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(54) **METHODS FOR CONTROLLING THE RHEOLOGICAL PROPERTIES OF VISCOELASTIC SURFACTANTS BASED FLUIDS**

VERFAHREN ZUR KONTROLLE DER RHEOLOGISCHEN EIGENSCHAFTEN VON  
VISCOELASTISCHEN TENSID-BASIERTE BOHRFLÜSSIGKEITEN

PROCEDES PERMETTANT DE CONTROLER LES PROPRIETES RHEOLOGIQUES DE FLUIDES  
A BASE DE TENSIOACTIFS VISCOELASTIQUES

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**EP 1 573 169 B1**

**Description****Technical Field of the Invention**

5     **[0001]** This invention relates to compositions and methods used in adjusting the rheological properties of viscoelastic surfactant (VES) fluids, especially for use in treatment of subterranean formations and oil and gas wells.

**Background of the Invention**

10    **[0002]** Viscoelastic surfactant fluids are normally made by mixing in appropriate amounts suitable surfactants such as anionic, cationic, nonionic and zwitterionic surfactants in an aqueous medium. The rheology of viscoelastic surfactant fluids, in particular the increase in viscosity of the solution, is attributed to the three dimensional structure formed by the components in the fluids. When the surfactant concentration significantly exceeds a critical level, and eventually subject to the presence of an electrolyte, the surfactant molecules aggregate and form structures such as micelles that can  
15    interact to form a network exhibiting viscoelastic behavior. In the remaining part of this description, the term "micelle" will be used as a generic term for organized interacting species.

20    **[0003]** Viscoelastic surfactant solutions are usually formed by the addition of certain reagents to concentrated solutions of surfactants, frequently consisting of long-chain quaternary ammonium salts such as cetyltrimethylammonium bromide (CTAB). Common reagents that generate viscoelasticity in the surfactant solutions are salts such as ammonium chloride, potassium chloride, sodium salicylate and sodium isocyanate and non-ionic organic molecules such as chloroform. The electrolyte content of surfactant solutions is also an important control on their viscoelastic behavior.

25    **[0004]** There has been considerable interest in using such viscoelastic surfactants in wellbore-service applications. Reference is made for example to U.S. Patents Nos. 4,695,389; 4,725,372; 5,551,516, 5,964,295, and 5,979,557.

30    **[0005]** The rheological properties of aqueous mixtures of surfactants are determined by their tendency to seclude their hydrophobic part, and expose their hydrophilic part, toward the solvent. This behavior typically results in the formation of three-dimensional network structure, called micelles. Depending in particular upon the structure of these micelles, the fluid viscosity is more or less increased, and the fluid may exhibit both viscous and elastic behavior.

35    **[0006]** The common approach to develop new viscoelastic-surfactant systems is to screen a large number of surfactants - and surfactant mixtures - until one meets specific performance specifications. This approach is obviously time-consuming. Moreover, wellbore services fluids tend to be used under a large variety of conditions, notably temperature, salinity and shear stress. Unfortunately, viscoelastic-surfactants based-fluids are typically very sensitive to variations of the above-mentioned parameters. Therefore the "screening" approach-tends to result in numerous systems that are tailored for specific conditions. This presents logistical issues and requires extensive training of field personnel.

40    **[0007]** Consequently, it would be desirable to have one system whose properties could be adjusted to meet a variety of specifications. For example, consider the possibility of using a particular viscoelastic surfactant system throughout a broad temperature range. It is known that the micelles responsible for the rheological properties of viscoelastic surfactant-based fluids are normally stable within a narrow temperature range. Surfactants with longer carbon-atom hydrophobic chains (more than 18 carbon atoms) offer fluid stability at higher temperatures. However, increasing the chain length is also detrimental to the surfactant's hydrophilic properties; therefore, complete dissolution of the surfactant requires considerably more time than that of shorter chain counterparts. There is therefore a need for means to "boost" the viscosity of shorter-chain systems at higher temperatures. It should be further emphasized that some relatively inexpensive viscoelastic surfactants may provide an increase of viscosity that is less than it would be desirable for some applications. Providing means to boost the viscosity would be a way of allowing the use of "less than perfect" product - or to limit the quantity of surfactant to be added to the systems and therefore decreasing the total cost of the system.

45    **[0008]** Another property of viscoelastic surfactant-based systems is their shear sensitivity. For instance, in the oil industry, it is often favorable to provide fluids that exhibit high viscosity at little or no shear and low viscosity at high shear. Such fluids are easy to pump but will be highly viscous after placement in the well. Though the shear-sensitivity is an intrinsic property of most viscoelastic systems, an independent aspect is the degree of viscosity-recovery or re-healing once the fluid is no more subject to high shear. Controlling the degree of reassembling (re-healing) is necessary  
50    to maximize performance of the surfactant system for different applications. For example, in hydraulic fracturing it is critical for the fluid to regain viscosity as quickly as possible after exiting the high-shear region in the tubulars and entering the low-shear environment in the hydraulic fracture. On the other hand, it is beneficial in coiled tubing cleanouts to impart a slight delay in regaining full viscosity in order to more efficiently "jet" the solids from the bottom of the wellbore into the annulus. Once in the annulus the regained viscosity will ensure that the solids are effectively transported to the surface.  
55    Improving the viscosity-recovery and minimizing the time required for such recovery is therefore desirable.

60    **[0009]** Finally, it is well known that the introduction of certain components to a viscoelastic surfactant-based system can cause a dramatic decrease in the fluid viscosity, called "breaking". Breaking can also occur by varying the amount of water or electrolyte or other components that may already be present in the fluid. For example, in oilfield applications,

the viscosity of viscoelastic surfactant fluids is reduced or lost upon exposure to formation fluids (e.g., crude oil, condensate and/or water). The viscosity reduction effectuates cleanup of the reservoir, fracture, or other treated area.

[0010] However, in some circumstances, it would be suitable to have a better control of that breaking, for instance, when breaking of the fluid is desired at a particular time or condition, when it is desired to accelerate viscosity reduction or when the natural influx of reservoir fluids (for example, in dry gas reservoirs) does not break or breaks incompletely the viscoelastic surfactant fluid.

[0011] This disclosure describes compositions and methods employed to modify the rheology of aqueous solutions comprising a thickening amount of a viscoelastic surfactant.

[0012] UK Patent GB2332223, "Viscoelastic surfactant based gelling composition for wellbore service fluids" by Hughes, Jones and Tustin describes methods to delay and control the build-up of viscosity and gelation of viscoelastic surfactant based gelling compositions. These methods are used to facilitate placement of the delayed ("pre-gel") fluid into a porous medium and then to trigger formation of the viscoelastic gel in-situ.

[0013] Rose et. al. describe in U.S. Patent No. 4,735,731 several methods to reversibly break the viscosity of viscoelastic-surfactant based solutions through an intervention at surface. These methods include heating/cooling the fluid, adjusting the pH or contacting the fluid with an effective amount of a miscible or immiscible hydrocarbon and then, subjecting the fluid to conditions such that the viscosity of the fluid is substantially restored. The reversible treatment of Rose is useful for drilling fluids so that the fluid pumped into the well is viscous enough to carry cuttings to the surface but able to be broken at surface for solids removal. The breaking methods discussed in Rose are not used to break a viscoelastic solution down a well and further appear to have an immediate impact on the viscosity of the fluid.

[0014] U.S. Patent application No. 09/826,127 filed April 4, 2001 and published under No. 20020004464 discloses different types of breaking agents and different means to achieve a delayed release of the breaking agents downhole so that the rheological properties of the aqueous fluids are not altered at surface or during the injection phase. U.S. Application No. 10/194,522 filed July 12, 2002 further discloses that some polymers, in particular some polyelectrolytes, can be used as breaking agents.

[0015] However, it was further found that the same types of polymers could also have completely different effects on the rheology of aqueous solutions comprising thickening amount of viscoelastic surfactants. Therefore, there exists a need for methods for breaking/enhancing/healing viscoelastic surfactant fluids after subterranean oil- or gas-well treatments, at predetermined times or conditions.

## Summary of the Invention

[0016] The authors of the present invention have found that, at given viscoelastic-surfactant concentration, a polymer can perform different functions (breaker, viscosity enhancer or viscosity recovery enhancer), depending upon its molecular weight and its concentration in the fluid, or more precisely, depending on the ratio of the concentration of added polymer and the concentration of viscoelastic surfactant.

[0017] More particularly, it was found that if the added polymer has a low molecular weight, the polymer mainly acts as a breaking agent.

[0018] According to the present invention, there is provided a method of treating a subterranean formation, the method comprising injecting down a well an aqueous fluid comprising a thickening amount of a viscoelastic surfactant, and being characterized by adding to the aqueous fluid a polymer additive or precursor thereof that causes a substantial reduction in viscosity of the thickened fluid and has a molecular weight of less than 25,000.

[0019] The methods of the present inventions are focused upon but not limited to rheology modifiers for viscoelastic surfactant systems based upon cationic surfactants such as erucyl methyl bis(2-hydroxyethyl) ammonium chloride ("EM-HAC"); zwitterionic surfactants such as betaine surfactants; and anionic surfactants such as the oleic acid derivatives. However, the methods and compositions described herein are also presented for adjusting the viscosity of viscoelastic surfactant fluids based on anionic, cationic, nonionic and zwitterionic surfactants.

[0020] It is a feature of the invention to provide methods and compositions for the delayed adjustment of the viscosity of the viscoelastic surfactant gelling compositions without significantly or substantially compromising the initial fluid properties required for proppant suspension and transport during a fracturing treatment. The invention thus concerns a method of treating a subterranean formation by injecting down a well an aqueous fluid comprising a thickening amount of a viscoelastic surfactant and also comprising a viscosity-adjustor or a precursor thereof. Optimized formulations ensure that the viscoelastic gel rapidly formed under surface conditions remains stable during pumping and placement into the fractures. Then, at a later time, the gel viscosity is significantly altered by the added polymer.

[0021] The shear sensitivity and hydration of the viscoelastic system fluid can be fine-tuned based on the need for the application.

[0022] A preferred feature of the present invention relates to the use of polyelectrolytes as breakers of viscoelastic-surfactants based solutions. Polyelectrolytes useful in the invention may be anionic, cationic, nonionic or zwitterionic. Although it should be understood that any suitable polyelectrolyte may be used, the following are preferred: sulfonated

polynaphthalenes, sulfonated polystyrenes and sulfonated styrene/maleic anhydride polymers. More specifically, polyethylene glycol PEG, polypropylene glycol (PPG), block co-polymers of PEG and PPG, polynaphthalene sulfonate and polystyrene sulfonate are preferred.

**[0023]** It should be understood that the fracturing compositions of the invention may contain components in addition to water, electrolytes, surfactants and breakers. Such additional components are, for example, acids, bases, buffers, chelating agents for the control of multivalent cations, freezing point depressants, etc.

**[0024]** Even if the present application is focused on treatments of hydrocarbon wells, the methods and compositions of the invention can also be employed for other applications, including but not limited to water wells, recovery of coalbed methane, and the containment or remediation of ground or groundwater contamination.

## Brief Description of the Drawings

**[0025]** Figure 1 shows a plot of the polymer molecular weight vs. the weight ratio of polymer to viscoelastic surfactant and the effect of the addition of polymer to the aqueous solution comprising said viscoelastic surfactant.

**[0026]** Figure 2 is a graph showing the effect of a low molecular weight sodium polystyrene sulfonate on a betaine based VES fluid at temperature ranging from 21° C (70°F) to 154°C (310°F).

**[0027]** Figure 3 is a graph showing the effect of a low molecular weight sodium polystyrene sulfonate on a betaine based VES fluid at temperature ranging from 21° C (70°F) to 154°C (310°F).

**[0028]** Figure 4 is a graph showing the effect of a high molecular weight sodium polynaphthalene sulfonate at high concentration on a betaine based VES fluid at temperature ranging from 10°C (50°F) to 149°C (300°F).

**[0029]** Figure 5 is a graph showing the effect of a high molecular weight sodium polystyrene sulfonate at high concentration on a betaine based VES fluid at temperature ranging from 66°C (150°F) to 110°C (230°F).

**[0030]** Figure 6 is a graph showing the effect of a high molecular weight sodium polystyrene sulfonate at high concentration on a cationic quaternary amine based VES fluid at temperature ranging from 10°C (50°F) to 127°C (260°F).

**[0031]** Figure 7 is a graph showing the effect of a high molecular weight sodium polystyrene sulfonate at low concentration on a cationic quaternary amine based VES fluid after shear degradation of the viscosity.

**[0032]** Figure 8 is a graph showing the effect of a high molecular weight sodium polynaphthalene sulfonate at low concentration on a betaine based VES fluid after shear degradation of the viscosity.

**[0033]** Figure 9 is a graph showing the effect of a high molecular weight sodium polynaphthalene sulfonate at low concentration on a betaine based VES fluid after shear degradation of the viscosity.

**[0034]** Figure 10 is a graph showing the effect of a short molecular weight polypropylene glycol on a betaine based VES fluid at temperature ranging from 21° C (70°F) to 154°C (310°F).

## Detailed Description

### EXAMPLE 1: Additions of Polymers to adjust the viscosity

**[0035]** In the application of viscoelastic surfactant based gelling compositions comprising viscoelastic surfactants in combination with polymers, drastically different effects may be achieved depending on the molecular weight of the polymer and the weight ratio of added polymers to the viscoelastic surfactant.

**[0036]** For examples, to an aqueous solution comprising a surfactant (in test A to F and I a zwitterionic surfactant noted Z1, Erucic amidopropyl dimethyl betaine) and in test G and H, a cationic quaternary amine) is added different type of polymers (PNS: polynaphthalene sulfonate or PSS: polystyrene sulfonate or PPG: polypropylene glycol), whose molecular weight is listed in table I below. The weight ratio of polymer to the surfactant is noted by Wt%. Table I

Table I

| Test# | Surfactant | Polymer | Molecular Weight | Wt%  | Effect  |
|-------|------------|---------|------------------|------|---------|
| A     | Z1         | PNS     | 30000            | 17.5 | Booster |
| B     | Z1         | PNS     | 100000           | 10   | Booster |
| C     | Z1         | PNS     | 30000            | 2.5  | Healer  |
| D     | Z1         | PNS     | 70000            | 2.5  | Healer  |
| E     | Z1         | PSS     | 15000            | 2.5  | Breaker |
| F     | Z1         | PSS     | 3000             | 2.5  | Breaker |
| G     | C1         | PNS     | 30000            | 10   | Booster |

(continued)

| Test# | Surfactant | Polymer | Molecular Weight | Wt%  | Effect  |
|-------|------------|---------|------------------|------|---------|
| H     | C1         | PNS     | 30000            | 5    | Healer  |
| I     | Z1         | PPG     | 4000             | 12.5 | Breaker |

[0037] Based on the above data, figure 1 was drawn by plotting the different tests using as X-axis the molecular weight of the added polymer and as Y-axis the weight ratio Wt%. This diagram can be divided in three sectors. With polymers of lower molecular weights, the viscosity is broken. With polymers of higher molecular weights, the viscosity of the solution is either enhanced (if the concentration of polymer is high enough) or the additive promotes the recovery of the viscosity after shear-degradation (healer effect).

#### EXAMPLE 2: Polymer as Breaking Agent

[0038] A base fluid was prepared by adding to water 2.4 weight percent of erucic amidopropyl dimethyl betaine. 0.06wt% (weight percent) of polystyrene sulfonate (having a molecular weight estimated between 15000 and 20000) is added to the solution so that the value of Wt% is equal to 2.5%. The fluid viscosities with and without the polymer additive were determined at 100 sec<sup>-1</sup> from 21° C (70°F) to 154°C (310°F) and plotted figure 2. Figure 2 shows that a substantial decrease in fluid viscosity is observed when the polystyrene sulfonate breaker is present. This reduction in fluid viscosity is permanent.

#### EXAMPLE 3: Polymer as Breaking Agent

[0039] A base fluid was prepared by adding to water 2.4weight percent of erucic amidopropyl dimethyl betaine. 0.06wt% of polystyrene sulfonate (having a molecular weight estimated between 3000 and 5000) is added to the solution so that the value of Wt% is equal to 2.5%. The fluid viscosities with and without the polymer additive were determined at 100 sec<sup>-1</sup> from 21° C (70°F) to 154°C (310°F) and plotted figure 2. Figure 3 shows that a substantial decrease in fluid viscosity is achieved upon adding the polystyrene sulfonate breaker. This reduction in fluid viscosity is permanent.

#### EXAMPLE 4: Polymer as Viscosity Booster

[0040] A base fluid was prepared by adding to water 2.4wt% of erucic amidopropyl dimethyl betaine. 0.42wt% of polynaphthalene sulfonate (having a molecular weight estimated to be about 30000) is added to the solution so that the value of Wt% is equal to 17.5%. The fluid viscosities with and without the polymer additive were determined at 100 sec<sup>-1</sup> from 10°C (50°F) to 149°C (300°F) and plotted figure 4. Figure 4 shows that a substantial increase in fluid viscosity is achieved by adding the polymer, and this increase is especially significant at temperatures between about 93°-104°C (200°-220°F).

#### EXAMPLE 5: Polymer as Viscosity Booster

[0041] A base fluid was prepared by adding to water 2.wt% of erucic amidopropyl dimethyl betaine. 0.24wt% % of polystyrene sulfonate (having a molecular weight estimated of about 1,000,000) is added to the solution so that the value of Wt% is equal to 10%. The fluid viscosities with and without the polymer additive were determined at 100 sec<sup>-1</sup> from 66°C (150°F) to 110°C (230°F) and plotted Figure 5. Again, a substantial increase in fluid viscosity is achieved by the addition of the polymer.

#### EXAMPLE 6: Polymer as Viscosity Booster

[0042] A base fluid was prepared by adding to water 4 weight percent of cationic quaternary amine and 4% potassium chloride. Polystyrene sulfonate (having a molecular weight of about 30,000) is added at a concentration of 2.4 kg/1000 litres (20lb/1000 gal) of base fluid, corresponding to a weight ratio of 10%. The viscosity of the base fluid with/without the polymer additive were determined at 100 sec<sup>-1</sup> from 10°C (50°F) to 127°C (260°F) and plotted Figure 6. A significant increase of the viscosity was observed at the lower and higher temperatures. Some viscosity reduction was observed within the intermediate temperature range.

**EXAMPLE 7: Polymer as Healer**

**[0043]** A base fluid was prepared by adding to water 4wt% of cationic quaternary amine and 4% potassium chloride. Polystyrene sulfonate (having a molecular weight of about 30,000) is added at a concentration of 1.2 kg/1000 litres (10lb/1000 gal) of base fluid, corresponding to a weight ratio of 5%. The fluid is subject to a shear of  $5,000 \text{ sec}^{-1}$  for 3 minutes. The viscosity of the base fluid with/without the polymer additive was determined at  $1 \text{ sec}^{-1}$  and  $21^\circ\text{C}$  ( $70^\circ\text{F}$ ) and is plotted along time Figure 7. The addition of the polymer provides a quick recovery of the viscosity when the high shear was terminated.

**EXAMPLE 8: Polymer as Healer**

**[0044]** A base fluid was prepared by adding to water 2.wt% of erucic amidopropyl dimethyl betaine. 0.06 wt% of polystyrene sulfonate (having a molecular weight estimated to be about 30,000) is added to the solution so that the value of Wt% is equal to 2.5%. The fluid is subject to a shear of  $5,000 \text{ sec}^{-1}$  for 3 minutes.. The viscosity of the base fluid with/without the polymer additive were determined at  $1 \text{ sec}^{-1}$  and  $21^\circ\text{C}$  ( $70^\circ\text{F}$ ) is plotted along time Figure 8. The addition of the polymer provides a quick recovery of the viscosity when the high shear was terminated.

**EXAMPLE 9: Polymer as Healer**

**[0045]** A base fluid was prepared by adding to water 2.4wt% of erucic amidopropyl dimethyl betaine. 0.06wt% of polynaphthalene sulfonate (having a molecular weight estimated of about 70,000) is added to the solution so that the value of Wt% is thus equal to 2.5%. The fluid is subject to a shear of  $5,000 \text{ sec}^{-1}$  for 3 minutes. The viscosity of the base fluid with/without the polymer additive were determined at  $1 \text{ sec}^{-1}$  and  $21^\circ\text{C}$  ( $70^\circ\text{F}$ ) is plotted along time Figure 9. The addition of the polymer provides a quick recovery of the viscosity at no shear.

**EXAMPLE 10: Polymer as Breaker**

**[0046]** A base fluid was prepared by adding to water 6 weight percent of erucic amidopropyl dimethyl betaine VES fluid. 0.3 weight percent % of polypropylene glycol (having a molecular weight estimated of about 4,000) is added to the solution so that the value of Wt% is thus equal to 12.5%. The viscosity of the base fluid with/without the polymer additive were determined at  $100 \text{ sec}^{-1}$  and plotted along temperature Figure 10. The addition of the polymer provides a decrease of viscosity.

**[0047]** The preceding description of specific embodiments of the present invention is not intended to be a complete list of every possible embodiment of the invention. Persons skilled in this field will recognize that modifications can be made to the specific embodiments described here that would be within the scope of the present invention. In particular, though the different embodiments of the present invention were optimised for hydraulic fracturing applications, the invention is also applicable to numerous other oil field applications using surfactant-based complex fluids such as acidizing, gravel packing, coiled tubing cleanup, and other novel chemical treatments.

**Claims**

1. A method of treating a subterranean formation from a well, the method comprising the steps of:

- a. providing an aqueous fluid comprising a thickening amount of a viscoelastic surfactant; and a polymer additive that causes a reduction in viscosity of the thickened fluid compared to viscosity of the aqueous fluid with the thickening amount of the viscoelastic surfactant alone; wherein the polymer additive has a molecular weight of less than 25,000; and
- b. injecting the aqueous fluid down the well.

2. The method according to claim 1, wherein said polymer additive is a polyelectrolyte.

3. The method according to claim 2, wherein said polyelectrolyte is encapsulated.

4. The method according to claim 2 or claim 3, wherein said polymer additive is selected from the group consisting of sulfonated polynaphthalenes, sulfonated polystyrenes and sulfonated styrene/maleic anhydride polymers.

5. The method according to any of claims 1 to 4, wherein said viscoelastic surfactant is selected from the group

consisting of anionic surfactant, cationic surfactant and, zwitterionic surfactant.

6. The method according to claim 5, wherein said viscoelastic surfactant is selected from the group consisting of erucyl methyl bis(2-hydroxyethyl)ammonium chloride ("EMHAC"), betaine surfactants; and oleic acid derivatives.

## Patentansprüche

1. Verfahren zum Behandeln einer unterirdischen Formation aus einem Brunnen, umfassend:

- a. die Bereitstellung eine wässrige Flüssigkeit aus einer Verdickung Höhe der viskoelastischen Tensid und Additiv ein Polymer, das eine Verringerung der Viskosität der Flüssigkeit im Vergleich zu dicker Viskosität der wässrigen Flüssigkeit mit der Verdickung Höhe der viskoelastischen Tensid allein, daß das Polymer Zusatzstoff eine Molekulargewicht von weniger als 25.000, und  
b. Injektion der wässrigen Flüssigkeit in die Brunnen.

2. Verfahren nach Anspruch 1, bei dem das Polymer Additiv ist ein Polyelektrolyt.

3. Verfahren nach Anspruch 2, bei dem die Polyelektrolyt ist gekapselt.

4. Verfahren nach Anspruch 2 oder Anspruch 3, bei dem das Polymer Zusatzstoff ausgewählt ist aus der Gruppe bestehend aus sulfonierten polynaphthalenes, sulfonierten Polystyrole und sulfonierten Styrol / Maleinsäureanhydrid-Polymere.

5. Verfahren nach einem der Ansprüche 1 bis 4, wobei gesagt viskoelastische Tensid ausgewählt ist aus der Gruppe bestehend aus anionischen Tensiden, kationischen Tensiden und zwitterionic Tensids.

6. Verfahren nach Anspruch 5, **dadurch** sagte viskoelastische Tensid ausgewählt ist aus der Gruppe bestehend aus erucyl Methyl-bis (2-hydroxyethyl) ammoniumchlorid ( "EMHAC"), Betain Tenside und Ölsäure Derivate.

## Revendications

1. Méthode de traitement d'une formation souterraine d'un puits, la méthode comprenant les étapes:

- a. fournir un fluide aqueux comprenant une quantité épaississante d'un agent tensioactif viscoélastique, et d'un additif polymère provoquant une réduction de la viscosité du fluide épaissi par rapport à la viscosité du fluide aqueux avec la quantité épaississante de l'agent tensioactif viscoélastique seul; dans lequel l'additif est un polymère de poids moléculaire inférieur à 25000, et  
b. injecter le fluide aqueux dans le puits.

2. La méthode selon la revendication 1, **caractérisé en ce que** l'additif est un polymère polyelectrolyte.

3. La méthode selon la revendication 2, **caractérisé en ce que** le polyelectrolyte est encapsulée.

4. La méthode selon la revendication 2 ou la revendication 3, **caractérisé en ce que** l'additif polymère est choisi parmi le groupe: polynaphthalenes sulfonés, polystyrènes sulfonés et polymères styrène/anhydride maléique sulfonés.

5. La méthode selon l'une quelconque des revendications 1 à 4, **caractérisé en ce que** l'agent tensioactif viscoélastique est choisi parmi le groupe d'agent tensioactif anionique, cationique et zwitterionique.

6. La méthode selon la revendication 5, **caractérisé en ce que** l'agent tensioactif viscoélastique est choisi parmi le groupe : erucyl méthyl bis (2-hydroxyéthyl) ammonium chlorure ( "EMHAC"), betaine et les dérivés de l'acide oléique.

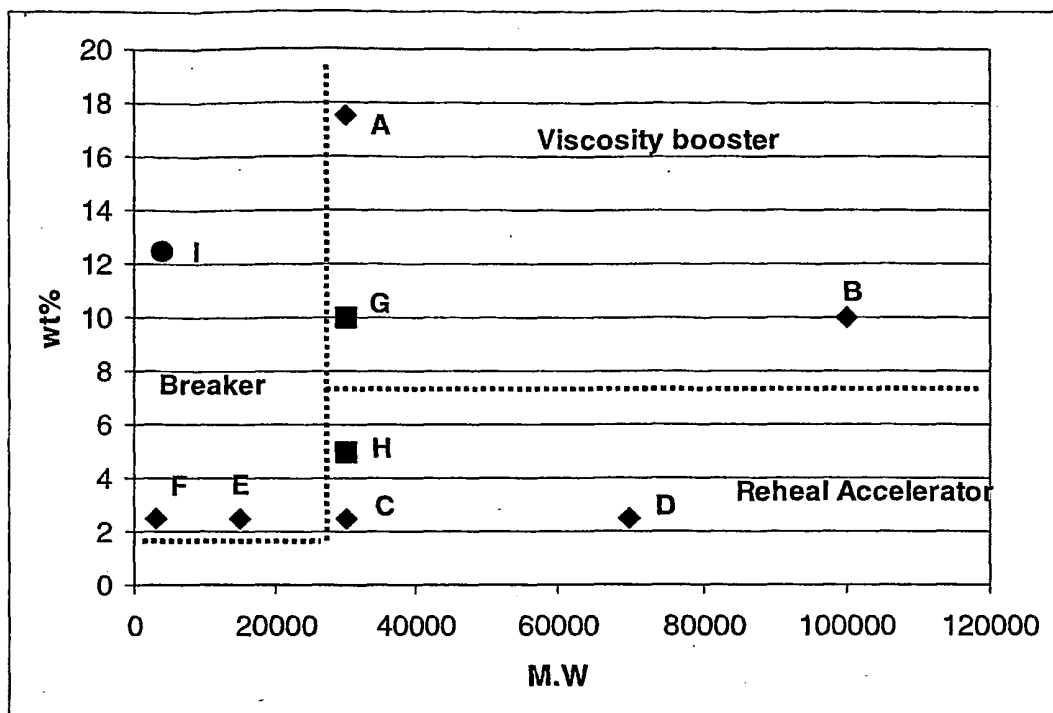


Figure 1

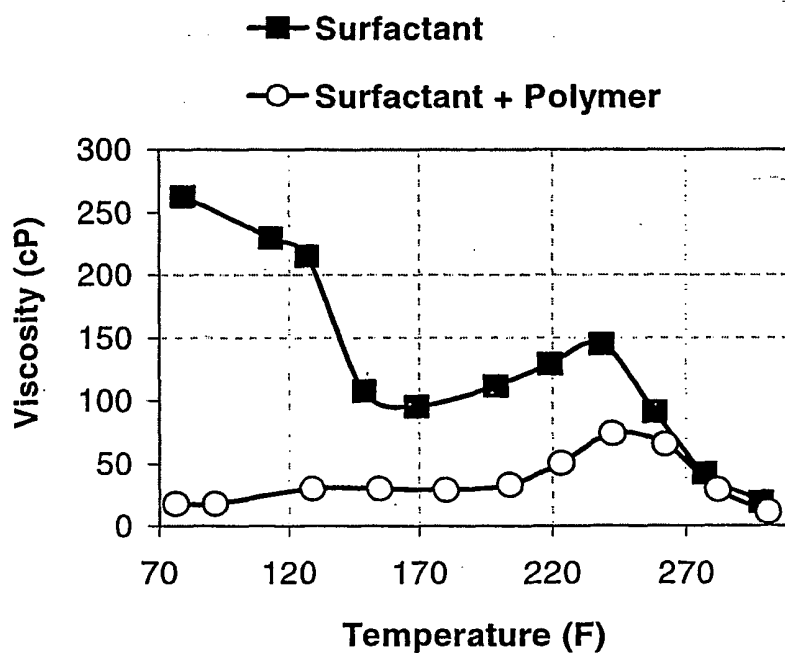


Figure 2

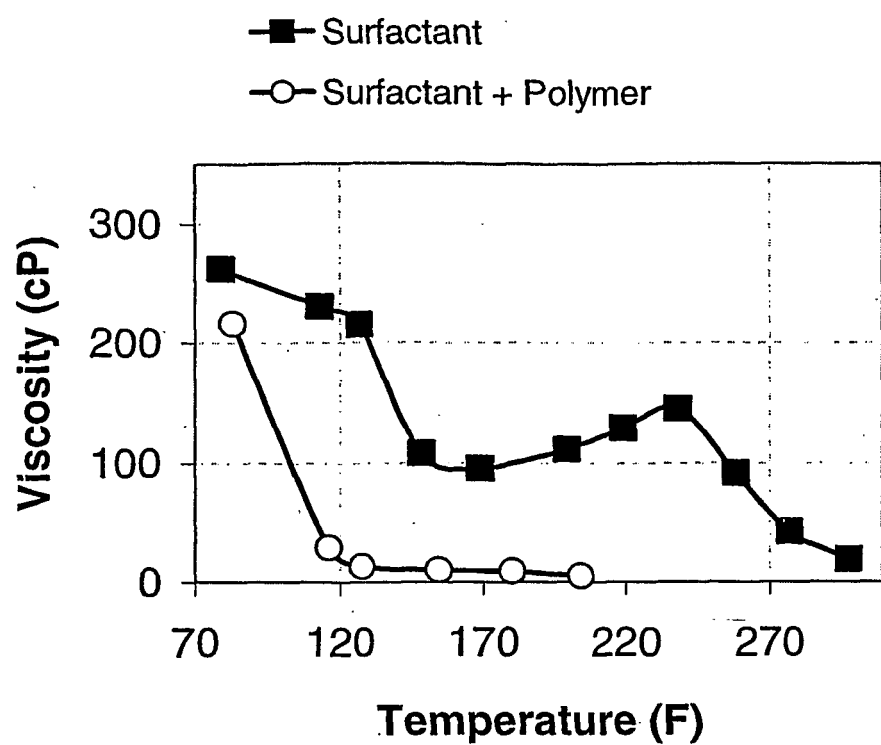


Figure 3

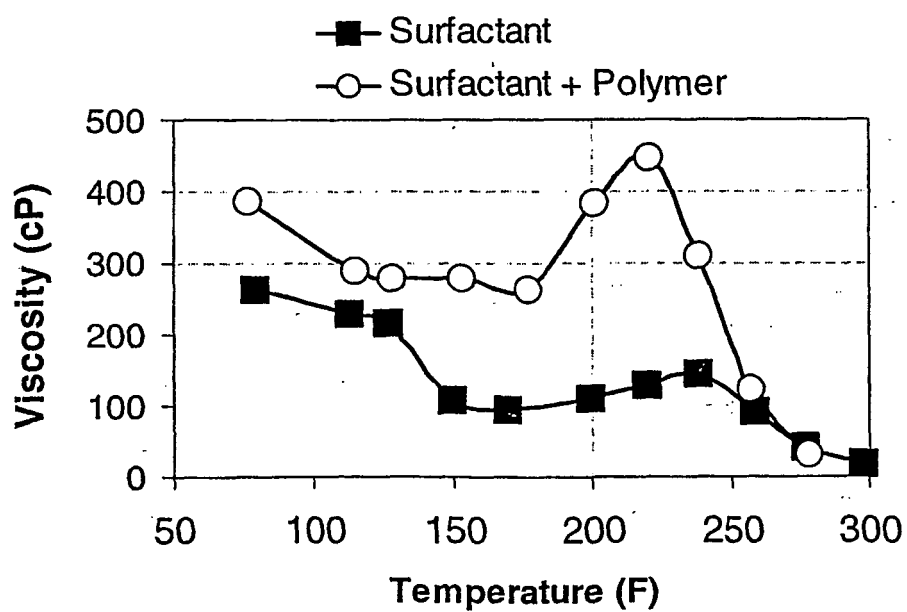


Figure 4

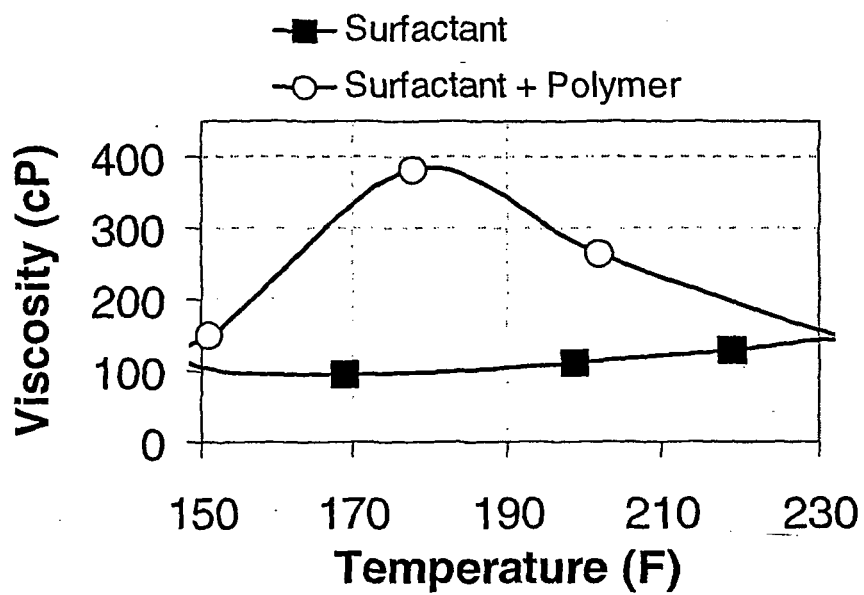


Figure 5

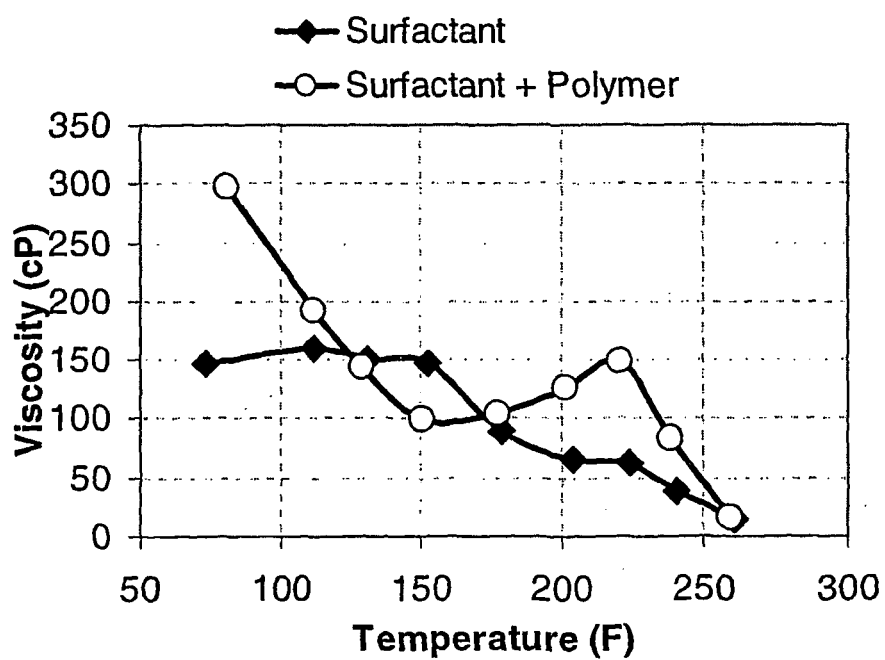


Figure 6

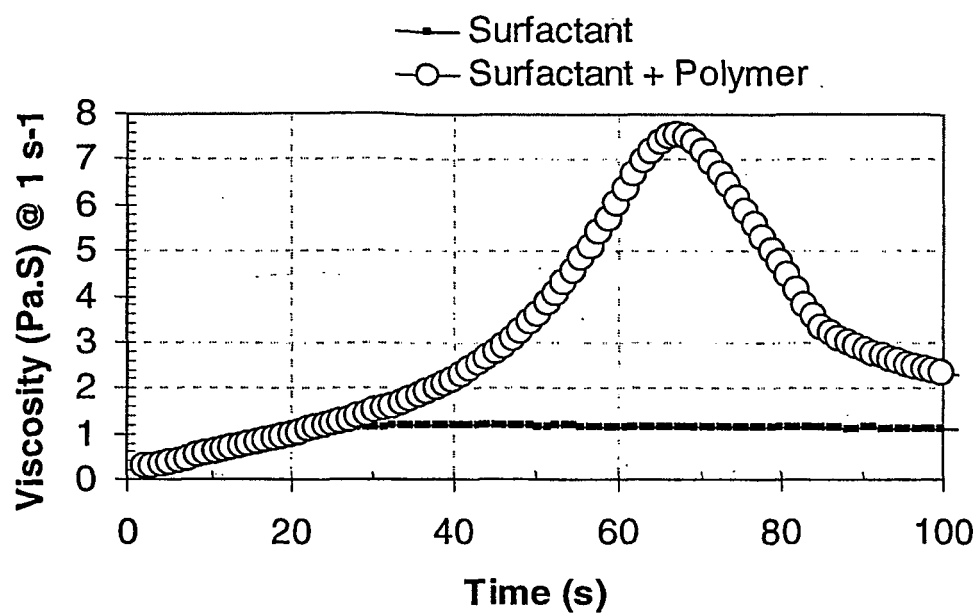


Figure 7

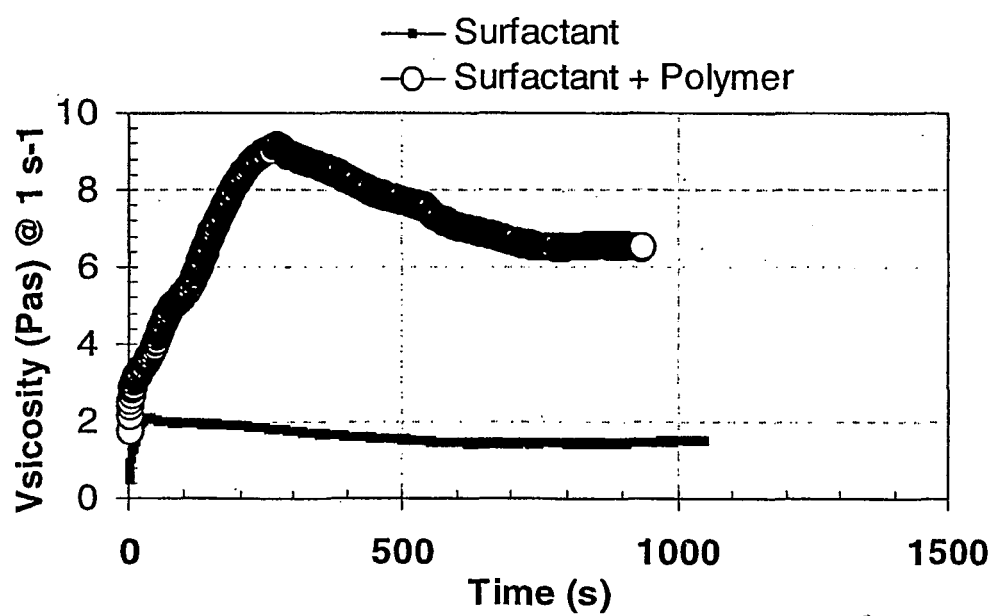


Figure 8

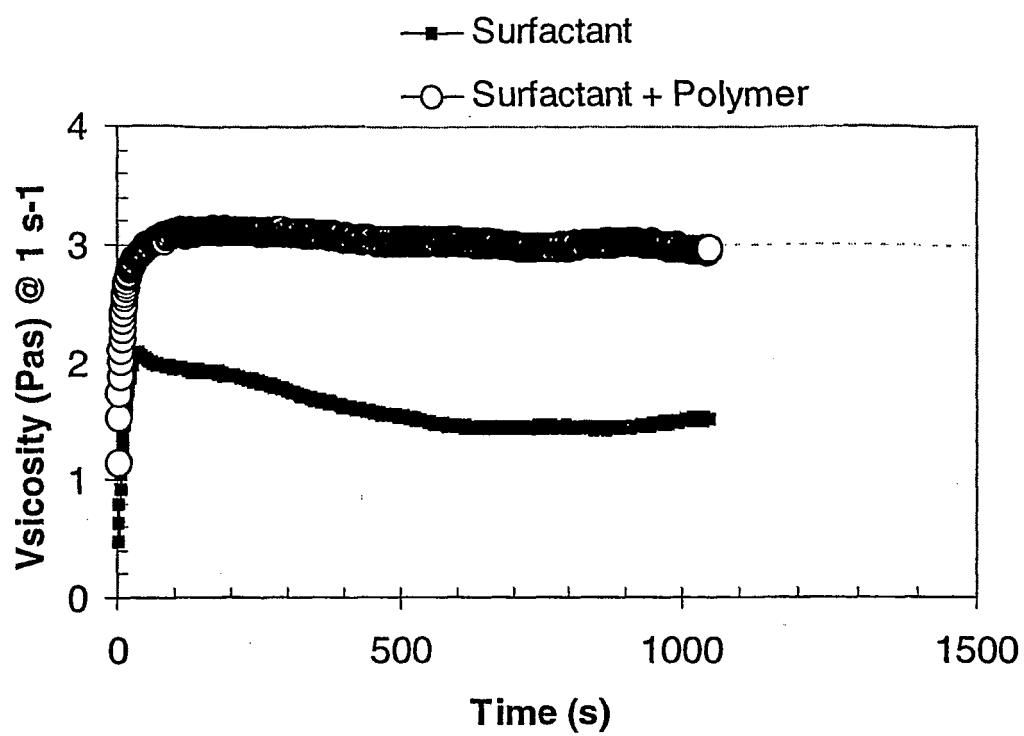


Figure 9

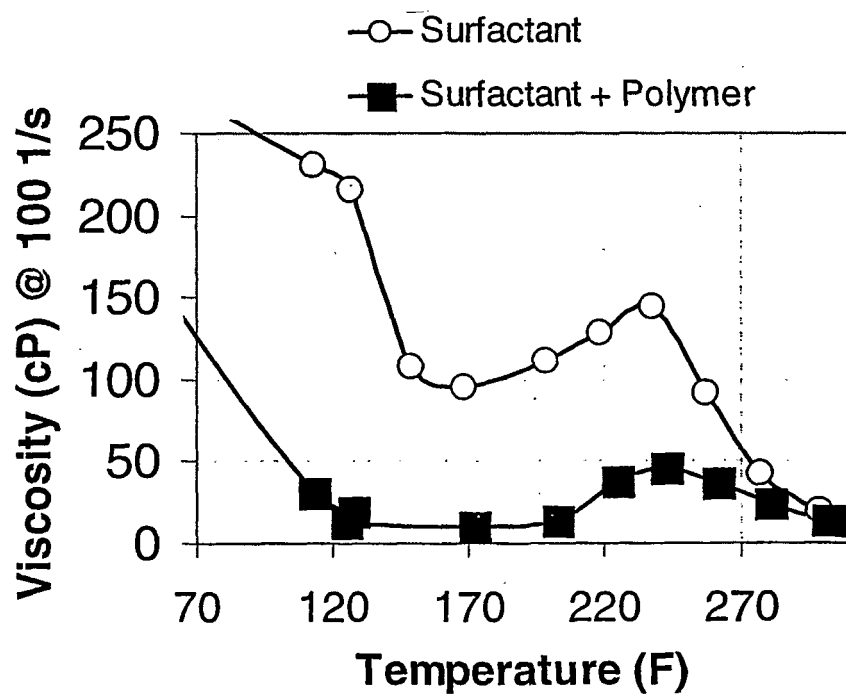


Figure 10

**REFERENCES CITED IN THE DESCRIPTION**

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