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(54) HEAT TRANSFER MATERIALS AND METHODS OF MAKING AND USING THE SAME

WÄRMETRANSFERMATERIALIEN UND VERFAHREN ZU IHRER HERSTELLUNG UND VERWENDUNG

MATÉRIAUX DE TRANSFERT THERMIQUE ET PROCÉDÉS DE FABRICATION ET D'UTILISATION DESDITS MATÉRIAUX

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Description

Background

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[0001] In recent years, a significant industry has developed which involves the application of customer-selected designs, messages, illustrations, and the like (referred to collectively hereinafter as "images") to substrates through the use of heat transfer papers. The images are transferred from the heat transfer paper to the substrate through the application of heat and pressure, after which the release or transfer paper is removed. Typically, a heat transfer material includes a cellulosic base sheet and an image-receptive coating on a surface of the base sheet. The image-receptive coating usually contains one or more thermoplastic polymeric binders, as well as, other additives to improve the transferability and printability of the coating. For example US 2009/280250 discloses a heat transfer material comprising an image-receptive coating which comprises thermoplastic polyolefin wax microparticles and a thermoplastic binder.

[0002] The quality of the image formed on the image-receptive coating on the heat transfer material directly correlates to the quality of the image formed on the final substrate (e.g., an article of clothing). Digital electrographic toner printing (often referred to as laser printing) is a well-known method of printing high quality images onto a paper sheet. Another type of digital toner printing is called digital offset printing.

[0003] When utilizing a toner ink printing process, the printable surface (e.g., an image-receptive coating of a heat transfer sheet) is specially designed to fuse with the toner ink at the printing temperatures (e.g., typically from about 50 °C to about 120 °C but sometimes may reach as high as about 200 °C). This printable surface is designed to attract and adhere the toner ink from the printer. However, due to this affinity for the toner ink, the printable surface often picks up unwanted, stray toner ink from the printer. This stray toner ink can blur the image and provide unwanted background "noise" on the printable surface. When utilized with a heat transfer paper, any stray toner ink on the heat transfer paper will be transferred to the substrate.

[0004] As such, a need exists for a heat transfer paper which improves the quality of an image printed onto the image-receptive coating of a heat transfer paper.

Summary

[0005] The present invention is directed to, in one embodiment, a method of making a heat transfer material. According to the method, a splittable layer is formed to overlie a base sheet. An image-receptive coating is formed to overlie the splittable layer. The image-receptive coating includes thermoplastic polystyrene microparticles, a thermoplastic binder, and a humectant. The thermoplastic polystyrene microparticles have an average particle size of from about 5 μ m (microns) to about 80 μ m (microns) and melt at temperatures between about 90°C and about 115°C.

In an alternative embodiment thermoplastic polyester microparticles having an average particle size of from $5\mu m$ to 80 μm and melting at temperatures between $90^{\circ}C$ and $115^{\circ}C$ can be substituted for the polystyrene microparticles for use in the image receptive coating.

[0006] A second thermoplastic microparticle (e.g., thermoplastic polyamide microparticles) can also be included in the image-receptive coating. Alternatively, a combination of thermoplastic polyester microparticles and thermoplastic polyamide microparticles can be included in the image-receptive coating. The heat transfer material is then dried. The humectant is configured to draw moisture back into the heat transfer sheet after drying.

[0007] The present invention is also generally directed to, in another embodiment, a heat transfer material configured for hot peel heat transfer of an image to a substrate. Additionally, the present invention is directed to a method of transferring an image to a substrate using the heat transfer material presently described.

[0008] Other features and aspects of the present invention are discussed in greater detail below.

Brief Description of the Drawings

[0009] A full and enabling disclosure of the present invention, including the best mode thereof, directed to one of ordinary skill in the art, is set forth more particularly in the remainder of the specification, which makes reference to the appended figures in which:

Figure 1 shows a cross-sectional view of an exemplary heat transfer sheet made in accordance with the present invention; and

Figures 2-4 sequentially show an exemplary method of transferring an image to a substrate using the heat transfer sheet of Fig. 1.

[0010] Repeat use of reference characters in the present specification and drawings is intended to represent same or analogous features or elements of the invention.

Definitions

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[0011] As used herein, the term "printable" is meant to include enabling the placement of an image on a material by any means, such as by direct and offset gravure printers, silk-screening, typewriters, laser printers, laser copiers, other toner-based printers and copiers, dot-matrix printers, and ink jet printers, by way of illustration. Moreover, the image composition may be any of the inks or other compositions typically used in printing processes.

[0012] The term "toner ink" is used herein to describe an ink adapted to be fused to the printable substrate with heat.
[0013] The term "molecular weight" generally refers to a weight-average molecular weight unless another meaning is clear from the context or the term does not refer to a polymer. It long has been understood and accepted that the unit for molecular weight is the atomic mass unit, sometimes referred to as the "dalton." Consequently, units rarely are given in current literature. In keeping with that practice, therefore, no units are expressed herein for molecular weights.

[0014] As used herein, the term "cellulosic nonwoven web" is meant to include any web or sheet-like material which contains at least about 50 percent by weight of cellulosic fibers. In addition to cellulosic fibers, the web may contain other natural fibers, synthetic fibers, or mixtures thereof. Cellulosic nonwoven webs may be prepared by air laying or wet laying relatively short fibers to form a web or sheet. Thus, the term includes nonwoven webs prepared from a papermaking furnish. Such furnish may include only cellulose fibers or a mixture of cellulose fibers with other natural fibers and/or synthetic fibers. The furnish also may contain additives and other materials, such as fillers, e.g., clay and titanium dioxide, surfactants, antifoaming agents, and the like, as is well known in the papermaking art.

[0015] As used herein, the term "polymer" generally includes, but is not limited to, homopolymers; copolymers, such as, for example, block, graft, random and alternating copolymers; and terpolymers; and blends and modifications thereof. Furthermore, unless otherwise specifically limited, the term "polymer" shall include all possible geometrical configurations of the material. These configurations include, but are not limited to isotactic, syndiotactic, and random symmetries.

Detailed Description

[0016] Reference will now be made in detail to embodiments of the invention, one or more examples of which are provided herein. Each example is provided by way of explanation of the invention and not meant as a limitation of the invention. For example, features illustrated or described as part of one embodiment may be utilized with another embodiment to yield still a further embodiment. It is intended that the present invention include such modifications and variations as come within the scope of the appended claims.

[0017] Generally speaking, the present invention is directed to a heat transfer paper configured to reduce the amount of stray toner on the image-receptive coating, especially when the image is formed via a laser printer or laser copier. Although the composition of the toner ink can vary (e.g., according to its color, the printing process utilized, etc.), the toner ink generally adheres to the image-receptive coating at the elevated printing temperatures. These toner printing processes result in the toner ink fusing to the image-receptive coating, which can increase the durability of the transferred image on the substrate. Additionally, the heat transfer paper can provide superior color quality to transferred images as well as wash durability in that image.

[0018] In order to produce an image on a substrate, a toner ink is first applied (e.g., printed) onto an image-receptive coating of a heat transfer sheet to form an image. The image printed onto the image-receptive coating is a mirror image of the image to be transferred to the final substrate. One of ordinary skill in the art would be able to produce and print such a mirror image, using any one of many commercially available software picture/design programs. Due to the vast availability of these printing processes, nearly every consumer easily can produce his or her own image to make a coated image on a substrate. Essentially, any design, character, shape, or other image that the user can print onto the image-receptive layer coating can be transferred to the substrate. The image formed on the image-receptive coating of the heat transfer sheet can be either a "positive" or "negative" image. A "positive" image is an image that is defined by the ink applied to the image-receptive coating. On the other hand, a "negative" image is an image that is defined by the area of the image-receptive coating that is free of ink.

[0019] Referring to Fig. 1, an exemplary heat transfer sheet 10 is shown having a toner ink 12 applied to its image-receptive coating 14. In Fig. 1, an image is positively defined by the toner ink 12 on the image-receptive coating 14, with the remainder of the surface area of the image-receptive coating 14 being substantially free of toner ink 12. As stated, the image defined by toner ink 12 is a mirror image of the desired coated image to be applied to the final substrate.

[0020] The image-receptive coating 14 overlies a splittable layer 16 and a base sheet 18. In the exemplary embodiment shown, the image-receptive coating 14 is adjacent to and directly overlies the splittable layer 16, without any intermediate layers. In turn, the splittable layer 16 is adjacent to and directly overlies the base sheet 18, also without any intermediate layers. However, in other embodiments, intermediate layers may be positioned between the image-receptive coating 14, the splittable layer 16, and/or the base sheet 18. For example, a conformable layer may be positioned between the base sheet 18 and splittable layer 16 to facilitate the contact between the heat transfer sheet 10 and the substrate 20 to which the image is to be transferred. An example of a suitable conformable layer is disclosed in U.S. Pat. No. 4,863,781

to Kronzer.

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[0021] The toner ink 12 is, in one particular embodiment, printed on the image-receptive coating 14 via the use of a laser printer or laser copier. These printing processes typically operate at temperatures ranging from about 50° C to about 120° C, but may sometimes be as high as 200°C, to ensure that the toner ink 12 melts and adheres to the surface to which it is printed. The image-receptive coating 14 resists melting at the printing temperatures to inhibit damage to the coating and to resist leaving residual coating material on the printer/copier machinery.

[0022] After the toner ink 12 has been printed onto the image-receptive coating 14, the heat transfer sheet 10 is positioned adjacent to a substrate 20. The heat transfer sheet 10 is positioned such that the image-receptive coating 14 and the toner ink 12 are adjacent to the substrate 20, as shown in Fig. 2. The substrate 20 can be any surface to which the image is to be transferred. The substrate can be a fabric cloth, nonwoven web, film, or any other surface. Desirable substrates include, for example, fabrics such as 100% cotton T-shirt material, and so forth.

[0023] Heat (H) and pressure (P) are then applied to the exposed base sheet 18 of the heat transfer sheet 10 adjacent to the substrate 20. The heat (H) and pressure (P) can be applied to the heat transfer sheet 10 via a heat press, an iron (e.g., a conventional hand iron), etc. The heat (H) and pressure (P) can be applied to the heat transfer sheet 10 for a time sufficient to cause the image-receptive coating 14 and the splittable layer 16 to soften and melt. Temperatures at the transfer can be from about 150° C or greater, such as from about 150° C to about 250° C, and can be applied for a period of a few seconds to a few minutes (e.g., from about 5 seconds to about 5 minutes).

[0024] At the transfer temperature, both the image-receptive coating 14 and the splittable layer 16 soften and melt. The image-receptive coating 14 softens and flows directly onto or into the substrate 20. Once the heat (H) and pressure (P) are removed from the heat transfer sheet 10, the base sheet 18 is removed before the heat transfer sheet 10 can substantially cool (i.e., while the heat transfer sheet 10 is still hot). Removing the base sheet occurs by separating the splittable layer 16. A first portion (16A) of the splittable layer 16 remains on the base sheet 18 and is removed from the substrate 20, while a second portion (16B) of the splittable layer 16 is transferred to the substrate 20 along with the image-receptive coating 14. This process is an example of a hot peelable transfer process. As used herein, the phrase "hot peelable transfer process" refers to a process wherein one or more meltable layers is still in a molten state when a non-transferable portion of a heat transfer sheet is removed. Such a process allows release of the heat transfer sheet via splitting of the meltable layer(s).

[0025] Thus, as discussed above, the image-receptive coating 14 of the present invention does not appreciably melt and/or soften at the printing temperatures in the laser printer and/or copier. However, the image-receptive coating 14 does melt and soften at the transfer temperatures during the heat transfer of the image to the substrate 20.

I. Image-Receptive Coating

[0026] The image-receptive coating 14 is configured to melt and conform to the surface of the substrate 20 to which the image is applied. In addition, the image-receptive coating 14 provides a print surface for the heat transfer sheet 10 and is formulated to minimize feathering of the printed image and bleeding or loss of the image when the transferred image is exposed to water.

[0027] According to one embodiment of the present invention, thermoplastic polystyrene microparticles having a narrow melting range are present in the image-receptive coating 14. The thermoplastic polystyrene microparticles provide a porous structure to the image-receptive coating 14 enabling better absorption of the toner ink 12 to the image-receptive coating 14. Additionally, the image-receptive coating 14 is constructed to reduce or eliminate the attraction of stray toner ink to the heat transfer sheet 10.

[0028] Polystyrenes are polymers that can acquire a negative charge during the printing process. Typically, when utilizing a laser printer/copier to apply a toner ink to a printable surface, a static charge is created on the printable surface through contact with the various rollers utilized in the laser printer/copier. While at the printing temperature, the toner ink is attracted to and adheres to this charged surface. The printing surface and the toner ink then cool off quickly, drying the toner ink in place on the printable surface. Without wishing to be bound by theory, the present inventor believes that the thermoplastic polystyrene microparticles can quickly dissipate any static charge that is built up in the image-receptive coating 14. The loss of this static charge inhibits the image-receptive coating 14 from attracting any stray toner ink from the laser printer/copier, which would otherwise be attracted to a charged image-receptive coating 14.

[0029] It is believed that this ability to dissipate the charge created during the printing process can be attributed to the nature of the polystyrenes to acquire a negative static charge by attracting electrons when contact other materials. For example, according to the Triboelectric Series, which is a list of materials showing which have a greater tendency to become positive (give away electrons) and which have a greater tendency to become negative (acquire electrons), polystyrene tends to attract electrons. Triboelectricity is the physics of charge generated through friction. The triboelectric series is a list that ranks various materials according to their tendency to gain or lose electrons. It usually lists materials in order of decreasing tendency to charge positively (lose electrons), and increasing tendency to charge negatively (gain electrons). Somewhere in the middle of the list are materials that do not show strong tendency to behave either way.

Note that the tendency of a material to become positive or negative after triboelectric charging has nothing to do with the level of conductivity (or ability to discharge) of the material. Due to complexities involved in experiments that involve controlled charging of materials, different researchers sometimes get different results in determining the rank of a material in the triboelectric series. One of the reasons for this is the multitude of factors and conditions that affect a material's tendency to charge. However, the listing shown in Table 1, is a commonly used Triboelectric Series (shown from the most positive to neutral to the most negative).

TABLE 1: Triboelectric Series

	OUDEACE MATERIAL	
10	SURFACE MATERIAL	CHARGE
10	Human skin	Large Positive
	Leather	
	Rabbit's fur	
	Acetate	
15	Glass	
	Quartz	
	Mica	
	Human hair	
00	Polyamide	
20	Wool	
	Lead	
	Silk	
	Aluminum	
25	Paper	Small Positive
	Cotton	None
	Steel	None
	Wood	Small Negative
	Lucite	
30	Amber	
	Sealing wax	
	Acrylic	
	Polystyrene	
35	Rubber balloon	
	Hard rubber	
	Nickel, Copper	
	Sulfur	
	Brass, Silver	
40	Gold, Platinum	
	Acetate, Rayon	
	Synthetic rubber	
	Polyester	
45	Styrene (Styrofoam)	
	Orlon	
	Polyvinylidene chloride	
	Polyurethane	
	Polyethylene	
50	Polypropylene	
	Vinyl (PVC)	
	Silicon	
	Teflon	
55	Silicone rubber	
55	Ebonite	Large Negative
		5 0

[0030] Polystyrene is an aromatic polymer made from the aromatic monomer styrene. Pure polystyrene is generally a long chain hydrocarbon with every other carbon connected to a phenyl group "Isotactic polystyrene" generally refers to an isomer of polystyrene where all of the phenyl groups are on the same side of the hydrocarbon chain. Metallocenecatalyzed polymerization of styrene can produce an ordered "syndiotactic polystyrene" with the phenyl groups on alternating sides. This syndiotactic polystyrene is highly crystalline with a melting point of about 270 °C.

[0031] "Atactic polystyrene" generally refers to an isomer of polystyrene where the phenyl groups are randomly distributed on both sides of the hydrocarbon chain. This random positioning prevents the polymeric chains from ever aligning with sufficient regularity to achieve any significant crystallinity. As such, atactic polystyrene has no true melting point and generally melts over a relatively large temperature range, such as between about 90°C and about 115°C. This relatively large melting temperature range allows the thermoplastic polystyrene microparticles to resist melting and flowing at the temperatures briefly encountered during printing by the laser printer/copier, but sufficiently melt at the transfer temperature encountered during heat transfer of the image to the substrate. According to the present invention the thermoplastic polystyrene microparticles melt at a temperature range between about 90°C and about 115°C. In one particular embodiment, the thermoplastic polystyrene microparticles melt at a temperature range between about 95°C and about 105°C.

[0032] The melting point of the thermoplastic polystyrene microparticles can be influenced by the molecular weight of the thermoplastic polystyrene microparticles, although the melting point can be influenced by other factors. In one embodiment, the weight average molecular weight (M_w) of the thermoplastic polystyrene polymer in the microparticles can be from about 10,000 g/mol to about 15,000 g/mol and the number average molecular weight (determined by measuring the molecular weight of n polymer molecules, summing the weights, and dividing by n) can be from about 2,500 to about 10,000.

[0033] The present inventor has found that control of the particle size of the thermoplastic polystyrene microparticles is particularly important in controlling the affinity of the image-receptive coating 14 to unwanted stray toner ink. According to the present invention the thermoplastic polystyrene microparticles have an average particle size (diameter) of about 5 micrometers (microns) to about 80 μ m (microns), such as from about 15 μ m (microns) to about 50 μ m (microns). For example, the thermoplastic polystyrene microparticles can be polystyrene particles having an average diameter of about 20 μ m (microns) (e.g., a diameter range of about 18 μ m (microns) to about 22 μ m (microns)) and an average molecular weight of 12,000 g/mol, such as the polystyrene particles available under the trade name DYNOSEED TS-20 (Microbeads AS, Skedsmokorset, Norway). Another example of suitable thermoplastic polystyrene microparticles can be polystyrene particles having an average diameter of about 40 μ m (microns) (e.g., a diameter range of about 38 μ m (microns) to about 42 μ m (microns)) and an average molecular weight of 15,500 g/mol, such as the polystyrene particles available under the trade name DYNOSEED TS-40 (Microbeads AS, Skedsmokorset, Norway).

based on the dry weight of the image-receptive coating 14, such as from about 25% to about 85%. In one particular embodiment, the thermoplastic polystyrene microparticles can be present in the image-receptive coating 14 from about 30% to about 80% based on the dry weight of the image-receptive coating 14, such as from about 35% to about 80%. [0035] In one embodiment, another type of thermoplastic polymer microparticles can be included in the image-receptive coating 14 along with the thermoplastic polystyrene microparticles. Like the thermoplastic polystyrene microparticles, the second thermoplastic polymer microparticles can provide a porous structure to the image-receptive coating 14 enabling better absorption of the toner ink 12 into the image-receptive coating 14. The second type of thermoplastic polymer microparticles can also add gloss, abrasion resistance, and/or another quality to the image-receptive coating 14 transferred to the heat transfer sheet 10. The second thermoplastic polymer microparticles can be present in an amount of from about 10% to about 75% based on the dry weight of the image-receptive coating 14, such as from about 25% to about 50%. In one particular embodiment, the thermoplastic polystyrene microparticles can be present in the image-receptive coating 14 from about 30% to about 45% based on the dry weight of the image-receptive coating 14, such as from about 35% to about 40%. The second thermoplastic polymer microparticles can be present in a dry weight percentage that is substantially equal to the thermoplastic polystyrene microparticles.

[0036] The second thermoplastic polymer microparticles may be polyamide, polyester, polyolefin, ethylene-vinyl acetate copolymer, or mixtures thereof, and can have an average particle size ranging from about 2 to about 50 μm (microns), such as from about 5 to about 20 μm (microns). In one particular embodiment, the second thermoplastic polymer microparticles are polyamide microparticles. Suitable polyamide microparticles are those 6/12 copolyamide particles (believed to be a copolymer of a 6C diamine and a 12C diacid, sometimes referred to as a 6/12 nylon) available commercially under the trade name Orgasol® 3501 EXD (Atofina Chemicals, Inc., Philadelphia, I.), which have an average particle size (measured as the diameter) of 10 μm (microns) with a variation of about +/- 3 and Orgasol® 3502 EXD (Atofina Chemicals, Inc., Philadelphia, I.), which have an average particle size (measured as the diameter) of 20 μm (microns) with a variation of about +/- 3. Other microparticles suitable as the second thermoplastic polymer microparticles are commercially available under the trade name PropylTex 200S (Micro Powders, Inc., Tarrytown, NY), which are believed to be polypropylene particles having an average diameter of about 35 μm (microns) to about 45 μm (microns)

and a maximum particle size of 74 μ m (microns).

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[0037] In an alternative embodiment, thermoplastic polyester microparticles can be substituted for the polystyrene microparticles, for use in the image-receptive coating 14 either alone or in combination with thermoplastic polyamide microparticles, such as those described above. According to the present invention the thermoplastic polyester microparticles have an average particle size of from about 5 μ m (microns) to about 80 μ m (microns) and melt at temperatures between about 90°C and about 115°C.

[0038] Additionally, the image-receptive coating 14 includes a thermoplastic binder. The thermoplastic binder can act as an anchor to hold the thermoplastic polystyrene microparticles in the image-receptive coating 14. Thus, the thermoplastic binder can provide cohesion and mechanical integrity to the image-receptive coating 14. In general, any thermoplastic binder may be employed which meets the criteria specified herein. Suitable thermoplastic thermoplastic binders include, but are not limited to, polyamides, polyolefins, polyesters, polyurethanes, poly(vinyl chloride), poly(vinyl acetate), polyethylene oxide, polyacrylates, polystyrene, polyacrylic acid, and polymethacrylic acid. Copolymers and mixtures thereof also can be used. As a practical matter, water-dispersible ethylene-acrylic acid copolymers have been found to be particularly effective thermoplastic binders. The thermoplastic binder can be present from about 5% to about 40% based on the dry weight of the image-receptive coating 14, such as from about 10% to about 30%.

[0039] In one particular embodiment, the thermoplastic binder can be "polar" in nature. Differences in polarity between two substances (such as a polymer and a solvent) are directly responsible for the different degrees of-intermolecular stickiness from one substance to another. For instance, substances that have similar polarities will generally be soluble or miscible in each other but increasing deviations in polarity will make solubility increasingly difficult. Without wishing to be bound by theory, it is believed that if the binder used in the image-receptive coating 14 is more polar, the toner ink 12 can adhere better and with more durability to the thermoplastic binder having some degree of polarity. As such, the image-receptive coating may lose less of the toners after several wash and dry cycles than similar coatings made with non-polar binders.

[0040] In general, any polar thermoplastic binder can be utilized in accordance with the present invention. In one embodiment, polymers containing carboxy groups can be utilized. The presence of carboxy groups can readily increase the polarity of a polymer because of the dipole created by the oxygen atom. For example, in some embodiments, carboxylated (carboxy-containing) polyacrylates can be used as the acrylic latex binder. Also, other carboxy-containing polymers can be used, including carboxylated nitrile-butadiene copolymers, carboxylated styrenebutadiene copolymers, carboxylated ethylene-vinylacetate copolymers, and carboxylated polyurethanes. Also, in some embodiments, a combination of polar thermoplastic binders can be utilized within the transfer coating.

[0041] In one embodiment, the polar thermoplastic binder can be an acrylic latex binder. Suitable polyacrylic latex binders can include polymethacrylates, poly(acrylic acid), poly(methacrylic acid), and copolymers of the various acrylate and methacrylate esters and the free acids; ethylene-acrylate copolymers; vinyl acetate-acrylate copolymers, and the like. Suitable acrylic latex polymers that can be utilized as the thermoplastic binder include those acrylic latexes sold under the trade name HYCAR® by Noveon, Inc. of Cleveland, Ohio, such as HYCAR® 26684 and HYCAR® 26084.

[0042] The image-receptive coating 14 also includes a humectant configured to draw moisture back into the image-receptive coating 14 after drying. The moisture can help preserve the image-receptive coating 14 (along with the heat transfer sheet 10) during production and storage. However, due to the strict melting characteristic demands of the image-receptive coating 14, the humectant does not melt at the printing temperature, so as to avoid any processing problems during the printing process. Thus, the humectant has a melting point of greater than about 120°C.

[0043] The image-receptive coating 14 can, in one particular embodiment, include urea (also known as diaminometh-anal) as the humectant. Urea has a melting point of 132.7°C, which is generally above the temperatures associated with the printing process. Urea decomposes upon heating at temperatures higher than 132.7°C. Thus, at the transfer temperature, the urea can decompose and form by-products, such as ammonia, oxides of nitrogen, and carbon dioxide. This decomposition of urea at the transfer temperature acts to remove the urea from the transferred image-receptive coating 14. This result is particularly useful since the humectant serves no purpose after the image-receptive coating 14 is transferred to the substrate 20 and the base sheet 18 is removed.

[0044] A second humectant can also be present in the image-receptive coating 14 to facilitate the return of moisture into the image-receptive coating 14 after drying.

[0045] In one particular embodiment, the second humectant can be a hydrophilic polymer, such as polyethylene glycol or polypropylene glycol. However, polyethylene glycol melts at temperatures encountered during the printing process. The amount of this hydrophilic polymer (e.g., polyethylene glycol) included within the image-receptive coating 14 is therefore limited. If too much of this meltable hydrophilic polymer is included in the image-receptive coating 14, then the image-receptive coating 14 can stick to the fuser section of some laser printer/copier machines. For example, the hydrophilic polymer can be included in an amount of less than about 3% by weight based on the dry weight of the image-receptive coating 14, such as from about 0.01 % to about 2%.

[0046] This hydrophilic polymer, particularly polyethylene glycol, can double as a plasticizer when included in the image-receptive coating 14. One suitable polyethylene glycol that can be included in the image-receptive coating 14 as

the second humectant, and as a plasticizer, is available under the name Carbowax E-300 from Dow Chemical Company, Midland, Mich.

[0047] Processing aids can also be included in the image-receptive coating 14, including, but not limited to, thickeners (e.g., sodium polyacrylate such as Paragum 231 from Para-Chem Southern, Inc., Simpsonville, South Carolina), dispersants, viscosity modifiers, etc. Surfactants can also be present in the image-receptive coating 14. In one embodiment, the surfactant can be a non-ionic surfactant, such as the non-ionic surfactant available under the trade name Triton X100 (Dow Chemical Company, Midland, Mich.).

[0048] Additionally, pigments and other coloring agents may be present in the image-receptive coating 14. For decoration of dark fabrics, the image-receptive coating 14 may further include an opacifier with a particle size and density well suited for light scattering (e.g., aluminum oxide particles, titanium oxide particles, and the like). However, when it is desired to have a relatively clear or transparent coating, the image-receptive coating 14 can be substantially free from pigments, opacifying agents, and other coloring agents (e.g., free from metal particles, metalized particles, clay particles, etc.).

[0049] In one embodiment, the image-receptive coating 14 does not contain a cross-linking agent or other catalyst that would promote crosslinking in the image-receptive coating 14, especially between the polymeric materials in the coating (i.e., the thermoplastic polystyrene microparticles, the thermoplastic binder, the second thermoplastic microparticles, etc.). In this regard, the melt properties of the image-receptive coating 14 can remain substantially unchanged through the various heating and cooling processes to which it is subjected (e.g., the printing process and the image transfer process). Thus, the polymeric material of the image-receptive coating 14 can be substantially cross-link free. For example, the polystyrene is not, in one particular embodiment, a copolymer containing divinylbenzene for cross-linking the polystyrene chains. The polymeric material can, for example, have less than about 10% of its polymeric chains crosslinked to each other through inter-polymer chain covalent bonding, such as less than about 5%, or less than about 2%. In this embodiment, the thermoplastic binder can include only non-crosslinking polymeric materials (e.g., a non-crosslinking acrylic).

[0050] The image-receptive coating 14 can have a thickness of from about 20,3 μ m-76,2 μ m (0,8-3 mils) to ensure that the image-receptive coating 14 provides a sufficient coating on the heat transfer sheet 10 and subsequently to the substrate 20, while a coating thickness of from 25,4 μ m-63,5 μ m (about 1.0 to about 2.5 mils) is desired. However, if the image-receptive coating 14 is too thick or stiff, it will impart too much stiffness to the substrate 20 after it is transferred. [0051] The image-receptive coating 14 may be formed on the heat transfer sheet 10 by known coating techniques, such as by roll, blade, Meyer rod, and air-knife coating procedures. The resulting heat transfer material then may be dried by means of, for example, steam-heated drums, air impingement, radiant heating, or some combination thereof.

II. Splittable Layer

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[0052] The splittable layer 16 of the heat transfer material 10 is configured to allow the base sheet 18 to be removed (e.g., peeled away) from the substrate 20 while still hot (i.e., a hot peel) after the application of heat (H) and pressure (P) in the transfer process. The splittable layer 16 generally softens and melts at temperatures lower than those causing the image-receptive coating 14 to melt. For example, the splittable layer 16 can melt at temperatures of from about 80°C to about 130°C. The polymer can have, in one embodiment, a melt index, as determined in accordance with ASTM Test Method D-1238-82, of at least about 25 g/10 minutes. However, since the splittable layer 16 is concealed within the construction of the heat transfer material 10 by the base sheet 18 and the image-receptive coating 14, the splittable layer 16 is protected from melting during the printing process. Additionally, the period which the heat transfer material 10 is exposed to higher temperatures during the printing process, as explained above, is generally too short to cause the splittable layer 16 to melt.

[0053] The splittable layer 16 can be constructed of any polymeric material that meets the criteria above. Polymeric materials suitable for forming the splittable layer 16 include, but are not limited to, copolymers of ethylene and acrylic acid, methacrylic acid, vinyl acetate, ethyl acetate, or butyl acrylate. Other polymers that may be employed include polyesters, polyamides, and polyurethanes. Waxes, plasticizers, rheology modifiers, antioxidants, antistats, antiblocking agents, release agents, and other additives may be included as either desired or necessary. In one particular embodiment, the polymeric material includes a combination of ethylene-methacrylic acid copolymer (EMAA) and ethylene-acrylic acid copolymer (EAA).

[0054] In one embodiment, the splittable layer 16 is an extruded film layer. For example, the splittable layer 16 may be applied to the base sheet 18 with an extrusion coater that extrudes molten polymer through a screw into a slot die. The film exits the slot die and flows by gravity onto the base sheet 18. The resulting coated material is passed through a nip to chill the extruded film and bond it to the underlying base sheet 18. For less viscous polymers, the molten polymer may not form a self-supporting film. In these cases, the material to be coated may be directed into contact with the slot die or by using rolls to transfer the molten polymer from a bath to the heat transfer material.

III. Base Sheet

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[0055] The heat transfer material 10 of the present invention includes base sheet 18 that acts as a backing or support layer for the heat transfer sheet 10. The base sheet 18 is flexible and has first and second surfaces, and is typically a film or a cellulosic nonwoven web. In addition to flexibility, the base sheet 18 also provides strength for handling, coating, sheeting, other operations associated with the manufacture thereof, and for removal after transfer of the image-receptive coating 14 to a substrate 20. The basis weight of the base sheet 18 generally may vary, such as from about 30 to about 150 g/m². Suitable base sheets 18 include, but are not limited to, cellulosic nonwoven webs and polymeric films. A number of suitable base sheets 18 are disclosed in U.S. Pat. Nos. 5,242,739; 5,501,902; and U.S. Pat. No. 5,798,179. [0056] Desirably, the base sheet 18 comprises paper. A number of different types of paper are suitable for the present invention including, but not limited to, common litho label paper, bond paper, and latex saturated papers. In some embodiments, the base sheet 18 will be a latex-impregnated paper such as described, for example, in U.S. Pat. No. 5,798,179. The base sheet 18 is readily prepared by methods that are well known to those having ordinary skill in the art. [0057] Although the description above is directed to a hot peel heat transfer material, the heat transfer material of the present invention could be utilized in a cold peel material. In this embodiment, a release coating layer (not shown) is present on the surface of the base sheet 18 that contacts the splittable layer 16 (e.g., between the base sheet 18 and the splittable layer 16). The release coating layer separates the transferable material (i.e., the image-receptive coating 14 and the splittable layer 16) of the heat transfer material 10 from the non-transferable material (i.e., the base sheet 18). The release coating layer does not transfer to a coated substrate. Consequently, the release coating layer may comprise any material having release characteristics, which is also conformable when heated. Desirably, the release coating layer does not melt or become tacky when heated, and provides release of an image bearing coating during a hot or cold peelable transfer process.

[0058] A number of release coating layers are known to those of ordinary skill in the art, any of which may be used in the present invention. Typically, the release coating layer comprises a cross-linked polymer having essentially no tack at transfer temperatures (e.g. 177° C.) and a glass transition temperature of at least about 0° C. As used herein, the phrase "having essentially no tack at transfer temperatures" means that the release coating layer does not stick to an overlaying layer to an extent sufficient to adversely affect the quality of the transferred image. Suitable polymers include, but are not limited to, silicone-containing polymers, acrylic polymers and poly(vinyl acetate). Further, other materials having a low surface energy, such as polysiloxanes and fluorocarbon polymers, may be used in the release coating layer, particularly in cold peel applications. Desirably, the release coating layer comprises a cross-linked silicone-containing polymer or a cross-linked acrylic polymer. Suitable silicone-containing polymers include, but are not limited to, SYL-OFF® 7362, a silicone-containing polymer available from Dow Corning Corporation (Midland, Mich.). Suitable acrylic polymers include, but are not limited to, HYCAR® 26672, an acrylic latex available from B.F. Goodrich, Cleveland, Ohio; MICHEM® Prime 4983, an ethylene-acrylic acid copolymer dispersion available from Michelman Chemical Company, Cincinnati, Ohio; HYCAR® 26684, an acrylic latex also available from B.F. Goodrich, Cleveland, Ohio; and RHO-PLEX® SP 100, an acrylic latex available from Rohm & Haas, Philadelphia, Pa.

[0059] The release coating layer may further contain additives including, but not limited to, a cross-linking agent, a release-modifying additive, a curing agent, a surfactant and a viscosity-modifying agent. Suitable cross-linking agents include, but are not limited to, XAMA 7, an aziridine cross-linker available from B.F. Goodrich. Suitable release-modifying additives include, but are not limited to, SYL-OFF® 7210, a release modifier available from Dow Corning Corporation. Suitable curing agents include, but are not limited to, SYL-OFF® 7367, a curing agent available from Dow Coming Corporation. Suitable surfactants include, but are not limited to, TERGITOL® 15-S40, available from Union Carbide; TRITON® X100, available from Union Carbide; and Silicone Surfactant 190, available from Dow Coming Corporation. In addition to acting as a surfactant, Silicone Surfactant 190 also functions as a release modifier, providing improved release characteristics, particularly in cold peel applications.

[0060] The release coating layer may have a layer thickness, which varies considerably depending upon a number of factors including, but not limited to, the substrate to be coated, the thickness of the splittable layer 16, the press temperature, and the press time. Desirably, the release coating layer has a thickness, which does not restrict the flow of the splittable layer 16 and the image-receptive coating 14. Typically, the release coating layer has a thickness of less than about 26 μ m (microns) (1 mil). More desirably, the release coating layer has a thickness of from 1,27 μ m (about 0.05 mil. to about 0.5 mil). Even more desirably, the release coating layer has a thickness of from 2,03 μ m-8,38 μ m (about 0.08 mil. to about 0.33 mil).

[0061] The thickness of the release coating layer may also be described in term of a coating weight. Desirably, the release coating layer has a dry coating weight of less than 22.5 gsm (about 6 lb./144 yd 2). More desirably, the release coating layer has a dry coating weight of from 11.3 gsm (about 3.0 lb./144 yd 2) to 1.1 gsm (about 0.3 lb./144 yd 2). Even more desirably, the release coating layer has a dry coating weight of from 7.5 gsm (about 2.0 lb./144 yd 2) to 1.9 gsm (about 0.5 lb./144 yd 2).

[0062] The present invention may be better understood with reference to the examples that follow. Such examples,

however, are not to be construed as limiting the present invention which is defined by the appended claims. In the examples, all parts are parts by weight unless stated otherwise.

Examples

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[0063] The following materials were used in these Examples:

Hycar 26684 (Noveon, Inc., Cleveland, Ohio) is an acrylic latex polymer;

Triton X-100 (Dow Chemical Company, Midland, Mich.) is a surfactant;

Urea:

Carbowax E-300 (Dow Chemical Company, Midland, Mich.) is a polypropylene glycol having an average molecular weight of 300;

Paragum 231 (Para-Chem Southern, Inc., Simpsonville, S.C.) is sodium polyacrylate useful as a thickener.

Example 1:

[0064] A base paper (10,88 kg (24 lb.) super smooth base paper available under the trade name Classic Crest® from Neenah Paper, Inc., Alpharetta, GA) was first coated with an acrylic splitting layer by extruded 33,02 μ m (1.3 mils) EMAA (ethylene-methacrylic acid) and 12,7 μ m (0.5 mils) of EAA (ethylene-acrylic acid) onto the base paper. Then, an image-receptive coating was applied to the splitting layer. The image-receptive coating was applied in an amount of 2.5 pounds per ream (144 yards²), which is about 9.4 gsm, using a Myer rod. The coating was applied as an aqueous dispersion/mixture and then dried to remove the water.

[0065] The following dispersion:

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	%	Dry Parts	% Dry Weight
Water			
Triton X-100	33	5	4.8
Dynoseeds TS-20	100	100	95.2

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was used to make the image-receptive coating according to the formula:

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	%	Dry Parts	% dry wt.
Water			
Particle Dispersion	25	105	77.9
Hycar 26684	48.9	23	17.1
Carbowax E-300	100	1.75	1.3
Urea	22	3.5	2.6
Paragum 231	13.8	1.5	1.1

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[0066] The resulting coated sheets were printed using four different color laser printers (Brother HL-4040CN, Minolta 2300, Okidata C5150, Hewlett Packard 3600) with each yielding a clean print.

Example 2:

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[0067] Different image-receptive coatings were prepared and then applied to the splitting layer of a base paper as described above in Example 1. The compositions of each image-receptive coating tested were essentially consistent, except for the type of particles included in the coatings (except where noted). Table 2 shows the types of particles used in each sample image receptive coating.

5		polyester 0-75 μm micron	Griltex 6E																						100%	%09	25%
10		polyester 0-35 µm micron	Griltex 6E																								
15 20		polystyrene 80 µm micron	Dynoseed TS-80														10%	25%									
25		polystyrene 40 μm micron	Dynoseed TS-40										25%	%09	%52	100%			25%	%09							
30	Table 2	polystyrene 20 µm micron	Dynoseed TS-20	25%	%09	75%	100%														20%		100%				
35 40		polystyrene 10 µm micron	Dynoseed TS-10					72%	%09	%52	100%	100%															
45		polyamide 20 ⊾m micron	Orgasol 3502																75%	%09	20%			100%			
50		polyamide 10 μm micron	Orgasol 3501	75%	%09	25%		%52	%09	25%			%52	%09	25%		%06	%52				100%				%09	75%
55			SAMPLE	4	В	Э	O	Ш	Ь	9	т	-	¥	Τ	M	Z	0	Ь	Ø	Я	S	-	Π	^	M	×	>

5	polyester 0-75 µm micron Griltex 6E	10%				
10	polyester 0-35 µm micron Griltex 6E		100%	%09	25%	10%
15	polystyrene 80 µm micron Dynoseed TS-80					
20						
25	polystyrene 40 µm micron Dynoseed TS-40					
30 (continued)	polystyrene 20 μm micron Dynoseed TS-20					
35	polystyrene 10 µm micron Dynoseed TS-10					
40	poly; µn Dyno;					
45	polyamide 20 µm micron Orgasol 3502					
50	polyamide 10 μm micron Orgasol 3501	%06		%09	%52	%06
55	SAMPLE	Z	AA	BB	၁၁	QQ

[0068] The particles were included in the coating as a dispersion, created by mixing the particles with water and a surfactant (Triton X-100 available from Dow Chemical Company, Midland, Mich.), as shown above in Example 1 (i.e., 5 dry parts Triton X-100 to 100 dry parts particles). In addition to the particle dispersions, each coating contained an acrylic latex polymer (Hycar 26684 available from Noveon, Inc., Cleveland, Ohio), a propylene glycol having an average molecular weight of 300 (Carbowax E-300 available from Dow Chemical Company, Midland, Mich.), sodium polyacrylate useful as a thickener (Paragum 231 available from Para-Chem Southern, Inc., Simpsonville, S.C.), and urea as shown above in Example 1 (except where noted).

[0069] In the samples shown in Table 2, Sample U is identical to Sample D except that Sample U did not include Carbowax E300, resulting in the peel force for Sample U being siightly higher.

[0070] After printing, the printed sheets were used to transfer an image to a cloth (Hanes ® Beefy-T 100% cotton t-shirt). Results are shown in Table 3. All heat transfers in these examples were hot peel transfers as described above. Printing was performed using the Okidata C5150 laser printer.

[0071] The Sheffield smoothness of the coated sheet increases in value as the roughness increases.

[0072] Wash Color refers to how well the transfer on fabric retained color following 5 wash cycles. The wash color was rated on a scale of 1 - 4, with 4 being the best and 1 the worst.

[0073] Hunter L refers to a color meter machine test that assigns a value on the level of whiteness of the transfer. To that end, an area of each printed image was purposely left blank so that it could be used for doing a Hunter test. In theory, the more scattered toner attracted to the sheet during printing, the less white the final transfer will be - resulting in a less clean transfer. The higher the Hunter L value, the cleaner the transfer. Table 3 has a column for how the transfer looks (after it is applied to the cloth) and another column on the table for how the printed sheet looks before transfer. For the heat transfer, how the transfer on the fabric looks is more important since this is the end product. The peel force was measured on a scale of 1-5 as perceived by the end user. Color densisty was determined using an X-Rite Specrodensitometer and color 100% cyan color block and reported as Response T (US standard) visual density.

25 Table 3

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			i abie 5				
SAMPLE	Transfer Hunter L	Perceived Peel Force	Sheffield Smoothness	Print Hunter L	Transfer Color Visual Den ^T	Wash Color	Wash Color Visual Den ^T
Α	89	2	60	93	0.90	4	0.87
В	91	3	100	94	0.91	3	0.85
С	92	2	120 - 130	94	0.95	2	0.82
D	94	2	175	94	0.91	2	0.81
Е	88	4	35 - 40	92	1.01	4	0.93
F	89	3	40 - 45	92	1.00	4	0.91
G	91	3	60 - 75	93	0.97	3	0.86
Н	93	2	125 - 135	94	0.96	3	0.84
I	92	2	72 - 75	93	0.95	2	0.82
K	94	2	290 - 320	96	0.90	3	0.84
L	95	3	370	96	0.86	3	0.84
М	95	4	380 - 400	96	0.86	2	0.81
N	94	5	380 - 400	95	0.80	1	0.75
0	95	5	400 +	96	0.77	1	0.74
Р	95	5	400 +	96	0.78	1	0.66
Q	94	5	350	95	0.95	4	0.87
R	95	3	380	95	0.88	3	0.85
S	91	3	135 - 140	93	0.96	4	0.88
Т	90	4	30	93	0.93	4	0.88

(continued)

SAMPLE	Transfer Hunter L	Perceived Peel Force	Sheffield Smoothness	Print Hunter L	Transfer Color Visual Den ^T	Wash Color	Wash Color Visual Den ^T
U	94	3	175	94	0.96	2	0.82
V	90	5	120 - 130	93	0.96	4	0.90
W	95	3	400 +	94	0.82	1	0.79
Х	94	3	360	95	0.86	3	0.85
Y	94	3	250 - 270	95	0.86	3	0.86
Z	91	4	80 - 110	93	0.87	3	0.85
AA	94	3	330	94	0.85	2	0.83
ВВ	92	3	150 - 160	94	0.89	3	0.86
CC	91	3	85 - 95	94	0.87	3	0.86
DD	90	4	50 - 65	93	0.91	3	0.85

[0074] These and other modifications and variations to the present invention may be practiced by those of ordinary skill in the art, without departing from the scope of the present invention, which is defined in the appended claims. In addition, it should be understood the aspects of the various embodiments may be interchanged both in whole or in part. Furthermore, those of ordinary skill in the art will appreciate that the foregoing description is by way of example only, and is not intended to limit the invention so further described in the appended claims.

Claims

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- 1. A method of making a heat transfer material, the method comprising:
 - forming a splittable layer overlying a base sheet;
 - forming an image-receptive coating overlying the splittable layer to form the heat transfer material, wherein the image-receptive coating comprises thermoplastic polystyrene microparticles, a thermoplastic binder, and a humectant, wherein the thermoplastic polystyrene microparticles have an average particle size of from 5 μ m (microns) to 80 μ m (microns) and melt at temperatures between 90°C and 115°C; and
 - drying the heat transfer material, wherein the humectant is configured to draw moisture back into the heat transfer sheet after drying.
- 2. The method as in claim 1, wherein the thermoplastic polystyrene microparticles melt at temperatures between 95°C and 105 °C.
- 3. The method as in any preceding claim, wherein the thermoplastic polystyrene microparticles have an average particle size of from 38 μm (microns) to 42 μm (microns) or from 18 μm (microns) to 22 μm (microns).
 - **4.** The method as in any preceding claim, wherein the image-receptive coating further comprises a plurality of second thermoplastic polymer microparticles having an average particle size of from 2 μm (microns) to 50 μm (microns).
 - 5. The method as in claim 4, wherein the image-receptive coating comprises the thermoplastic polystyrene microparticles in an amount from 10% to 75% by weight based on the dry weight of the image-receptive coating, and wherein the image-receptive coating comprises the second thermoplastic polymer microparticles in an amount from 10% to 75% by weight based on the dry weight of the image-receptive coating.
- The method as in claim 4 or 5, wherein the second thermoplastic polymer microparticles comprise polyamide microparticles.

- 7. The method as in any preceding claim, wherein the splittable layer directly overlies the base sheet, and wherein the image-receptive coating directly overlies the splittable layer.
- 8. The method as in any preceding claim, wherein the splittable layer is melt extruded directly onto the base sheet, wherein the splittable layer comprises a polymeric material that melts at temperatures between 80 °C and 130 °C.
 - **9.** A method of making a heat transfer material, the method comprising:

forming a splittable layer overlying a base sheet;

forming an image-receptive coating overlying the splittable layer to form the heat transfer material, wherein the image-receptive coating comprises thermoplastic polyester microparticles, a thermoplastic binder, and a humectant, wherein the thermoplastic polyester microparticles have an average particle size of from 5 µm (microns) to 80 µm (microns) and melt at temperatures between 90°C and 115°C; and

drying the heat transfer material, wherein the humectant is configured to draw moisture back into the heat transfer sheet after drying.

- 10. The method of claim 9, wherein the image-receptive coating further comprises thermoplastic polyamide microparticles, and wherein the thermoplastic polyamide microparticles have an average particle size of from 2 µm (microns) to 50 µm (microns).
- 11. The method as in any preceding claim, wherein the humectant comprises urea.
- 12. The method as in any preceding claim, wherein the image-receptive coating is substantially free from a cross-linking agent.
- 13. The method as in any preceding claim, wherein the image-receptive coating further comprises a hydrophilic polymer present in the image-receptive coating from a positive amount to 3% by weight based on the dry weight of the imagereceptive coating.
- 30 14. A heat transfer material configured for hot peel heat transfer of an image to a substrate, the heat transfer material being formed according to the method of any preceding claim.
 - **15.** A method of transferring an image to a substrate, the method comprising:

printing toner ink onto the image-receptive coating of the heat transfer material of claim 14 to form an image; positioning the heat transfer material adjacent the substrate, wherein the image-receptive coating contacts the substrate;

heating the heat transfer material to a temperature of 150°C to 250°C under a pressure force; and peeling the base sheet from the substrate while the heat transfer material is still warm.

Patentansprüche

1. Verfahren zum Herstellen eines Thermotransfermaterials, wobei das Verfahren umfasst:

Ausbilden einer abspaltbaren Schicht, die über einer Grundbahn liegt;

Ausbilden einer Abbild-aufnehmenden Beschichtung, die über der abspaltbaren Schicht liegt, um das Thermotransfermaterial zu bilden, wobei die Abbild-aufnehmende Beschichtung thermoplastische Polystyrol-Mikroteilchen, ein thermoplastisches Bindemittel und ein Feuchthaltemittel umfasst, wobei die thermoplastischen Polystyrol-Mikroteilchen eine durchschnittliche Teilchengröße von 5 µm (Mikrometern) bis 80 µm (Mikrometern) aufweisen und bei Temperaturen zwischen 90 °C und 115 °C schmelzen; und

Trocknen des Thermotransfermaterials, wobei das Feuchthaltemittel ausgebildet ist, nach dem Trocknen Feuchtigkeit zurück in die Thermotransferbahn zu ziehen.

- 2. Verfahren nach Anspruch 1, wobei die thermoplastischen Polystyrol-Mikroteilchen bei Temperaturen zwischen 95 °C und 105 °C schmelzen.
 - 3. Verfahren nach einem beliebigen vorangehenden Anspruch, wobei die thermoplastischen Polystyrol-Mikroteilchen

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eine durchschnittliche Teilchengröße von 38 μm (Mikrometern) bis 42 μm (Mikrometern) oder von 18 μm (Mikrometern) bis 22 µm (Mikrometern) aufweisen.

- Verfahren nach einem beliebigen vorangehenden Anspruch, wobei die Abbild-aufnehmende Beschichtung weiter 5 eine Vielzahl zweiter thermoplastischer Polymer-Mikroteilchen mit einer durchschnittlichen Teilchengröße von 2 µm (Mikrometern) bis 50 μm (Mikrometern) umfasst.
 - 5. Verfahren nach Anspruch 4, wobei die Abbild-aufnehmende Beschichtung die thermoplastischen Polystyrol-Mikroteilchen in einer Menge von 10 bis 75 Gew.-% auf Grundlage des Trockengewichts der Abbild-aufnehmenden Beschichtung umfasst, und wobei die Abbild-aufnehmende Beschichtung die zweiten thermoplastischen Polymer-Mikroteilchen in einer Menge von 10 bis 75 Gew.-% auf Grundlage des Trockengewichts der Abbild-aufnehmenden Beschichtung umfasst.
 - 6. Verfahren nach Anspruch 4 oder 5, wobei die zweiten thermoplastischen Polymer-Mikroteilchen Polyamid-Mikroteilchen umfassen
 - 7. Verfahren nach einem beliebigen vorangehenden Anspruch, wobei die abspaltbare Schicht direkt über der Grundbahn liegt, und wobei die Abbild-aufnehmende Beschichtung direkt über der abspaltbaren Schicht liegt.
- 20 8. Verfahren nach einem beliebigen vorangehenden Anspruch, wobei die abspaltbare Schicht direkt auf die Grundbahn schmelzextrudiert wird, wobei die abspaltbare Schicht ein Polymermaterial umfasst, das bei Temperaturen zwischen 80 °C und 130 °C schmilzt.
 - 9. Verfahren zum Herstellen eines Thermotransfermaterials, wobei das Verfahren umfasst:

Ausbilden einer abspaltbaren Schicht, die über einer Grundbahn liegt;

Ausbilden einer Abbild-aufnehmenden Beschichtung, die über der abspaltbaren Schicht liegt, um das Thermotransfermaterial zu bilden, wobei die Abbild-aufnehmende Beschichtung thermoplastische Polyester-Mikroteilchen, ein thermoplastisches Bindemittel und ein Feuchthaltemittel umfasst, wobei die thermoplastischen Polyester-Mikroteilchen eine durchschnittliche Teilchengröße von 5 μm (Mikrometern) bis 80 μm (Mikrometern) aufweisen und bei Temperaturen zwischen 90 °C und 115 °C schmelzen; und

Trocknen des Thermotransfermaterials, wobei das Feuchthaltemittel ausgestaltet ist, nach dem Trocknen Feuchtigkeit zurück in die Thermotransferbahn zu ziehen.

- 35 10. Verfahren nach Anspruch 9, wobei die Abbild-aufnehmende Beschichtung weiter thermoplastische Polyamid-Mikroteilchen umfasst, und wobei die thermoplastischen Polyamid-Mikroteilchen eine durchschnittliche Teilchengröße von 2 μm (Mikrometern) bis 50 μm (Mikrometern) aufweisen.
 - 11. Verfahren nach einem beliebigen vorangehenden Anspruch, wobei das Feuchthaltemittel Harnstoff umfasst.
 - 12. Verfahren nach einem beliebigen vorangehenden Anspruch, wobei die Abbild-aufnehmende Beschichtung im Wesentlichen frei von einem Vernetzungsmittel ist.
- 13. Verfahren nach einem beliebigen vorangehenden Anspruch, wobei die Abbild-aufnehmende Beschichtung ein hy-45 drophiles Polymer umfasst, das in der Abbild-aufnehmenden Beschichtung von einer nachweisbaren Menge bis zu 3 Gew.-% auf Grundlage des Trockengewichts der Bild-aufnehmenden Beschichtung vorhanden ist.
 - 14. Thermotransfermaterial, ausgestaltet für Heißablösungs-Thermotransfer eines Abbilds auf ein Substrat, wobei das Thermotransfermaterial gemäß dem Verfahren nach einem beliebigen vorangehenden Anspruch ausgebildet ist.
 - 15. Verfahren zum Übertragen eines Bildes auf ein Substrat, wobei das Verfahren umfasst:

Drucken von Tonertinte auf die Abbild-aufnehmenden Beschichtung des Thermotransfermaterials nach Anspruch 14, um ein Abbild zu bilden;

Positionieren des Thermotransfermaterials benachbart zum Substrat, wobei die Abbild-aufnehmende Beschichtung das Substrat berührt;

Erwärmen des Thermotransfermaterials auf eine Temperatur von 150 °C bis 250 °C unter einer Andruckkraft; und Ablösen der Grundbahn von dem Substrat, während das Thermotransfermaterial noch warm ist.

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Revendications

- 1. Méthode pour fabriquer un matériau de transfert de chaleur, la méthode comprenant les étapes suivantes :
 - former une couche divisible susjacente à une feuille de base ;

former un revêtement réceptif d'image susjacent à la couche divisible pour former le matériau de transfert de chaleur, dans laquelle le revêtement récepteur d'image comprend des microparticules de polystyrène thermoplastique, un liant thermoplastique et un humectant, dans laquelle les microparticules de polystyrène thermoplastique ont une dimension de particule moyenne de 5 μ m (microns) à 80 μ m (microns) et fondent à des températures situées entre 90°C et 115°C ; et

sécher le matériau de transfert de chaleur, dans laquelle l'humectant est configuré pour attirer l'humidité dans la feuille de transfert de chaleur après le séchage.

- 2. Méthode selon la revendication 1, dans laquelle les microparticules de polystyrène thermoplastique fondent à des températures situées entre 95°C et 105°C.
- 3. Méthode selon l'une des revendications précédentes, dans laquelle les microparticules de polystyrène thermoplastique ont une dimension de particule moyenne de 38 μ m (microns) à 42 μ m (microns) ou de 18 μ m (microns) à 22 μ m (microns).
- 4. Méthode selon l'une des revendications précédentes, dans laquelle le revêtement réceptif d'image comprend en outre une multitude de secondes microparticules de polymère thermoplastique possédant une dimension de particules moyenne de 2 μm (microns) à 50 μm (microns).
- 5. Méthode selon la revendication 4, dans laquelle le revêtement réceptif d'image contient les microparticules de polystyrène thermoplastique en une quantité de 10% à 75% du poids sur la base du poids à sec du revêtement réceptif d'image, et dans laquelle le revêtement réceptif d'image contient les secondes microparticules de polymère thermoplastique en une quantité de 10% à 75% du poids sur la base du poids à sec du revêtement réceptif d'image.
- **6.** Méthode selon la revendication 4 ou 5, dans laquelle les secondes microparticules de polymère thermoplastique contiennent des microparticules polyamides.
 - 7. Méthode selon l'une des revendications précédentes, dans laquelle la couche divisible est directement susjacente à la feuille de base, et dans laquelle le revêtement réceptif d'image est directement susjacent à la couche divisible.
 - **8.** Méthode selon l'une des revendications précédentes, dans laquelle la couche divisible est appliquée par extrusion de matière fondue directement sur la feuille de base, dans laquelle la couche divisible contient une matière polymérique qui fond à des températures situées entre 80°C et 130°C.
- **9.** Méthode pour fabriquer un matériau de transfert de chaleur, la méthode comprenant les étapes suivantes :

former une couche divisible susjacente à une feuille de base ;

former un revêtement réceptif d'image susjacent à la couche divisible pour former le matériau de transfert de chaleur, dans laquelle le revêtement récepteur d'image comprend des microparticules de polyester thermoplastique, un liant thermoplastique et un humectant, dans laquelle les microparticules de polyester thermoplastique ont une dimension de particule moyenne de 5 μ m (microns) à 80 μ m (microns) et fondent à des températures situées entre 90°C et 115°C; et

sécher le matériau de transfert de chaleur, dans laquelle l'humectant est configuré pour attirer l'humidité dans la feuille de transfert de chaleur après le séchage.

- **10.** Méthode selon la revendication 9, dans laquelle le revêtement réceptif d'image comprend en outre des microparticules de polyamide thermoplastique, et dans laquelle les microparticules de polyamide thermoplastique ont une dimension de particule moyenne de 2 μm (microns) à 50 μm (microns).
- 55 **11.** Méthode selon l'une des revendications précédentes, dans laquelle l'humectant contient de l'urée.
 - 12. Méthode selon l'une des revendications précédentes, dans le revêtement réceptif d'image est substantiellement exempt d'agent de réticulation.

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- 13. Méthode selon l'une des revendications précédentes, dans laquelle le revêtement réceptif d'image contient en outre un polymère hydrophile présent dans le revêtement réceptif d'image à partir d'une quantité positive à 3% du poids sur la base du poids à sec du revêtement réceptif d'image.
- 5 14. Matériau de transfert de chaleur configuré pour le transfert de chaleur par exfoliation à chaud d'une image sur un substrat, le matériau de transfert de chaleur étant formé selon la méthode selon l'une des revendications précédentes.
 - 15. Méthode pour transférer une image sur un substrat, la méthode comprenant les étapes suivantes :

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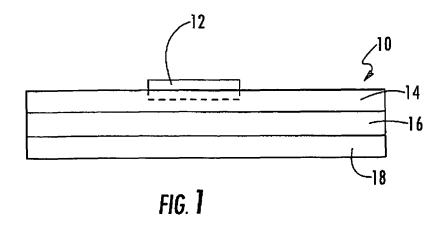
40

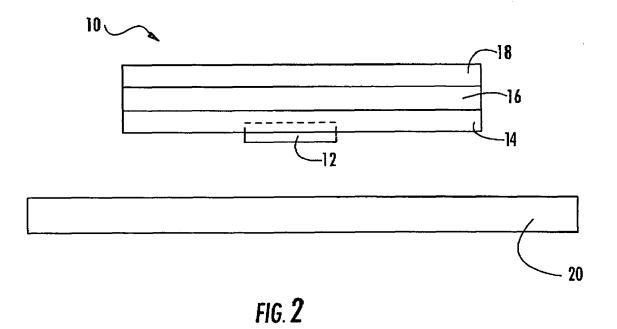
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- 10 imprimer de l'encre sur le revêtement réceptif d'image du matériau de transfert de chaleur selon la revendication 14 pour former une image;
 - positionner le matériau de transfert de chaleur de manière adjacente au substrat, dans laquelle le revêtement réceptif d'image est en contact avec le substrat ;
 - chauffer le matériau de transfert de chaleur à une température de 150°C à 250°C sous une force de pression ; et exfolier la feuille de base depuis le substrat alors que le matériau de transfert de chaleur est encore chaud.





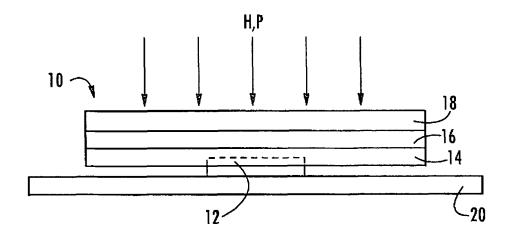
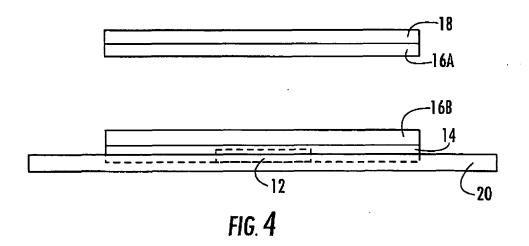


FIG. 3



REFERENCES CITED IN THE DESCRIPTION

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