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(54) **Membranes of (per)fluorinated amorphous polymers**

(57) Porous membranes of (per)fluorinated amorphous polymers having a porosity in the range 5-500 nm, preferably 20-100 nm, measured by an atomic force electronic microscope; the membrane pore average size distribution being narrow, about 80%-90% of the pores have a size ranging from minus 5 nm to plus 5 nm of the value of the distribution maximum peak.

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Description

[0001] The present invention relates to porous membranes of amorphous fluoropolymers for separation processes.

[0002] Specifically, the invention relates to membranes having improved porosity, in the range 5-500 nm, derived from amorphous (per)fluorinated polymeric films having a flat or cylindrical shape, and usable in separation processes such as ultrafiltration, nanofiltration, in particular of non aqueous solutions of organic solvents containing a solute, for example a dissolved polymer. Another use is in the contactor field wherein an aqueous liquid phase containing a dissolved gas, for example O₂ or CO₂, is contacted with the membrane, through which the gas permeation takes place allowing the control of the gas concentration in the liquid phase.

[0003] The preparation of films and dense non porous membranes of amorphous fluoropolymers obtained by dissolving an amorphous fluoropolymer in a solvent, filtering the solution and then using the spin coating method to coat glass substrata having a low thickness, is known in the prior art. This procedure is described in the European Patent 416,528, wherein as fluoropolymer an amorphous perfluorinated copolymer formed of perfluoro-2,2-dimethyl-1,3-dioxole (PDD) in an amount between 30% and 99% by moles and of tetrafluoroethylene (TFE), is used. The obtained films have a substantially uniform thickness in the range 0.3-20 µm.

[0004] Dense membranes of this type have a chemical-structural porosity, pore size of about some Angstrom (about 10). See W.J. Davies and R.A. Pethrick, Eur. Polym. J. 30 (1994) 1289. These membranes can be used for the separation of a gas from a liquid. The productivity is rather good as regards the gas passing through the membrane. However for the membranes to be used in ultrafiltration processes or for contactors, an even improved productivity is desirable.

[0005] USP 4,948,851 describes articles obtained by the same amorphous perfluorinated copolymers of EP '528, specifically self-supported films having a thickness of about 100-400 µm, prepared by molding of the melted amorphous polymer. With these copolymers it is possible to prepare self-supported films by casting having a thickness of the order of 20-25 µm. The obtained films are dense and do not show porosity as above said, therefore they have the drawbacks mentioned in EP '528.

[0006] USP 5,051,114 describes films obtained by deposition from casting having a thickness of the order of 25 µm. Such films can be used supported on porous substrata to form composite membranes useful for enriching and/or separating gaseous mixtures. In this patent selectivity values for the couple O₂/N₂ from 2.0 to 2.4 are reported, depending on the dioxole PDD content in the PDD/TFE copolymer. It is shown that by increasing the dioxole content, the oxygen permeability increases but the selectivity decreases. Further data relating to the selectivity to gases are given in the Chapter 22 of "Modern Fluoropolymers" (1997) by P.R. Resnick and W.H. Buck, wherein in the case of membranes of copolymers PDD/TFE in molar ratio 87:13 obtained from casting, a value of 5.0 for the couple CO₂/N₂ and 4.4 for the couple H₂/N₂ is mentioned. The membranes described in this patent, even though they have a good gas-permeability, have no porosity and therefore they have the drawbacks mentioned in EP '528.

[0007] In the European Patent Publication 969,025 in the name of the Applicant the preparation of non porous manufactured articles of amorphous fluoropolymers formed of a copolymer of 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxoles (TTD) with tetrafluoroethylene, is described. The obtained membranes are dense and asymmetric and have selectivity towards gases, optionally used on supports suitable for obtaining composite membranes. In order to obtain said membranes under the form of flat dense films the solution is deposited on a smooth flat support with a stratifying knife and the solvent is eliminated by thermal treatment at high temperature, as mentioned in said patent. Tests carried out by the Applicant have shown that in order to obtain said selective membranes, thermal treatments at a temperature close to the boiling point of the used solvent and a subsequent treatment at a temperature near the amorphous polymer T_g, are used. In order to achieve the asymmetric membranes the phase inversion technique is used. According to this technique, after the solution deposition the support coated by the polymer is dipped into a coagulation bath maintained at a predefined temperature and formed of a non-solvent, preferably n-pentane. Such membranes show an improved selectivity in the gaseous separation, but since they have the layer effective to permeation with porosity of chemical-structural type (some Angstrom), they are not very suitable to be used as contactor membranes, for example for the control of the gas concentration in a liquid. Therefore also these membranes show the above mentioned drawbacks.

[0008] Polytetrafluoroethylene (PTFE) porous films (of the order of thousands of Angstrom) are commercially known, obtained by a complex lubricated extrusion process at high temperature, wherein the film porosity is induced by a complex stretching mechanical treatment of the extruded film. Films of this kind are known for example by GORE-TEX®. The membranes obtained from these products are used for ultrafiltration processes. The drawback of these membranes is that they are obtained with an expensive and complex process from the processing point of view.

[0009] The need was therefore felt to have available films of (per)fluorinated amorphous polymers characterized by an improved porosity and obtainable by a cheap industrial process, without using the complex thermomechanical treatments used for the porous films based on foam PTFE available on the market.

[0010] The Applicant has surprisingly found that it is possible to obtain films and porous membranes of (per)fluorinated amorphous polymers suitable for ultrafiltration, nanofiltration processes and in the contactor field, not showing the

prior art drawbacks.

[0011] An object of the invention are porous membranes of (per)fluorinated amorphous polymers having a porosity in the range 5-500 nm, preferably 20-100 nm, determined by an atomic force electronic microscope (FEM).

[0012] The membrane pore average size distribution of the (per)fluorinated amorphous polymers of the invention is very narrow, about 80%-90% of the pores have a size ranging from minus 5 nm to plus 5 nm of the value of the distribution maximum peak. Generally the pore distributions are monodispersed for 80-90% close to the maximum peak value. For example, for a membrane having the distribution maximum peak close to about 60 nm, the pore sizes are for about 80-90% in the range 55-65 nm.

[0013] The pore number for mm² in correspondence of the distribution maximum peak ranges for example from about 20 to 150.

[0014] The porous membranes of (per)fluorinated amorphous polymers of the present invention are obtainable from copolymers of the following monomers:



wherein: Z is selected from F, R_f, OR_f, preferably OR_f; R_f is a perfluoroalkyl radical C₁-C₅; X₁ and X₂ are selected from F and CF₃; Y₁ and Y₂ are selected from F, Cl, CF₃, OR_f, preferably F.

[0015] Also homopolymers of the monomers of structure (I) are usable.

[0016] Other amorphous (per)fluorinated polymers which can be used are those obtained from the cyclopolymerization of monomers having structure (II) with bisvinyloxymethanes having structure (III):



wherein X¹ and X², equal to or different from each other, are F, Cl, preferably F; X³ and X⁴, equal to or different from each other, are F or CF₃. The copolymers of monomer (II) with monomers (III) show cyclic repeating units in the backbone. Said copolymers are described in EP 683,181 in the name of the Applicant, herein incorporated by reference. Also homopolymers of the monomers of structure (III) are usable.

[0017] Alternatively, other amorphous (per)fluorinated polymers which can be used in the present invention are those obtained by the cyclopolymerization of monomers having structure (II) with dienes having structure (IV):



wherein n = 1-5, preferably 1-2.

[0018] Also in this case copolymers showing cyclic repeating units, are obtained. Also homopolymers of the monomers of structure (IV) are usable. Said polymers are described in USP 4,910,276.

[0019] Also amorphous copolymers of the monomers having structure (I) and/or (III) and/or (IV) can be used.

[0020] The amount of one or more comonomers having structure (I), (III), (IV), is to be such to bring to amorphous (per)fluorinated polymers. The skilled in the field is able to easily determine the amount of such comonomers.

[0021] The copolymers obtained from the structures (I) and (II) wherein Z=OR_f with R_f=CF₃, X₁,X₂,Y₁,Y₂=F, are particularly preferred for the present invention results. The dioxole percentage having structure (I) is generally in the range 40%-90% by moles, preferably 50%-85% by moles.

[0022] The dioxole class having structure (I) preferably used in the present invention is mentioned in EP 633,256; still more preferably 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole (TTD) is used.

[0023] The compounds having structure (II) used in the polymerization are tetrafluoroethylene, perfluoroalkylvinylethers (C₁-C₅), hexafluoropropene, chlorotrifluoroethylene, preferably tetrafluoroethylene (TFE) is used.

[0024] The films and membranes of the (per)fluorinated amorphous polymers are obtainable by a casting process

comprising dissolving a copolymer derived from the above mentioned structures in a fluorinated solvent and effecting the solvent evaporation under specific operating conditions.

[0025] Specifically, the process for preparing the porous membranes of the present invention comprises the preparation at room temperature, generally in the range 15°C-25°C, of a solution of the amorphous polymer in a fluorinated solvent; the solution viscosity at 23°C being in the range 5-5,000 cP (centipoise), preferably 10-300 cP; then the solution is spread on an inert support, preferably by a stratifying knife having a defined thickness; it is let evaporate at a constant temperature, preferably equal to the spreading one, lower than 10°C-45°C with respect to the solvent boiling temperature, preferably lower than 20°C-35°C, for a sufficient time to allow a slow solvent evaporation and the consequent formation inside the film of pores having the above mentioned sizes; said time being in the range of about 1-10 days, preferably 3-6 days.

[0026] The fluorinated solvent has generally a boiling temperature in the range 50°C-300°C, preferably 50°C-150°C, still more preferably 50°C-80°C.

[0027] Preferred conditions to carry out the invention process are those wherein the polymeric solution spreading and the solvent evaporation are carried out at a temperature in the range 10°C-40°C and using a dihydrofluoropolyether solvent as defined hereunder having a boiling temperature in the range 55°C-60°C.

[0028] Generally the polymer concentration in the solution is in the range 1-20% by weight, preferably 1-10% by weight.

[0029] With the invention process porous membranes are surprisingly and unexpectedly obtained, whose porosity ranges from 5 to 500 nm and depends on the selected operating conditions. Such porosity is remarkably higher than the porosity of the prior art perfluorinated dense membranes, which only have a porosity of chemical-structural type, with pore size of about some Angstrom (about 10, see the comparative Examples).

[0030] Without to be bound to any theory, the Applicant keeps that the different pore sizes depend on various parameters, such as the solution vapour pressure at the spreading temperature and at the subsequent evaporation temperature, the polymer concentration in the solution which determines the casting solution viscosity.

[0031] As fluorinated solvents for the present invention casting process, (per)fluoropolyethers (Galden[®], Fomblin[®], Krytox[®], Demnum[®]), dihydrofluoropolyethers (H-Galden[®]), fluorinated and perfluorinated ethers Fluorinert[®] (series FC and HFE) optionally containing one or more hydrogen atoms in the end groups, perfluoroalkanes, optionally containing nitrogen and/or oxygen atoms instead of one or more carbon atoms in the backbone, can be used. Generally all the solvents having a solubility parameter similar to that of the used amorphous polymer, are suitable.

[0032] The above mentioned (per)fluoropolyethers comprise one or more of the following units, statistically distributed along the chain, selected from: (C₃F₆O), (C₂F₄O), (CFXO) wherein X is equal to F or CF₃, (CR₁R₂CF₂CF₂O) wherein R₁ equal to or different from R₂ is H, F, perfluoroalkyl C₁-C₃.

[0033] In particular (per)fluoropolyethers containing the following units can be mentioned:

a) -O(C₃F₆O)_m(CFXO)_n- wherein the units (C₃F₆O) and (CFXO) are perfluoroalkylene units statistically distributed along the chain; m' and n' are integers such as to give products having boiling point generally in the range 60°C-300°C, preferably 60°C-150°C, and m'/n' is in the range 5-40, when n' is different from 0; X is equal to F or CF₃; n' can also be 0;

b) -O(C₂F₄O)_p(CFXO)_q-(C₃F₆O)_t wherein p', q' and t' are integers such as to give products having the boiling point mentioned in a), p'/q' ranges from 5 to 0.3, preferably from 2.7 to 0.5; t' can be 0 and q'/(q'+p'+t') lower than or equal to 1/10 and the t'/p' ratio is from 0.2 to 6;

c) -(CR₁R₂CF₂CF₂O)_n- wherein R₁, R₂, equal to or different from each other are H, F, perfluoroalkyl C₁-C₃; n is an integer such as to give products having the boiling point mentioned in a).

[0034] The (per)fluoropolyether end groups are selected from -CF₃, -C₂F₅, -C₃F₇, optionally containing one or two chlorine atoms, -CF₂H, -CFHCF₃.

[0035] The indicated fluoropolyethers are obtainable with the well known processes in the prior art, for example USP 3,665,041, USP 2,242,218, USP 3,715,378, USP 4,954,271 and the European patents EP 239,123, EP 148,482, WO 95/26218.

[0036] The hydrofluoropolyethers contain as end groups one or two end groups selected from -CF₂H, -CFHCF₃. The boiling points of the solvents having hydrogenated end groups are preferably in the range 50°C-250°C, still more preferably 50°C- 150°C.

[0037] Preferably in order to obtain the porous membranes of the present invention for the casting process as a solvent a dihydrofluoropolyether type b) is used, wherein t' is equal to 0, X=F or CF₃ and both the end groups are -CF₂H. Preferably the boiling point is between 50°C and 80°C.

[0038] As a support for the polymer film spreading, glass/quartz, polymethylmethacrylate, polycarbonate, poly-

urethane, polystyrene, ceramic and metal supports, thermoplastic fluoropolymers, preferably glass and polyurethane, can for example be used.

[0039] The porous membranes of the present invention amorphous polymers can be used in separation processes such as ultrafiltration, nanofiltration, in particular of non aqueous solutions of organic solvents containing a solute, for example a dissolved polymer.

[0040] Due to the characteristics of high chemical resistance and hydrophobicity due to the (per)fluorinated structure of the amorphous polymer of the invention, said membranes can be used as contactor membranes having a high productivity. As application of contactor membranes, purification of a fluid containing gaseous impurities put into contact with the side of a membrane, which can be in the supported or self-supported form, can be mentioned. For example in the semiconductor industry, contactor membranes can be used for obtaining ultrapure water free from dissolved gases. The gases to be eliminated, differently from the liquid, pass through the membrane allowing to obtain a purified liquid. Due to the polymer high chemical resistance the invention membranes can be used with aggressive liquids and/or gases.

[0041] Finally the porous membranes of the invention represent an alternative to the foam PTFE films used in multilayer textiles, known by GORETEX[®], but obtainable with a simplified process from the industrial point of view.

[0042] The present invention will now be better illustrated by the following examples, which have a merely indicative purpose and not limitative of the scope of the invention itself.

EXAMPLES

EXAMPLE 1

[0043] The copolymer used in this Example, formed of 60% by moles of 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole (TTD) and of 40% by moles of TFE, is prepared according to EP 633,257.

[0044] A solution of the above mentioned copolymer is prepared at room temperature in a hydrofluoropolyether solvent of type b) wherein t' is equal to 0, X=F and both end groups are -CF₂H and having a boiling point of 55°C (Galden[®] HT55). The solution having a polymer concentration equal to 10% by weight, is let under stirring for 2 days at room temperature, and is prepared by using a jar-rotor. The solution viscosity is 100 cP.

[0045] Then the polymeric film spreading is carried out on a glass sheet by operating at a temperature of 13°C using a stratifying knife Braive Instruments, whose thickness is fixed at 250 µm.

[0046] The solution is then let evaporate for four days at a constant temperature of 13°C.

[0047] A symmetric membrane having a porous structure with pore average size equal to 34 nm is obtained. The pore distribution comprises values ranging from about 20 to 60 nm, about 90% of the pores has sizes in the range 30-38 nm. The pore number per mm² having a 34 nm size is about 20. The obtained membrane thickness is about 15 µm.

[0048] The pore size and number per mm² are determined by an atomic force electronic microscope.

[0049] Such membrane is characterized by permeability tests to separated gaseous flows of O₂, N₂, CO₂. The gas pressure is 5 Kg/cm².

[0050] The gas-permeability has been determined by thermostated volumetric cell at the temperature of 25°C and is expressed in Barrer:

$$1 \text{ Barrer} = (10^{-10} \text{ cm}^3 \text{ xcm}) / (\text{cm}^2 \text{ xcmHgsec})$$

[0051] The permeability value in the various tests is reported in Table 2. The data in the Table show that the invention membrane gives high permeability values and has no selectivity.

EXAMPLE 2

[0052] Example 1 was repeated but carrying out the polymeric film spreading on a glass sheet at a temperature of 24°C using a stratifying knife Braive Instruments, whose thickness is fixed at 250 µm.

[0053] The solution is then let evaporate for four days at a constant temperature of 24°C.

[0054] A symmetric membrane having a porous structure with pore average size equal to 40 nm is obtained. The pore distribution comprises values ranging from about 25 to 50 nm, about 90% of the pores has sizes in the range 38-42 nm. The pore number per mm² having a 40 nm size is about 90. The obtained membrane thickness is about 15 µm.

[0055] Such membrane is characterized by permeability tests to separated gaseous flows of O₂, N₂, CO₂.

[0056] The gas-permeability has been determined by thermostated volumetric cell at the temperature of 25°C.

[0057] The permeability value in the various cases is reported in Table 2. The data in the Table show that the invention membrane gives high permeability values and has no selectivity.

EXAMPLE 3

[0058] Example 1 was repeated but carrying out the polymeric film spreading on a glass sheet at a temperature of 36°C using a stratifying knife Braive Instruments, whose thickness is fixed at 250 µm.

[0059] The solution is then let evaporate for four days at a constant temperature of 36°C.

[0060] A symmetric membrane having a porous structure with pore average size equal to 59 nm is obtained. The pore distribution comprises values ranging from about 40 to 100 nm, about 90% of the pores has sizes in the range 55-64 nm. The pore number per mm² having a 59 nm size is about 120 and represents 80% of the total of the pores. The obtained membrane thickness is about 15 µm.

[0061] Such membrane is characterized by permeability tests to separated gaseous flows of O₂, N₂, CO₂.

[0062] The gas-permeability has been determined by thermostated volumetric cell at the temperature of 25°C and is reported in Table 2.

[0063] The data in the Table show that the invention membrane gives high permeability values and has no selectivity.

EXAMPLE 4

[0064] The copolymer used in this Example, formed of 80% by moles of 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole (TTD) and of 20% by moles of TFE, is prepared according to EP 633,257.

[0065] A solution of the above mentioned copolymer is prepared at room temperature in the perfluoropolyether solvent of Example 1. The solution having a polymer concentration equal to 10% by weight, is let under stirring for 2 days at room temperature, and is prepared by using a jar-rotor. The solution viscosity is 80 cP.

[0066] Then the polymeric film spreading is carried out on a glass sheet operating at a temperature of 18°C using a stratifying knife Braive Instruments, whose thickness is fixed at 250 µm.

[0067] The solution is then let evaporate for four days at a constant temperature of 18°C. A symmetric membrane having a porous structure with pore average size equal to 36 nm is obtained. The obtained membrane thickness is about 15 µm.

[0068] Such membrane is characterized by permeability tests to separated gaseous flows of O₂, N₂, CO₂.

[0069] The permeability values in the various cases are reported in Table 2. The data in the Table show that the invention membrane gives high permeability values and has no selectivity.

EXAMPLE 5 (comparative)

[0070] A solution of the copolymer of Example 1 is prepared, formed of 60% of TTD and of 40% by moles of TFE, in a Galden[®] solvent HT55. The copolymer concentration is equal to 10% by weight. The solution viscosity is 100 cP.

[0071] The solution is deposited at room temperature on a glass support by the stratifying knife of Example 1 having a defined thickness equal to 250 µm. Then the support with the just deposited polymer is dipped in a coagulation bath formed of n-pentane at a temperature of 20°C, obtained by cooling the coagulation bath with ice. In this way an asymmetric membrane is obtained according to the phase inversion technique described in the European Patent Publication 969,025. The thickness of the obtained membrane is about 20 µm.

[0072] The so obtained membrane is characterized by permeability tests to separated gaseous flows of O₂, N₂, CO₂. The data reported in Table 2 show that the membrane has a porosity of structural type, indeed the pore average sizes are of about 5 Å (0.5 nm). Furthermore the membrane of this Example is selective towards different gases. Furthermore from the data it results that the permeability is remarkably lower in comparison with the ones of the membranes of the invention.

[0073] In Table 1 the operating conditions of Examples 1-5 are reported and in Table 2 the gas-permeability values of the various Examples.

TABLE 1

Example	TTD % moles	Solvent	Polymer Concentr. % by wt.	Operating Temperature (°C)
1	60	GALDEN [®] HT55	10	13°C
2	60	GALDEN [®] HT55	10	24°C
3	60	GALDEN [®] HT55	10	36°C

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TABLE 1 (continued)

Example	TTD % moles	Solvent	Polymer Concentr. % by wt.	Operating Temperature (°C)
4	80	GALDEN® HT55	10	18°C
5(comp)	60	GALDEN® HT55	10	20°C

TABLE 2

Example	Average Porosity (nm)	Permeability (Barrer)			Selectivity	
		O ₂	N ₂	CO ₂	O ₂ /N ₂	CO ₂ /N ₂
1	34	1500	1500	1500	1	1
2	40	9000	9000	9000	1	1
3	59	25000	25000	25000	1	1
4	36	3500	3500	3500	1	1
5(comp)	0.5	111	34.1	269	3.2	7.9

25 **Claims**

1. Porous membranes of (per)fluorinated amorphous polymers having a porosity in the range 5-500 nm, preferably 20-100 nm, determined by an atomic force electronic microscope.
2. Porous membranes of (per)fluorinated amorphous polymers according to claim 1, wherein the membrane pore average size distribution is narrow, about the 80%-90% of the pores having a size ranging from minus 5 nm to plus 5 nm of the value of the distribution maximum peak.
3. Porous membranes of (per)fluorinated amorphous polymers according to claims 1-2 obtainable from the polymerization:

A) polymers of one or more monomers having structure (II):



wherein: Y₁ and Y₂ are selected from F, Cl, CF₃, OR_f wherein R_f is a C₁-C₅ perfluoroalkyl radical; with one or more comonomers having the following structures:



wherein: Z is selected from F, R_f, OR_f, preferably OR_f; R_f is a perfluoroalkyl radical C₁-C₅; X₁ and X₂ are selected from F and CF₃; bisvinyloxymethanes having structure (III):



wherein X¹ and X², equal to or different from each other, are F, Cl, preferably F; X³ and X⁴, equal to or different from each other, are F or CF₃; dienes having structure (IV):



wherein n = 1-5, preferably 1-2;
or

B) homopolymers of monomers having structure (I) or (III) or (IV);

C) copolymers of monomers having structure (I) or (III) or (IV).

4. Porous membranes of (per)fluorinated amorphous polymers according to claim 3, wherein the copolymer is derived from the structures (I) and (II) wherein Z=OR_f with R_f=CF₃, X₁,X₂,Y₁,Y₂=F.
5. Porous membranes of (per)fluorinated amorphous polymers according to claims 3-4, wherein the dioxole percentage having structure (I) is in the range 40%-90% by moles, preferably 50%-85% by moles.
6. Porous membranes of (per)fluorinated amorphous polymers according to claims 3-5, wherein as dioxole having structure (I) 2,2,4-trifluoro-5-trifluoromethoxy-1,3-dioxole (TTD) is used.
7. Porous membranes of (per)fluorinated amorphous polymers according to claims 3-6, wherein the monomers having structure (II) are selected from tetrafluoroethylene, perfluoroalkylvinylethers (C₁-C₅), hexafluoropropene, chlorotrifluoroethylene, tetrafluoroethylene (TFE) is preferably used.
8. Process for obtaining the porous membranes of (per)fluorinated amorphous polymers according to claims 1-7, comprising:
- the preparation at room temperature, in the range 15°C-25°C, of a solution of the amorphous polymer in a fluorinated solvent; the solution viscosity at 23°C being in the range 5-5.000 cP (centipoise), preferably 10-300 cP;
 - the solution is spread on an inert support, preferably by a stratifying knife having a defined thickness;
 - it is let evaporate at a constant temperature, preferably equal to that of the spreading, inferior of 10°C-45°C with respect to the solvent boiling temperature, preferably inferior of 15°C-35°C, for such a time as to allow a slow evaporation of the solvent and the consequent formation inside the film of pores having the above mentioned sizes; said time being in the range from about 1 to 10 days, preferably 3-6 days.
9. A process according to claim 8, wherein the fluorinated solvent has preferably a boiling temperature in the range 50°C-300°C, preferably 50°C-150°C.
10. A process according to claims 8-9, wherein the polymer concentration in the solution is in the range 1-20% by weight, preferably 1-10% by weight.
11. A process according to claims 8-10, wherein the fluorinated solvent is selected from (per)fluoropolyethers (Galden[®], Fomblin[®], Krytox[®], Demnum[®]), hydrofluoropolyethers (H-Galden[®]), fluorinated and perfluorinated ethers Fluorinert[®] (series FC and HFE) optionally containing one or more hydrogen atoms in the end groups, perfluoroalkanes.
12. A process according to claims 8-11, wherein the fluorinated solvent is selected from (per)fluoropolyethers containing the following units:
- a) -O(C₃F₆O)_m(CFXO)_n- wherein the units (C₃F₆O) and (CFXO) are perfluoroalkylene units statistically distributed along the chain; m' and n' are integers such as to give products having boiling point generally in the range 60°C-300°C, preferably 60°C-150°C, and m'/n' is in the range 5-40, when n' is different from 0; X is equal to F or CF₃; n' can also be 0;
 - b) -O(C₂F₄O)_p(CFXO)_q-(C₃F₆O)_t wherein p', q'and t' are integers such as to give products having the boiling point mentioned in a), p'/q' ranges from 5 to 0.3, preferably from 2.7 to 0.5; t' can be 0 and q'/(q'+p'+t') lower than or equal to 1/10 and the t'/p' ratio is from 0.2 to 6;
 - c) -(CR₁R₂CF₂CF₂O)_n- wherein R₁, R₂, equal to or different from each other are H, F, perfluoroalkyl C₁-C₃; n

is an integer such as to give products having the boiling point mentioned in a);

the end groups are selected from $-\text{CF}_3$, $-\text{C}_2\text{F}_5$, $-\text{C}_3\text{F}_7$, optionally containing one or two chlorine atoms, $-\text{CF}_2\text{H}$, $-\text{CFHCF}_3$.

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13. A process according to claim 12, wherein the fluorinated solvent is dihydrofluoropolyether of type b) wherein t' is equal to 0, X=F or CF_3 and both end groups are $-\text{CF}_2\text{H}$; the boiling point being in the range 50°C - 80°C .

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14. A process according to claims 8-13, wherein the polymeric solution spreading and the solvent evaporation are carried out at a temperature between 10°C and 40°C and using a fluorinated solvent having a boiling temperature between 55°C and 60°C .

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15. A process according to claims 8-14, wherein the support for the spreading of the polymeric film is selected from: glass/quartz, polymethylmethacrylate, polycarbonate, polyurethane, polystyrene, ceramic and metal supports, thermoplastic fluoropolymers, preferably glass and polyurethane.

16. Use of the porous membranes of (per)fluorinated amorphous polymers according to claims 1-7 in separation processes such as ultrafiltration, nanofiltration and as contactor membranes.

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European Patent
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EUROPEAN SEARCH REPORT

Application Number
EP 00 10 9957

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int.Cl.7)
X	US 5 032 274 A (MILLIPORE CORPORATION) 16 July 1991 (1991-07-16) * column 2, line 41 - line 58 * * column 4, line 65 - column 5, line 13 * * column 6, line 53 - line 63 * * claims 1,2; examples 1-6 *	1,3-10	B01D67/00 B01D71/32 B01D71/34 B01D71/36
X	US 5 489 406 A (MEMTEC LIMITED) 6 February 1996 (1996-02-06) * column 1, line 26 - line 34; examples 1-8 *	1,3,7-10	
X	WO 97 35905 A (H. CHAOUK) 2 October 1997 (1997-10-02) * page 16, paragraph 2 - paragraph 4 * * page 17, paragraph 3 - page 18, paragraph 1 *	1,3	
			TECHNICAL FIELDS SEARCHED (Int.Cl.7)
			B01D
The present search report has been drawn up for all claims			
Place of search MUNICH		Date of completion of the search 7 September 2000	Examiner Luethe, H
CATEGORY OF CITED DOCUMENTS			
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	

EPO FORM 1503 03 82 (P04C01)

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

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07-09-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 5032274 A	16-07-1991	US 4902456 A	20-02-1990
		US 4990294 A	05-02-1991
		DE 68923319 D	10-08-1995
		DE 68923319 T	04-01-1996
		EP 0340732 A	08-11-1989
		JP 2548092 B	30-10-1996
		JP 8034874 A	06-02-1996
		JP 2208329 A	17-08-1990
		JP 2573524 B	22-01-1997
		US 4906377 A	06-03-1990
US 5489406 A	06-02-1996	US 5318417 A	07-06-1994
		US 5277851 A	11-01-1994
		US 5698101 A	16-12-1997
		AU 653528 B	06-10-1994
		AU 7880091 A	27-11-1991
		WO 9117204 A	14-11-1991
		CA 2082511 A	10-11-1991
		DE 69122754 D	21-11-1996
		DE 69130879 D	18-03-1999
		DE 69130879 T	29-07-1999
		EP 0527913 A	24-02-1993
		EP 0734759 A	02-10-1996
		ES 2093705 T	01-01-1997
		ES 2129245 T	01-06-1999
US 5395570 A	07-03-1995		
WO 9735905 A	02-10-1997	AU 716787 B	09-03-2000
		AU 2028997 A	17-10-1997
		AU 716443 B	24-02-2000
		AU 2289797 A	17-10-1997
		BR 9708355 A	03-08-1999
		BR 9708363 A	03-08-1999
		CA 2248003 A	02-10-1997
		CA 2248045 A	02-10-1997
		CN 1214708 A	21-04-1999
		WO 9735906 A	02-10-1997
		EP 0889923 A	13-01-1999
		EP 0889925 A	13-01-1999
		JP 2000508006 T	27-06-2000
		JP 2000508682 T	11-07-2000
		NO 984479 A	27-11-1998
		NO 984480 A	27-11-1998
		US 6060530 A	09-05-2000
AT 187971 T	15-01-2000		
AU 717979 B	06-04-2000		

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

**ANNEX TO THE EUROPEAN SEARCH REPORT
ON EUROPEAN PATENT APPLICATION NO.**

EP 00 10 9957

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

07-09-2000

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
WO 9735905 A		AU 2158297 A	17-10-1997
		AU 707836 B	22-07-1999
		AU 5149096 A	23-10-1996
		BR 9604857 A	26-05-1998
		BR 9708357 A	03-08-1999
		CN 1217003 A	19-05-1999
		DE 69605778 D	27-01-2000
		DE 69605778 T	20-07-2000
		WO 9735904 A	02-10-1997
		EP 0819141 A	21-01-1998
		EP 0889924 A	13-01-1999
		JP 11503473 T	26-03-1999
		JP 2000508005 T	27-06-2000
		NO 974583 A	26-11-1997
		NO 984478 A	27-11-1998
		NZ 304330 A	28-05-1999
		US 5994133 A	30-11-1999
	CA 2248162 A	02-10-1997	
	US 6015609 A	18-01-2000	