

# (11) EP 2 223 936 A1

(12)

## **EUROPEAN PATENT APPLICATION**

(43) Date of publication:

01.09.2010 Bulletin 2010/35

(51) Int Cl.: **C07K 14/37**<sup>(2006.01)</sup>

(21) Application number: 09153889.2

(22) Date of filing: 27.02.2009

(84) Designated Contracting States:

AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO PL PT RO SE SI SK TR

**Designated Extension States:** 

**AL BA RS** 

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## (54) Cellobiose Dehydrogenase

(57) The present invention relates to cellobiose dehydrogenases (CDH) having glucose oxidation activity at a pH of 7.4 or above, modifications to modify the pH dependency of the enzymes activity, uses for these

CDHs, in particular electrode sensors and electrochemical cells.

## Description

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[0001] The present invention relates to cellobiose dehydrogenase (CDH) enzymes, modifications thereof and electrochemical uses

[0002] Cellobiose dehydrogenase (EC 1.1.99.18, CDH) was first discovered 1974 in the extracellular enzyme system of *Phanerochaete chrysosporium* and later on in several other basidiomycete fungi. A special characteristic of this enzyme is its composition: the combination of a catalytically active flavin domain, hosting a non-covalently bound FAD, and a haem domain, with a haem b as a cofactor. Both domains are connected by a flexible linker. By its catalytic activity, the substrate, e.g. cellobiose, is oxidised in a reaction which reduces the FAD of the flavin domain. Subsequently, FAD can be reoxidised by the action of the haem domain. The spectral characteristics of a typical CDH clearly show the presence of both cofactors. Another characteristic described is the strong glucose discrimination of all well characterised enzymes. Until the discovery of the ascomycete fungus *Myriococcum thermophilum* (Stoica et al., 2005, Biosensors and Bioelectronics 20: 2010-2018; Harreither et al., 2007, Electroanalysis 19: 172-180), CDH was believed to strongly inhibit the conversion of glucose (Henriksson et al., 1998, Biochimica Biophysica Acta 1383: 48-54). Similarly, for a long time only CDHs exhibiting an acidic activity optimum were known, especially when the haem domain is involved in catalysis as it depends on intramolecular electron transfer (IET), which is necessary to transfer electrons via the haem to the electron acceptor. This is the case with cytochrome c in enzymatic assays, as well as on electrode surfaces where the haem domain enables direct electron transfer (DET) to the electrode.

[0003] Electrochemical applications described in the literature are the detection of cellobiose, cello-oligosaccharides, lactose and maltose soluble cellodextrins, ortho- and para-diphenolic compounds (Lindgren et al., 1999, Analyst 124: 527-532) and catecholamines (Stoica et al., 2004, Analytical Chemistry, 76: 4690-4696) mostly by mediated electron transfer (MET). So far the application in glucose biosensors based on the direct electron transfer (DET) properties of CDH was prevented by i) a very low or no glucose turnover, ii) the acidic pH optimum of most known CDHs and iii) a bad performance of some CDHs on electrodes.

[0004] Although, one CDH with well functioning IET at neutral or alkaline pH values is known (from the fungus *Humicola insolens*), it was shown not to convert glucose (Schou et al., 1998, Biochemical Journal 330: 565-571). One CDH currently known to convert glucose with significant turnover numbers was found in cultures of *Myriococcum thermophilum* (Harreither et al., 2007, Electroanalysis 19: 172-180). However, this enzyme has an acidic pH optimum for the IET and shows no activity under physiological pH conditions (pH 7.4). Another obstruction is the sometimes bad electronic communication of a CDH with an electrode surface, like the *Humicola insolens* and *Sclerotium rolfsii* CDHs (Lindgren et al., 2001, Journal of Electroanalytical Chemistry 496: 76-81), which results in very low current densities and therefore low signals even with high concentrations of the natural substrate cellobiose.

**[0005]** Thus CDH activity on glucose under neutral conditions, which is necessary for applications in e.g. physiological fluids is not satisfying, in particular not for the electrochemical measurement of glucose or the generation of electricity in biofuel cells. It is therefore a goal of the present invention to provide an alternative enzyme suitable to convert glucose at physiological pH ranges.

**[0006]** Therefore, in a first embodiment the present invention provides a CDH having glucose oxidation activity at a pH of 7.4 or above, selected from a CDH isolated from *Chaetomium atrobrunneum*, *Corynascus thermophilus*, *Hypoxylon haematostroma*, *Neurospora crassa* or *Stachybotris bisbyi* or being a modified CDH of *Myriococcum thermophilum*. According to the invention it has been surprisingly found that certain CDHs have a suitable activity under physiological pH conditions. Furthermore, the invention provides the modification of acidic CDHs to increase their activity at pH 7.4 and above

[0007] The term "cellobiose dehydrogenase" is defined herein as an enzyme consisting of a flavin domain and a haem domain connected by a peptide linker, which oxidises carbohydrates like its natural substrates cellobiose and cello-oligosaccharides and others like lactose, maltose and in particular glucose for preferred inventive uses. The reoxidation of the flavin domain cofactor can be achieved by direct oxidation by two-electron acceptors including quinones like 2,6-dichloroindophenol, o- or p-benzoquinone or derivatives thereof, methylene blue, methylene green, Meldola's blue or one-electron acceptors like potassium ferricyanide, ferricenium hexafluorophosphate,  $FeCl_3$  or by intramolecular electron transfer (IET) to the haem domain cofactor and further to a terminal electron acceptor like cytochrome c (cyt c) or an electrode surface.

**[0008]** The flavin domain of the CDH, which is responsible for the glucose oxidation activity and the haem domain, responsible for the regeneration of the flavin domain, may have two different pH optima, such as in the case of the natural CDH of *Myriococcum thermophilum*. In principle, the haem domain can be bypassed by providing the flavin domain with oxidants such as 2,6-dichloroindophenol which can directly reoxidise the flavin domain without the haem domain. According to the present invention, however, it should be understand that the CDH has a glucose oxidation activity at a pH of 7.4 by the action of both the flavin and the haem domain as can e.g. be measured by the cyt c assay or by measurement after immobilisation on an electrode surface. The haem domain acts as intermediate electron transmitter between cyt c and the flavin domain or between the electrode surface and the flavin domain, respectively.

[0009] The natural, wild-type CDH of *M. thermophilum* does not have the inventive glucose oxidation activity at a pH of 7.4 by action of both the flavin and the haem domain. The present invention has now for the first time provided a modification of the CDH of *M. thermophilum* which has the desired glucose oxidation activity. This modification according to the present invention should now be understood in that the inventive *M. thermophilum* CDH deviates from the wild-type *M. thermophilum* CDH by the substantially increased glucose oxidation activity at a pH of 7.4. This modification can be facilitated by increasing the interaction between the flavin and the haem domains, e.g. by modifying specific key amino acids responsible for that interaction as is further described herein. Preferably increasing the interaction includes increasing contacts or interaction energy between the domains. A prediction of such modifications can be easily made *in silico* using e.g. force field based interaction energy calculations. Furthermore interaction can be determined by measuring protein activity as described herein. Furthermore, it is possible to increase the pH dependency of the haem domain to a more basic pH. The pH optimum of the flavin domain of the natural CDH of *M. thermophilum* could in principle have the required pH properties to oxidise glucose, as is e.g. shown in **Fig. 2f** (by measurement of the 2,6-dichloroin-dophenol (DCIP)assay).

**[0010]** Also provided are enzyme preparations comprising novel CDHs. The term "enzyme" or "enzyme preparation" as used herein refers to a cellobiose dehydrogenase from a specified organism which is at least about 20% pure, preferably at least about 40% pure, even more preferably at least about 60% pure, even more preferably at least 80% pure and most preferably at least 90% pure as determined by polyacrylamide gel electrophoresis (PAGE).

[0011] The present invention relates to cellobiose dehydrogenases isolated from novel producers or genetically engineered from existing protein scaffolds, which are able to oxidise glucose more efficiently than the currently known cellobiose dehydrogenases. The kinetic constants of the enzymes responsible for this effect are a preferably lower  $K_M$  value and a higher  $k_{cat}$  value for glucose than the currently characterised enzymes (e.g., *Phanerochaete chrysosporium* CDH:  $K_M$  = 1600 mM,  $k_{cat}$  = 2.64 s<sup>-1</sup>, Henriksson et al., 1998, Biochimica and Biophysica Acta 1383: 48-54; Humicola insolens CDH: no glucose conversion detected, Schou et al., 1998, Biochemical Journal 330: 565-571; *Trametes villosa* CDH:  $K_M$  = 1300 mM,  $k_{cat}$  = 1.92 s<sup>-1</sup>, Ludwig et al., 2004, Applied Microbiology and Biotechnology 64: 213-222). In addition, the  $k_m$  and  $k_{cat}$  values for glucose oxidation of the inventive enzymes shall be at a pH of 7.4.

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[0012] In a further aspect the present invention provides a cellobiose dehydrogenase of SEQ ID NO: 5 (*Chaetomium atrobrunneum*), SEQ ID NO: 7 (*Corynascus thermophilum*), SEQ ID NO: 3 (*Hypoxylon haematostroma*), SEQ ID NO: 11 (*Neurospora crassa*) and SEQ ID NO: 9 (*Stachybotrys bisbyi*). Furthermore homologuous enzymes are provided having glucose oxidation activity at a pH of 7.4 or above comprising an amino acid sequence being at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 75%, at least 80%, at least 85%, at least 90%, at least 95%, at least 98%, in particular preferred at least 99%, identical to any one of SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11.

[0013] In preferred embodiments the CDH of *Chaetomium atrobrunneum* comprises an amino acid sequence of SEQ ID NO: 5. The CDH is readily available from *C. atrobrunneum* using the isolation methods described herein. In a further related aspect the present invention also provides a CDH comprising an amino acid sequence of SEQ ID NO: 5, or an amino acid sequence being at least 83%, preferably at least 85%, at least 88%, at least 90%, at least 92%, at least 93%, at least 94%, at least 95%, at least 96%, at least 97%, at least 98%, in particular preferred at least 99%, identical to SEQ ID NO: 5.

**[0014]** In preferred embodiments the CDH of *Corynascus thermophilum* comprises an amino acid sequence of SEQ ID NO: 7. The CDH is readily available from *C. thermophilum* using the isolation methods described herein. In a further related aspect the present invention also provides a CDH comprising an amino acid sequence of SEQ ID NO: 7, or an amino acid sequence being at least 76%, preferably at least 78%, at least 80%, at least 83%, at least 85%, at least 88%, at least 90%, at least 92%, at least 94%, at least 95%, at least 98%, in particular preferred at least 99%, identical to SEQ ID NO: 7.

[0015] In preferred embodiments the CDH of Hypoxylon haematostroma comprises an amino acid sequence of SEQ ID NO: 3. The CDH is readily available from H. haematostroma using the isolation methods described herein. In a further related aspect the present invention also provides a CDH comprising an amino acid sequence of SEQ ID NO: 3, or an amino acid sequence being at least 68%, preferably at least 70%, at least 72%, at least 74%, at least 76%, at least 78%, at least 80%, at least 80%, at least 83%, at least 85%, at least 88%, at least 90%, at least 95%, at least 98%, in particular preferred at least 99%, identical to SEQ ID NO: 3.

[0016] In preferred embodiments the CDH of *Neurospora crassa* comprises an amino acid sequence of SEQ ID NO: 11. The CDH is readily available from *N. crassa* using the isolation methods described herein. In a further related aspect the present invention also provides a CDH comprising an amino acid sequence of SEQ ID NO: 11, or an amino acid sequence being at least 72%, preferably at least 74%, at least 76%, at least 78%, at least 80%, at least 82%, at least 84%, at least 86%, at least 88%, at least 90%, at least 92%, at least 95%, at least 98%, in particular preferred at least 99%, identical to SEQ ID NO: 11.

[0017] In preferred embodiments the CDH of *Stachybotrys bisbyi* comprises an amino acid sequence of SEQ ID NO: 9. The CDH is readily available from *S. bisbyi* using the isolation methods described herein. In a further related aspect

the present invention also provides a CDH comprising an amino acid sequence of SEQ ID NO: 9, or an amino acid sequence being at least 59%, preferably at least 60%, at least 62%, at least 65%, at least 70%, at least 72%, preferably at least 74%, at least 76%, at least 78%, at least 80%, at least 82%, at least 84%, at least 86%, at least 88%, at least 90%, at least 95%, at least 98%, in particular preferred at least 99%, identical to SEQ ID NO: 9.

**[0018]** Preferably, a homologous or modified CDH of has 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, at least 11, at last 13, at least 15, at least 17, at least 20, at least 25, at least 30, at least 40, at least 50, at least 60, at least 80, at least 100 and/or up to 100, up to 80, up to 60, up to 50, up to 40, up to 30, up to 20, up to 15 amino acid substitutions, deletions, insertions or modifications, and any ranges between these values, as compared to any one of the CDHs of SEQ ID NOs 3, 5, 7, 9 or 11.

[0019] The present invention provides novel sequences of CDHs from *Chaetomium atrobrunneum* (SEQ ID NO: 5), *Corynascus thermophilum* (SEQ ID NO: 7), *Hypoxylon haematostroma* (SEQ ID NO: 3), *Neurospora crassa* (SEQ ID NO: 11) and *Stachybotrys bisbyi* (SEQ ID NO: 9). The CDHs of these sequences, as well as homologues with at least 50% sequence identity thereto are novel CDHs which also fulfill the inventive properties of having a glucose oxidation activity at a pH of 7.4. The modification of homologuous enzymes thereto with at least 50% sequence identity are preferably of amino acids which do not lower the pH requirement on the glucose oxidation acitivity. Any such modification can easily be tested by a glucose oxidation test on e.g. an electrode surface or by a cyt c assay, using cyt c to reoxidise the haem and subsequently the flavin domain of the CDH. Homologues can be readily identified by sequence comparisons such as by sequence alignment using publicly available tools, such as BLASTP.

**[0020]** The inventive CDH may be a modified CDH of *Myriococcum thermophilum*, comprising a flavin and a haem domain, wherein electron transfer from the flavin to the haem domain is increased as compared to wild type CDH of *M. thermophilum*, preferably as measured by the cyt c assay. Thus the invention relates to genetic engineering of a CDH to improve the enzymes activity further in the direction of high IET under neutral or alkaline pH conditions. The methods for the modification may be any known in the art such as amino acid mutations, including amino acid substitutions, deletions or additions but also chemical modification/derivatisation of amino acid side chains, in particular acidic amino acid side chains.

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[0021] As mentioned above, the invention includes homologuous sequences to the inventive CDHs of SEQ ID NOs 1 (*M. thermophilum* - with further modifications to improve the pH dependency as mentioned above), SEQ ID Nos. 3, 5, 7, 9 or 11 with at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 75%, at least 80%, at least 85%, preferably at least 90%, at least 98%, or at least 99%, sequence identity to the above sequences of SEQ ID NOs 1, 3, 5, 7, 9, or 11. Preferably the catalytic site has a minimum of modifications of e.g. 1, 2, 3, 4, 5, 6, 7, 8, 9 or 10 amino acid deletions, substitutions or additions or is even exempted from modifications as compared to the wild type catalytic sites. The catalytic site is from Phe 251 to Ala 287 (Rossman Fold, flavin binding), Val 334 to Leu 345 and Met 724 to Asp 732 of the *M. thermophilum* sequence of SEQ ID NO: 1. Corresponding amino acids also exist for the CDHs of *Chaetomium atrobrunneum* (SEQ ID NO: 5), *Corynascus thermophilum* (SEQ ID NO: 7), *Hypoxylon* haema*tostroma* (SEQ ID NO: 3), *Neurospora crassa* (SEQ ID NO: 11) and *Stachybotrys bisbyi* (SEQ ID NO: 9).

**[0022]** Preferably any one of the inventive CDHs shows IET (the transfer from electrons from the flavin to the haem domain) under neutral, alkaline or preferentially physiological (pH 7.4) pH conditions. To ensure a sufficiently high electrocatalytic activity of the enzyme under those conditions the IET as measured with the cyt c assay at pH 7.4 should be at least about 10% of the value maximum IET value measured under acidic pH conditions, or more preferably about 20%, or more preferably about 40%, or more preferably about 60%, or even more preferably about 80%, or most preferably should the pH optimum of IET be already neutral or alkaline.

**[0023]** The cellobiose dehydrogenases show preferably a sufficiently high direct electron transfer (DET) rate from the enzyme to the electrode to obtain a sufficiently high response at low substrate concentrations for a low detection limit and a high sensitivity. Only enzymes exhibiting high enough a DET current at the applied overpotential of +300 mV vs. Ag|AgCl (in 0.1 M KC1) to result in a detection limit of glucose (the lower limit of the linear range of the electrode was defined as the detection limit) with a spectrographic graphite electrode setup below 4 mM (the usual blood glucose concentration in a healthy human is 4-7 mM).

[0024] Preferably, the inventive CDH has an glucose oxidation activity at a pH of 7.4 or above and comprises an amino acid sequence of amino acids 22 to 828 of SEQ ID NO: 1 or an amino acid sequence being at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 75%, at least 80%, at least 85%, at least 90%, at least 94%, at least 95%, at least 96%, at least 98%, in particular preferred at least 99%, identical to amino acids 22 to 828 of SEQ ID NO: 1, **characterised in that** the amino acid sequence has at least one additional amino acid substitution, deletion or insertion to the sequence of SEQ ID NO:1 increasing electron transfer from the flavin to the haem domain as compared to wild type CDH of *Myriococcum thermophilum* of SEQ ID NO:1, preferably as measured by the cyt *c* assay. Given in SEQ ID NO: 1 is the wild type *M. thermophilum* CDH which does not have the required glucose oxidation activity at a pH of 7.4. The sequence of SEQ ID NO: 1 has further a signal peptide up to amino acid 21 which may not be present on the final processed enzyme. It could now be shown that according to the present invention by a single (or more)

amino acid substitution, deletion or insertion of the CDH with the final *M. thermophilum* CDH sequence the pH optimum of the glucose oxidation activity can be shifted to a more basic pH, in particular to a physiologically relevant pH of 7.4. Those skilled in the art can readily chose from possible amino acid modifications given the extensive sequence and functional information depicted herein and furthermore, can without undue burden test the modified CDH by a simple cyt c assay as described herein. Preferably, the inventive modified CDH of *M. thermophilum* has 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, at least 11, at last 13, at least 15, at least 17, at least 20, at least 25, at least 30, at least 40, at least 50, at least 60, at least 80, at least 100 and/or up to 100, up to 80, up to 60, up to 50, up to 40, up to 30, up to 30, up to 20, up to 15 amino acid substitutions, deletions, insertions or modifications, and any ranges between these values.

**[0025]** In a further embodiment the electron transfer of the modified CDH is increased by increasing electrostatic interaction between the flavin and the haem domain, preferably at a pH of 7.4. The electrostatic interaction can be increased by optimising charge interactions at pH 7.4 of basic and acidic amino acids of the flavin and haem domain by site-directed mutagenesis, electrostatic repulsion can be reduced or electrostatic attraction increased. From the available sequence and structural information those skilled in the art can readily chose from a vast amount of such possible mutations which increase the interaction at pH 7.4, and preferably results in an increased activity in a cytic assay.

[0026] As has been pointed out, the intramolecular electron transfer (IET) rate between the flavin domain and the haem domain depends heavily on the pH. E.g. in the basidiomycete *Trametes villosa* CDH IET is fast at pH 3.5, slows down significantly at pH 5.0, and is virtually absent above pH 6.0. Contrary, the IET of the ascomycete *H. insolens CDH* is not affected by alkaline conditions, having a pH optimum of around 8.0. From the kinetic data of *Humicola insolens CDH* an alkaline pH optimum for cyt c reduction is obvious and the DET measured for that enzyme was highest at pH >7 and is thus an exception so far for CDHs. Interestingly, although an ascomycete CDH, *Myriococcum thermophilum CDH* has an IET behaviour similar to basidiomycete enzymes. Preferably, the above amino acids are modified in the *M. thermophilum* CDH to increase the activity at a pH of 7.4 as measured in a cyt c assay. Furthermore it has been found that these amino acids are of particular interest for the activity of the enzyme. Amino acids corresponding to these amino acids in CDHs of *Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Neurospora crassa* or *Stachybotris bisb*yi with preferably modifications, according to the homologues to the sequences of the SEQ IDs NO: 3, 5, 7, 9 and 11 in other amino acids than in those corresponding to the above amino acids of *Myriococcum thermophilum* of SEQ ID NO: 1.

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**[0027]** The changes in the above mentioned amino acids of SEQ ID NO: 1 in order to increase the activity at pH 7.4 are preferably to increase electrostactic interaction between the flavin and the haem domain as mentioned above.

**[0028]** In particular preferred embodiments the modification of the CDH is a modification of the haem domain of any one of amino acids 90-100, 115-124, 172-203, preferably of any one of amino acids 176, 179-182, 195, 196, 198, 201 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1 and/or of the flavin domain of any one of amino acids 311-333, 565-577, 623-625, 653-664, 696-723, preferably of any one of amino acids 318, 325, 326, 328, 568, 571, 574, 575, 624, 654, 663, 702, 709, 712, 717, correspond to the *M. thermophilum* CDH of SEQ ID NO: 1, or any combination thereof.

[0029] Possible modifications include (i) the exchange of acidic amino acids by neutral (polar or apolar) residues (e.g. Ser, Thr, Ala) to decrease the number of negative charges and weaken the electrostatic force field at either the haem or the flavin domain at neutral/alkaline pH values. (ii) The exchange of acidic amino acids by alkaline residues (Lys, Arg) to increase the number of positive charges and weaken the electrostatic force field at either the haem or the flavin domain at neutral/alkaline pH values, and (iii) the introduction of alkaline residues (Lys, Arg) instead of neutral residues (Hydrophobic or hydrophilic) to increase the number of positive charges and weaken the negative electrostatic force field at neutral/alkaline pH.

**[0030]** Particularly the modification may be an increase of positive charge in the of amino acids 172-203 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1, preferably of amino acid 181, in particular preferred a D181K mutation, and/or a decrease of a negative charge of amino acids 565-577 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1, preferably of amino acid(s) 568 and/or 571, in particular preferred a D568S and/or E571S mutation, or any combination thereof.

**[0031]** In further embodiments the activity of the inventive CDH is a glucose dehydrogenase activity and may be an electrocatalytic oxidation of glucose.

**[0032]** The inventive CDH may be isolated, in particular from Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Neurospora crassa or Stachybotris bisbyi or any genetically modified cell to recombinantly express the inventive CDH. Isolation may be performed by diafiltration and subsequent ion exchange chromatography by collecting fractions with CDH activity. The CDH can be further purified, e.g. using hydrophobic interaction chromatography.

[0033] The inventive CDH may also comprise a linker or be a part of a fusion protein. An inventive CDH polypeptide comprising the inventive sequences may be up to 500 kDa, up to 400 kDa, up to 300 kDa, up to 200 kDa or even up to 150 kDa.

[0034] In another aspect the present invention provides a nucleic acid molecule encoding a CDH of the invention. A

preferred embodiment of the invention is a nucleic acid molecule encoding a cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above and comprising

- a nucleotide sequence of SEQ ID NOs 4, 6, 8, 10 or 12, or
- the open reading frame of SEQ ID NOs 4, 6, 8, 10 or 12 or

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- a nucleotide sequence with at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 75%, at least 80%, at least 90%, at least 95%, at least 98%, in particular preferred at least 99%, identity to SEQ ID NO: 2, 4, 6, 8, 10 or 12 or the open reading frame of SEQ ID NOs: 2, 4, 6, 8, 10 or 12, further comprising a nucleotide mutation, substitution, deletion or insertion,
- a nucleotide sequence that hybridizes with any one of SEQ

**[0035]** ID NO: 2, 4, 6, 8, 10 or 12 under stringent condition. "Stringent conditions" relate to hybridisation reactions under defined hybridisation conditions which is a function of factors as concentration of salt or formamide in the hybridisation buffer, the temperature of the hybridisation and the post-hybridisation wash conditions. Such conditions are for example hybridisation at 68°C in a standard SSC hybridisation buffer containing 0.1%SDS followed by stringent washing in wash buffer at the same temperature. Stringent washing can be performed for example by two times washing with 2xSSC buffer followed by two wash steps with 0.5xSSC buffer. Stringent hybridisation conditions will preferably involve a temperature of 15°C to 25°C below the melting temperature (Tm), whereby the Tm of a hybridisation product of a nucleic acid probe can be calculated using a formula based on the g + c contained in the nucleic acids and that takes chain lengths into account, such as the formula Tm = 81.5 to 16.6 (log [na+]) + 0.41 (% G + C) - 600/N), wherein N = chain length (Sambrook et al. (1989), which is incorporated herein by reference). In practice an estimated Tm for an oligonucleotide probe is often confirmed and thus a person skilled in the art can calculate the Tm for any chosen probe whose nucleotide sequence is known.

**[0036]** A nucleic acid sequence of *M. thermophilum* may be defined by the SEQ ID NO: 2 or the open reading frame of SEQ ID NO: 2, including homologs with at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 85%, at least 95%, at least 95%, at least 98%, in particular preferred at least 99%, identity to SEQ ID NO: 2 or the open reading frame of SEQ ID NO: 2, further comprising a codon mutation, substitution, deletion or insertion to encode a CDH with glucose oxidation activity at a pH of 7.4 or above, preferably of amino acids 22 to 828 of SEQ ID NO: 1 with an additional amino acid substitution, deletion or insertion. Preferably the encoded CDH is of the amino acid sequence with the above mentioned amino acid modification.

**[0037]** The inventive nucleic acid molecules encoding a CDH with glucose oxidating activity at pH 7.4 may be isolated or purified. The inventive nucleic acid molecules, in particular their open reading frame may be comprised in a vector, preferably an expression or cloning vector. An inventive nucleotide molecule might further contain regulatory elements such as promotors and enhancers. Such a nucleic acid molecule comprising the inventive sequences may consist of up to 1,000,000 nucleotides, up to 900,000 nucleotides, up to 800,000 nucleotides, up to 700,000 nucleotides, up to 600,000 nucleotides, up to 500,000 nucleotides, up to 300,000 nucleotides, up to 200,000 nucleotides, up to 100,000 nucleotides, up to 50,000 nucleotides or up to 25,000 nucleotides.

**[0038]** Particular benefits of the inventive CDHs are i) a high glucose turnover rate, ii) sufficient activity of the flavin domain and IET at pH 7.4 and iii) good DET characteristics. It was further found that these CDHs and other known CDHs are in particular suitable as anodic material in a bioelectrode as e.g. biosensor or biofuel cell.

**[0039]** One approach was to identify CDHs of different fungal strains from the phylum of ascomycota. It was found that some CDHs, e.g. from *Chaetomium atrobrunneum*, *Corynascus thermophilus*, *Hypoxylon haematostroma*, *Neurospora crassa* or *Stachybotris bisbyi* are able to convert glucose with high turnover rates at pH 7.4 and have additionally very good DET properties. Some of the found CDHs are new *per* se as mentioned above. A further aspect of the present invention is the use of all CDHs described herein on an electrode, in an electrochemical cell, in particular in a biosensor to measure glucose, preferably at a physiological pH such as pH 7.4.

**[0040]** Furthermore it was found that CDH from *M. thermophilum*, which is known to oxidise glucose and have good DET properties but shows no IET at pH values above pH 7.0, can be genetically engineered to increase the pH range of the IET to more alkaline conditions by means of site-directed mutagenesis. These CDHs were particularly suitable for electrochemical devices.

[0041] Thus, in a further aspect the present invention provides an electrode comprising an immobilised CDH having glucose oxidation activity at a pH of 7.4 or above and having at least 10%, at least 12%, at least 14%, at least 16%, preferably at least 18%, at least 19%, in particular preferred at least 20%, at least 21%, or even at least 22%, at least 23%, at least 24%, at least 25%, at least 26%, at least 27%, at least 28%, at least 29% or at least 30% glucose, lactose or cellobiose oxidising activity at a pH of 7.4 as compared to their maximal activity at a lower pH as determined by the cyt c assay. Preferably the immobilised CDH also has a glucose oxidation activity at a pH of 7.4 or above and having at least 10%, at least 12%, preferably at least 18%, at least 19%, in particular preferred at least 20%, at least 21%, or

even at least 22%, at least 23%, at least 24%, at least 25%, at least 26%, at least 27%, at least 28%, at least 29% or at least 30%, glucose, lactose or cellobiose oxidising activity at a pH of 7.4 as compared to their maximal activity at a lower pH as determined by a 2,6-dichloroindophenol (DCIP) assay. An electrode is generally a conducting surface, e.g. suitable for an electro-chemical element.

**[0042]** The activity measurement by either the cyt c or DCIP assay can be readily facilitated in a model system. The CDH can be directly used with the substrate (glucose but also lactose or cellobiose) and a reoxidising agent being either cyt c for reoxidation at the haem domain or DCIP for reoxidation at the flavin domain. The CDH is tested at a pH of 7.4 in any suitable buffer, e.g. a potassium or natrium phosphate buffer. To determine the maximum of the CDH activity, the activity is continuously measured at different pH values, e.g. ranging from pH 3 to pH 7.4 or higher and determining the maximum activity. The pH of 7.4 is then compared with this maximum activity and should have the required activity fraction mentioned above. The activity can e.g. be given as absolute values in U/mg or as relative values. Preferably, the inventive CDH has the required activity portion as compared to the maximum activity in both a cyt c and a DCIP assay. These activity values preferably also apply to the new CDHs described above, as such, independent of their fixation on an electrode. Preferably, the assay to determine the inventive CDH on the electrode is performed by glucose oxidation. Alternatively, also using lactose is possible.

[0043] The electrode may be of any material suitable to immobilise the CDH, e.g. carbon such as graphite, glassy carbon, boron doped diamond, gold electrodes modified with promoters e.g., thiols, screen-printed electrodes, screen printed electrodes containing carbon nanotubes (single or multi-walled). It may contain other nanoparticles to increase the specific surface area. Particular uses of the inventive electrodes are in the provision of biosensors and enzymatic biofuel cells, more specifically to glucose biosensors and glucose oxidizing biofuel cell anodes using the direct electron transfer properties (DET) of cellobiose dehydrogenase (CDH) to measure the glucose concentration at neutral, alkaline or, preferentially, physiological pH (in human body fluids, e.g., 7.4 in blood) or use glucose for the generation of an electric current in biofuel cells under the same pH conditions.

**[0044]** In particular preferred embodiments the specific activity for glucose oxidation by using the cyt c assay at pH 7.4 is higher than 0.5 U/mg, preferably at least 0.6 U/mg, at least 0.7 U/mg, at least 0.8 U/mg, at least 0.9 U/mg, at least 1 U/mg, or at least 1.2 U/mg CDH, or a current density higher than 80 nA/cm<sup>2</sup> at pH 7.4.

**[0045]** In further preferred embodiments the apparent  $K_M$  value of the CDH for glucose in solution (DCIP assays at optimum activity) is lower than 1.7 M, preferably lower than 1.5, lower than 1.2, preferably lower than 1 M or, when measured on electrodes an apparent  $K_M$  value below 200 mM, preferably below 150 mM.

**[0046]** In another embodiment the present invention provides an electrode, wherein the CDH is of *Chaetomium atro-brunneum*, *Corynascus thermophilus*, *Hypoxylon haematostroma*, *Neurospora crassa* or *Stachybotris bisbyi* or a modified CDH of *Myriococcum thermophilum* with an increased activity at pH of 7.4 as defined in above, or homologues with certain sequence identities, amino acid modifications, etc. as defined above.

[0047] On the electrode, the CDH may be immobilised by adsorption, preferably also physical entrapment, complex formation, preferably via an additional complexing linker, covalent binding, in particular cross linking, or ionic linkage and/or the immobilized cellobiose dehydrogenase can be cross-linked, in particular by bifunctional agents, to increase stability or activity. It has been shown that crosslinking with bifunctional agents, such as agents with two reactive groups making a connection with the CDH, can stabilize the CDH and even increase its activity (measurable e.g. by the cyt c assay described herein). This advantage can lead to a increased sensitivity and lowering the detection limit for glucose. Such a cross-linking agent is e.g. glutaraldehyde or any other dialdehydes.

**[0048]** The electrodes might be used in form of a single electrode or electrode stacks. More specifically, the application of these enzymes is in (bio)electrochemical devices such as glucose biosensors or biofuel cells anodes. The electrode may be used as biosensor or as biofuel cell anode.

**[0049]** In another aspect the present invention provides an electro-chemical cell comprising an electrode as described above as an anodic element and a cathodic element.

**[0050]** In preferred embodiments of the electrochemical cell the anodic fluid can be glucose containing solution. Preferably the electrode is suitable for measurement in blood, serum and other body fluids.

**[0051]** The electrochemical cell may further comprise a solution of at least pH 6.0, preferably at least pH 6.5 or at least pH 6.7, in particular preferred at least pH 7.0, even more preferred at least pH 7.1, or at least pH 7.2, or at least pH 7.3, especially preferred at least 7.4, as anodic fluid.

[0052] According to another aspect a method of detecting or quantifying glucose in a sample is provided comprising

- providing a CDH having glucose oxidation activity at a pH of 7.4 or above,
- contacting a fluid sample having a pH of at least 6.0, preferably at least 6.5, or at least 6.7, more preferred at least 7.0, at least 7.1, at least 7.2, in particular preferred at least 7.3, especially preferred at least 7.4, with the CDH, and
- detecting an oxidation of glucose of the sample by the CDH.

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[0053] Preferably the oxidation is detected electrochemically, preferably with an immobilised CDH on an electrode,

in particular preferred as defined above.

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[0054] One of the world-wide leading causes of death and disability is diabetes. The diagnosis and management of diabetes mellitus requires continuous monitoring of blood glucose levels. Amperometric enzyme electrodes, based on glucose oxidase, play an increasingly important role and have been a target of substantial research. Most sensors are used for individual, daily diabetes monitoring, but the demand for continuous in vivo monitoring of patients is also significant. Real-time measurements are highly desired in intensive care units, during surgery, or for the management of diabetes, where rapid biochemical changes can be missed by discrete measurements. Such monitoring requires miniaturized, biocompatible, and stable sensors. Although research has reached the level of short-term implantation, an implantable glucose sensor possessing long-term stability has not yet been realised. Besides the obvious biocompatability challenge, some sensors are prone to errors due to low oxygen tension or electroactive interferences. Third generation biosensors depend on enzymes that are able to permit direct electron transfer (DET) between the electrode material and the redox active centre. Usually this is hindered by the encapsulation of the redox center by the protein structure. However, as has been shown herein, the inventive CDH can exhibit electrical communication with electrode supports.

**[0055]** In certain embodiments the CDH has at least 10%, or at least 12%, preferably at least 14%, or at least 16%, in particular preferred at least 18%, or at least 20%, at least 21%, or even at least 22%, at least 23%, at least 24%, at least 25%, at least 26%, at least 27%, at least 28%, at least 29%, or at least 30%, glucose, lactose or cellobiose oxidising activity at a pH of 7.4 as compared to the maximal activity at a pH below 7.4 as determined by a cyt c assay and/or DCIP assay.

[0056] The fluid sample may be any fluid which potentially comprises glucose, including blood, serum and other body fluids.

**[0057]** In particularly preferred embodiments the CDH is of *Chaetomium atrobrunneum*, *Corynascus thermophilus*, *Hypoxylon* haema*tostroma*, *Neurospora crassa* or *Stachybotris bisbyi* or a modified CDH of *Myriococcum thermophilum* with an increased activity at pH of 7.4 as defined above.

[0058] The cellobiose dehydrogenase of the present invention may be obtained from microorganisms of any genus. For purposes of the present invention the term "obtained from" as used herein in connection with a given source shall mean that the enzyme is produced by the source or by a cell in which the nucleic acid sequence of the cellobiose dehydrogenase gene from the source has been inserted. The enzyme or its nucleic acid sequence may be obtained from any fungal source and in a preferred embodiment from the genus *Chaetomium, Corynascus, Hypoxylon, Myriococcum, Neurospora* or *Stachybotrys*. In a more preferred embodiment the enzymes or the nucleic acid sequences are obtained from the species *Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Myriococcum thermophilum, Neurospora crassa* or *Stachybotrys bisbyi*.

[0059] In the most preferred embodiment the enzymes or the nucleic acid sequences are obtained from the strains Chaetomium atrobrunneum CBS 238.71, Corynascus thermophilus CBS 405.69, Hypoxylon haematostroma CBS 255.63, Myriococcum thermophilum CBS 208.89, Neurospora crassa DSMZ 2968 or Stachybotrys bisbyi DSMZ 63042. [0060] It will be understood that for the aforementioned species, the invention encompasses both the perfect and imperfect states and other taxonomic equivalents, e.g., anamorphs, regardless the species name by which they are

known. Those skilled in the art will readily recognise the identity of appropriate equivalents.

[0061] It is understood that one of skills in the art may engineer the mentioned or other cellobiose dehydrogenases to obtain the outlined specifications of the enzymes and enzyme variants described herein to obtain modified enzymes using the principles outlined herein like the rational approach via site-directed mutagenesis or directed evolution approaches (e.g., gene shuffling, error-prone PCR) and subsequent screening of the generated diversity. The techniques to introduce a mutation into the nucleic acid sequence to exchange one nucleotide for another nucleotide with the aim to exchange one amino acid for another in the resulting protein may be accomplished by site-directed mutagenesis using any of the methods known in the art.

[0062] Preferably the present invention is defined as follows:

Definition 1. A CDH having glucose oxidation activity at a pH of 7.4 or above, selected from an CDH isolated from *Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Neurospora crassa* or *Stachybotris bisbyi* or being a modified CDH of *Myriococcum thermophilum*, e.g. selected from the group of a CDH of *Chaetomium atrobrunneum, Hypoxylon haematostroma, Stachybotris bisbyi* or a modified CDH of *Myriococcum thermophilum*.

Definition 2. Cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprising an amino acid sequence of SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11.

Definition 3. Cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprising an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11.

Definition 4. CDH according to definition 1 being a modified CDH of Myriococcum thermophilum, comprising a flavin

and a haem domain, wherein electron transfer from the flavin to the haem domain is increased as compared to wild type CDH of *Myriococcum thermophilum*, preferably as measured by a cyt c assay.

Definition 5. CDH having glucose oxidation activity at a pH of 7.4 or above comprising an amino acid sequence of amino acids 22 to 828 of SEQ ID NO: 1 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to amino acids 22 to 828 of SEQ ID NO: 1, characterised in that the amino acid sequence has at least one additional amino acid substitution, deletion or insertion to the sequence of SEQ ID NO:1 increasing electron transfer from the flavin to the haem domain as compared to wild-type CDH of *M. thermophilum* of SEQ ID NO:1, preferably as measured by a cyt c oxidation assay.

Definition 6. CDH according to definition 4 or 5, **characterised in that** the electron transfer is increased by increasing, preferably optimising, electrostatic interactions between the flavin and the haem domain.

Definition 7. CDH according to any one of definition 4 to 6, **characterised in that** the modification is a modification of the haem domain of any one of amino acids 90-100, 115-124, 172-203, preferably of any one of amino acids 176, 179-182, 195, 196, 198, 201 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1 and/or of the flavin domain of any one of amino acids 311-333, 565-577, 623-625, 653-664, 696-723, preferably of any one of amino acids 318, 325, 326, 328, 568, 571, 574, 575, 624, 654, 663, 702, 709, 712, 717, correspond to the *M. thermophilum* CDH of SEQ ID NO: 1, or any combination thereof.

Definition 8. CDH according to definition 7, **characterised in that** the modification is an increase of positive charge in the of amino acids 172-203 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1, preferably of amino acid 181, in particular preferred a D181K mutation, and/or a decrease of a negative charge of amino acids 565-577 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1, preferably of amino acid(s) 568 and/or 571, in particular preferred a D568S and/or E571S mutation, or any combination thereof.

Definition 9. CDH according to any one of definition 1 to 8, **characterised in that** the glucose oxidation activity is a glucose dehydrogenase activity.

Definition 10. CDH according to any one of definition 1 to 9, **characterised in that** the activity is an electrocatalytic oxidation of glucose.

Definition 11. CDH according to any one of definition 1 to 10, being isolated by diafiltration, ion exchange chromatography by collecting fractions with CDH activity, preferably being further purified by hydrophobic interaction chromatography.

Definition 12. A nucleic acid molecule encoding a cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above and comprising

a nucleotide sequence of SEQ ID NOs 4, 6, 8, 10 or 12, or

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- the open reading frame of SEQ ID NOs 4, 6, 8, 10 or 12 or
- a nucleotide sequence with at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 75%, at least 80%, at least 85%, at least 90%, at least 95%, at least 98%, in particular preferred at least 99%, identity to SEQ ID NO: 2, 4, 6, 8, 10 or 12 or the open reading frame of SEQ ID NOs: 2, 4, 6, 8, 10 or 12, further comprising a nucleotide mutation, substitution, deletion or insertion, preferably a codon mutation, substitution, deletion or insertion,
- a nucleotide sequence that hybridizes with any one of SEQ ID NO: 2, 4, 6, 8, 10 or 12 under stringent condition, preferably encoding a cellobiose dehydrogenase as defined in any one of definition 1 to 11.

Definition 13. An electrode comprising an immobilised CDH having glucose oxidation activity at a pH of 7.4 or above and having at least 10%, preferably at least 14%, in particular preferred at least 18%, or at least 20%, at least 21%, or even at least 22%, at least 23%, at least 24%, at least 25%, at least 26%, at least 27%, at least 28%, at least 29%, or at least 30%, glucose, lactose or cellobiose oxidising activity at a pH of 7.4 as compared to their maximal activity at a lower pH as determined by the cyt c assay, preferably also as compared to their maximal activity at a lower pH as determined by the DCIP assay, and/or an electrode comprising an immobilised CDH of any one of definitions 1 to 11.

Definition 14. Electrode according to definition 13, characterised in that the cyt c assay is performed by cellobiose oxidation.

Definition 15. Electrode according to definition 13 or 14, characterised in that the glucose oxidation activity at pH 7.4 is at least 0.5 U/mg CDH.

Definition 16. Electrode according to any one of definition 13 to 15, **characterised in that** the Km value of the CDH for an glucose oxidation reaction at pH 7.4 is below 1M.

Definition 17. Electrode according to any one of definition 13 to 16, **characterised in that** the CDH is of *Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Neurospora crassa, Stachybotrys bisbyi,* or a modified CDH of *Myriococcum thermophilum* with an increased activity at pH of 7.4 as defined in any one of definition 4 to 8, or

the cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprises an amino acid sequence of SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11,

preferably the CDH is further defined as in any one of definition 1 to 11.

Definition 18. Electrode according to any one of definition 13 to 16, **characterised in that** the CDH is immobilised by adsorption, complex formation, preferably via an additional complexing linker, covalent or ionic linkage, and/or wherein the immobilized cellobiose dehydrogenase is cross-linked, in particular by bifunctional agents, to increase stability or activity. Definition 19. An electrochemical cell comprising an electrode according to any one of definition 13 to 18 as an anodic element and a cathodic element.

Definition 20. Electrochemical cell according to definition 19, comprising a glucose containing solution as anodic fluid. Definition 21. Electrochemical cell according to definition 18 or 20, comprising a solution of at least pH 6.0, preferably at least pH 6.5, in particular preferred at least pH 7.0, even more preferred at least pH 7.2, as anodic fluid. Definition 22. Method of detecting or quantifying glucose in a sample using any of the previously described CDHs or electrodes or the method comprising

- providing a CDH having glucose oxidation activity at a pH of 7.4 or above,
- contacting a fluid sample having a pH of at least 6.5, preferably at least 7.0, in particular preferred at least 7.3, especially preferred at least 7.4, with the CDH, and
- detecting an oxidation of glucose of the sample by the CDH.

Definition 23. Method according to definition 22, wherein the oxidation is detected electrochemically, preferably with an immobilised CDH on an electrode, in particular preferred as defined in any one of definition 13 to 21.

Definition 24. The method of definition 22 or 23, **characterised in that** the CDH has at least 10%, preferably at least 14%, in particular preferred at least 18%, or at least 20%, at least 21%, or even at least 22%, at least 23%, at least 24%, at least 25%, at least 26%, at least 27%, at least 28%, at least 29%, or at least 30%, glucose, lactose or cellobiose oxidising activity at a pH of 7.4 as compared to the maximal activity at a pH below 7.4 as determined by a cyt c assay.

Definition 25. The method of definition 22 to 24, **characterised in that** the CDH is of *Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Neurospora crassa* or *Stachybotris bisbyi* or a modified CDH of *Myriococcum thermophilum* with an increased activity at pH of 7.4 as defined in any one of definition 4 to 8, preferably the CDH is further defined as in any one of definition 1 to 11.

Definition 26. The method of any one of claims 22 to 25, characterized in that the cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprises an amino acid sequence of SEQ ID NO: 3, SEQ ID NO: 5, SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to SEQ ID NO: 3, SEQ ID NO: 5, SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11.

[0063] The present invention is further illustrated by the following figures and examples without being restricted thereto.

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#### [0064]

Figure 1 gives the codon-optimised nucleotide sequence (SEQ ID NO: 2) and the corresponding amino acid sequence (SEQ ID NO: 1) of *Myriococcum thermophilum* CDH used for site-directed mutagenesis. Non-limiting preferred mutations sites are indicated by "\*" and particular non-limiting preferred sites are marked by "+".

Figure 2 shows the pH profiles of screened ascomycete CDHs Chaetomium atrobrunneum CDH, 2.a; Corynascus thermophilus CDH, 2.b; Hypoxylon haematostroma CDH, 2.c; Neurospora crassa CDH, 2.d; Stachybotrys bisbyi CDH, 2.e; Myriococcum thermophilum CDH, 2.f; and the genetically engineered enzyme variants Myriococcum thermophilum CDH variant D160K, 2.g; Myriococcum thermophilum CDH variant D547S/E550S, 2.h using lactose and a soluble electron acceptor, 2,6-dichloroindophenol, (DCIP, dotted, grey lines) or cyt c (solid, black lines) as substrates and 50 mM citrate-phosphate buffer (pH 3.0-8.0).

Figure 3 is a sequence alignment of amino acid sequences of CDHs from *Chaetomium atrobrunneum* (SEQ ID NO: 5), *Corynascus thermophilum* (SEQ ID NO: 7), *Hypoxylon haematostroma* (SEQ ID NO: 3), *Myriococcum thermophilum* (SEQ ID NO: 1), *Neurospora crassa* (SEQ ID NO: 11) and *Stachybotrys bisbyi* (SEQ ID NO: 9).

Figure 4 shows a setup of the wall jet electrode and auxiliary instruments. The sensor assembly (A) was continuously flushed with buffer and samples were applied through an ultrafast injection valve. The obtained current at a potential of 300 mV was recorded. The flow-jet system (A) consisted of a carbon working electrode (WE), a platin counter electrode (CE) and a silver reference electrode (RE) connected to a potentiostat.

Figure 5: Measurement setup of the flow-cell system

Examples:

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### 10 Example 1: Materials

[0065] Chemicals used in buffers and fermentation media were commercial products and at least of analytical grade if not otherwise stated. Peptone from meat and microcrystalline cellulose were from VWR International (Vienna, Austria), alpha-cellulose from Sigma-Aldrich (Vienna, Austria). Substrates for kinetic studies were lactose, glucose, 2,6-dichloroindophenol (DCIP) and cytochrome c from horse heart(cyt c) from Sigma-Aldrich in the highest grade of purity available. Buffers were prepared using water purified and deionised (18  $M\Omega$ ) with a Milli-Q system (Millipore, Bedford, MA, USA), fermentation media contained reversed osmosis water (0.1  $M\Omega$ ).

#### **Example 2: Enzyme Assays**

[0066] Enzymatic activity of cellobiose dehydrogenase was detected by two assays. The DCIP assay, measuring the activity of the flavin domain was performed by measuring the time-dependent reduction of 300  $\mu$ M DCIP in 50 mM citrate-phosphate buffer at the indicated pH (3.0-8.0), containing 30 mM lactose at 520 nm and 30 °C. The absorption coefficient for DCIP is pH dependent but differs at 520 nm only about 3% within pH 3.0 to 8.0 and was determined to be 6.8 mM<sup>-1</sup> cm<sup>-1</sup> (Karapetyan et al., 2005 Journal of Biotechnology 121: 34-48).

[0067] Alternatively, enzymatic activity was determined by the reduction of cytochrome c at 30°C and 550 nm (cyt c,  $\epsilon_{550}$  = 19.6 mM<sup>-1</sup> cm<sup>-1</sup>, Canevascini et al., 1991, European Journal of Biochemistry 198: 43-52) in an assay containing 20  $\mu$ M cyt c and 30 mM lactose, which specifically detects the activity of the whole enzyme (flavin and haem domain). The cyt c assay gives thereby also a measure of the efficiency of the intramolecular electron transfer (IET) between both domains as an indication of the enzyme's response on electrodes in a pH range of 3.0 to 8.0 (50 mM sodium citrate-phosphate buffer). For the detection of activity with glucose the above mentioned assays were used, but lactose was exchanged for 100 mM glucose.

[0068] One unit of enzymatic activity was defined as the amount of enzyme that oxidises 1  $\mu$ mol of lactose per min under the assay conditions. Lactose was chosen instead of the natural substrate cellobiose, as it shows no substrate inhibition with CDH. The reaction stoichiometry with carbohydrates is 1 for the two-electron acceptor DCIP, but 2 for the one-electron acceptor cyt c.

#### **Example 3: Enzyme Kinetics**

40 [0069] Carbohydrate stock solutions used for measuring activity and kinetic constants with the DCIP and cyt c assays were prepared in the appropriate buffer several hours before the experiment to allow mutarotation to reach equilibrium. pH profiles were determined using 50 mM citrate-phosphate buffer (3.0-8.0). To ensure an assay temperature of 30 °C the cuvettes were incubated in a thermostated chamber for at least 20 min. After the measurement, the pH was again checked in the cuvettes. Kinetic constants were calculated by fitting the observed data to the Henri-Michaelis-Menten equation or to the adapted model for substrate inhibition using nonlinear least-squares regression and the program SigmaPlot (Systat Software, San Jose, CF, USA).

### **Example 4: Protein characterisation**

[0070] The protein concentration was determined by the dye-staining method of Bradford using a pre-fabricated assay from Bio-Rad Laboratories Hercules, CA, USA) and bovine serum albumin as standard according to the manufacturers recommendations.

**[0071]** For spectral characterisation apparently homogeneous CDH (in the oxidised state) was diluted to an absorption of ~1 at 280 nm and the spectrum from 260 to 700 nm taken with an Hitachi U3000 spectrophotometer (Tokyo, Japan). After reduction with lactose (final concentration 1 mM) the reduced spectrum was taken.

**[0072]** For electrophoretic characterisation SDS-PAGE was carried out on a Hoefer SE 260 Mighty Small II vertical electrophoresis unit. Gels (10.5x10 cm; 10% T, 2.7% C) were cast and run according to the manufacturers' modifications of the Laemmli system. Isoelectric focusing in the range of pH 2.5 to 6.5 was performed on a Multiphor II system using

precast, dry gels rehydrated with Ampholytes (GE Healthcare Biosiences, Vienna, Austria). Protein bands on the SDS-PAGE were stained with silver, bands on the IEF gel with Coomassie blue R-250, according to the instructions.

### **Example 5: Screening for Suitable Cellobiose Dehydrogenases**

[0073] Fungal strains (*Chaetomium atrobrunneum* CBS 238.71, Corynascus thermophilus CBS 405.69, *Hypoxylon haematostroma* CBS 255.63, *Myriococcum thermophilum* CBS 208.89, *Neurospora crassa* DSMZ 2968 and *Stachybotrys bisbyi* DSMZ 63042) were obtained from the Centraalbureau voor Schimmelcultures (CBS, Utrecht, The Netherlands) and Deutsche Sammlung von Mikroorganismen und Zellkulturen (DSMZ, Braunschweig, Germany) in freeze dried or actively growing form on agar slants and were periodically subcultured on potato dextrose agar (PDA) plates. Freshly inoculated agar plates were grown at 25 or 30 °C, depending on the published growth temperatures of the cultures until reaching a diameter of 5 cm and then used to inoculate shaking flasks. The medium used for submersed cultures contained (per litre): 20 g of alpha-cellulose, 5 g of peptone from meat and 0.3 ml of a trace element solution. The trace element solution contained (per litre): 1 g of ZnSO<sub>4</sub>·7H<sub>2</sub>O, 0.3 g of MnCl<sub>2</sub>·4H<sub>2</sub>O, 3 g of H<sub>3</sub>BO<sub>3</sub>, 2 g of CoCl<sub>2</sub>·6H<sub>2</sub>O, 0.1 g of CuSO<sub>4</sub>·5H<sub>2</sub>O, 0.2 g of NiCl<sub>2</sub>·6H<sub>2</sub>O, 4 ml of H<sub>2</sub>SO<sub>4</sub> (Sachslehner et al., 1997, Applied Biochemistry and Biotechnology 6365: 189-201). For the cultivation in shaking flasks, 1 L Erlenmeyer flasks were filled with 0.3 L of medium. After sterilisation the flasks were inoculated with 3 cm<sup>2</sup> of finely cut mycelium from PDA plates and incubated in a rotary shaker (110 rpm, eccentricity = 1.25 cm) at 25 or 30 °C. Samples were taken regularly and the production of CDH was monitored.

## **Example 6: CDH Production and Purification from Fungal Sources**

[0074] CDH production was performed in up to 16 parallel shaking flask cultures per strain using identical conditions as in the screening procedure. Cultures were harvested on the day exhibiting maximum cyt c activity. The culture supernatant was separated from residual cellulose and fungal biomass by centrifugation (20 min, 6000 x g) and concentrated and diafiltrated using a polyethersulfone hollow fibre cross-flow module with a 10 kDa cut-off (Microza UF module SLP-1053, Pall Corporation) until a conductivity of 2 mS cm<sup>-1</sup> was reached. The concentrated enzyme preparation was applied to a DEAE Sepharose column (chromatography equipment from GE Healthcare Biosciences) mounted on an ÄKTA Explorer system and equilibrated with 50 mM sodium acetate buffer, pH 5.5. The column was eluted with a linear salt gradient (0 to 0.5 M NaCl in the same buffer) in 10 column volumes (CV). Fractions with a high specific CDH activity were pooled, saturated ammonium sulphate solution was slowly added at 4 °C to 20% final saturation and applied to a PHE-Source column equilibrated with 100 mM sodium acetate buffer, pH 5.5 containing (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> (20% saturation) and 0.2 M NaCl. The column was eluted with a linear gradient (0 to 100% of 20 mM sodium acetate buffer, pH 5.5) in 10 CV. The purest CDH fractions were pooled, desalted with 20 mM sodium acetate buffer, pH 5.5, concentrated and frozen at -70 °C for further use.

## **Example 7: Obtaining Nucleotide and Protein Sequences of New CDHs.**

[0075] Mycelium for nucleic acid isolations was harvested from cellulose induced growing cultures after 5 days. The mycelium was frozen in liquid nitrogen and homogenized using mortar and pestle. Portions of 100 mg mycelium were used for DNA extraction (Liu et al., 2000, Journal of Clinical Microbiology, 38: 471). Total RNA was isolated using TriFast (Peqlab, Erlangen, Germany). cDNA synthesis was performed with the First Strand cDNA Synthesis Kit (Fermentas, Vilnius, Lithuania) and the anchor primer (5'-GGCCACGCGTCGACTAGTACTTTTTTTTTTTT-3'). Degenerated primer on the basis of known ascomycete CDH sequences were used to amplify fragments of genomic DNA encoding for CDH. For the amplification of the adjacent upstream region the DNA Walking SpeedUp Premix Kit (Seegene, Seoul, Korea) was used. For the amplification of the 3' region cDNA was used as a template. To obtain full-length cDNA clones encoding the CDH proteins a nested PCR with two specific forward primer upstream of the putative start codon and two reverse primer, one specific for a sequence shortly downstream of the stop codon and the universal primer (5'-GTACTAGTCGACGCGTGGCC-3') complementary to the anchor primer, was done. Names in the following primer table are abbreviated as follows: *Chaetomium atrobrunneum*, CA; Corynascus thermophilus, CT; *Hypoxylon haematostroma*, HH; *Neurospora crassa*, NC; *Stachybotrys bisbyi*, SB.

	FORWARD PRIMER		REVERSE PRIMER
5'-HH-1	ATGCCTCTCTTGTTTGGACCG	UNIVERSAL	
5'-HH-2	TCAACTCTCATACTTGGCTTGG	3'-HH-1	TACATCCAGCTTACCGGCACTG
5'-CA-1	TAGAGTCGAGGCGAACCAG	UNIVERSAL	

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(continued)

	FORWARD PRIMER		REVERSE PRIMER
5'-CA-2	TTGCTGCTGTGCTCCTATGC	3'-CA-1	TTCCTTCCCTCCATCAACTCC
5'-SB-1	TCTTGCTACGCACTTCGGTATTG	UNIVERSAL	
5'-SB-2	TGTGTACCCTGTTTACTCACC	3'-SB-1	GTACCCATTAAGTACACTGCCAG
5'-CT-1	TCTTATAAGCCTTTGGCTCC	UNIVERSAL	
5'-CT-2	TTGGCTCCGTTGGAACAATG	3'-CT-1	TTCCCCCTTCGAATTCGGTC
5'-NC-1	CGCACCAACCGTGTGAAGTG	UNIVERSAL	
5'-NC-2	TACAAGATGAGGACCACCTCG	3'-NC-1	AGCTACCTATCACCCTCTGTC

[0076] The obtained PCR products were then fully sequenced to obtain the complete nucleic acid sequence of the respective cdh gene.

### Example 8: Generation of Myriococcum thermophilum CDH variants by site-directed mutagenesis

**[0077]** For enhanced production of recombinant *Myriococcum thermophilum* CDH (Zamocky et al., 2008,) in *Pichia pastoris* the gene (gene bank accession code EF 492052, GI:164597963) was codon optimised (**FIG 1**) for expression in P. *pastoris* and synthesized by GenScript (Piscataway, NJ, USA). The gene shows a maximum similarity with CDH from *Thielavia heterothallica* (74% identity) and only 63% identity with the gene from *Humicola insolens*. On the protein level, the similarity is highest to *Thielavia heterothallica* CDH (93% identity, 97% positives, 0% gaps) and quite low for *Humicola insolens* CDH (61 % identity, 71% positives, 2% gaps).

[0078] The synthetic M. thermophilum CDH gene was mutated by a two-step site-directed mutagenesis protocol using PCR and Dpn/digestion. The yeast vector pPICZ A carrying the synthetic CDH gene was used as template for mutagenic PCR. For the replacement of Asp160 with Lys the primers 5'-TCCAAGCTTTTAAAGATCCAGGTAAC-3' (Mt CDH-D160K-fw) and 5'-AAAAGCTTGGACCCAACCAAG-3' (Mt CDH-D160K-rv) were used. For the double mutant D547S/E550S primers 5'-GTCTTCTATTCTTTTTACTCTGCTTGGGATG -3' (Mt CDH- D547S/E550S -fw) and 5'- ATAGAAGACCACATCAGGG -3' (Mt CDH- D547S/E550S -rv) were used. The mutation sites are indicated by bold letters in the mutagenic forward primers. PCR was performed under the following conditions: 98 °C for 30 s, then 32 cycles of 98 °C for 10 s; 62 °C for 20 s; 72 °C for 2 min, with a 10 min final extension at 72 °C. The 50  $\mu$ l reaction mix contained Phusion HF Buffer (New England Biolabs, Ipswich, MA, USA), 0.1  $\mu$ g of plasmid DNA, 1 unit of Phusion DNA polymerase (New England Biolabs), 10  $\mu$ M of each dNTP and 5 pmol of each primer.

[0079] PCR reactions were separated by agarose gel electrophoresis and bands at 6 kB purified using the Wizard SV Gel and PCRClean-Up System (Promega, Madison, WI, USA). The purified PCR fragment was digested with DpnI (Fermentas, Vilnius, Lithuania) to remove methylated DNA. 10  $\mu$ I of this reaction was used to transform chemically competent NEB-5-Alpha E. *coli* cells (New England Biolabs) according to the manufacturer. For each mutation 3 colonies were checked by sequencing for the presence of the correct mutation. The purified plasmid of a positive clone was linearized with *SacI* and used to transform competent X-33 *P. pastoris* cells. Colonies growing on YPD zeocin agar plates (100 mg/L) were checked by PCR for the integration of the construct. Two positive clones of each mutation were further cultivated under induced condition and analysed for CDH production. The clones with the highest yield were selected for fermentation.

#### **Example 9: Production of recombinant CDH**

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[0080] An overnight pre-culture of a *Pichia pastoris* transformant (selected from a YPD plate with 100 mg/L Zeocin) was inoculated into 0.3 L of production stage medium in a Infors HT multifermenter (Bottmingen, Switzerland). The production stage medium contained per litre: 26.7 ml of H<sub>3</sub>PO<sub>4</sub> (85%); 0.93 g of CaSO<sub>4</sub>·2H<sub>2</sub>O; 14.9 g of MgSO<sub>4</sub>·7H<sub>2</sub>O; 18.2 g of K<sub>2</sub>SO<sub>4</sub>; 4.13 g of KOH; 4% (v/v) glycerol; 1.45 ml of PTM<sub>1</sub> trace element solution for P. *pastoris* according to the Invitrogen manual 053002 Ver. B (Carlsbad, CF, USA). The PTM<sub>1</sub> trace element solution contains per litre: 6 g of CuSO<sub>4</sub>·5H<sub>2</sub>O, 0.08 g of NaI, 3 g of MnSO<sub>4</sub>·H<sub>2</sub>O, 0.2 g of NaMoO<sub>4</sub>·2H<sub>2</sub>O, 0.02 g of H<sub>3</sub>BO<sub>3</sub>, 0.5 g of CoCl<sub>2</sub>, 20 g of ZnCl<sub>2</sub>, FeSO<sub>4</sub>·7H<sub>2</sub>O, 0.2 g of biotin, 5 ml of sulfuric acid. A glycerol feed was performed with an addition of 9 gL<sup>-1</sup>h<sup>-1</sup> until the wet cell weight exceeded 150 g per litre. As soon as the residual glycerol was used up (determined by monitoring the increase of the dissolved oxygen tension), a methanol feed (100% methanol containing 12 ml PTM<sub>1</sub> trace element solution per litre) with an average addition of 3 gL<sup>-1</sup>h<sup>-1</sup> was started and continued for 72 h at 30 °C and 20% oxygen

tension. The culture supernatant was separated from residual biomass by centrifugation (20 min, 6000 x g) and concentrated and purified by hydrophobic interaction chromatography. To that purpose, saturated ammonium sulphate solution was slowly added to the clear culture supernatant at 4 °C to 20% final saturation. After a second centrifugation step (30 min, 30,000 x g) the solution was applied to a PHE-Source column (GE Healthcare Biosciences) equilibrated with 100 mM sodium acetate buffer, pH 5.5 containing (NH $_4$ ) $_2$ SO $_4$  (20% saturation) and 0.2 M NaCl. The column was eluted with a linear gradient (0 to 100% of 20 mM sodium acetate buffer, pH 5.5) in 10 CV. The purest CDH fractions were pooled, desalted with 20 mM sodium acetate buffer, pH 5.5, concentrated and stored for further use.

### **Example 10: Electrochemical Equipment**

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[0081] A three electrode flow through amperometric wall jet cell was used (Appelqvist et al, Anal. Chim. Acta, 169 (1985) 237-47.) and contained the working electrode (graphite electrode modified with CDH), a reference electrode (Ag|AgCl in 0.1 M KCl) and a counter electrode made of a platinum wire, connected to a potentiostat (Zäta Elektronik, Höör, Sweden). The enzyme modified electrode was pressfitted into a Teflon holder and inserted into the wall jet cell and kept at a constant distance (ca. 1 mm) from the inlet nozzle. The response currents were recorded on a strip chart recorder (Kipp & Zonen, Delft, The Netherlands). The electrochemical cell was connected on-line to a single line flow injection (FI) system, in which the carrier flow was maintained at a constant flow rate of 0.5 ml min⁻¹ by a peristaltic pump (Gilson, Villier-le-Bel, France). The injector was an electrically controlled six-port valve (Rheodyne, Cotati, CA, USA), and the injection loop volume was 50 μl.

[0082] For the screen-printed electrodes a special methacrylate wall jet flow for flow injection analysis (FIA) from DropSense (Oviedo, Spain) was used. The electrochemical cell consists of a carbon working electrode (4 mm diameter), a carbon counter electrode and silver reference electrode connected to a potentiostat (Zäta Elektronic). The response currents were recorded on a strip chart recorder (Kipp & Zonen). The electrochemical cell was connected on-line to a single flow injection (FI) system, in which the carrier flow was maintained at a constant flow rate of 0.5 ml mim¹ by a peristaltic pump (Gilson). For injection an electronically controlled six-port valve (Rheodyne) and a injection loop (50 μl) was used.

#### **Example 11: Preparation of Enzyme Modified Graphite Electrodes**

[0083] CDH was immobilised through simple chemo-physical adsorption onto the surface of solid spectroscopic graphite electrodes (diameter = 3.05 mm, Ringsdorff Spektralkohlestäbe, SGL Carbon Sigri Greatlakes Carbon Group Ringsdorff-Werke GmbH, Bonn Germany). The electrode was cut and polished on wet emery paper (Tufbak, Durite, P400) and afterwards carefully rinsed with Milli-Q water and dried. Then 5 µl of enzyme solution was spread onto the entire active surface of the electrode (0.0731 cm²). The electrode was dried at room temperature and then stored overnight at 4 °C. Before use, the electrode was thoroughly rinsed with Milli-Q water in order to remove any weakly adsorbed enzyme and plugged into in the wall jet cell already containing buffer. Then, the required potential was applied until a stable background current was obtained before any substrate was injected into the flow system.

### **Example 12: Preparation of Enzyme-Modified Screen Printed Electrodes**

**[0084]** Five  $\mu$ I of enzyme solution was placed on the carbon-based electrode (DropSens, Oviedo, Spain) so that the whole area was entirely coated with solution. The immobilisation was allowed to proceed overnight at 4°C. Before use the electrodes were thoroughly rinsed with water. Cross-linking of the biocomponent was carried out by chemical modification with glutaraldehyde where 1  $\mu$ I of an aqueous 1% glutaraldehyde solution was applied on the enzyme layer at 37°C for 10-15 min. After rinsing the electrodes were allowed to dry at room temperature.

[0085] The optimum for the applied potential was determined with a 10 mM lactose solution. The potential was varied stepwise from - 250 to +600 mV vs. Ag|AgCl in 0.1 M KCl and +300 mV chosen for further experiments.

## Example 13: pH Profiles of CDH Immobilised on Electrodes

[0086] The activity versus pH-profile for direct electron transfer (DET) of the adsorbed enzyme was determined electrochemically using a flow injection system. The substrate was lactose with a concentration of 5 mM. As enzyme assays should proceed under saturating substrate conditions so that slight variations in the absolute concentration have no influence on the reaction rate an amount at least 10 times the  $K_M$ -value should be present. The following buffers were used in the experiments: 50 mM sodium citrate buffer (pH 3.0-6.5), 50 mM sodium phosphate buffer (pH 6.0-9.0). The buffers were degassed before use to prevent micro bubbles in the flow system.

## **Example 14: Heterogeneous Enzyme Kinetics on Electrodes**

[0087] The kinetic parameters  $K_M$  (Michaelis-Menten constant) and  $V_{max}$  (maximum volumetric activity), in this case equal to  $I_{max}$  (maximum response in current), were determined for a number of substrates in the DET mode (the electron acceptor being the graphite electrode). All kinetic parameters were calculated by nonlinear least-square regression, fitting the observed data to the Henri-Michaelis-Menten equation. These calculations were done after correcting the substrate concentration values using the dispersion factor of the flow system used including the wall jet cell by dividing the steady state current registered for a 50 mM ferrocyanide solution with that of the peak current for the injected sample having an equal concentration of ferrocyanide and using an applied potential of 400 mV (Ruzicka and Hansen, Flow Injection Analysis, 2nd ed., Wiley, New York 1988). In our case, for a 1 mm distance between electrode and inlet nozzle and 0.5 ml min<sup>-1</sup> flow rate, the dispersion factor D was equal to 1.18 (Fig. 5).

#### Example 15: CDH of Chaetomium atrobrunneum

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[0088] A cellobiose dehydrogenase with high glucose turnover rates and activity under physiological pH conditions was obtained from liquid cultures of Chaetomium atrobrunneum. The culture was grown and screened as described. The maximum activity under the chosen conditions was 90 U/L (cyt c assay, pH 6.0, 11th day). For enzyme production and purification the outlined procedures were applied and resulted in an CDH preparation with a specific activity of 11.7 U/mg (DCIP assay, pH 6.0), an apparent molecular weight of 90 kDa as determined by SDS-PAGE and an isoelectric point of 4.6. The calculated molecular weight of the obtained protein sequence is 86.047 kDa and fits well to the native CDH. The calculated isoelectric point is 5.0. The spectrum of Chaetomium atrobrunneum CDH is typical and shows the haem alpha-, beta- and gamma-bands of the reduced enzyme at 563, 533 and 430 nm. In the oxidised enzyme the gamma-band has its absorption maximum at 421 nm with a shoulder at 450 nm, which disappears after reduction with lactose and corresponds to the absorption peak of the FAD cofactor. Kinetic characterisation with the cyt c assay and lactose as electron donor resulted in a neutral pH profile with an activity maximum at pH 5.0 and still 18% relative activity at pH 7.4 (FIG 2.a). The specific activity at pH 7.4 was 0.88 U/mg using the cyt c assay and glucose as substrate. The pH optimum of the flavin domain was obtained with the DCIP assay and shows a more acidic pH optimum, however, the flavin domain is sufficiently active at pH 7.4 also with this electron acceptor. Kinetic constants for glucose (obtained with the cyt c assay at pH 5.0) are a  $K_M$  of 240 mM and a  $k_{cat}$  of 17.5 s<sup>-1</sup> for glucose, which shows in comparison to currently known enzymes a far better suitability of this enzyme for the proposed application.

[0089] To test the electrochemical behaviour of *Chaetomium atrobrunneum* CDH on electrodes, the purified enzyme preparation was immobilised by adsorption on a spectroscopic graphite electrode surface. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the current at pH 7.4. The optimum pH under the chosen conditions is 5.6 and 48% of the maximum current was obtained at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_M$  value of the heterogenised enzyme at optimum pH on the electrode surface was determined to be 80 mM and  $I_{max} = 30$  nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_M$  value. The DET current density obtained at pH 7.4 was 233 nA/cm² and the linear range for glucose detection at pH 7.4 with the chosen setup within 3-15 mM.

#### **Example 16: CDH from Corynascus thermophilus**

**[0090]** A cellobiose dehydrogenase with high glucose turnover rates and activity under physiological pH conditions was obtained from liquid cultures of *Corynascus thermophilus*. The culture was grown as described in (Gautsch, diploma thesis, Universität für Bodenkultur, Wien, 2004). The maximum activity obtained was 1400 U/L (cyt c assay, pH 6.0, 6<sup>th</sup> day). For enzyme production and purification the outlined procedures were applied and resulted in a CDH preparation with a specific activity of 17.9 U/mg, an apparent molecular weight of 87 kDa as determined by SDS-PAGE and an isoelectric point of 4.1. The calculated molecular weight of the obtained protein sequence is 81.946 kDa and fits well to the native CDH. The calculated isoelectric point is 4.64. The spectrum of *C. thermophilus* CDH is typical and shows the haem  $\alpha$ -,  $\beta$ - and  $\gamma$ -bands of the reduced enzyme at 562, 533 and 429 nm. In the oxidised enzyme the  $\gamma$ -band has its absorption maximum at 420 nm with a shoulder at 450 nm, which disappears after reduction with lactose and corresponds to the absorption peak of the FAD cofactor. Kinetic characterisation with the cyt c assay resulted in a pH profile with an activity maximum at pH 7.5 and 98% relative activity at pH 7.4 (**FIG 2.b**). The specific activity at pH 7.4 was 3.6 U/mg using the cyt c assay and glucose as substrate. The pH optimum of the flavin domain was obtained with the DCIP assay and shows a more acidic pH optimum, however, the flavin domain is sufficiently active at pH 7.4 also with this electron acceptor. Kinetic constants for glucose (obtained with the cyt c assay at pH 5.0) are a K<sub>M</sub> of 950 mM and a k<sub>cat</sub> of 32 s<sup>-1</sup> for glucose.

[0091] To test the electrochemical behaviour of C. thermophilus CDH on electrodes, the purified enzyme preparation

was immobilised by adsorption on a spectroscopic graphite electrode surface. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the current at pH 7.4. The optimum pH under the chosen conditions is 8.5 and 96% of the maximum current was obtained at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_M$  value of the heterogenised enzyme at optimum pH on the electrode surface was determined to be 188 mM and  $I_{max}$  = 190 nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_M$  value. The DET current density obtained at pH 7.4 was 3500 nA/cm² and the linear range for glucose detection at pH 7.4 with the chosen setup within 1-15 mM.

## 10 Example 17: CDH of Hypoxylon haematostroma

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[0092] A cellobiose dehydrogenase with high glucose turnover rates and activity under physiological pH conditions was obtained from liquid cultures of Hypoxylon haematostroma. The culture was grown and screened as described. The maximum activity under the chosen conditions was 65 U/L (cyt c assay, pH 6.0, 9th day). For enzyme production and purification the outlined procedures were applied and resulted in an CDH preparation with a specific activity of 15.3 U/mg (DCIP assay, pH 6.0), an apparent molecular weight of 85 Da as determined by SDS-PAGE and an isoelectric point of 4.1. The calculated molecular weight of the obtained protein sequence is 87.514 kDa and fits well to the native CDH. The calculated isoelectric point is 6.37. The spectrum of Hypoxylon haematostroma CDH is typical and shows the haem alpha-, beta- and gamma-bands of the reduced enzyme at 563, 533 and 429 nm. In the oxidised enzyme the gamma-band has its absorption maximum at 421 nm with a shoulder at 450 nm, which disappears after reduction with lactose and corresponds to the absorption peak of the FAD cofactor. Kinetic characterisation with the cyt c assay and lactose as electron donor resulted in a neutral pH profile with an activity maximum at pH 5.5 and still 65% relative activity at pH 7.4 (FIG 2.c). The specific activity at pH 7.4 was 2.73 U/mg using the cyt c assay and glucose as substrate. The pH optimum of the flavin domain was obtained with the DCIP assay and shows a more acidic pH optimum, however, the flavin domain is sufficiently active at pH 7.4 also with this electron acceptor. Kinetic constants for glucose (obtained with the cyt c assay at pH 5.5) are a  $K_M$  of 260 mM and a  $k_{cat}$  of 8.8 s<sup>-1</sup> for glucose, which shows in comparison to currently known enzymes a far better suitability of this enzyme for the proposed application.

[0093] To test the electrochemical behaviour of *Hypoxylon haematostroma* CDH on electrodes, the purified enzyme preparation was immobilised by adsorption on a spectroscopic graphite electrode surface. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the current at pH 7.4. The optimum pH under the chosen conditions is 7.5 and the maximum current was obtained at this pH and at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_M$  value of the heterogenised enzyme at optimum pH (7.4) on the electrode surface for glucose was determined to be 49 mM and  $I_{max}$  = 55 nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_M$  value. The DET current density obtained at pH 7.4 was 383 nA/cm² and the linear range for glucose detection at pH 7.4 with the chosen setup within 2-20 mM.

#### **Example 18: CDH of Neurospora crassa**

[0094] A cellobiose dehydrogenase with high glucose turnover rates and activity under physiological pH conditions was obtained from liquid cultures of Neurospora crassa. The culture was grown and screened as described. The maximum activity under the chosen conditions was 156 U/L (cyt c assay, pH 6.0, 18th day). For enzyme production and purification the outlined procedures were applied and resulted in an CDH preparation with a specific activity of 10.6 U/mg (DCIP assay, pH 6.0), an apparent molecular weight of 90 kDa as determined by SDS-PAGE and an isoelectric point of 4.3. The calculated molecular weight of the obtained protein sequence is 86.283 kDa and fits well to the native CDH. The calculated isoelectric point is 6.68. The spectrum of Neurospora crassa CDH is typical and shows the haem alpha-, beta- and gamma-bands of the reduced enzyme at 563, 533 and 430 nm. In the oxidised enzyme the gamma-band has its absorption maximum at 421 nm with a shoulder at 450 nm, which disappears after reduction with lactose and corresponds to the absorption peak of the FAD cofactor. Kinetic characterisation with the cyt c assay and lactose as electron donor resulted in a neutral pH profile with an activity maximum at pH 6.0 and 52% relative activity at pH 7.4 (FIG 2.d). The specific activity at pH 7.4 was 1.04 U/mg using the cyt c assay and glucose as substrate. The pH optimum of the flavin domain was obtained with the DCIP assay and shows a more acidic pH optimum, however, the flavin domain is sufficiently active at pH 7.4 also with this electron acceptor. Kinetic constants for glucose (obtained with the cyt c assay at pH 5.5) are a K<sub>M</sub> of 1680 mM and a kcat of 15.9 s<sup>-1</sup> for glucose, which shows in comparison to other known enzymes a far better suitability of this enzyme for the proposed application.

**[0095]** To test the electrochemical behaviour of *Neurospora crassa* CDH on electrodes, the purified enzyme preparation was immobilised by adsorption on a spectroscopic graphite electrode surface. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the

current at pH 7.4. The optimum pH under the chosen conditions is 5.0 and 31% of the maximum current was obtained at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_M$  value of the heterogenised enzyme at optimum pH on the electrode surface was determined to be 90 mM and  $I_{max}$  = 5 nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_M$  value. The DET current density obtained at pH 7.4 was 82 nA/cm² and the linear range for glucose detection at pH 7.4 with the chosen setup within 2-10 mM.

#### Example 19: CDH of Stachybotris bisbyi

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[0096] A cellobiose dehydrogenase with high glucose turnover rates and activity under physiological pH conditions was obtained from liquid cultures of Stachybotris bisbyi. The culture was grown and screened as described. The maximum activity under the chosen conditions was 154 U/L (cyt c assay, pH 6.0, 24th day). For enzyme production and purification the outlined procedures were applied and resulted in a CDH preparation with a specific activity of 7.9 U/mg (DCIP assay, pH 6.0), an apparent molecular weight of 100 kDa as determined by SDS-PAGE and an isoelectric point of 4.5. The calculated molecular weight of the obtained protein sequence is 86.212 kDa and fits well to the native CDH when considering a glycosylation of 14% of S. bisbyi CDH, a value which lies within the observed range (2-15%, Zámocky et al., 2006, Current Protein and Peptide Science, 7: 255-280). The calculated isoelectric point is 6.37. The spectrum of Stachybotris bisbyi CDH is typical and shows the haem alpha-, beta- and gamma-bands of the reduced enzyme at 562, 533 and 430 nm. In the oxidised enzyme the gamma-band has its absorption maximum at 420 nm with a shoulder at 450 nm, which disappears after reduction with lactose and corresponds to the absorption peak of the FAD cofactor. Kinetic characterisation with the cyt c assay resulted in a pH profile with an activity maximum at pH 5.5 and 60% relative activity at pH 7.4 (FIG 2.e). The specific activity at pH 7.4 was 0.58 U/mg using the cytic assay and glucose as substrate. The pH optimum of the flavin domain was obtained with the DCIP assay and shows a similar trend indicating that substrate oxidation by the enzyme is efficient at pH 7.4. Kinetic constants for glucose (obtained with the cyt c assay at pH 5.5) are a  $K_M$  of 950 mM and a  $k_{cat}$  of 14.1 s<sup>-1</sup> for glucose, which shows in comparison to currently known enzymes a far better suitability of this enzyme for the proposed application.

**[0097]** To test the electrochemical behaviour of *Stachybotris bisbyi* CDH on electrodes, the purified enzyme preparation was immobilised by adsorption on a spectroscopic graphite electrode surface. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the current at pH 7.4. The optimum pH under the chosen conditions is 5.0 and 27% of the maximum current was obtained at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_M$  value of the heterogenised enzyme at optimum pH on the electrode surface was determined to be 131 mM and  $I_{max}$  = 65 nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_M$  value. The DET current density obtained at pH 7.4 was 237 nA/cm<sup>2</sup> and the linear range for glucose detection at pH 7.4 with the chosen setup within 3-15 mM.

## Example 20: CDH from Myriococcum thermophilum

**[0098]** CDH from *Myriococcum thermophilum* was found to oxidise glucose very efficiently (Harreither et al., 2007, Electroanalysis 19: 172-180), but not under physiological conditions. It was used as a protein scaffold for which DET at neutral pH was developed by means of genetic engineering. The enzyme variants D160K and D547S/E550S were obtained according to the described methods to increase the IET at pH 7.4 and thereby the electrode response in order to optimise the enzyme for applications under neutral or alkaline pH conditions.

[0099] For enzyme production and purification of the enzyme from the native producer, the protocol given in (Harreither et al., 2007, Electroanalysis 19: 172-180) was followed and resulted in a CDH preparation with a specific activity of 10.7 U/mg (DCIP assay, pH 6.0), an apparent molecular weight of 94 kDa and an isoelectric point of 3.8. The calculated molecular weight of the protein sequence is 86.701 kDa and fits well to the native CDH. The calculated isoelectric point is 4.62. The spectrum of CDH obtained from M. thermophilum is typical and shows the haem alpha-, beta- and gammabands of the reduced enzyme at 563, 533 and 429 nm. In the oxidised enzyme the gamma-band has its absorption maximum at 421 nm with a shoulder at 450 nm, which disappears after reduction with lactose and corresponds to the absorption peak of the FAD cofactor. Kinetic characterisation with the cyt c assay and lactose as electron donor resulted in a neutral pH profile with an activity maximum between pH 4.0-4.5 and 0% relative activity at pH 7.4 (FIG 2.f). The specific activity at pH 7.4 was also 0 U/mg using the cyt c assay and glucose as substrate. The pH optimum of the flavin domain was obtained with the DCIP assay and shows a far less acidic pH optimum (6.0), indicating that substrate oxidation at the flavin domain is performed even at neutral and slightly alkaline conditions efficiently, but the IET is rate limiting. The obtained kinetic constants for glucose (DCIP assay, pH 6.0; K<sub>M</sub> = 250 mM, kcat = 14.2 s<sup>-1</sup>) show that although glucose conversion is very efficient, M. thermophilum CDH is not suitable for the proposed application because no IET was measured above pH 7.0.

**[0100]** The recombinant enzyme variants were produced heterologeously in *P. pastoris* according to the explained routines. The molecular weights, isoelectric points or spectral properties did not differ significantly from the native enzyme produced by the fungus.

**[0101]** Kinetic characterisation of D160K with the cyt c assay and lactose as electron donor resulted in a pH with an activity maximum at 5.0 and 24% relative activity at pH 7.4 (**FIG 2.g**). The specific activity at pH 7.4 was 1.01 U/mg using the cyt c assay and glucose as substrate. To test the electrochemical behaviour of D160K on electrodes, the purified enzyme preparation was immobilised by adsorption on a screen printed electrode. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the current at pH 7.4. The optimum pH under the chosen conditions is 5.5 and 52% of the maximum current was obtained at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_{\rm M}$  value of the heterogenised enzyme at optimum pH on the electrode surface was determined to be 133 mM and  $I_{\rm max}$  = 105 nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_{\rm M}$  value. The DET current density obtained at pH 7.4 was 513 nA/cm<sup>2</sup> and the linear range for glucose detection at pH 7.4 with the chosen setup within 0.5-20 mM.

**[0102]** Kinetic characterisation of D547S/E550S with the cyt c assay and lactose as electron donor resulted in a pH with an activity maximum between 4.5 and 5.0 and 13% relative activity at pH 7.4 (**FIG 2.h**). The specific activity at pH 7.4 was 0.70 U/mg using the cyt c assay and glucose as substrate. To test the electrochemical behaviour of D547S/E550S on electrodes, the purified enzyme preparation was immobilised by adsorption on a graphite electrode surface. Using a flow cell and subsequent injections of 50 mM glucose, DET currents were determined to determine the pH optimum, the current at the pH optimum and the current at pH 7.4. The optimum pH under the chosen conditions is 5.5 and 24% of the maximum current was obtained at pH 7.4 in 10 mM phosphate buffered saline (PBS) containing 100 mM NaCl. The  $K_M$  value of the heterogenised enzyme at optimum pH on the electrode surface was determined to be 55 mM and  $I_{max}$  = 30 nA. The currents obtained in glucose measurements should therefore follow a nearly linear relationship for concentrations approx. five-fold below the  $K_M$  value. The DET current density obtained at pH 7.4 was 241 nA/cm<sup>2</sup> and the linear range for glucose detection at pH 7.4 with the chosen setup within 1-20 mM.

### **Example 21: Sequences**

### [0103]

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>M.thermophilum (SEQ ID NO: 1)

mrtssrligalaaallpsalaqnnvpntftdpdsgitfntwgldedspqtqggftfgvalpsdalttdasefigylkcarndesgw cgislggpmtnsllitawphedtvytslrfatgyampdvyegdaeitqvsssvnsthfslifrcknclqwshggssggastsggvl vlgwvqafddpgnptcpeqitlqqhdngmgiwgaqlntdaaspsytdwaaqatktvtgdcegptetsvvgvpvptgvsfdyivvgg gaggipaadklseagksvlliekgfastantggtlgpewleghdltrfdvpglcnqiwvdskgiacedtdqmagcvlgggtavnag lwfkpysldwdylfpdgwkyndvqpainralsripgtdapstdgkryyqegfevlskglaaggwtsvtannapdkknrtfahapfm faggerngplgtyfqtakkrnnfdvwlntsvkrviregghitgvevepfrdggyegivpvtkvtgrvilsagtfgsakillrsgig pedqlevvaasekdgptmignsswinlpvgynlddhlntdtvishpdvvfydfyeawddpiesdknsylesrtgilaqaapnigpm fweeivgadgivrqlqwtarvegslgapnghtmtmsqylgrgatsrgrmtitpslttivsdvpylkdpndkeaviqgiinlqnalq nvanltwlfpnstitpreyvesmvvspsnrrsnhwmgtnklgtddgrkggsavvdldtrvygtdnlfvidasifpgvpttnptsyi vvaaehassrilalpdlepvpkygqcggrewtgsfvcadgstceyqnewysqcl

>H.haematostroma (SEQ ID NO: 3)

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20 >C.attrobruneum (SEQ ID NO: 5)

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>C.thermophilum (SEQ ID NO: 7)

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	>S.bisbyi (SEQ ID NO: 9)
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>M.thermophilum (SEQ ID NO: 2)

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>H.haematostroma (SEQ ID NO: 4)
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### 45 Claims

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- 1. A cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above, selected from a cellobiose dehydrogenase being a modified cellobiose dehydrogenase of *Myriococcum thermophilum*, a cellobiose dehydrogenase isolated from *Chaetomium atrobrunneum*, *Hypoxylon haematostroma or Stachybotrys bisbyi*.
- 2. Cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprising an amino acid sequence of SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 9 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to SEQ ID NO: 3, SEQ ID NO: 5 or SEQ ID NO: 9.
- 3. Cellobiose dehydrogenase according to claim 1 being a modified cellobiose dehydrogenase of Myriococcum ther-mophilum, comprising a flavin and a haem domain, wherein electron transfer from the flavin to the haem domain is increased as compared to wild type cellobiose dehydrogenase of Myriococcum thermophilum, preferably as measured by the cyt c assay.

- 4. Cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprising an amino acid sequence of amino acids 22 to 828 of SEQ ID NO: 1 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to amino acids 22 to 828 of SEQ ID NO: 1, characterised in that the amino acid sequence has at least one additional amino acid substitution, deletion or insertion to the sequence of SEQ ID NO:1 increasing electron transfer from the flavin to the haem domain as compared to wild type cellobiose dehydrogenase of Myriococcum thermophilum of SEQ ID NO:1, preferably as measured by the cytic assay.
- 5. Cellobiose dehydrogenase according to claim 4, characterised in that the modification is a modification of the haem domain of any one of amino acids 90-100, 115-124, 172-203, preferably of any one of amino acids 176, 179-182, 195, 196, 198, 201 corresponding to the *M. thermophilum* CDH of SEQ ID NO: 1 and/or of the flavin domain of any one of amino acids 311-333, 565-577, 623-625, 653-664, 696-723, preferably of any one of amino acids 318, 325, 326, 328, 568, 571, 574, 575, 624, 654, 663, 702, 709, 712, 717, correspond to the *M. thermophilum* cellobiose dehydrogenase of SEQ ID NO: 1, or any combination thereof, preferably comprising a modification with an increase of positive charge in the of amino acids 172-203 corresponding to the *M. thermophilum* cellobiose dehydrogenase of SEQ ID NO: 1, preferably of amino acid 181, in particular preferred a D181K mutation, and/or a decrease of a negative charge of amino acids 565-577 corresponding to the *M. thermophilum* cellobiose dehydrogenase of SEQ ID NO: 1, preferably of amino acid(s) 568 and/or 571, in particular preferred a D568S and/or E571S mutation, or any combination thereof.
- 20 6. Cellobiose dehydrogenase according to any one of claims 1 to 5, being isolated by diafiltration, ion exchange chromatography by collecting fractions with cellobiose dehydrogenase activity, preferably being further purified by hydrophobic interaction chromatography.
  - 7. A nucleic acid molecule encoding a cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above and comprising
    - a nucleotide sequence of SEQ ID NOs 4, 6, 8, 10 or 12, or

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- the open reading frame of SEQ ID NOs 4, 6, 8, 10 or 12 or
- a nucleotide sequence with at least 50%, preferably at least 55%, at least 60%, at least 65%, at least 70%, at least 75%, at least 80%, at least 85%, at least 95%, at least 98%, in particular preferred at least 99%, identity to SEQ ID NO: 2, 4, 6, 8, 10 or 12 or the open reading frame of SEQ ID NOs: 2, 4, 6, 8, 10 or 12, further comprising a nucleotide mutation, substitution, deletion or insertion, preferably a codon mutation, substitution, deletion or insertion,
- a nucleotide sequence that hybridizes with any one of SEQ ID NO: 2, 4, 6, 8, 10 or 12 under stringent condition,

preferably encoding a cellobiose dehydrogenase as defined in claims 1 to 6.

- **8.** An electrode comprising an immobilised cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above and having at least 10%, preferably at least 14%, in particular preferred at least 18%, glucose, lactose or cellobiose oxidising activity at a pH of 7.4 as compared to their maximal activity at a lower pH as determined by a cyt c assay, preferably also as compared to their maximal activity at a lower pH as determined by a DCIP assay.
- **9.** Electrode according to claim 8, **characterised in that** the glucose oxidation activity at pH 7.4 is at least 0.5 U/mg cellobiose dehydrogenase.
- **10.** Electrode according to any one of claims 8 or 9, **characterised in that** the K<sub>M</sub> value of the cellobiose dehydrogenase for an glucose oxidation reaction at pH 7.4 is below 1 M.
- 11. Electrode according to any one of claims 8 to 10, characterised in that the cellobiose dehydrogenase is of Chaetomium atrobrunneum, Corynascus thermophilus, Hypoxylon haematostroma, Neurospora crassa or Stachybotrys bisbyi or a modified cellobiose dehydrogenase of Myriococcum thermophilum with an increased activity at pH of 7.4 as defined in any one of claims 3 to 5, or the cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprises an amino acid sequence of SEQ ID NO: 3, SEQ ID NO: 5, SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to SEQ ID NO: 3, SEQ ID NO: 5, SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11, preferably wherein the cellobiose dehydrogenase is defined as in any one of claims 1 to 6 or encoded by the nucleic acid molecule of claim 7.

- 12. Electrode according to any one of claims 8 to 11, **characterised in that** the cellobiose dehydrogenase is immobilised by adsorption, complex formation, preferably via an additional complexing linker, covalent or ionic linkage, preferably wherein the immobilized cellobiose dehydrogenase is cross-linked, in particular by bifunctional agents, to increase stability or activity.
- **13.** An electrochemical cell comprising an electrode according to any one of claims 8 to 12 as an anodic element and a cathodic element, preferably comprising a glucose containing solution as anodic fluid, wherein further preferred the anodic fluid is a solution of at least pH 6.0, preferably at least pH 6.5, in particular preferred at least pH 7.0, even more preferred at least pH 7.2 or most preferred at pH 7.4 or above.
- 14. Method of detecting or quantifying glucose in a sample comprising

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- providing a cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above,
- contacting a fluid sample having a pH of at least 6.5, preferably at least 7.0, in particular preferred at least 7.3, especially preferred at least 7.4, with the cellobiose dehydrogenase, and
- detecting an oxidation of glucose of the sample by the cellobiose dehydrogenase,

wherein preferably the oxidation is detected electrochemically, preferably with an immobilised cellobiose dehydrogenase on an electrode, in particular preferred as defined in any one of claims 8 to 12.

- **15.** The method of claim 14, **characterised in that** the cellobiose dehydrogenase is of *Chaetomium atrobrunneum*, *Corynascus thermophilus*, *Hypoxylon haematostroma*, *Neurospora crassa or Stachybotrys bisbyi* or a modified cellobiose dehydrogenase of *Myriococcum thermophilum* with an increased activity at pH of 7.4 as defined in any one of claims 3 to 5,
- or the cellobiose dehydrogenase having glucose oxidation activity at a pH of 7.4 or above comprises an amino acid sequence of SEQ ID NO: 3, SEQ ID NO: 5, SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11 or an amino acid sequence being at least 50%, preferably at least 75%, in particular preferred at least 90%, identical to SEQ ID NO: 3, SEQ ID NO: 5, SEQ ID NO: 7, SEQ ID NO: 9 or SEQ ID NO: 11,
- preferably the cellobiose dehydrogenase is further defined as in any one of claims 1 to 6.

Fig. 1

↓ start signal pept	R L	I G A		A A 14
ATG AGA ACT TCT TCT		rc GGT GCC ↓ start hae		CA GCT 42
L L P S A TTG CTT CCT TCT GCC	L A	Q N N	V P	N T 28 AC ACC 84
F T D P D TTT ACT GAC CCT GAC		I T F TC ACT TTC		W G 42 GG GGA 126
L D E D S CTT GAC GAG GAT TCT	~	T Q G CT CAG GGT		T F 56 CT TTC 168
G V A L P GGT GTT GCT TTG CCA	S D Z TCC GAC G			A S 70 CA TCT 210
E F I G Y GAG TTC ATC GGT TAC	TTG AAG T	C A R GT GCT AGA * * *	AAT GAC G	E S 84 AG TCC 252 * *
G W C G I GGA TGG TGT GGT ATT * *		G G P GT GGT CCT		N S 98 AC TCC 294
L L I T A TTG TTG ATT ACT GCT * * * *	TGG CCT C	H E D AC GAG GAC * * *		Y T 112 AC ACT 336
S L R F A TCC TTG AGA TTT GCT		Y A M AC GCC ATG		V Y 126 TT TAC 378
E G D A E GAG GGT GAT GCT GAA		Q V S AA GTC TCT		V N 140 TC AAT 420
S T H F S TCC ACT CAT TTC TCT		F R C TT AGA TGT		C L 154 GT TTG 462
Q W S H G CAA TGG TCC CAC GGA * *	GGT TCC T	S G G CT GGT GGT + * *	GCT TCT A	T S 168 CC TCC 504 + +
G G V L V GGT GGT GTT CTT GTT * * * * * *	L G !	W V Q GG GTC CAA * * *	A F C	D D 182 AC GAT 546
P G N P T CCA GGT AAC CCA ACC * + * * +	TGT CCA G	E Q I AA CAG ATT	T L (	Q Q 196
H D N G M CAC GAC AAT GGA ATG	G I	W G A GG GGT GCA	Q L I CAA TTG A	N T 210 AT ACC 630
D A A S P GAT GCT GCA TCT CCA  ↓ start lir	TCC TAT A	CC GAC TGG		Q A 224 AA GCT 672
	G D	C E G	P T I	

Fig. 1 Continuation

TCT	$\operatorname{GTG}$			V GTT							S	$\mathbf{F}$	fla D GAT	252
doma Y TAC	I	V GTT	V GTC	G GGA				G GGA			P CCA		A GCT	266 798
D GAC			S TCT	E GAG				S TCC					E GAG	280 840
K AAG	G GGT	F TTC	A GCT	S TCT		A GCT		T ACC			T ACT	L TTG	G GGT	294 882
P CCA	E GAG	W TGG *	L TTG *	E GAG *		H CAC *	D GAT *		T ACT +		F TTC *	D GAC *	V GTT *	308 924
P CCA *	G GGT *	L CTT +	C TGC +	N AAC *	Q CAA +		W TGG *	V GTG *	D GAC *	S TCT *	K AAG	G GGA	I ATC	322 966
A GCT	C TGC	E GAG	D GAT	T ACT			M ATG			C TGT	V GTT	L CTT	G GGT	336 1008
G GGA	G GGT	T ACC	GCA		AAT	GCT	GGT	CTT	TGG	TTC	K AAG	P CCA	Y TAT	350 1050
S TCT	L TTG	D GAT		D GAT							W TGG	K AAG	Y TAC	364 1092
N AAC		V GTC		P CCT						L TTG	S TCT	R CGT	I ATT	378 1134
P CCT	G GGT	T ACT	D GAC	A GCT			T ACT		G GGA	K AAG	R AGA	Y TAC	Y TAC	392 1176
Q CAG	E GAA	G GGT	F TTT	E GAG	V GTT	L CTT	S TCT	K AAA	G GGT	L TTG	A GCC	A GCT	G GGT	406 1218
G GGA	W TGG	T ACC	S TCT	V GTG	ACT			TAA	GCT		D GAC	K AAG	K AAG	420 1260
				A GCT	CAC	GCA	CCT		ATG					434 1302
													K AAA	
			AAC	TTC		GTC	TGG		AAC	ACT		GTT	AAG	
	V GTT			E GAA									V GTG	476 1428

# Fig. 1 Continuation

$\mathbf{T}$	P CCT K AAG	V	AGA T	GAT G	GGA R	GGT V	I	GAG L	GGT S	ATC A	GTG G	$_{\rm T}^{\rm CCT}$	GTG F	490 1470 504 1512
G GGT	S TCC						R CGT							518 1554
D GAC	Q CAA	L TTG					A GCT				D GAT	G GGA	P CCT	532 1596
T ACC	M ATG	I ATC	G GGT	N AAC	S TCC	S TCT	W TGG	I ATT	N AAC	L TTG	P CCT	V GTG	G GGA	546 1638
Y TAC	N AAC	L TTG	D GAC	D GAT *	H CAC *	L TTG *			D GAC *	T ACC	V GTG *		S TCT +	560 1680
H CAC +	P CCT *	D GAT *	V GTG	V GTC	F TTC	Y TAT	D	F TTT	Y TAC	E	A GCT	W TGG	D	574 1722
D	P CCA	I ATT	E GAA				N AAC				E GAA	S TCT	R AGA	588 1764
T ACC	G GGA	I ATC	L TTG	A GCT	Q CAA		A GCT				G GGT	P CCA		602 1806
177	F.T	_	_											
F TTC	W TGG	E GAA	E GAG	I ATT	V GTG		A GCT +		G GGT	I ATT	V GTC	R AGA	Q CAA	616 1848
TTC L	TGG Q	GAA W	GAG T	ATT A	GTG R	GGA * V	GCT + E	GAC * G	GGT S	ATT L	GTC G	AGA A		1848
TTC L TTG N	TGG Q	GAA W TGG H	GAG T ACC T	ATT A GCC M	GTG R AGA T	GGA  V GTT	GCT + E GAG	GAC * G GGT	GGT S TCT Y TAC	ATT L TTG L	GTC G GGT G	AGA A GCA R	CÃA P CCT G	1848
TTC L TTG N AAC	TGG Q CAG G	GAA W TGG H CAT	GAG T ACC T ACC	ATT A GCC M ATG	R AGA T ACC	GGA  V GTT  M ATG	GCT E GAG S TCT	GAC  GGT  Q CAA  * I	S TCT Y TAC + T	L TTG L CTT *	GTC GGT GGT * S	AGA A GCA R CGT * L	P CCT G GGT + T	1848 630 1890 644
TTC  L TTG  N AAC  A GCC  * T	TGG Q CAG GGA T ACT	GAA W TGG H CAT S TCT * V	TACC TACC RAGA * S	ATT  A GCC  M ATG  GGT + D	R AGA T ACC R AGA *	GGA  * V GTT  M ATG  M ATG	GCT  E GAG  S TCT  T ACC	GAC  * G GGT  Q CAA  * I ATC	STCT YTAC + TAC ACT	L TTG L CTT * P CCA D	GTC  GGT  GGT  * STCT  P	AGA A GCA R CGT * L TTG	CÃA  P CCT  G GGT + T ACC	1848 630 1890 644 1932
TTC  L TTG  N AAC  A GCC  * T ACT	TGG Q CAG GGA T ACT * I ATT	GĀA W TGG H CAT S TCT * V GTT	GAG T ACC R AGA * S TCC	ATT  A GCC  M ATG  GGT + D GAC	GTG R AGA T ACC R AGA * V GTC	GGA  * V GTT  M ATG  M ATG  G  G	GCT  + E GAG  S TCT  T ACC  Y TAC	GAC  * GGT  Q CAA  * I ATC  L CTT	S TCT Y TAC + T ACT K AAA	L TTG  L CTT  * P CCA  D GAC	GTC GGT GGT * STCT PCCA Q	AGA A GCA R CGT * L TTG N AAC	CÃA  P CCT  G GGT + T ACC  D GAC	1848 630 1890 644 1932 658 1974 672 2016
TTC  L TTG  N AAC  A GCC  * T ACT  K AAA	TGG Q CAG GGA T ACT * I ATT	GAA WTGG HCAT STCT * VGTT AGCC	GAG TACC ACC RAGA * STCC VGTG	ATT  A GCC  M ATG  GGT  + D GAC  I ATT	GTG R AGA T ACC R AGA * V GTC Q CAA	GGA  * V GTT  M ATG  P CCT  G GGT	GCT  + E GAG  S TCT  T ACC  Y TAC  I ATT	GAC  * GGT  QCAA  * IATC  LCTT  IATC	S TCT Y TAC + T ACT K AAA N AAAC * L	L CTT * P CCA D GAC L TTG * F	GTC GGGT * GGGT * STCT PCCA QCAG * P	AGA  A GCA  R CGT  * L TTG  N AAC  N AAT  * N	CÃA  P CCT  G GGT + T ACC  D GAC  A GCT * S	1848 630 1890 644 1932 658 1974 672 2016
TTC  L TTG  N AAC  A GCC  * T ACT  K AAA  L TTG	TGG  Q CAG  GGA  T ACT  * I ATT  E GAA  Q CAG + I	GĀĀ W TGG H CAT S TCT V GTT A GCC N AAC *	GAG TACC RAGA * STCC VGTG VGTT	ATT  A GCC  M ATG  GGT  D GAC  I ATT  A GCC  * R	GTG  R AGA  T ACC  R AGA  * V GTC  Q CAA  N AAT  * E	GGA  * V GTT  M ATG  P CCT  G GGT  L TTG  * Y	GCT  + E GAG  S TCT  T ACC  Y TAC  I ATT  T ACC	GAC  * GGT  Q CAA  * ATC  L CTT  I ATC  W TGG + E	GGT S TCT Y TAC + T ACT K AAA N AAC * L TTG * S	L TTG  L CTT  * P CCA  D GAC  L TTG  * F  TTC  * M	GTC GGT GGT * STCT PCCA QCAG * PCCA + V	AGA  A GCA  R CGT  L TTG  N AAC  N AAC  * N AAC  V	CÃA  P CCT  G GGT + T ACC  D GAC  A GCT * S TCT * S	1848 630 1890 644 1932 658 1974 672 2016 686 2058

# Fig. 1 Continuation

L TTG	G GGT	T ACT	D GAT		R AGA					742 2226
D GAC			T ACT							756 2268
	D GAT		S TCC							770 2310
T ACT	-		I ATT							784 2352
	L CTT		L TTG							798 2394
			G GGA							812 2436
			T ACC						Q CAA	826 2478
C TGT	L TTG		83 24	30 190						

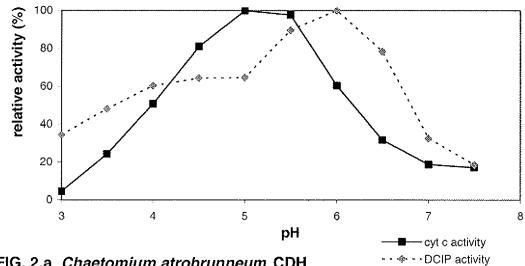


FIG. 2.a Chaetomium atrobrunneum CDH

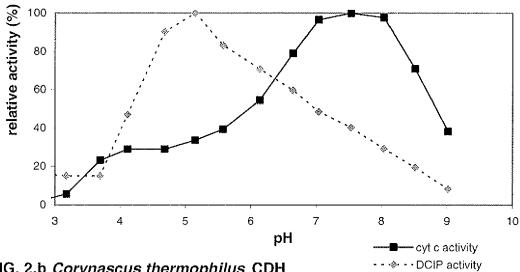
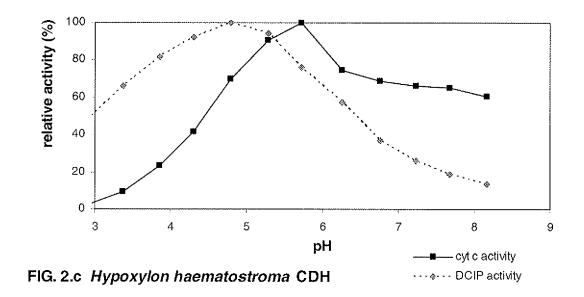
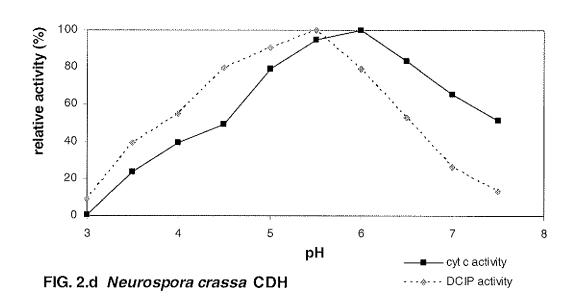
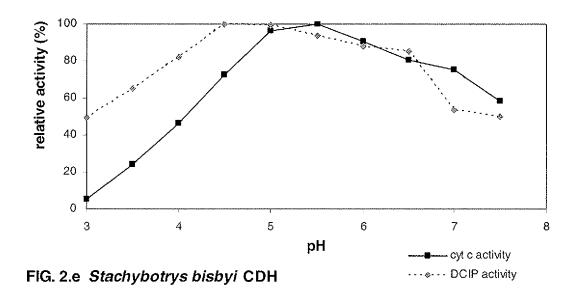
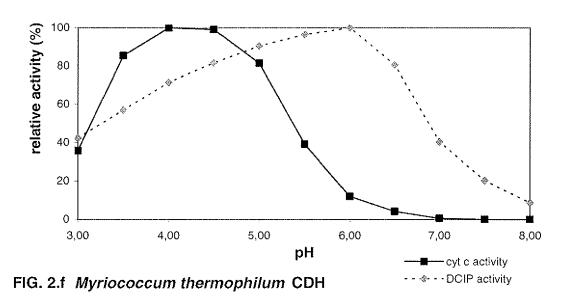


FIG. 2.b Corynascus thermophilus CDH









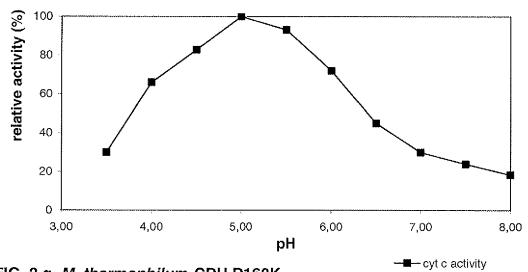


FIG. 2.g M. thermophilum CDH D160K

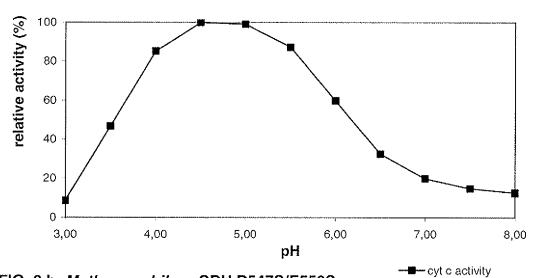


FIG. 2.h M. thermophilum CDH D547S/E550S

136 133 135

70 64 69 67 67

Fig. 3

205

275 263 265

261

203

MRPSSRFVGALAAAASFLPSALAQNNAAVTFTDPDTGIVFNSWGLANGAPQTQGGFTFGVALPSDALTTD MKLLSRVGATALAATLSLKQCAAQM-TEGTYTHEATGITFKTWTPSDG-----STFTFGLALPGDALTND MGRLGSLAKLLLAVGLNVQQCFGQNGPPTPYTDSETGITFATWSGGNG-LAPWGGLTFGVALPENALTTD : MRTSSRIIGALAAAL--LPSALAQNNVPNTFTDPDSGITFNTWGLDEDSPQTQGGFTFGVALPSDALTTD MRTTSAFLSGLAAVASLLSPAFAQT-APKTFTHPDTGIVFNTWSASD--SQTKGGFTVGMALPSNALTTD --MLFKLSNWLLALALFVGNVVAQLEGPTPYTDPDTGIVFQSWVNPAG-----TLKFGYTYPANAATVA ATEFIGYLECASADN---QGWCGVSMG--GPMTNSLLITAWPHEDNVYTSLRFATGYAMPDVYSGDATIT ATEYIGLIRCQITDP-SSPGYCGISHGQSGQMTQALLIVAWASEDVVYTSFRYATGYTLPELYTGDAKLT ATELIGYLKCGSNGT-TTDAWCGLSFG--GPMTNSLLIMAWPHEDEILTSFRFASGYTRPDLYTGDAKLT ASEFIGYLKCARNDE---SGWCGISLG--GPMTNSLLITAWPHEDTVYTSLRFATGYAMPDVYEGDAEIT : ATEFIGYLECSSAKNGANSGWCGVSLR--GAMTNNLLITAWPSDGEVYTNLMFATGYAMPKNYAGDAKIT ATEFIGFLECQ-----GAGWCSVSLG-GSMLNXPLVVAYPSGDEVLASLKWATGYANPEPYGGNHKLS QISSSINATHFKLIFRCONCLOWTHDGASGGASTSAGVLVLGWVQAFPSPGNPTCPDQITLEQHNNGMGI QIASSVSGDSFEVLFRCENCFSWDQNGATGSVSTSNGALVLGYAASKSGLTGATCPDTAEFGFHNNGFGQ QVSSSVNSTHFSLIFRCKNCLQWSHGGSSGGASTSGCVLVLGWVQAFDDPGNPTCPEQITLQQHDNGMGI QIASSVNATHFTLVFRCQNCLSWDQDGVTGGISTSNKGAQLGWVQAFPSPGNPTCPTQITLSQHDNGMGQ QISSSVTSAGFRVVYRCEGCLAWNYQGIEGGSPTNGASMPIGWAYSASSVLNGDCVDNTVLIQHDT-FGN WGAVMDSNVANPSYTEWAAQATKTVEAECDG--PSETDIVGVPVPTGT-----TFDYIVVGGGAGGI QISSTIDKDHFTLIFRCONCLAWNQDGASGSASTSAGSLILGWASALRAPTNAGCPAEINFNNGGMI WGAVLEG-ATSDSYEEWAQLATITPPTTCDG--NGPGDKVCVPAPEDT------YDYIVVGAGAGGI WGATLDESAANPSYSEWAAKATATVTGDCGGATPTTTTTTTTSVPTATGIPVPT-GTYDYIVVGAGAGGI WGAQLNTDAASPSYTDWAAQATKTVTGDCEG--PTETSVVGVPVPTGV-----SFDY1VVGGGAGGI WGAAFDSNIANPSYTAWAAKATKTVTGTCSG--PVTTSIAATPVPTGV-----SFDYIVVGGGAGGI YGFVPDESSLRTEYNDWTELPTRVVRGDCGGSTTTSSVPSSTAPPQGTGIPVPTGASYDYIVVGSGAGGI 9 × 20 ٠. C.attro C.therm C.therm C.attro C.therm M.therm N.crass S.bisby C.attro C.therm H.haema M.therm N.crass S.bisby M.therm S.bisby H.haema M.therm H.haema H.haema S.bisby

Fig. 3 Continuation

8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	405 401 415 403 405	4775 4785 473 473 473	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
C.attro: PTADKLSEAGKSVLLIEKGIASTAEHGGTLGPEWLEGNDLTRFDVFGLCNQIWVDSKGIACEDTDQMAGC: C.therm: TVADKLSEAGKSVLLIEKGIASTAEHGGTLGPEWLEGNDLTRFDVFGLCNQIWVDSKGIACEDTDQMAGC: H.haema: PLADKLSEAGKSVLLIEKGPPSSGRWGGTLKPEWLKDTNLTRFDVFGLCNQIWVNSAGVACTDTDQMAGC: M.therm: PAADKLSEAGKSVLLIEKGFASTANTGGTLGPEWLEGHDLTRFDVFGLCNQIWVDSKGIACEDTDQMAGC: N.crass: PVADKLSESGKSVLLIEKGFASTGEHGGTLKPEWLNNTSLTRFDVFGLCNQIWKDSDGIACSDTDQMAGC: S.bisby: PIADKLTEAGKKVLLIEKGPPSSGRYDGKLKPTWLEGTNLTRFDVPGLCNQIWVDSAGIACRDTDQMAGC:	360 * 420 C.attro : VLGGGTAVNAGLWFKPYSLDWDYLFPSGWKYRDIQAAIGRVFSRIPGTDAPSTDGKRYYQQGFDVLAGGL : C.therm : VLGGGTAVNAGLWWKPHPADWDDNFPHGWKSSDLADATERVFSRIPGTWHPSQDGKLYRQEGFEVISQGL : H.haema : VLGGGTAVNAGLWWKPYRLDWDYNFPRGWKSRDMAAATRRVFSRIPGTDNPSMDGKRYLQQGFEILAGGL : M.therm : VLGGGTAVNAGLWFKPYSLDWDYLFPDGWKYNDVQPAINRALSRIPGTTPPSQDGKRYYQEGFEVLSKGL : N.crass : VLGGGTAVNAGLWYKPYTKDWDYLFPSGWKSSEMIGATNRVFSRIGGTTVPSQDGKRYYQQGFEVLANGL : S.bisby : VLGGGTAVNAGLWWKPNPIDWDYNFPSGWKSSEMIGATNRVFSRIGGTTVPSQDGKTYYQQGFNVLSSGL :	* 440 * 480 * * 480 * C.attro : SAGGWNKVTANSSPDKKNRTESNAPFMFSGGERGGPLATYLISAKRRSNFNLWLNTSVRRVIREGGHVTG : C.therm : ANAGWREVDANQEPSEKNRTYSHSVFMFSGGERGGPLATYLASAAQRSNFNLWVNTSVRRAIRTGPRVSG : H.haema : KAAGWTEVTANDAPNKKNHTYSHSPFMFSGGERGGPMGTYLVSASRRKNFHLWTGTAVKRVVRTGGHITG : M.therm : AAGGWTSVTANNAPDKKNRTFAHAPFMFAGGERNGPLGTYFQTAKRNNFDVWLNTSVRRVIREGGHITG : N.crass : KASGWKEVDSLKDSEQKNRTFSHTSYMYINGERGGPLATYLVSAKKRSNFKLWLNTAVKRVIREGGHITG : S.bisby : KAAGWTSVSLNNAPAQRNRTYGAGPFMFSGGERGGPLATYLATAKRGNFDLWTNTQVKRVIRQGGHVTG :	C.attro : VEVEPFRIGGYQGIVNVTAVSGRVVLSAGTFGSAKILLRGGIGPADQLEVVKASKIDGPTMISNASWIPL : C.therm : VELECLADGGFNGTVNLKEGGG-VIFSAGAFGSAKILLRSGIGPEDQLEIVASS-KDGETFISKNDWIKL : H.haema : LEVEPFVNGGYTGVVNVTSITGRVVLSAGAFGSAKILLRSGIGPEDQLEIVKSS-TDGPTMISDSSWITL : M.therm : VEVEPFRDGGYEGIVPVTKVTGRVILSAGTFGSAKILLRSGIGPEDQLEVVAASEKDGPTMIGNSSWINL : N.crass : VEVEAFRNGGYSGIIPVTNTTGRVVLSAGTFGSAKILLRSGIGPKDQLEVVKAS-ADGPTMVSNSSWIDL : S.bisby : VEVENYNGDGYKGTVKVTPVSGRVVLSAGTFGSAKILLRSGIGPKDQLAIVKNS-TDGPTMASERDWINL :
			OOHZZS

Fig. 3 Continuation

615 609 623 613 614 612	684 679 682 683 683	753 747 761 750 751	823 831 820 821 821
: PVGYNLDDHLNTDTVITHPDVAFYDFYEAWNTPIEADKNSYLSSRTGILAQAAPNIGPMMWEEIKGADGI : PVGHNLIDHLNTDLIITHPDVVFYDFYAAWDNPITEDKEAXINSRSGILAQAAPNIGPLMWEEVTPSDGI : PVGYNLEDHTNTDTVVTHPDVVFYDFYEAG-HPNVTDKDLYINSRAGILAQAAPNIGPMFWEEIKGRDGV : PVGYNLDDHLNTDTVISHPDVVFYDFYEAWDDPIESDKNSYLESRTGILAQAAPNIGPMFWEEIVGADGI : PVGHNLVDHTNTDTVIQHNNVTFYDFYKAWDNPNTTDMNLXLNGRSGIFAQAAPNIGPLFWEEITGADGI : PVGYNLEDHTNTDIVISHPDVVHYDFYEAWTASIESDKTAYLGKRSGILAQAAPNIGPLFFDEVRGADNI	640 * 680 * 700  : VRQLQWTARVEG-SFDTPNGQAMTISQYLGRGATSRGRMTITPSLTTVVSDVPYLKDPNDKEAVIQGIVN  : TRQFQWTCRVEGDSSKTNSTHAMTLSQYLGRGVVSRGRMGITSGLTTTVAEHPYLHNDGDLEAVIQGIQN  : VRQLQWTARVEG-SAGTPNGYAMTMSQYLGRGAKSRGRMTITFALTTVVSTVPYLQDRNDVEAVIQGIKN  : VRQLQWTARVEG-SIGAPNGHTMTMSQYLGRGATSRGRMTITPSLTTIVSDVPYLKDPNDKEAVIQGIIN  : VRQLHWTARVEG-SFETPDGYAMTMSQYLGRGATSRGRMTLSPTLNTVVSDLPYLKDPNDKAAVVQGIVN  : VRSIQYTARVEG-NSVVPNGKAMVISQYLGRGAVSRGRMTISQGLNTIVSTAPYLSNVNDLEAVIKSLEN	* 760 * 760 * 760 * 760 * 740 * 740 * 760	TBULFVVDASIFPGTPTTNPSAYIVTAAEHASQRILGLAAPKPVGKWGQCGGRQWTGSFQCVSGTKCEVV  TDNLFVVDASIFPGMSTGNPSAMIVIVAEQAAQRILSLRY  TDNLFVVDAGIFPGHITTNPTSYIVIAAERASERILDLPPARAQPRFAQCGGRTWTGSFQCAAPYTCQYR  TDNLFVIDASIFPGVPTTNPTSYIVVAAEHASSRILALPDLEPVPKYGQCGGREWTGSFVCADGSTCEYQ  TDNLFVVDASIFPGWVTTNPTAYIVVAAEHAAAKILAQPANEAVPKWGWCGGPTYTGSQTCQAPYKCEKQ  TDNLFVVDASIFPGWVTTNPSAYIVIAAEHAASKILSLPTAKAAAKYEQCGGLEYNGNFQCASGLTCTWL
ro ma ma cm cm	ro na na cm cm cm	ro ra ra ra ra ra ra ra ra ra ra ra ra ra	
C.attro C.therm H.haema M.therm N.crass S.bisby	C.attro C.therm H.haema M.therm N.crass	C.attro C.therm H.haema M.therm N.crass S.bisby	C.attro C.therm H.haema M.therm N.crass

# Fig. 3 Continuation

831	1	839	828	829	829
• •	• •	• •	••	• •	••
NEWYSQCL		NERYSQCR	NEWYSOCL	NDWYWQCV	NDYYWQCT
••	• •	••	• •	• •	••
C.attro	C.therm	H.haema	M.therm	N.crass	S.bisby

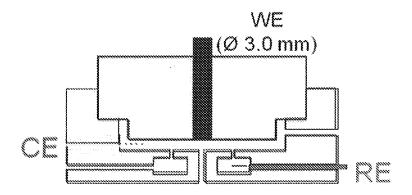


Fig. 4

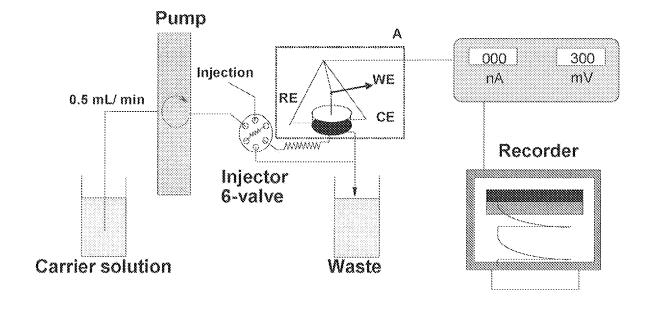


Fig. 5



# **EUROPEAN SEARCH REPORT**

Application Number EP 09 15 3889

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X : part Y : part docu A : tech O : non	ATEGORY OF CITED DOCUMENTS icularly relevant if taken alone icularly relevant if combined with anot iment of the same category inclogical background written disclosure rmediate document	L : document cited fo	ument, but publise the application rother reasons	hed on, or



# **EUROPEAN SEARCH REPORT**

Application Number EP 09 15 3889

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