
- [0005]** Therefore, bleaching compositions are often formulated comprising bleach and a fabric protective system. Compositions comprising bleach and one of various fabric protective systems are well-known in the art. An example is a bleaching composition based on hypohalite bleach comprising a polymer and a borate as a fabric protective system (EP-A-0 824 147), or bleaching compositions based on hypochlorite comprising a metaborate as a fabric protective sys-
- 25 tem (EP-A-0 781 840).
- [0006]** However, there are some limitations to the convenience of said bleaches comprising a fabric protective system. In particular, it is well-known from consumer research that said compositions may still damage and/or yellow the fabrics which are being bleached, although said compositions comprise a fabric protective system.
- [0007]** It is therefore an object of the present invention to provide a bleach-containing composition, suitable for use in laundry applications, whereby the fabric yellowing-prevention is further improved.
- 30 **[0008]** It is a further object of the present invention to provide a bleaching composition, suitable for use in laundry applications, whereby fabric safety is further improved.
- [0009]** Fabrics made of cotton and/or synthetic materials have free hydroxyl groups. It is believed that fabric damage and/or fabric yellowing occurs due to the oxidation of the hydroxyl groups by the bleach. The oxidation of the hydroxyl groups is believed to be further emphasised at lower pHs. Indeed, low pH shifts the chemical equilibrium between hypochlorite and hypochlorous acid in favor of the latter. Hypochlorous acid is a more aggressive bleach, which has been found to further contribute to fabric yellowing and fabric damage. Moreover, it is further believed that the hydroxyl groups of the fabrics are more easily oxidised at lower pHs.
- 35 **[0010]** It has been found that fabric protection can be achieved using a combination of a borate and/or boric acid and a metal oxide as described herein in bleach-containing compositions. The bleaching compositions of the present invention provide further improved fabric yellowing-prevention and/or further improved fabric safety on fabrics bleached therewith, as compared to bleaching compositions comprising other fabric protective systems, e.g., a metaborate alone, or bleaching compositions comprising no fabric protective system. It is believed that the hydroxyl groups of a fabric are protected by the fabric protective system according to the present invention from oxidation by a bleach, especially a hypochlorite bleach, thus, providing fabric safety and/or fabric yellowing-prevention.
- 40 **[0011]** Advantageously the bleaching compositions of the present invention are suitable for various laundry bleaching applications both when used in diluted conditions, e.g., as a detergent additive or a fully formulated laundry detergent composition, and when used in neat condition, e.g., as a liquid pretreater (spotter).
- [0012]** A further advantage of the bleaching compositions according to the present invention is that they are able to perform in a variety of conditions, i.e., in hard and soft water as well as when used neat or diluted.
- 50 **[0013]** Yet another advantage of the bleaching compositions of the present invention is that said bleaching compositions may also be suitable for the bleaching of different types of surfaces including hard-surfaces like floors, walls, tiles, glass, kitchen surfaces, bathrooms surfaces, toilet bowls and/or dishes and the like, as well as fabrics. More particularly, the bleaching compositions of the present invention are suitable for bleaching any type of fabrics including natural fabrics (e.g., fabrics made of cotton, viscose, linen, silk and wool), synthetic fabrics such as those made of polymeric fibers
- 55 of synthetic origin as well as those made of both natural and synthetic fibers. Indeed, the bleaching compositions of the present invention may be used on synthetic fibers despite a standing prejudice against the use of hypohalite bleaches, especially hypochlorite bleaches, on synthetic fibers, as evidenced by warning on labels of commercially available

hypochlorite bleaches and clothes.

Background art

- 5 **[0014]** Bleach-containing compositions based on hypochlorite bleach or peroxygen bleach suitable for bleaching surfaces, e.g., fabrics or hard-surfaces, have been described in the art.
- [0015]** JP 59036198 discloses bathroom cleaner compositions comprising a hypochlorite bleach, a partially crosslinked polyacrylic acid, an inorganic alkaline agent (e.g., alkali metal borates) and from 5%-70% of a water-insoluble abrasive (e.g., TiO_2 and Al_2O_3). However, compositions comprising less than 5% of a metal oxide as described
- 10 herein, are not disclosed.
- [0016]** EP-A-0 781 840 discloses compositions comprising hypochlorite and a metaborate as a pH buffering means. However, compositions comprising a metal oxide as described herein are not disclosed.
- [0017]** EP-A-0 824 147 discloses liquid bleaching compositions comprising a hypohalite bleach and a polymer. The composition may further comprise a boron salt as pH buffering component. However, compositions comprising a metal
- 15 oxide as described herein are not disclosed.

Summary of the invention

- [0018]** The present invention encompasses a bleaching composition comprising a bleach, a borate or boric acid or a mixture thereof and less than 5% by weight of the total composition of a metal oxide selected from the group consisting of oxides of Sn, Ti, Zn, Al, Bi, Ga, Ge, Sb and Mg, or a mixture thereof.
- [0019]** The present invention also encompasses a process of bleaching surfaces, e.g., fabrics, wherein said surfaces are contacted with a bleaching composition as defined herein.
- [0020]** The present invention further encompasses the use of a combination of a metal oxide and a borate or boric acid or a mixture thereof, in a fabric bleaching composition comprising a bleach to treat a surface whereby good fabric
- 25 yellowing-prevention performance and/or fabric safety are achieved.

Detailed description of the invention

The bleaching composition

- [0021]** The bleaching compositions may be formulated either as solids or liquids.
- [0022]** In the case where the bleaching compositions are formulated as solids, for example as granules, tablets or powders, they are preferably applied on the fabrics to be bleached according to the present invention in a liquid form,
- 35 i.e., solid compositions are used in combination with an appropriate solvent, typically water.
- [0023]** In the case where the bleaching compositions are formulated as liquids including gel and paste form, the bleaching compositions are preferably but not necessarily formulated as aqueous compositions. Liquid bleaching compositions are preferred herein for convenience of use. Preferred liquid bleaching compositions of the present invention are aqueous and therefore, preferably comprise water in an amount of from 60% to 98%, more preferably of from 80%
- 40 to 97% and most preferably 85% to 97% by weight of the total composition.

Bleach

- [0024]** As a first essential ingredient, the compositions of the present invention comprise a bleach. Any bleach known to those skilled in the art may be suitable for use herein. Preferred bleaches include any hypohalite bleach, as well as any peroxygen bleach.
- [0025]** Suitable hypohalite bleaches may be provided by a variety of sources, including bleaches that lead to the formation of positive halide ions and/or hypohalite ions, as well as bleaches that are organic based sources of halides such as chloroisocyanurates.
- 50 **[0026]** Suitable hypohalite bleaches for use herein include the alkali metal and alkaline earth metal hypochlorites, hypobromites, hypoiodites, chlorinated trisodium phosphate dodecahydrates, potassium and sodium dichloroisocyanurates, potassium and sodium trichlorocyanurates, N-chloroimides, N-chloroamides, N-chloroamines and chlorohydrantoin.
- [0027]** For the bleaching compositions herein, the preferred hypohalite bleaches among those described above are the alkali metal or alkaline earth metal hypochlorites selected from the group consisting of sodium, potassium, magnesium, lithium and calcium hypochlorites, and mixtures thereof. Sodium hypochlorite is the most preferred hypohalite bleach.
- 55 **[0028]** Suitable peroxygen bleaches to be used herein are hydrogen peroxide, water soluble sources thereof or mix-

tures thereof. A hydrogen peroxide source refers to any compound that produces perhydroxyl ions when said compound is in contact with water.

[0029] Suitable water-soluble sources of hydrogen peroxide for use herein include: percarbonates; persulfates; persulphates, such as monopersulfate; perborates; peroxyacids, such as diperoxydodecandioic acid (DPDA), magnesium perphthalic acid, perlauric acid, perbenzoic and alkylperbenzoic acids, hydroperoxides, aliphatic and aromatic diacyl peroxides; and mixtures thereof. Preferred peroxygen bleaches herein are hydrogen peroxide, hydroperoxide and/or diacyl peroxide. Hydrogen peroxide is the most preferred peroxygen bleach herein.

[0030] Suitable hydroperoxides for use herein are tert-butyl hydroperoxide, cumyl hydroperoxide, 2,4,4-trimethylpentyl-2-hydroperoxide, di-isopropylbenzenemonohydroperoxide, tert-amyl hydroperoxide and 2,5-dimethyl-hexane-2,5-dihydroperoxide.

[0031] Suitable aliphatic diacyl peroxides for use herein are dilauroyl peroxide, didecanoyl peroxide, dimyristoyl peroxide, or mixtures thereof. Suitable aromatic diacyl peroxide for use herein is for example dibenzoyl peroxide.

[0032] Accordingly, the bleaching compositions according to the present invention may typically comprise from 0.01% to 20%, preferably from 0.25% to 15%, more preferably from 0.5% to 8% and most preferably from 1% to 6% by weight of the total composition of a bleach.

Borate or boric acid

[0033] As a second essential ingredient, the bleaching compositions according to the present invention comprise a borate or boric acid or a mixture thereof.

[0034] Suitable borates include alkali metal salts of borates and mixtures thereof. Suitable alkali metal salts of borates include alkali metal salts of metaborate, tetraborate, octaborate, pentaborate, dodecaborate, borontrifluoride and alkyl borates containing from 1 to 12, preferably from 1 to 4 carbon atoms. Preferred alkyl borate includes methyl borate, ethyl borate and propyl borate.

[0035] Particularly preferred herein are the alkali metal salts of metaborate, such as sodium metaborate, potassium metaborate, or mixtures thereof.

[0036] Borates like sodium metaborate and sodium tetraborate are commercially available from Borax and Societa Chimica Larderello under the name sodium metaborate and Borax®.

[0037] The bleaching compositions according to the present invention may comprise from 0.01% to 10%, more preferably from 0.5% to 5%, even more preferably from 0.7% to 3%, and most preferably from 0.7% to 1.5% by weight of the total composition of said borate or boric acid or a mixture thereof.

Metal oxide

[0038] A third essential ingredient of the bleaching compositions of the present invention is a metal oxide selected from the group consisting of oxides of Sn, Ti, Zn, Al, Bi, Ga, Ge, Sb and Mg, or a mixture thereof.

[0039] Examples of said metal oxides include SnO, SnO₂, TiO₂, TiO, ZnO, Al₂O₃, Bi₂O₃, Bi₂O₅, Ga₂O₃, GeO₂, GeO, Sb₂O₃, Sb₃O₅ and MgO or a mixture thereof. Particularly preferred herein is TiO₂.

[0040] TiO₂ is commercially available from Bayer, Chemapol, Kronos, ICI, Solvay, Rhone Poulenc or Laporte.

[0041] The bleaching composition according to the present invention comprises less than 5%, preferably from 0.001% to 3%, more preferably from 0.01% to 2%, even more preferably from 0.015% to 0.8%, and most preferably from 0.02% to 0.2% by weight of the total composition of a metal oxide as described herein.

[0042] Cotton fabrics contain cellulose (i.e., (C₆H₁₀O₅)_n). When a bleach-containing composition, e.g., a hypochlorite-containing composition, is used to bleach a cotton fabric, the bleach may react with the hydroxyl substituents of cellulose in position 6, 2, 3 and 5 of the cellulosic ring and thus may introduce new functional groups like carbonylic, aldehydic and/or carboxylic groups. Synthetic fabrics may be made of various synthetic materials like polyamide-elasthane that also comprise hydroxyl groups capable of being oxidized by bleach.

[0043] It has now been found that a combination of a borate and/or boric acid, and a metal oxide as described herein, in the compositions according to the present invention act together as a fabric protective agent. Compared to a bleach-containing composition comprising a fabric protective system other than that described by the present invention or no fabric protective system at all, the fabric yellowing-prevention performance and/or the fabric safety performance is further improved.

[0044] Whilst not wishing to be bound by theory, it is believed that the fabric protective system consisting of a borate and/or boric acid and of a metal oxide as described herein, of a bleaching composition according to the present invention is capable of protecting, i.e., esterifying and/or complexing, the hydroxyl groups present on the fibers of a treated fabric. Otherwise, said hydroxyl groups would be subject to oxidation by the bleach, e.g., hypochlorite bleach, which leads to fabric yellowing and/or fabric damages.

[0045] The fabric yellowing-prevention and/or fabric safety performance benefit can be achieved when using the com-

positions of the present invention neat, e.g., as a pretreater, as well as diluted, e.g., typical dilution with water in conventional washing conditions.

[0046] Indeed, the yellowing-prevention effect and/or safety effect of the present invention can be evaluated by comparing the composition according to the present invention to a bleach-containing composition comprising a fabric protective system other than that claimed by the present invention.

[0047] By "further improved yellowing-prevention", it is meant herein that the yellowing of fabrics, i.e., the appearance of a yellowish tone of said fabrics, that occurs by using the bleach-containing compositions of the present invention (e.g., those containing a hypohalite bleach, e.g., hypochlorite) is reduced, compared to the yellowing that occurs by using a bleach-containing composition comprising a fabric protective system other than that described by the present invention.

[0048] The degree of yellowing can be determined by both visual and instrumental grading. Visually, the difference in yellowing between items treated with different compositions can be determined by a team of expert panelists. Instrumentally, the assessment can be determined with the help of Colorimeters such as Ganz Griesser® instruments (e.g., Datacolor® Spectraflash® SF 500, Machbet White-eye® 500) or a ZEISS ELREPHO® or others which are available for instance from Hunterlab® or Gardner®.

[0049] By "further improved fabric safety", it is meant herein that the damage caused on fabric by using the bleach-containing compositions of the present invention (e.g., those containing a hypohalite bleach, e.g., hypochlorite) is reduced, compared to the damage caused by using a bleach-containing composition comprising a fabric protective system other than that described by the present invention.

[0050] Fabric safety may be evaluated by different test methods including the degree of polymerisation test method according to UNI (Ente Nazionale Italiano di Unificazione) official method UNI 8282-Determinazione della viscosità intrinseca in soluzione di cuprietilendiammina (CED).

[0051] Another aspect of the present invention is the use of a combination of a metal oxide and a borate and/or boric acid, in a fabric bleaching composition comprising a bleach to improve the fabric yellowing-prevention performance and/or fabric safety of the bleaching composition.

pH

[0052] Where the bleaching compositions according to the present invention are liquid compositions, the pH can be from 0 to 14. The borate may also act as a pH buffering means.

[0053] In a preferred embodiment of the present invention wherein the bleaching composition herein is a composition comprising an alkali metal hypochlorite as the bleach, the borate may provide a prolonged through-the-bleach buffering action, i.e., maintains the pH of the bleaching solution at a pH of at least 8, preferably at least 8.5, more preferably at least 9.5 for a longer period of time, throughout the treatment, e.g., at a dilution level of 200:1 (water:composition), as compared to the buffering action obtained with the same composition without said borate.

[0054] It has been found that the buffering action of the borate also affects the hypochlorite/ hypochlorous acid such that the shift of hypochlorite into hypochlorous acid is reduced. It is believed that by controlling this equilibrium, fabric damage and/or fabric yellowing can be reduced.

[0055] In an embodiment, wherein the bleaching compositions herein are liquid and comprise a hypohalite bleach, typically an alkali metal hypochlorite, the recommended pH range of the bleaching composition is from 8 to 14, preferably 8.5 to 14, more preferably from 9 to 13.5, and even more preferably from 9.5 to 13.5. It is in this alkaline pH range that the optimum stability and performance of the hypohalite bleach, e.g., hypochlorite, is obtained.

[0056] If appropriate, the compositions of the present invention may comprise an alkalinity source to adjust the pH of said compositions. The bleaching compositions herein may comprise up to 10%, preferably of from 0.04% to 5% and more preferably of from 0.1% to 2% by weight of the total composition of said alkalinity source.

[0057] Suitable alkalinity sources for use herein are the caustic alkalis, such as sodium hydroxide, potassium hydroxide and/or lithium hydroxide, and/or the alkali metal oxides such, as sodium and/or potassium oxide or mixtures thereof.

[0058] A preferred alkalinity source is a caustic alkali, more preferably sodium hydroxide and/or potassium hydroxide.

[0059] Other suitable alkalinity sources include ammonia, ammonium carbonate and hydrogen carbonate.

[0060] In an embodiment, wherein the bleaching compositions herein are liquid and comprise a peroxygen bleach, the recommended pH range of the bleaching composition to achieve good stability is from 1 to 9, preferably between pH 1 and 8, more preferably between pH 1 and 7 and most preferably between pH 1 and 6.

[0061] If appropriate, the bleaching compositions herein may further comprise an acid to adjust pH of said compositions. The bleaching compositions of the present invention may comprise up to 10%, preferably of from 0.04% to 5% and more preferably of from 0.1% to 2% by weight of the total composition of said acid.

[0062] Suitable acids for use herein may be organic acids, inorganic acids, or mixtures thereof. Preferred organic acids are citric, maleic, oxalic succinic, tartaric acids or mixtures thereof. A preferred inorganic acid is sulphuric acid.

Optional pH buffering components

[0063] The bleaching compositions according to the present invention may further comprise as an optional, but preferred ingredient, a pH buffering component in addition to the pH buffering action that the borate may have. Particularly useful are alkali metal salts of carbonates, polycarbonates, sesquicarbonates, silicates, polysilicates, phosphonates, stannates, alluminates or mixtures thereof. The preferred alkali metal salts to be used herein are sodium and potassium salts. Particularly preferred are alkali metal salts of carbonate. The preferred alkali metal salt of carbonate is sodium carbonate.

[0064] The bleaching compositions of the present invention may comprise up to 10%, preferably from 0.01% to 5% and more preferably from 0.02% to 3% by weight of the total composition of a pH buffering component.

Other optional ingredients

[0065] The bleaching compositions according to the present invention may further comprise other optional ingredients, such as bleach-stable surfactants, pigments, dyes, optical brighteners, solvents, stabilizing agents, enzymes, chelating agents, radical scavengers, polymers, bleach activators, builders, hydrotropes and the like.

Bleach activators

[0066] In an embodiment of the present invention where the bleaching compositions herein comprise a peroxygen bleach, said bleaching compositions may comprise a bleach activator up to a level of 10% by weight of the total composition. Suitable bleach activators to be used herein include those belonging to the class of esters, amides, imides, or anhydrides. Examples of suitable compounds of this type are disclosed in British Patent GB 1 586 769 and GB 2 143 231 and a method for their formation into a prilled form is described in European Published Patent Application EP-A-62 523. Suitable examples of such compounds to be used herein are tetracetyl ethylene diamine (TAED), sodium 3,5,5 trimethyl hexanoyloxybenzene sulphonate, diperoxy dodecanoic acid as described for instance in US 4 818 425 and non-ylamide of peroxyadipic acid as described for instance in US 4 259 201 and n-nonanoyloxybenzenesulphonate (NOBS). Also suitable are N-acyl caprolactams selected from the group consisting of substituted or unsubstituted benzoyl caprolactam, octanoyl caprolactam, nonanoyl caprolactam, hexanoyl caprolactam, decanoyl caprolactam, undecenoyl caprolactam, formyl caprolactam, acetyl caprolactam, propanoyl caprolactam, butanoyl caprolactam pentanoyl caprolactam or mixtures thereof. A particular family of bleach activators of interest was disclosed in EP 624 154, and particularly preferred in that family is acetyl triethyl citrate (ATC). Acetyl triethyl citrate has the advantage that it is environmental friendly as it eventually degrades into citric acid and alcohol. Furthermore, acetyl triethyl citrate has a good hydrolytical stability in the product upon storage and it is an efficient bleach activator.

Stabilizing agents

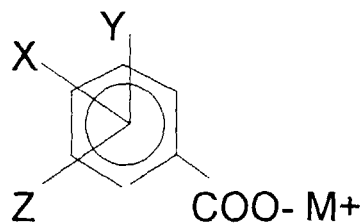
[0067] The bleaching compositions according to the present invention may further comprise a stabilizing agent, preferably a radical scavenger, a chelating agent or a mixture thereof. Naturally, for the purpose of the invention, the stabilizing agents have to be stable to the bleach used.

[0068] Highly preferred stabilizing agents are radical scavengers alone or in combination with a chelating agent.

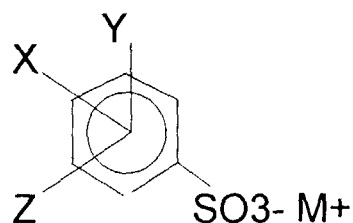
[0069] Suitable radical scavengers for use herein include aromatic radical scavengers comprising an unsaturated ring system of from 3 to 20, preferably of from 3 to 18 and more preferably of from 5 to 14 carbon atoms and having a double bond set comprising a total of $4n+2$ electrons, wherein n is an integer of from 0 to 4, preferably of from 1 to 3. Indeed said aromatic radical scavengers include benzene derivatives, naphthalene derivatives, annulene derivatives, cyclopentadiene derivatives, cyclopropene derivatives and the like, especially aryl carboxylates and/or aryl sulfonates.

[0070] Particularly suitable radical scavengers (aryl carboxylates, aryl sulphonate and derivatives thereof) for use in the present invention have one of the following formulae :

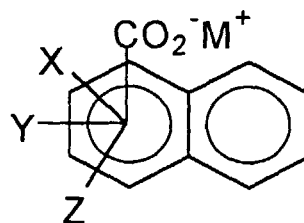
(a)



(b)



c)



d)



wherein each X, Y, and Z are -H, -COO⁻ M⁺, -Cl, -Br, -SO₃⁻ M⁺, -NO₂, -OCH₃, or a C₁ to C₁₀ primary and secondary alkyl groups and M is H or an alkali metal, or mixtures thereof. Examples of these components include pyromellitic acid, i.e., where X, Y and Z are -COO⁻ H⁺; hemimellitic acid, trimellitic acid, i.e., where X and Y are -COO⁻ H⁺ and Z is H. Preferred

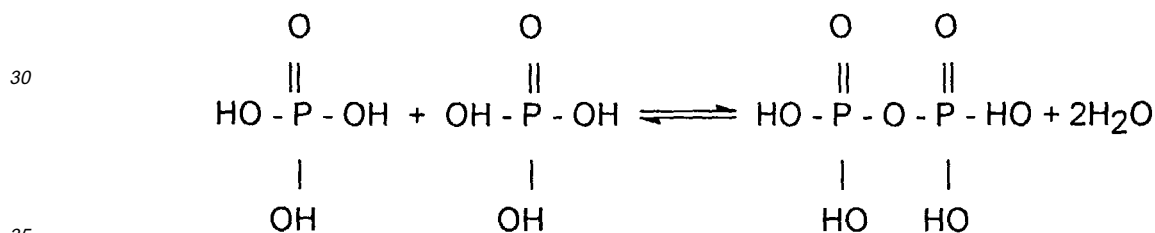
to be used in the present invention as radical scavengers are phthalic acid; sulfophthalic acid; other mono-substituted phthalic acids; di-substituted benzoic acids; alkyl-, chloro-, bromo-, sulfo-, nitro- and alkoxy- benzoic acids, i.e., where Y and Z are -H and X is a C₁ to C₁₀ primary and secondary alkyl groups, -Cl, -Br, -SO₃⁻H⁺, -NO₂ or -OCH₃ (anisic acid) respectively and substituted sulfonic acids. Highly preferred examples of the radical scavengers useful in the present invention are benzoic acid, toluic acid, 4-toluene sulfonic acid, 3-nitro benzoic acid, 2 n-octyl benzoic acid, 2 n-octyl sulfonic acid, anisic acid or mixtures thereof. Most preferred herein are n-anisic acid, benzoic acid, methoxy benzoic acid and/or 3-nitrobenzoic acid.

[0071] The radical scavengers described above are the acidic forms of these species, i.e., M is H. It is intended that the present invention also covers the salt derivatives of these species, i.e., M is an alkali metal, preferably sodium or potassium. In fact, where the pH of the bleaching compositions of the present invention is in the alkaline range, the radical scavengers of the present invention exist primarily as the ionized salt in the aqueous composition herein. The anhydrous derivatives of certain species described herein above can also be used in the present invention, e.g., pyromellitic dianhydride, phthalic anhydride, sulphthalic anhydride and the like.

[0072] Suitable chelating agents for use herein may be any of those known to those skilled in the art such as the ones selected from the group comprising phosphonate chelating agents, phosphate chelating agents, polyfunctionally-substituted aromatic chelating agents, ethylenediamine N,N'-disuccinic acids, or mixtures thereof.

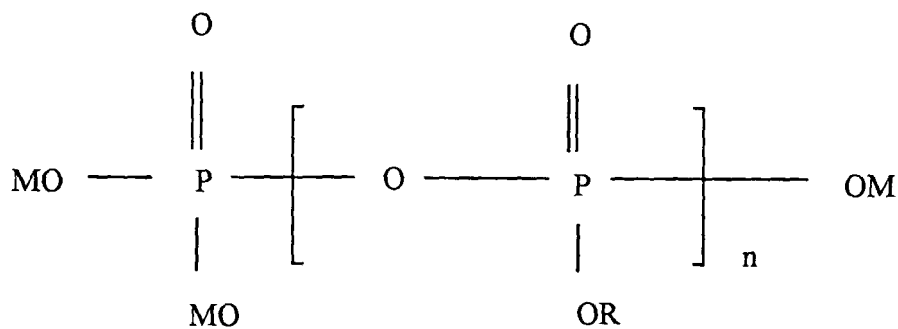
[0073] Suitable phosphonate chelating agents to be used herein may include alkali metal ethane 1-hydroxy diphosphonates (HEDP), alkylene poly (alkylene phosphonate), as well as amino phosphonate compounds, including amino aminotri(methylene phosphonic acid) (ATMP), nitrilo trimethylene phosphonates (NTP), ethylene diamine tetra methylene phosphonates, and diethylene triamine penta methylene phosphonates (DTPMP). The phosphonate compounds may be present either in their acid form or as salts of different cations on some or all of their acid functionalities. Preferred phosphonate chelating agents to be used herein are diethylene triamine penta methylene phosphonate (DTPMP) and ethane 1 -hydroxy diphosphonate (HEDP). Such phosphonate chelating agents are commercially available from Monsanto under the trade name DEQUEST®.

[0074] Suitable phosphate chelating agents are as follows: Phosphonic acid can be condensed in the reaction

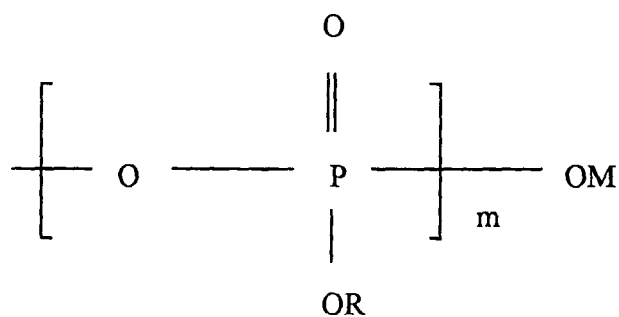


[0075] The reaction can be repeated with any of the reactive OH groups, and phosphate ingredients are obtained which can be

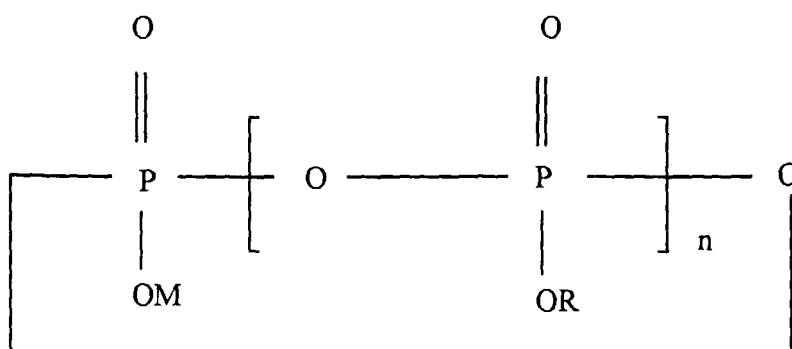
- linear or branched polyphosphates of structure



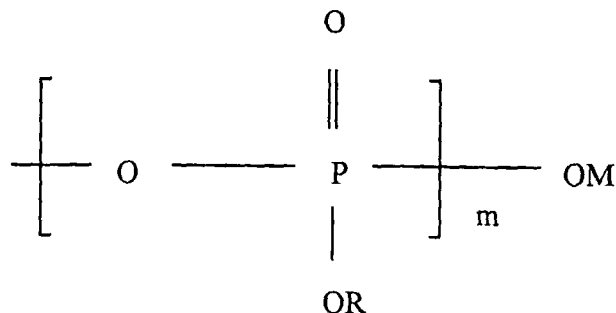
- when R is M or



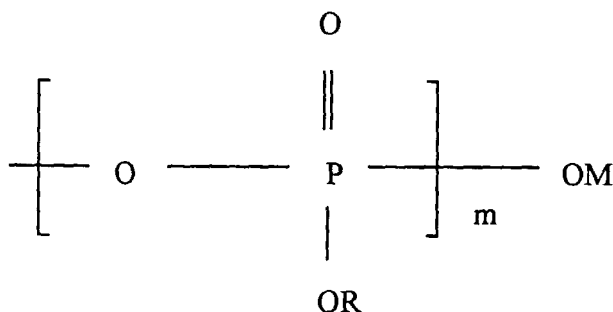
- when M is a counterion, preferably alkali metal;
- when $0 \leq n+m < 500$ (if $n+m = 0$ then the compound is phosphonic acid)
- cyclic polyphosphates (also referred to as metaphosphates), of structure



- when R is M or



- if R is



the phosphate compound contains both cycles and branched chain, and can be referred to as an ultraphosphate.)

- where M is a counterion, preferably an alkali metal
- where $0 \leq n+m < 500$

[0076] All such phosphate ingredients are suitable for use herein, and preferred are linear phosphate ingredients (i.e., R is M) where n is 1 (pyrophosphate) and n is 2 (tripolyphosphate (STPP)), most preferably where n is 2. The most commonly available form of this phosphate is where M is Sodium.

[0077] Phytic acid, which is a chelating agent particularly suitable for use herein, is a hexa-phosphoric acid that occurs naturally in the seeds of many cereal grains, generally in the form of the insoluble calcium-magnesium salt. It may also be derived from corn steep liquor. Commercial grade phytic acid is commercially available from J.T.Baker Co., e.g., as a 40% aqueous solution. It is intended that the present invention covers the acidic form of phytic acid as well as alkali metal salt derivatives thereof, particularly sodium or potassium salts thereof.

[0078] Sodium phytate is available from Jonas Chemical Co (Brooklyn, N.Y.). In fact since the typical pH of the bleaching compositions of the present invention are in the alkaline pH range, the phytic acid component exists primarily as the ionized salt in the liquid compositions herein even if it is added in its acidic form. Mixtures of such salts of phytic acid are also covered.

[0079] Polyfunctionally-substituted aromatic chelating agents may also be useful in the bleaching compositions herein. See U.S. patent 3,812,044, issued May 21, 1974, to Connor et al. Preferred compounds of this type in acid form are dihydroxydisulfobenzenes such as 1,2-dihydroxy-3,5-disulfobenzene.

[0080] A preferred biodegradable chelating agent for use herein is ethylene diamine N,N'-disuccinic acid, or alkali metal, or alkaline earth, ammonium or substitutes ammonium salts thereof or mixtures thereof. Ethylenediamine N,N'-disuccinic acids, especially the (S,S) isomer have been extensively described in US patent 4, 704, 233, November 3, 1987, to Hartman and Perkins. Ethylenediamine N,N'-disuccinic acid is, for instance, commercially available under the tradename ssEDDS® from Palmer Research Laboratories.

[0081] Particularly preferred chelating agents for use herein are phosphate chelating agents like sodium triphosphate, sodium pyrophosphate, phytic acid, and mixtures thereof.

[0082] Typically, the bleaching compositions according to the present invention may comprise from 0.01% to 10%, preferably from 0.01% to 8% by weight, more preferably from 0.1% to 5%, and most preferably from 0.2% to 3% by weight of the total composition of a stabilising agent.

Optional surfactants

[0083] The bleaching compositions of the present invention may further comprise a surfactant or a mixture thereof. Said surfactant may be present in the bleaching compositions according to the present invention in amounts up to 30%, preferably of from 0.1% to 20% and more preferably of from 1% to 10% by weight of the total composition.

[0084] Surfactants for use herein include anionic surfactants, nonionic surfactants, cationic surfactants, amphoteric surfactants, zwitterionic surfactants, and mixtures thereof. Naturally, for the purpose of the invention, the surfactants have to be stable to the used bleach.

[0085] Particularly preferred nonionic surfactants are alkoxyated nonionic surfactants. Suitable alkoxyated nonionic surfactants for use herein include capped and non-capped alkoxyated nonionic surfactants. Suitable non-capped alkoxyated nonionic surfactants are non-capped ethoxyated nonionic surfactants according to the formula $\text{RO}-(\text{C}_2\text{H}_4\text{O})_n\text{H}$, wherein R is a C_6 to C_{22} , preferably a C_8 to C_{22} alkyl chain, or a C_6 to C_{28} alkyl benzene chain, and wherein n is from 0 to 20, preferably from 1 to 15 and, more preferably from 2 to 15 and most preferably from 2 to 12. The preferred R chains for use herein are the C_8 to C_{22} alkyl chains. Propoxyated, butoxyated, ethoxy/butoxyated, ethoxy/propoxyated, butoxy/propoxyated and ethoxy/butoxy/propoxyated nonionic surfactants may also be used herein instead of the ethoxyated nonionic surfactants as defined herein above or together with said surfactants.

[0086] Preferred non-capped ethoxyated nonionic surfactants are according to the formula above and have an HLB (hydrophilic-lipophilic balance) below 16, preferably below 15, and more preferably below 14. Those ethoxyated nonionic surfactants have been found to provide good grease cutting properties.

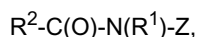
Accordingly suitable non-capped ethoxyated nonionic surfactants for use herein are Dobanol[®] 91-2.5 (HLB= 8.1; R is a mixture of C_9 and C_{11} alkyl chains, n is 2.5), or Lutensol[®] TO3 (HLB=8; R is a C_{13} alkyl chains, n is 3), or Lutensol[®] AO3 (HLB=8; R is a mixture of C_{13} and C_{15} alkyl chains, n is 3), or Tergitol[®] 25L3 (HLB= 7.7; R is in the range of C_{12} to C_{15} alkyl chain length, n is 3), or Dobanol[®] 23-3 (HLB=8.1; R is a mixture of C_{12} and C_{13} alkyl chains, n is 3), or Dobanol[®] 23-2 (HLB=6.2; R is a mixture of C_{12} and C_{13} alkyl chains, n is 2), or Dobanol[®] 45-7 (HLB=11.6; R is a mixture of C_{14} and C_{15} alkyl chains, n is 7) Dobanol[®] 23-6.5 (HLB=11.9; R is a mixture of C_{12} and C_{13} alkyl chains, n is 6.5), or Dobanol[®] 25-7 (HLB=12; R is a mixture of C_{12} and C_{15} alkyl chains, n is 7), or Dobanol[®] 91-5 (HLB=11.6; R is a mixture of C_9 and C_{11} alkyl chains, n is 5), or Dobanol[®] 91-6 (HLB=12.5; R is a mixture of C_9 and C_{11} alkyl chains, n is 6), or Dobanol[®] 91-8 (HLB=13.7 ; R is a mixture of C_9 and C_{11} alkyl chains, n is 8), Dobanol[®] 91-10 (HLB=14.2 ; R is a mixture of C_9 to C_{11} alkyl chains, n is 10), Dobanol[®] 91-12 (HLB=14.5 ; R is a mixture of C_9 to C_{11} alkyl chains, n is 12), or mixtures thereof. Preferred herein are Lutensol[®] TO3, or Lutensol[®] AO3, or Tergitol[®] 25L3, or Dobanol[®] 23-3, or Dobanol[®] 23-6.5, or Dobanol[®] 45-7, Dobanol[®] 91-8, or Dobanol[®] 91-10, or Dobanol[®] 91-12, or mixtures thereof. These Dobanol[®] surfactants are commercially available from SHELL. These Lutensol[®] surfactants are commercially available from BASF and these Tergitol[®] surfactants are commercially available from UNION CARBIDE.

[0087] Suitable capped alkoxyated non-ionic surfactants, having the terminal hydroxyl group capped, are according to the formula R(A)n-O-R1 where R and R1 are independently a C_6 to C_{22} , preferably a C_8 to C_{22} alkyl chain, or a C_6 to C_{28} alkyl benzene chain; A is an ethoxy or propoxy or butoxy unit; n is from 0 to 20, preferably from 1 to 15 and, more preferably from 2 to 15 and most preferably from 2 to 12.

[0088] Suitable chemical processes for preparing the alkoxyated nonionic surfactants for use herein include condensation of corresponding alcohols with alkylene oxide, in the desired proportions. Such processes are well-known to the person skilled in the art and have been extensively described in the art.

[0089] In a preferred embodiment of the present invention the bleaching compositions herein may comprise one of those ethoxyated nonionic surfactants or a mixture of those ethoxyated nonionic surfactants having different HLBs (hydrophiliclipophilic balance). In a preferred embodiment the bleaching compositions herein comprise an ethoxyated nonionic surfactant according to the above formula and having an HLB of up to 10 (i.e., a so called hydrophobic ethoxyated nonionic surfactant), preferably below 10, more preferably below 9, and an ethoxyated nonionic surfactant according to the above formula and having an HLB of above 10 to 16 (i.e., a so called hydrophilic ethoxyated nonionic surfactant), preferably from 11 to 14. Indeed, in this preferred embodiment the bleaching compositions of the present invention typically comprise from 0.01% to 15% by weight of the total composition of said hydrophobic ethoxyated nonionic surfactant, preferably from 0.5% to 10% and from 0.01% to 15% by weight of said hydrophilic ethoxyated nonionic surfactant, preferably from 0.5% to 10%. Such mixtures of ethoxyated nonionic surfactants with different HLBs may be desired as they allow optimum grease cleaning removal performance on a broader range of greasy soils having different hydrophobic/hydrophilic characters.

[0090] Other suitable nonionic surfactants to be used herein include polyhydroxy fatty acid amide surfactants, or mixtures thereof, according to the formula:



wherein R^1 is H, or C_1 - C_4 alkyl, C_1 - C_4 hydrocarbyl, 2-hydroxy ethyl, 2-hydroxy propyl or a mixture thereof, R^2 is C_5 - C_{31} hydrocarbyl, and Z is a polyhydroxyhydrocarbyl having a linear hydrocarbyl chain with at least 3 hydroxyls directly connected to the chain, or an alkoxyated derivative thereof.

[0091] Preferably, R^1 is C_1 - C_4 alkyl, more preferably C_1 or C_2 alkyl and most preferably methyl, R^2 is a straight chain C_7 - C_{19} alkyl or alkenyl, preferably a straight chain C_9 - C_{18} alkyl or alkenyl, more preferably a straight chain C_{11} - C_{18} alkyl or alkenyl, and most preferably a straight chain C_{11} - C_{14} alkyl or alkenyl, or mixtures thereof. Z preferably will be derived from a reducing sugar in a reductive amination reaction; more preferably Z is a glycityl. Suitable reducing sugars include glucose, fructose, maltose, lactose, galactose, mannose and xylose. As raw materials, high dextrose corn syrup, high fructose corn syrup, and high maltose corn syrup can be utilized as well as the individual sugars listed above. These corn syrups may yield a mix of sugar components for Z. It should be understood that it is by no means intended to exclude other suitable raw materials. Z preferably will be selected from the group consisting of $-CH_2-(CHOH)_n-CH_2OH$, $-CH(CH_2OH)-(CHOH)_{n-1}-CH_2OH$, $-CH_2-(CHOH)_2-(CHOR')(CHOH)-CH_2OH$, where n is an integer from 3 to 5, inclusive, and R' is H or a cyclic or aliphatic monosaccharide, and alkoxyated derivatives thereof. Most preferred are glycityls wherein n is 4, particularly $CH_2-(CHOH)_4-CH_2OH$.

[0092] In formula $R^2 - C(O) - N(R^1) - Z$, $N(R^1)$ can be, for example, N-methyl, N-ethyl, N-propyl, N-isopropyl, N-butyl, N-2-hydroxy ethyl, or N-2-hydroxy propyl. $R^2 - C(O) - N$ can be, for example, cocamide, stearamide, oleamide, lauramide, myristamide, capricamide, palmitamide, tallowamide and the like. Z can be 1-deoxyglucityl, 2-deoxyfructityl, 1-deoxymaltityl, 1-deoxylactityl, 1-deoxygalactityl, 1-deoxymannityl, 1-deoxymaltotriosityl and the like.

[0093] Suitable polyhydroxy fatty acid amide surfactants to be used herein are commercially available under the trade name HOE[®] from Hoechst.

[0094] Methods for making polyhydroxy fatty acid amide surfactants are known in the art. In general, they can be made by reacting an alkyl amine with a reducing sugar in a reductive amination reaction to form a corresponding N-alkyl polyhydroxyamine, and then reacting the N-alkyl polyhydroxyamine with a fatty aliphatic ester or triglyceride in a condensation/amidation step to form the N-alkyl, N-polyhydroxy fatty acid amide product. Processes for making compositions containing polyhydroxy fatty acid amides are disclosed for example in GB patent specification 809,060, published February 18, 1959, by Thomas Hedley & Co., Ltd., US patent 2,965,576, issued December 20, 1960 to E.R. Wilson, US patent 2,703,798, Anthony M. Schwartz, issued March 8, 1955, US patent 1,985,424, issued December 25, 1934, to Piggott and WO92/06070, each of which is incorporated herein by reference.

[0095] Suitable anionic surfactants for use herein are all those commonly known by those skilled in the art. Preferably, the anionic surfactants for use herein include alkyl sulphonates, alkyl aryl sulphonates, alkyl sulphates, alkyl alkoxyated sulphates, C_6 - C_{20} alkyl alkoxyated linear or branched diphenyl oxide disulphonates, or mixtures thereof.

[0096] Suitable alkyl sulphonates for use herein include water-soluble salts or acids of the formula RSO_3M wherein R is a C_6 - C_{20} linear or branched, saturated or unsaturated alkyl group, preferably a C_8 - C_{18} alkyl group and more preferably a C_{10} - C_{16} alkyl group, and M is H or a cation, e.g., an alkali metal cation (e.g., sodium, potassium, lithium), or ammonium or substituted ammonium (e.g., methyl-, dimethyl-, and trimethyl ammonium cations and quaternary ammonium cations, such as tetramethyl-ammonium and dimethyl piperdinium cations and quaternary ammonium cations derived from alkylamines such as ethylamine, diethylamine, triethylamine, and mixtures thereof, and the like).

[0097] Suitable alkyl aryl sulphonates for use herein include water-soluble salts or acids of the formula RSO_3M wherein R is an aryl, preferably a benzyl, substituted by a C_6 - C_{20} linear or branched saturated or unsaturated alkyl group, preferably a C_8 - C_{18} alkyl group and more preferably a C_{10} - C_{16} alkyl group, and M is H or a cation, e.g., an alkali metal cation (e.g., sodium, potassium, lithium, calcium, magnesium and the like) or ammonium or substituted ammonium (e.g., methyl-, dimethyl-, and trimethyl ammonium cations and quaternary ammonium cations, such as tetramethyl-ammonium and dimethyl piperdinium cations and quaternary ammonium cations derived from alkylamines such as ethylamine, diethylamine, triethylamine, and mixtures thereof, and the like).

[0098] By "secondary C_6 - C_{20} alkyl or C_6 - C_{20} alkyl aryl sulphonates", it is meant herein that in the formula as defined above, the SO_3M or aryl- SO_3M group is linked to a carbon atom of the alkyl chain being placed between two other carbons of the said alkyl chain (secondary carbon atom).

[0099] An example of a C_{14} - C_{16} alkyl sulphonate is Hostapur[®] SAS available from Hoechst. An example of commercially available alkyl aryl sulphonate is Lauryl aryl sulphonate from Su.Ma.. Particularly preferred alkyl aryl sulphonates are alkyl benzene sulphonates commercially available under trade name Nansa[®] available from Albright&Wilson.

[0100] Suitable alkyl sulphate surfactants for use herein are according to the formula R_1SO_4M wherein R_1 represents a hydrocarbon group selected from the group consisting of straight or branched alkyl radicals containing from 6 to 20 carbon atoms and alkyl phenyl radicals containing from 6 to 18 carbon atoms in the alkyl group. M is H or a cation, e.g.,

an alkali metal cation (e.g., sodium, potassium, lithium, calcium, magnesium and the like) or ammonium or substituted ammonium (e.g., methyl-, dimethyl-, and trimethyl ammonium cations and quaternary ammonium cations, such as tetramethyl-ammonium and dimethyl piperdinium cations and quaternary ammonium cations derived from alkylamines such as ethylamine, diethylamine, triethylamine, and mixtures thereof, and the like).

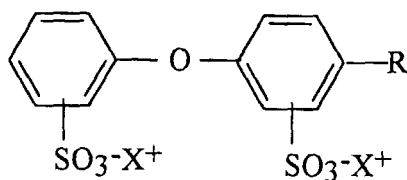
[0101] By "linear alkyl sulphate or sulphonate" it is meant herein a non-substituted alkyl sulphate or sulphonate wherein the alkyl chain comprises from 6 to 20 carbon atoms, preferably from 8 to 18 carbon atoms, and more preferably from 10 to 16 carbon atoms, and wherein this alkyl chain is sulphated or sulphonated at one terminus.

[0102] By "branched sulphonate or sulphate", it is meant herein an alkyl chain having from 6 to 20 total carbon atoms, preferably from 8 to 18 total carbon atoms, and more preferably from 10 to 16 total carbon atoms, wherein the main alkyl chain is substituted by at least another alkyl chain, and wherein the alkyl chain is sulphated or sulphonated at one terminus.

[0103] Particularly preferred branched alkyl sulphates to be used herein are those containing from 10 to 14 total carbon atoms like Isalchem 123 AS[®]. Isalchem 123 AS[®] commercially available from Enichem is a C₁₂₋₁₃ surfactant which is 94% branched. This material can be described as CH₃-(CH₂)_m-CH(CH₂OSO₃Na)-(CH₂)_n-CH₃ where n+m=8-9. Also preferred alkyl sulphates are the alkyl sulphates where the alkyl chain comprises a total of 12 carbon atoms, i.e., sodium 2-butyl octyl sulphate. Such alkyl sulphate is commercially available from Condea under the trade name Isofol[®] 12S. Particularly suitable linear alkyl sulphonates include C₁₂-C₁₆ paraffin sulphonate like Hostapur[®] SAS commercially available from Hoechst.

[0104] Suitable alkyl alkoxyated sulphate surfactants for use herein are according to the formula RO(A)_mSO₃M wherein R is an unsubstituted C₆-C₂₀ alkyl or hydroxyalkyl group having a C₆-C₂₀ alkyl component, preferably a C₁₂-C₂₀ alkyl or hydroxyalkyl, more preferably C₁₂-C₁₈ alkyl or hydroxyalkyl, A is an ethoxy or propoxy unit, m is greater than zero, typically between 0.5 and 6, more preferably between 0.5 and 3, and M is H or a cation which can be, for example, a metal cation (e.g., sodium, potassium, lithium, calcium, magnesium, etc.), ammonium or substituted-ammonium cation. Alkyl ethoxylated sulfates as well as alkyl propoxylated sulfates are contemplated herein. Specific examples of substituted ammonium cations include methyl-, dimethyl-, trimethyl-ammonium and quaternary ammonium cations, such as tetramethyl-ammonium, dimethyl piperdinium and cations derived from alkanolamines such as ethylamine, diethylamine, triethylamine, mixtures thereof, and the like. Exemplary surfactants are C₁₂-C₁₈ alkyl polyethoxylate (1.0) sulfate, C₁₂-C₁₈E(1.0)M, C₁₂-C₁₈ alkyl polyethoxylate (2.25) sulfate, C₁₂-C₁₈E(2.25)M, C₁₂-C₁₈ alkyl polyethoxylate (3.0) sulfate C₁₂-C₁₈E(3.0), and C₁₂-C₁₈ alkyl polyethoxylate (4.0) sulfate C₁₂-C₁₈E(4.0)M, wherein M is conveniently selected from sodium and potassium.

[0105] Suitable C₆-C₂₀ alkyl alkoxyated linear or branched diphenyl oxide disulphonate surfactants for use herein are according to the following formula:



wherein R is a C₆-C₂₀ linear or branched, saturated or unsaturated alkyl group, preferably a C₁₂-C₁₈ alkyl group and more preferably a C₁₄-C₁₆ alkyl group, and X⁺ is H or a cation, e.g., an alkali metal cation (e.g., sodium, potassium, lithium, calcium, magnesium and the like). Particularly suitable C₆-C₂₀ alkyl alkoxyated linear or branched diphenyl oxide disulphonate surfactants to be used herein are the C₁₂ branched diphenyl oxide disulphonic acid and C₁₆ linear diphenyl oxide disulphonate sodium salt respectively commercially available by DOW under the trade name Dowfax 2A1[®] and Dowfax 8390[®].

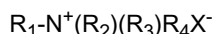
[0106] Other anionic surfactants useful herein include salts (including, for example, sodium, potassium, ammonium, and substituted ammonium salts such as mono-, di- and triethanolamine salts) of soap, C₈-C₂₄ olefinsulfonates, sulphonated polycarboxylic acids prepared by sulphonation of the pyrolyzed product of alkaline earth metal citrates, e.g., as described in British patent specification No. 1,082,179, C₈-C₂₄ alkylpolyglycolethersulfates (containing up to 10 moles of ethylene oxide); alkyl ester sulfonates such as C₁₄₋₁₆ methyl ester sulfonates; acyl glycerol sulfonates, fatty oleyl glycerol sulfates, alkyl phenol ethylene oxide ether sulfates, alkyl phosphates, isethionates such as the acyl isethionates, N-acyl taurates, alkyl succinamates and sulfosuccinates, monoesters of sulfosuccinate (especially saturated and unsaturated C₁₂-C₁₈ monoesters) diesters of sulfosuccinate (especially saturated and unsaturated C₆-C₁₄ diesters), acyl sarcosinates, sulfates of alkylpolysaccharides such as the sulfates of alkylpolyglucoside (the nonionic nonsulfated

compounds being described below), alkyl polyethoxy carboxylates such as those of the formula $RO(CH_2CH_2O)_kCH_2COO-M^+$ wherein R is a C_8-C_{22} alkyl, k is an integer from 0 to 10, and M is a soluble salt-forming cation. Resin acids and hydrogenated resin acids are also suitable, such as rosin, hydrogenated rosin, and resin acids and hydrogenated resin acids present in or derived from tall oil. Further examples are given in "Surface Active Agents and Detergents" (Vol. I and II by Schwartz, Perry and Berch). A variety of such surfactants are also generally disclosed in U.S. Patent 3,929,678, issued December 30, 1975, to Laughlin, et al. at Column 23, line 58 through Column 29, line 23.

[0107] Suitable amphoteric surfactants to be used herein include amine oxides having the following formula $R_1R_2R_3NO$ wherein each of R_1 , R_2 and R_3 is independently a saturated substituted or unsubstituted, linear or branched hydrocarbon chains of from 1 to 30 carbon atoms. Preferred amine oxide surfactants to be used according to the present invention are amine oxides having the following formula: $R_1R_2R_3NO$ wherein R_1 is an hydrocarbon chain comprising from 1 to 30 carbon atoms, preferably from 6 to 20, more preferably from 8 to 16, most preferably from 8 to 12; and wherein R_2 and R_3 are independently substituted or unsubstituted, linear or branched hydrocarbon chains comprising from 1 to 4 carbon atoms, preferably from 1 to 3 carbon atoms, and more preferably are methyl groups. R_1 may be a saturated substituted or unsubstituted linear or branched hydrocarbon chain. Suitable amine oxides for use herein are for instance natural blend C_8-C_{10} amine oxides as well as $C_{12}-C_{16}$ amine oxides commercially available from Hoechst.

[0108] Suitable zwitterionic surfactants for use herein contain both basic and acidic groups which form an inner salt giving both cationic and anionic hydrophilic groups on the same molecule at a relatively wide range of pH's. The typical cationic group is a quaternary ammonium group, although other positively charged groups like phosphonium, imidazolium and sulfonium groups can be used. The typical anionic hydrophilic groups are carboxylates and sulfonates, although other groups like sulfates, phosphonates, and the like can be used.

[0109] A generic formula for preferred zwitterionic surfactants for use herein (i.e., betaine and/or sulfobetaine) is :



wherein R_1 is a hydrophobic group; R_2 is hydrogen, C_1-C_6 alkyl, hydroxy alkyl or other substituted C_1-C_6 alkyl group; R_3 is C_1-C_6 alkyl, hydroxy alkyl or other substituted C_1-C_6 alkyl group which can also be joined to R_2 to form ring structures with the N, or a C_1-C_6 carboxylic acid group or a C_1-C_6 sulfonate group; R_4 is a moiety joining the cationic nitrogen atom to the hydrophilic group and is typically an alkylene, hydroxy alkylene, or polyalkoxy group containing from 1 to 10 carbon atoms; and X is the hydrophilic group which is a carboxylate or sulfonate group, preferably sulfonate group.

[0110] Preferred hydrophobic groups R_1 are aliphatic or aromatic, saturated or unsaturated, substituted or unsubstituted hydrocarbon chains that can contain linking groups such as amido groups, ester groups. More preferred R_1 is an alkyl group containing from 1 to 24 carbon atoms, preferably from 8 to 18, and more preferably from 10 to 16. These simple alkyl groups are preferred for cost and stability reasons. However, the hydrophobic group R_1 can also be an amido radical of the formula: $R_a-C(O)-NR_b-(C(R_c)_2)_m$, wherein R_a is an aliphatic or aromatic, saturated or unsaturated, substituted or unsubstituted hydrocarbon chain containing from 8 up to 20 carbon atoms, preferably an alkyl group containing from 8 up to 20 carbon atoms, preferably up to 18, more preferably up to 16; R_b is either a hydrogen a short chain alkyl or substituted alkyl containing from 1 to 4 carbon atoms, preferably a group selected from the group consisting of methyl, ethyl, propyl, hydroxy substituted ethyl or propyl and mixtures thereof, more preferably methyl or hydrogen; R_c is selected from the group consisting of hydrogen and hydroxy groups, and m is from 1 to 4, preferably from 2 to 3, more preferably 3, with no more than one hydroxy group in any $(C(R_c)_2)$ moiety.

[0111] Preferred R_2 is hydrogen, or an alkyl or substituted alkyl containing from 1 to 4 carbon atoms, preferably a group selected from the group consisting of methyl, ethyl, propyl, hydroxy substituted ethyl or propyl and mixtures thereof, more preferably methyl. Preferred R_3 is a C_1-C_4 carboxylic acid group, a C_1-C_4 sulfonate group, or an alkyl or substituted alkyl containing from 1 to 4 carbon atoms, preferably a group selected from the group consisting of methyl, ethyl, propyl, hydroxy substituted ethyl or propyl and mixtures thereof, more preferably methyl. Preferred R_4 is $(CH_2)_n$ wherein n is an integer from 1 to 10, preferably from 1 to 6, more preferably is from 1 to 3.

[0112] Some common examples of betaine/sulphobetaine are described in U.S. Pat. Nos. 2,082,275, 2,702,279 and 2,255,082, incorporated herein by reference.

[0113] Examples of particularly suitable alkyldimethyl betaines include coconut-dimethyl betaine, lauryl dimethyl betaine, decyl dimethyl betaine, 2-(N-decyl-N, N-dimethyl-ammonia)acetate, 2-(N-coco N, N-dimethylammonio) acetate, myristyl dimethyl betaine, palmityl dimethyl betaine, cetyl dimethyl betaine, stearyl dimethyl betaine. For example Coconut dimethyl betaine is commercially available from Seppic under the trade name of Amonyl 265[®]. Lauryl betaine is commercially available from Albright & Wilson under the trade name Empigen BB/L[®].

[0114] A further example of betaine is Lauryl-immino-dipropionate commercially available from Rhone-Poulenc under the trade name Mirataine H_2C-HA [®].

[0115] Particularly preferred zwitterionic surfactants for use in the bleaching compositions of the present invention are the sulfobetaine surfactants as they deliver optimum soap scum cleaning benefits.

[0116] Examples of particularly suitable sulfobetaine surfactants include tallow bis(hydroxyethyl) sulphobetaine, cocoamido propyl hydroxy sulfobetaines which are commercially available from Rhone Poulenc and Witco, under the trade name of Mirataine CBS® and Rewoteric AM CAS 15® respectively.

[0117] Further examples of amidobetaines/amidosulfobetaine include cocoamidoethylbetaine, cocoamidopropyl betaine or C₁₀-C₁₄ fatty acylamidopropylene(hydropropylene)sulfobetaine. For example C₁₀-C₁₄ fatty acylamidopropylene(hydropropylene)sulfobetaine is commercially available from Sherex Company under the trade name "Varion CAS® sulfobetaine".

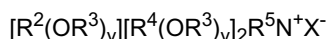
[0118] Suitable amines for use herein are according to the following formula RR'R"N wherein R is a saturated or unsaturated, substituted or unsubstituted, linear or branched alkyl groups containing from 1 to 30 carbon atoms, and preferably from 1 to 20 carbon atoms and wherein R' and R" are independently saturated or unsaturated, substituted or unsubstituted, linear or branched alkyl groups containing from 1 to 30 carbon atoms or hydrogen. Particularly preferred amines to be used according to the present invention are amines having the following formula RR'R"N wherein R is a saturated or unsaturated, linear or branched alkyl group containing from 1 to 30 carbon atoms, preferably from 8 to 20 carbon atoms, more preferably from 6 to 16, most preferably from 8 to 14 and wherein R' and R" are independently substituted or unsubstituted, linear or branched alkyl groups containing from 1 to 4 carbon atoms, preferably from 1 to 3 carbon atoms, and more preferably are methyl groups, or mixtures thereof.

[0119] Suitable amines for use herein are for instance C₁₂ dimethyl amine, coconut dimethyl amine, C₁₂-C₁₆ dimethyl amine. Said amines may be commercially available from Hoechst under the trade name Genamin®, AKZO under the trade name Aromox® or Fina under the trade name Radiamine®.

[0120] Suitable quaternary ammonium surfactants for use herein are according to the formula: R₁R₂R₃R₄N⁺ X⁻, wherein X⁻ is a counteranion such as halogen, methyl sulphate, methyl sulphonate, or hydroxide; R₁ is a saturated or unsaturated, substituted or unsubstituted, linear or branched alkyl group containing from 1 to 30 carbon atoms, preferably from 12 to 20, more preferably from 8 to 20; and R₂, R₃ and R₄ are independently hydrogen, or saturated or unsaturated, substituted or unsubstituted, linear or branched alkyl groups containing from 1 to 4 carbon atoms, preferably from 1 to 3 and more preferably methyl. In highly preferred quaternary ammonium surfactants herein R₁ is a C₁₀-C₁₈ hydrocarbon chain, most preferably C₁₂, C₁₄, or C₁₆; and R₂, R₃ and R₄ are all three methyl, and X is halogen, preferably bromide or chloride, most preferably bromide.

[0121] Examples of quaternary ammonium surfactants are myristyl trimethylammonium methyl sulphate, cetyl trimethylammonium methyl sulphate, lauryl trimethyl ammonium bromide, stearyl trimethyl ammonium bromide (STAB), cetyl trimethyl ammonium bromide (CTAB) and myristyl trimethyl ammonium bromide (MTAB). Highly preferred herein are lauryl trimethyl ammonium salts. Such trimethyl quaternary ammonium surfactants may be commercially available from Hoechst, or from Albright & Wilson under the trade name Empigen CM®.

[0122] Cationic surfactants suitable for use in bleaching compositions of the present invention are those having a long-chain hydrocarbonyl group. Examples of such cationic surfactants include the ammonium surfactants such as alkyl dimethylammonium halogenides, and those surfactants having the formula:



wherein R² is an alkyl or alkyl benzyl group having from 8 to 18 carbon atoms in the alkyl chain, each R³ is selected from the group consisting of -CH₂CH₂-, -CH₂CH(CH₃)-, -CH₂CH(CH₂OH)-, -CH₂CH₂CH₂-, and mixtures thereof; each R⁴ is selected from the group consisting of C₁-C₄ alkyl, C₁-C₄ hydroxyalkyl, benzyl ring structures formed by joining the two R⁴ groups, -CH₂CHOH-CHOHCOR⁶CHOHCH₂OH wherein R⁶ is any hexose or hexose polymer having a molecular weight less than 1000, and hydrogen when y is not 0; R⁵ is the same as R⁴ or is an alkyl chain wherein the total number of carbon atoms of R² plus R⁵ is not more than 18; each y is from 0 to 10 and the sum of the y values is from 0 to 15; and X is any compatible anion.

[0123] Other cationic surfactants useful herein are also described in U.S. Patent 4,228,044, Cambre, issued October 14, 1980, incorporated herein by reference.

Optional polymers

[0124] An optional component of the bleaching compositions herein is a polymer. Naturally, for the purpose of the invention, the polymer has to be stable in the presence of the used bleach.

[0125] Suitable polymers for use are polymers comprising monomeric units selected from the group consisting of unsaturated carboxylic acids, polycarboxylic acids, sulphonic acids, phosphonic acids and mixtures thereof. Co-polymerisation of the above monomeric units among them or with other co-monomers such as styrenesulfonic acid is also suitable.

[0126] Preferred examples of such polymers are the polymers and co-polymers of monomeric units selected from the group consisting of acrylic acid, maleic acid, vinylsulphonic acid and mixtures thereof. Also suitable for use herein are

the above mentioned polymers and co-polymers which are modified in order to contain other functional groups such as aminophosphonic and/or phosphonic units. More preferred polymers are selected from the group consisting of polyacrylate polymers, co-polymers of acrylic and maleic acid, co-polymers of styrene sulphonic acid and maleic acid, and mixtures thereof, preferably modified with aminophosphonic and/or phosphonic groups.

[0127] The molecular weight for these polymers and co-polymers is preferably below 100,000, most preferably between 500 and 50,000. In an embodiment of the present invention where the bleaching compositions herein comprise a hypohalite, preferably a hypochlorite bleach, the most suitable polymers and co-polymers for use herein will be soluble in an amount up to 0.1% by weight, in an aqueous bleaching composition comprising 5% by weight of sodium hypochlorite with its pH adjusted to 13 with sodium hydroxide.

[0128] Commercially available polymers, suitable for use herein, are the polyacrylate polymers sold under the tradename Good-Rite® from BF Goodrich, Acrysol® from Rohm & Haas, Sokalan® from BASF, Norasol® from Norso Haas. Also suitable for use herein are the co-polymers of styrene sulphonic acid and maleic acid, commercially available under the tradename Versaflex® from National Starch such as Versaflex 157, as well as Acumer® terpolymers from Rohm and Haas, in particular Acumer® 3100. Preferred commercially available polymers are the polyacrylate polymers, especially the Norasol® polyacrylate polymers and more preferred are the polyacrylate polymer Norasol® 410N (MW 10,000) and the polyacrylate polymer modified with phosphonic groups Norasol® 440N (MW 4000) and its corresponding acid form Norasol® QR 784 (MW 4000).

[0129] A preferred polymer for use herein is a polyacrylate polymer modified with phosphonic groups commercially available under the tradename Norasol® 440N (MW 4000) and its corresponding acid form Norasol® QR 784 (MW 4000) from Norso-Haas.

[0130] Mixtures of the polymers described herein may also be used in the present invention.

[0131] Polymers herein may be present in low amounts, i.e., in amounts of up to 10% by weight, preferably up to 1%, more preferably from 0.001% to 0.5% by weight, most preferably from 0.005% to 0.2% by weight of the total composition.

Optional brighteners

[0132] The bleaching compositions according to the present invention may also comprise a brightener as an optional ingredient. Naturally, for the purpose of the invention, the brightener has to be stable in the presence of the bleach used.

The brighteners may be desired herein to further enhance the whiteness performance of the bleaching compositions herein.

[0133] Brighteners are compounds which have the ability to fluoresce by absorbing ultraviolet wave-lengths of light and re-emitting visible light. Brighteners, also referred to as fluorescent whitening agent (FWA), have been extensively described in the art, see for instance EP-A-0 265 041, EP-A-0 322 564, EP-A-0 317 979 or "Fluorescent whitening agents" by A.K. Sarkar, published by MERROW, especially page 71-72.

[0134] Commercial optical brighteners which may be useful in the present invention can be classified into subgroups, which include, but are not necessarily limited to, derivatives of stilbene, pyrazoline, coumarin, carboxylic acid, methine-cyanines, dibenzothiophene-5,5-dioxide, azoles, 5- and 6-membered-ring heterocycles, and other miscellaneous agents. Examples of such brighteners are disclosed in "The Production and Application of Fluorescent Brightening Agents", M. Zahradnik, Published by John Wiley & Sons, New York (1982). Further optical brighteners which may also be used in the present invention include naphthlimide, benzoxazole, benzofuran, benzimidazole and any mixtures thereof.

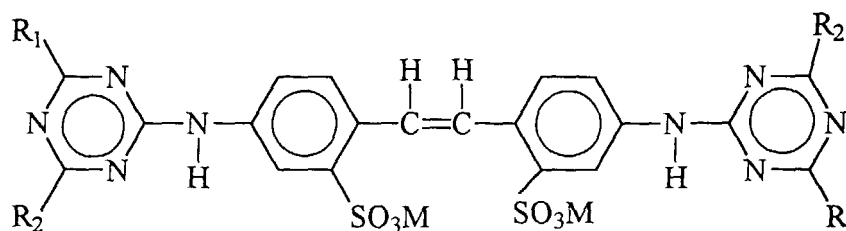
[0135] Examples of optical brighteners which are useful in the present bleaching compositions are those identified in U.S. Patent 4,790,856. These brighteners include the PHORWHITE® series of brighteners from Verona. Other brighteners disclosed in this reference include: Tinopal-UNPA®, Tinopal CBS® and Tinopal 5BM® available from Ciba-Geigy; Artic White CC® and Artic White CWD®; the 2-(4-styryl-phenyl)-2H-naphtho[1,2-d]triazoles; 4,4'-bis(1,2,3-triazol-2-yl)-stilbenes; 4,4'-bis(styryl)biphenyls; and the aminocoumarins.

[0136] Specific examples of brighteners useful herein include 4-methyl-7-diethylamino coumarin; 1,2-bis(-benzimidazol-2-yl)ethylene; 1,3-diphenyl-pyrazolines; 2,5-bis(benzoxazol-2-yl)thiophene; 2-styryl-naphtho-[1,2-d]oxazole; 2-(stilbene-4-yl)-2H-naphtho[1,2-d]triazole; 3-phenyl-7-(isoindolinyl) coumarin; 3-methyl-7-(isoindolinyl) coumarin; 3-chloro-7-(isoindolinyl) coumarin; 4-(isoindolinyl)-4'-methylstilbene; 4-(isoindolinyl)-4'-methoxystilbene; sodium 4-(isoindolinyl)-4'-stilbenesulfonate; 4-(isoindolinyl)-4'-phenylstilbene; 4-(isoindolinyl)-3-methoxy-4'-methylstilbene; 4-(2-chloroisoindolinyl)-4'-(2-methylisoindolinyl)-2,2'-stilbenedisulfonic acid; disodium 4,4'-diisoindolinyl-2,2'-stilbene disulfonate; 4,4'-diisoindolinyl-2,2'-stilbenedisulfonamide; disodium 4,4'-(7,8-dichloro-1-isoindolinyl)2,2'-stilbenedisulfonate; disodium 4,4'-(7-chloro-1-isoindolinyl)2,2'-stilbenedisulfonate; disodium 4,4'-(6-Isopropoxy-1-isoindolinyl)2,2'-stilbenedisulfonate; disodium 4,4'-(7-diisopropyl-1-isoindolinyl)2,2'-stilbenedisulfonate; disodium 4,4'-(7-butoxy-1-isoindolinyl)2,2'-stilbenedisulfonate; disodium 4,4'-(6-trifluoromethyl-1-isoindolinyl)2,2'-stilbenedisulfonate; disodium 4,4'-[6-(1,4,7-trioxanonyl)-1-isoindolinyl]2,2'-stilbenedisulfonate; disodium 4,4'-(7-methoxymethyl-1-isoindolinyl)2,2'-stilbenedisulfonate;

disodium 4,4'-(6-phenyl-1-isoindolyl)2,2-stilbenedisulfonate; disodium 4,4'-(6-naphthyl-1-isoindolyl)2,2-stilbenedisulfonate; disodium 4,4'-(6-methylsulfonyl-1-isoindolyl)2,2-stilbenedisulfonate; disodium 4,4'-(7-cyano-1-isoindolyl)2,2-stilbenedisulfonate; and disodium 4,4'-[7-(1,2,3-trihydroxypropyl)-1-isoindolyl]2,2-stilbenedisulfonate; disodium 4-isoindolyl-4'-ethoxy-2,2'-stilbenedisulfonate; disodium 4-isoindolyl-4'-methoxy-2,2'-stilbenedisulfonate; disodium 4-isoindolyl-4'-ethoxy-2,2'-stilbenedisulfonamide; disodium 4-isoindolyl-4'-methyl-2,2'-stilbenedisulfonamide; 4,4'-bis(4-phenyl-2H-1,2,3-triazol-2-yl)-2,2'-stilbenedisulfonic acid and mixture thereof. See also U.S. Patent 3,646,015, U.S. Patent 3,346,502 and U.S. Patent 3,393,153 for further examples of brighteners useful herein.

[0137] Indeed one of the functionally equivalent derivative salts of 4,4'-bis(4-phenyl-2H-1,2,3-triazol-2-yl)-2,2'-stilbenedisulfonic acid, namely its sodium salt is available from Mobay Chemical Corporation, a subsidiary of Bayer AG under the name Phorwite® CAN. The amine salt is available from Molay under the name Phorwite® CL solution. The potassium salt is available under the name Phorwite® BHC 766.

[0138] Other specific examples of optical brighteners useful in the present invention are those having the structural formula:



wherein R_1 is selected from anilino, N-2-bis-hydroxyethyl and NH-2-hydroxyethyl; R_2 is selected from N-2-bis-hydroxyethyl, N-2-hydroxyethyl-N-methylamino, morpholino, chloro and amino; and M is a salt-forming cation such as sodium or potassium.

[0139] When in the above formula, R_1 is anilino, R_2 is N-2-bis-hydroxyethyl and M is a cation such as sodium, the brightener is 4,4'-bis[(4-anilino-6-(N-2-bis-hydroxyethyl)-s-triazine-2-yl)amino]-2,2'-stilbenedisulfonic acid and disodium salt. This particular brightener species is commercially marketed under the tradename Tinopal-UNPA-GX by Ciba-Geigy Corporation. Tinopal-UNPA-GX is the preferred hydrophilic optical brightener useful in the detergent compositions herein.

[0140] When in the above formula, R_1 is anilino, R_2 is N-2-hydroxyethyl-N-2-methylamino and M is a cation such as sodium, the brightener is 4,4'-bis[(4-anilino-6-(N-2-hydroxyethyl-N-methylamino)-s-triazine-2-yl)amino]-2,2'-stilbenedisulfonic acid disodium salt. This particular brightener species is commercially marketed under the tradename Tinopal 5BM-GX by Ciba-Geigy Corporation.

[0141] When in the above formula, R_1 is anilino, R_2 is morpholino and M is a cation such as sodium, the brightener is 4,4'-bis[(4-anilino-6-morpholino-s-triazine-2-yl)amino]-2,2'-stilbenedisulfonic acid, sodium salt. This particular brightener species is commercially marketed under the tradename Tinopal AMS-GX by Ciba Geigy Corporation.

[0142] Other substituted stilbene 2,2'-disulfonic acid derivatives also include 4-4'-bis (2-2' styryl sulfonate) biphenyl, commercially available from Ciba-Geigy under the trade name Brightener 49® or other hydrophilic brighteners like for example Brightener 3® or Brightener 47®, also commercially available from Ciba-Geigy.

[0143] Further specific examples of brighteners useful in the present invention include the polycyclic oxazole derivatives such as benzo-oxazole derivatives, or mixtures thereof and particularly preferred herein the benzo-oxazole derivatives. An example of such a brightener is benzoxazole,2,2'-(thiophenaldyl)bis having the following formula $C_{18}H_{10}N_2O_2S$, commercially available from Ciba-Geigy under the trade name Tinopal SOP®. This brightener is almost insoluble in water, i.e., it has a solubility being lower than 1 gram per liter. Another example of such a brightener is bis(sulfobenzofuranyl)biphenyl, commercially available from Ciba-Geigy under the trade name Tinopal PLC®.

[0144] Typically the bleaching compositions according to the present invention may comprise up to 1.0%, preferably from 0.005% to 0.5%, more preferably from 0.005% to 0.3% and most preferably from 0.008% to 0.1% by weight of the total composition of a brightener.

Process of treating surfaces:

[0145] In the present invention, the bleaching composition of the present invention is used by applying the composition to the surface to be treated. The compositions according to the present invention are preferably used in liquid form.

[0146] By "in liquid form" it is meant that liquid compositions can be used per se in neat or in diluted form, and solid compositions, for example granules, tablets or powders, are dissolved in an appropriate solvent, typically water, before use, i.e., before being contacted to said surface.

[0147] By "in diluted form", it is meant herein that the compositions for the bleaching of fabrics according to the present invention may be diluted with a solvent by the user, the preferred solvent is water. Such dilution may occur for instance in hand laundry applications as well as by other means such as in a washing machine. Said compositions can be used at a dilution level of up to 500:1 (solvent:composition), preferably from 5:1 to 200:1 and more preferably from 10:1 to 80:1 (solvent:composition).

[0148] By "surfaces", it is meant herein any inanimate surface. These inanimate surfaces include, but are not limited to, hard-surfaces typically found in houses like kitchens, bathrooms, or in car interiors, e.g., tiles, walls, floors, chrome, glass, smooth vinyl, any plastic, plastified wood, table top, sinks, cooker tops, dishes, sanitary fittings such as sinks, showers, shower curtains, wash basins, WCs and the like, as well as fabrics including clothes, curtains, drapes, bed linens, bath linens, table cloths, sleeping bags, tents, upholstered furniture and the like, and carpets. Inanimate surfaces also include household appliances including, but not limited to, refrigerators, freezers, washing machines, automatic dryers, ovens, microwave ovens, dishwashers and so on.

[0149] By "treating a surface", it is meant herein bleaching and/or disinfecting said surfaces as the compositions of the present invention comprise a bleach.

[0150] Thus, the present invention also encompasses a process of treating (e.g., bleaching) a fabric, as the inanimate surface. In such a process a composition according to the present invention is contacted with the fabrics to be treated.

[0151] This can be done either in a so-called "pretreatment mode", where a liquid bleaching composition, as defined herein, is applied neat onto said fabrics before the fabrics are rinsed, or washed then rinsed, or in a "soaking mode" where a bleaching composition, as defined herein, is first diluted in an aqueous bath and the fabrics are immersed and soaked in the bath, before they are rinsed, or in a "through the wash mode", where a bleaching composition, as defined herein, is added on top of a wash liquor formed by dissolution or dispersion of a typical laundry detergent. It is also essential in both cases, that the fabrics be rinsed after they have been contacted with said composition, before said composition has completely dried off.

[0152] The compositions according to the present invention are preferably used on fabrics in a liquid form. More specifically, the process of bleaching fabrics according to the present invention comprises the steps of first contacting said fabrics with a bleaching composition according to the present invention, in its diluted form, then allowing said fabrics to remain in contact with said composition, for a period of time sufficient to bleach said fabrics, typically 1 to 60 minutes, preferably 5 to 30 minutes, then rinsing said fabrics with water. If said fabrics are to be washed, i.e., with a conventional composition comprising at least one surface active agent, said washing may be conducted by contacting said fabrics at the same time with a bleaching composition according to the present invention and said detergent composition, or said washing may be conducted before or after said fabrics have been bleached. Accordingly, said process allows the bleaching of fabrics and optionally the washing of fabrics with a detergent composition comprising at least one surface active agent before the step of contacting said fabrics with said bleaching composition and/or in the step where said fabrics are contacted with said bleaching composition and/or after the step where said fabrics are contacted with the bleaching composition and before the rinsing step and/or after the rinsing step.

[0153] In another embodiment of the present invention the process of bleaching fabrics comprises the step of contacting fabrics with a liquid bleaching composition according to the present invention, in its neat form) of allowing said fabrics to remain in contact with said bleaching composition for a period of time sufficient to bleach said fabrics, typically 5 seconds to 30 minutes, preferably 1 minute to 10 minutes and then rinsing said fabrics with water. If said fabrics are to be washed, i.e., with a conventional composition comprising at least one surface active agent, said washing may be conducted before or after that said fabrics have been bleached. In an embodiment of the present invention wherein the liquid bleaching composition of the present invention, is contacted to the fabrics in its neat form, it is preferred that the level of hypohalite bleach, is from 0.01% to 5%, preferably from 0.1% to 3.5%, more preferably from 0.2% to 2% and most preferably from 0.2% to 1%. Advantageously, the present invention provides liquid hypohalite bleach-containing compositions that may be applied neat onto a fabric to bleach, despite a standing prejudice against using hypochlorite-containing compositions neat on fabrics.

[0154] It is preferred to perform the bleaching process herein before said fabrics are washed. Indeed, it has been observed that bleaching said fabrics with the compositions according to the present invention (diluted and/or neat bleaching processes) prior to washing them with a detergent composition provides superior whiteness and stain removal with less energy and detergent than if said fabrics are washed first, then bleached.

[0155] Alternatively instead of following the neat bleaching process as described above (pretreatment application) with a rinsing step and/or a conventional washing step with a liquid or powder conventional detergent, the bleaching pre-treatment operation may also be followed by the diluted bleaching process as described above either in bucket (hand operation) or in a washing machine.

[0156] In another embodiment the present invention also encompasses a process of treating a hard-surface, as the

inanimate surface. In such a process a composition, as defined herein, is contacted with the hard-surfaces to be treated. The process of treating a hard-surface with a composition, as defined herein, comprises the step of applying said composition to said hard-surface, and optionally rinsing said hard-surface.

[0157] In the process of treating hard-surfaces according to the present invention the composition, as defined herein, may be applied to the surface to be treated in its neat form or may be applied at a dilution level of up to 200:1 (solvent:composition), preferably from 2:1 to 80:1 and more preferably from 2:1 to 60:1 (solvent:composition), preferably using water as the solvent.

[0158] When used as hard-surfaces bleaching/disinfecting compositions the compositions of the present invention are easy to rinse and provide good shine characteristics on the treated surfaces.

Examples

[0159] The following compositions which further illustrate the present invention were made by mixing the listed ingredients in the listed proportions:

Compositions (weight %)	I	II	III	IV	V	VI	VII
Sodium hypochlorite	5.0	5.0	5.0	2.5	2.5	5.0	5.0
Sodium hydroxide	1.3	1.3	1.3	0.7	1.4	0.7	1.4
Sodium carbonate	1.2	1.2	1.2	1.0	1.2	1.2	1.2
Sodium silicate	--	--	--	--	--	0.2	--
Sodium metaborate	1.0	--	0.75	1.0	0.75	0.5	--
Sodium tetraborate	--	--	1.0	--	--	--	--
Sodium borate	--	1.0	--	--	--	--	1.0
ZnO	0.1	0.1	0.2	--	0.2	0.1	--
Al ₂ O ₃	--	--	--	0.1	--	--	0.1
Minors and water	balance up to 100%						
Compositions (weight %)	VIII	IX	X	XI	XII	XIII	XIV
Sodium hypochlorite	5.0	5.0	5.0	2.5	2.5	5.0	5.0
Sodium hydroxide	1.3	1.3	1.3	0.7	1.4	0.7	1.4
Sodium carbonate	1.2	1.2	1.2	1.0	1.2	1.2	1.2
Sodium silicate	--	--	--	--	--	0.2	--
Sodium metaborate	1.0	--	0.75	1.0	0.75	0.5	--
Sodium tetraborate	--	--	1.0	--	--	--	--
Sodium borate	--	1.0	--	--	--	--	1.0
Tinopal PLC®	--	0.01	--	--	--	0.1	--
SnO ₂	0.1	--	0.2	--	--	0.1	--
Bi ₂ O ₃	--	0.1	--	--	0.2	--	--
TiO ₂	--	--	--	0.1	--	--	0.1
Minors and water	balance up to 100%						

Minors and water	balance up to 100%					
Compositions (weight %)	XV	XVI	XVII	XVIII	XIX	
Na Alkylsulphate	2.0	12.0	1.7	2.0	1.7	
Dobanol® 45-7	6.4	6.0	--	6.4	--	
Dobanol® 23-6.5	8.6	6.0	--	8.6	--	
Dobanol® 23-3	--	--	1.5	--	1.5	
Dobanol® 91-10	--	--	1.6	--	1.6	
Acetyl triethyl citrate	--	3.5	3.5	--	3.5	
Hydrogen peroxide	4.0	6.0	6.8	4.0	6.8	
Sodium metaborate	1.0	1.0	1.0	--	--	
Sodium borate	--	--	--	0.75	1.0	
ZnO	0.1	0.1	0.1	0.2	0.1	
Minors and water	balance up to 100%					
H ₂ SO ₄ up to pH 4						
Tinopal PLC® brightener is bis(sulfobenzofuranyl)biphenyl, commercially available from Ciba-Geigy. The Dobanol® surfactants are nonionic surfactants commercially available from SHELL.						

[0160] All the above bleaching compositions provide excellent fabric safety and/or excellent whiteness to fabrics when used to treat fabrics.

Claims

1. A bleaching composition comprising a bleach, a borate or boric acid or a mixture thereof and less than 5% by weight of the total composition of a metal oxide selected from the group consisting of oxides of Sn, Ti, Zn, Al, Bi, Ga, Ge, Sb and Mg or a mixture thereof.
2. A composition according to claim 1 wherein said bleach is a hypochlorite bleach or a peroxygen bleach.
3. A composition according to claim 2 wherein said hypochlorite bleach is an alkali metal or an alkaline earth metal hypochlorite.
4. A composition according to claim 2 wherein said peroxygen bleach is hydrogen peroxide or a source thereof.
5. A composition according to any of the preceding claims wherein said composition is solid.
6. A composition according to any of claims 1 to 4 wherein said composition is liquid.
7. A composition according to claim 6 comprising a hypochlorite bleach wherein said composition has a pH of from 8 to 14.
8. A composition according to any of the preceding claims comprising up to 10% by weight of the total composition of an alkalinity source.
9. A composition according to claim 6 comprising a peroxygen bleach wherein said composition has a pH of from 1 to 9.
10. A composition according to any of the preceding claims comprising up to 10% by weight of the total composition of an acid.

11. A composition according to any of the preceding claims which comprises from 0.01% to 20% by weight of the total composition of said bleach.
- 5 12. A composition according to any of the preceding claims which comprises from 0.01% to 10% by weight of the total composition of said borate or boric acid or a mixture thereof.
13. A composition according to any of the preceding claims wherein said borate is an alkali metal salt of borate or an alkyl borate or a mixture thereof.
- 10 14. A composition according to any of the preceding claims which comprises from 0.001% to 3% by weight of the total composition of said metal oxide.
- 15 15. A composition according to any of the preceding claims wherein said metal oxide is selected from the group consisting of SnO, SnO₂, TiO₂, TiO, ZnO, Al₂O₃, Bi₂O₃, Bi₂O₅, Ga₂O₃, GeO₂, GeO, Sb₂O₃, Sb₃O₅, MgO and a mixture thereof.
- 20 16. A composition according to any of the preceding claims which further comprises an optional ingredient selected from the group consisting of pH buffering components, bleach stable surfactants, pigments, dyes, optical brighteners, polymers, stabilizing agents, enzymes, chelating agents, radical scavengers, bleach activators, solvents, builders, hydrotropes and mixtures thereof.
17. A process of treating fabrics which comprises the step of contacting said fabrics with a composition according to any of the preceding claims, in its diluted form at a dilution level with water of up to 500:1 (water:composition).
- 25 18. A process according to claim 17 wherein said fabrics are washed with a detergent composition comprising at least one surface active agent before the step of contacting said fabrics with said composition and/or during the step where said fabrics are contacted with said composition and/or after a rinsing step after said composition has been removed.
- 30 19. A process of treating a fabric which comprises the steps of contacting said fabric with a composition according to any of claims 1 to 16, in its neat form, allowing said fabric to remain in contact with said composition for a period of time sufficient to treat said fabric, and then rinsing said fabric in water to remove said composition.
- 35 20. A process of treating a hard-surface with a composition according to any of the claims 1 to 16, wherein said process comprises the step of applying said composition to said hard-surface and optionally rinsing said hard-surface.
- 40 21. The use of a combination of a metal oxide and a borate or boric acid or a mixture thereof, in a fabric bleaching composition comprising a bleach to treat a fabric, whereby good yellowing-prevention performance and/or fabric safety is achieved.

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European Patent
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EUROPEAN SEARCH REPORT

Application Number
EP 98 87 0195

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.6)
D,A	DATABASE WPI Section Ch, Week 8414 Derwent Publications Ltd., London, GB; Class A14, AN 84-085858 XP002093763 & JP 59 036198 A (KAO CORP) , 28 February 1984 * abstract * ---	1-3,6-9, 11-13, 15,16,20	C11D3/02 C11D3/395 C11D3/39 C11D3/12 C11D7/54
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A	EP 0 825 250 A (PROCTER & GAMBLE) 25 February 1998 * claims 1-16 * * page 6, line 6 - line 21 * ---	1-4, 6-13, 16-21	TECHNICAL FIELDS SEARCHED (Int.Cl.6) C11D D06L
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A	DATABASE WPI Section Ch, Week 9745 Derwent Publications Ltd., London, GB; Class A97, AN 97-486663 XP002093764 & JP 09 227900 A (SUNSTAR CHEM IND CO LTD) , 2 September 1997 * abstract * -----	1,2,4, 11-13,16	
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Place of search		Date of completion of the search	Examiner
THE HAGUE		17 February 1999	Richards, M
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document			

EPO FORM 1503 03/82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 98 87 0195

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.
The members are as contained in the European Patent Office EDP file on
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17-02-1999

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