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(54)A method for making a protective coating on a metal substrate

(57)The method for making a protective coating on a metal substrate including nickel atoms comprises the steps of: electro-chemical deposition of an aluminum layer on the substrate using a bath consisting of a ionic liquid comprising a chloroaluminate anion and an organic cation, and vacuum heating of said substrate on which

the aluminum layer has been deposited, such that nickel atoms migrate from the substrate to the aluminum layer, with the formation of a nickel-aluminum alloy-based protective coating.

EP 2 330 233 A1

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Description

[0001] The present invention relates to a method for making a protective coating on a metal substrate including nickel atoms.

1

[0002] This metal substrate is particularly a turbine blade for use in the aeronautic field or in gas plants for producing electric power. At the high operating temperatures of the turbines of this kind, said coating acts as a barrier against substrate oxidation, as well as a "bond coat" for any possible subsequent coating of one or more further protective layers.

[0003] Said protective layer is known to be obtained by means of rather complicated and expensive methods, particularly the "chemical vapor deposition" (CVD).

[0004] The object of the present invention is to provide a method as stated in the preamble of the present description, which results in reduced energy consumption and is environment-friendly.

[0005] According to the invention, this object is achieved by means of a method comprising the steps of

- electro-chemical deposition of an aluminum layer on said substrate using a bath consisting of a ionic liquid comprising a chloroaluminate anion and an organic cation, and
- vacuum heating of said substrate on which the aluminum layer has been deposited, such that nickel atoms migrate from said substrate to the aluminum layer, with the formation of a nickel-aluminum alloybased protective coating.

[0006] The inventive method has the advantages of not requiring high temperatures or the use of dangerous gases and expensive plants, for being implemented.

[0007] The turbines coated by means of the inventive method has an improved energy performance. This improvement - though poor in absolute terms - is nevertheless very significant in view of the long operating life of a gas turbine. For example, an increase as low as 1 % in the performance of a 50 MW turbine may result in about 1000 tons of gas saved per year, which corresponds to about 200.000 Euro saved per year at present prices. The decrease of fuel consumption further results in the additional advantage of reducing the emission of undesired substances, such as nitrogen and carbon dioxide. [0008] The metal substrate being used preferably contains at least 10% by weight of nickel and can be particularly a nickel-based super-alloy, for example one of those that are commercially known as Hastelloy, Inconel, Waspaloy, Rene (e.g. Rene 41, Rene 80, Rene 95, Rene 104), Haynes, Incoloy, MP98T, TMS and the like.

[0009] The ionic liquid is for example a chloroaluminate of imidazole, pyridinium or ammonium.

[0010] Advantageously, the electrochemical deposition step is carried out at a temperature ranging between 20 and 50 °C using a current density ranging between 0,5 and 2,5 A/dm², whereas the vacuum heating step is

carried out at a temperature ranging between 900 and 1150 °C and at a pressure ranging between 1 * 10-5 and 133 * 10-5 Pa.

[0011] Typically, the aluminum layer deposited on the substrate can have a thickness ranging between 10 and 100 μ m.

[0012] An exemplary embodiment of the method according to the invention will be now provided by way of non-limiting illustration.

EXAMPLE

[0013] A substrate consisting of a nickel alloy made of Ni 72.0%, Cr 15.5%, Fe 8.0%, Si 0.5%, Mn 1.0%, C 0.15%, Cu 0.5%, S < 0.02% is dipped in a bath of 1-butyl, 3-methyl-imidazole hepta-chloroaluminate, made of AIC1 $_3$ and 1-butyl,3-methyl-imidazole chloride at 1:2 molar ratio. In the bath, in which an anode made of an AI plate with a purity of more than 99% is provided, an electric current is passed with a density of 1 A/dm 2 . This electrochemical treatment is carried out for 2 hours at ambient temperature thereby causing the formation of a 25 μ mthick aluminum coating layer on the substrate.

[0014] Subsequently, the coated substrate is removed from the bath and kept for 2 hours at a temperature of 1120 °C and a pressure of < 133*10⁻⁵ Pa. Thereby, a coating layer made of the A1₃Ni and A1Ni compounds is formed by interdiffusion.

[0015] Obviously, the principle of the invention being understood, the implementation details and the embodiments thereof may be widely changed relative to what has been described herein by way of example, without however departing from the scope of the invention as defined in the annexed claims.

Claims

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- A method for making a protective coating on a metal substrate including nickel atoms, comprising the steps of
 - electro-chemical deposition of an aluminum layer on said substrate using a ionic liquid bath comprising a chloroaluminate anion and an organic cation, and
 - vacuum heating of said substrate on which the aluminum layer has been deposited, such that nickel atoms migrate from said substrate to the aluminum layer, with the formation of a nickel-aluminum alloy-based protective coating.
- The method according to claim 1, wherein said step of electrochemical deposition is carried out at a temperature ranging between 20 and 50 °C.
- The method according to any preceding claim, wherein said step of electrochemical deposition pro-

2

vides for the use of a current density ranging between 0.5 and 2.5 A/dm 2 .

- **4.** The method according to any preceding claim, wherein said step of vacuum heating is carried out at a temperature ranging between 900 and 1150 °C and at a pressure ranging between 1 * 10⁻⁵ and 133 * 10⁻⁵ Pa.
- **5.** The method according to any preceding claim, wherein said metal substrate contains at least 10% nickel by weight.
- **6.** The method according to any preceding claim, wherein said metal substrate is a nickel-based super-alloy.
- 7. The method according to any preceding claim, wherein said ionic liquid is a chloroaluminate of imidazole, pyridinium or ammonium.
- 8. The method according to any preceding claim, wherein said aluminum layer has a thickness ranging between 10 and 100 μm .



EUROPEAN SEARCH REPORT

Application Number EP 09 42 5494

Category	Citation of document with ir of relevant passa	ndication, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (IPC)
Υ	EP 1 995 344 A1 (IN 26 November 2008 (2 * claims 1-3,6,11-1 * page 3, paragraph * page 4, paragraph * page 5, paragraph * page 6, paragraph	NCOA GMBH [DE]) 008-11-26) 6 * 12-13 * 133 * 15 42-47,56-57 *	1-8	INV. C25D3/66 C25D5/48 F02C7/00
Y	I.USOV, P.ARENDT, L H.WANG, S.FOLTYN AN "Characteristics of barrier films on Ha J.MATER. RES., vol. 19, no. 4, Apr pages 1175-1180, XP * page 1176, left-h * page 1180, left-h	D P.DOWDEN: alumina diffusion stelloy" il 2004 (2004-04), 2002577502 and column *	1-8	
Y	"Application of ion electrodeposition of PHYS. CHEM. CHEM. P 28 July 2006 (2006-4265-4279, XP002577 10.1039/b607329h * page 4265, left-h * * page 4265, right-paragraph - page 42 * page 4276, right-	of metals" PHYS, no. 8, 07-28), pages 503 DOI: Tand column, paragraph 2 Thand column, last 166, left-hand column *	1-8	TECHNICAL FIELDS SEARCHED (IPC) C25D F02C
	The present search report has l	peen drawn up for all claims		
	Place of search	Date of completion of the search		Examiner
	The Hague	21 April 2010	Gaı	ılt, Nathalie
X : parti Y : parti docu A : tech	ATEGORY OF CITED DOCUMENTS icularly relevant if taken alone icularly relevant if combined with another including the same category nological background written disclosure	T : theory or principle E : earlier patent doc after the filing dat D : document cited in L : document cited fo	sument, but publice e n the application or other reasons	shed on, or



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Application Number

EP 09 42 5494

	DOCUMENTS CONSIDERED				
Category	Citation of document with indication of relevant passages	n, where appropriate,	Relevant to claim	CLASSIFICATION OF THE APPLICATION (IPC)	
Υ	T.TSUDA, C.L.HUSSEY AND "Progress in Surface File Acidic Room-Temperature Ionic liquids" MEET.ABSTR. ELECTROCHEM 2037, 2006, XP002577504 ISSN: 1091-8213 * the whole document *	nishing with Lewis Chloroaluminate . SOC., no. 602,	1-8	TECHNICAL FIELDS SEARCHED (IPC)	
	The present search report has been dr	•			
Place of search		Date of completion of the search		Examiner	
The Hague		21 April 2010	Gau	lt, Nathalie	
CATEGORY OF CITED DOCUMENTS X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure		E : earlier patent doc after the filing dat D : document cited in L : document cited fo	T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons &: member of the same patent family, corresponding		

ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 09 42 5494

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

21-04-2010

cit	Patent document ed in search report		Publication date	Patent family member(s)	Publication date
EP	1995344	A1	26-11-2008	NONE	•
				pean Patent Office, No. 12/82	