



(11) **EP 2 752 902 B9**

(12) **CORRECTED EUROPEAN PATENT SPECIFICATION**

(15) Correction information:
Corrected version no 1 (W1 B1)
Corrections, see
Description Paragraph(s) 86, 132, 425

(51) Int Cl.:
H01L 51/50 ^(2006.01) **C07D 405/10** ^(2006.01)
C07D 405/14 ^(2006.01) **C07D 409/10** ^(2006.01)
C07D 409/14 ^(2006.01) **C09K 11/06** ^(2006.01)
H01L 51/00 ^(2006.01) **C09K 11/02** ^(2006.01)

(48) Corrigendum issued on:
30.08.2017 Bulletin 2017/35

(86) International application number:
PCT/JP2012/080156

(45) Date of publication and mention
of the grant of the patent:
10.05.2017 Bulletin 2017/19

(87) International publication number:
WO 2013/077352 (30.05.2013 Gazette 2013/22)

(21) Application number: **12851292.8**

(22) Date of filing: **21.11.2012**

(54) **AROMATIC HETEROCYCLIC DERIVATIVE, MATERIAL FOR ORGANIC ELECTROLUMINESCENT ELEMENT, AND ORGANIC ELECTROLUMINESCENT ELEMENT**

AROMATISCHES HETEROZYKLISCHES DERIVAT, MATERIAL FÜR EIN ORGANISCHES
ELEKTROLUMINESZENZELEMENT UND ORGANISCHES ELEKTROLUMINESZENZELEMENT

DÉRIVÉ HÉTÉROCYCLIQUE AROMATIQUE, MATIÈRE POUR ÉLÉMENT

ÉLECTROLUMINESCENT ORGANIQUE ET ÉLÉMENT ÉLECTROLUMINESCENT ORGANIQUE

(84) Designated Contracting States:
**AL AT BE BG CH CY CZ DE DK EE ES FI FR GB
GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO
PL PT RO RS SE SI SK SM TR**

(30) Priority: **22.11.2011 JP 2011255472**
03.09.2012 JP 2012193349

(43) Date of publication of application:
09.07.2014 Bulletin 2014/28

(60) Divisional application:
17162551.0 / 3 211 682

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Description

FIELD

5 **[0001]** Embodiment(s) described herein relates to an aromatic heterocyclic derivative and an organic-electroluminescence-device material comprising the aromatic heterocyclic derivative.

BACKGROUND

10 **[0002]** An organic electroluminescence device (hereinafter, occasionally referred to as an organic EL device) can be classified by the emission principle into two types: a fluorescent organic EL device and a phosphorescent organic EL device. When a voltage is applied to the organic EL device, holes are injected from an anode and electrons are injected from a cathode. The holes and the electrons are recombined in an emitting layer to form excitons. According to the electron spin statistics theory, singlet excitons and triplet excitons are generated at a ratio of 25% : 75%. In a fluorescent organic EL device which uses emission caused by singlet excitons, the limited value of an internal quantum efficiency is believed to be 25%. A technology for extending a lifetime of a fluorescent organic EL device using a fluorescent material has recently been improved and applied to a full-color display of a mobile phone, TV and the like. However, as compared with a phosphorescent organic EL device, a fluorescent organic EL device is required to be improved in efficiency.

20 **[0003]** In contrast, in relation to a technology of manufacturing a highly efficient fluorescent-organic-EL device, there has been disclosed a technology of extracting emissions derived from triplet excitons by a phenomenon (i.e., TTF (Triplet-Triplet Fusion) phenomenon) in which two triplet excitons collide and fuse with each other to generate singlet excitons. A blocking layer that effectively induces the TTF phenomenon requires to be made of a highly electron-resistant compound serving as a layer for transporting electrons as well as having a wide gap for increasing triplet energy. In view of this point, a compound formed of a hydrocarbon ring has been considered suitable.

25 **[0004]** Patent Literature 1 discloses an organic EL device having a pyrene skeleton or an anthracene skeleton and further a substituent selected from a carbazolyl group, dibenzofuranlyl group or dibenzothiophenyl group in an electron transporting layer adjacent to a fluorescent emitting layer.

30 **[0005]** Patent Literature 2 discloses an organic EL device using a fluoranthene derivative for a blocking layer in order to effectively induce The TTF phenomenon. KR 2012-0122812 and US 2013/256645 have an earlier priority date than the present application, but have been published during the priority year and after the filing date, respectively, and are thus no relevant prior art for the present application.

35 **[0006]** WO 2011/139055, WO 2011/136755, WO 2011/136520, KR 2011-0107680, KR 2011-0107681, WO 2011/071255 and JP 2010-040830 disclose organic heterocyclic compounds for organic optoelectronic and/or electro-luminescent devices. These heterocyclic compounds all comprise a nitrogen-containing heterocyclic fused ring, e.g. a carbazole ring.

[0007] WO 2010/136109 discloses heterocyclic compounds containing a nitrogen-fused heterocyclic ring as well. The compounds of the present invention, to the contrary, do not contain a nitrogen-containing fused ring.

40 **[0008]** WO 2007/069569 discloses a heterocyclic compound in which a substituent position of a fused heterocyclic ring (e.g. benzofurane) is 4-position or 6-position. The compounds of the present invention have different substituent positions.

[0009] WO 2010/044342 discloses small compounds including a nitrogen-containing six-membered ring. The compound either has a nitrogen-containing fused ring or does not include a linking group between the nitrogen-containing six-membered ring and the heterocyclic fused ring.

45 **[0010]** According to the present invention, however, a linking group is required.

CITATION LIST

PATENT LITERATURES

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[0011]

Patent Literature 1: International Publication No. WO2010/001817

Patent Literature 2: International Publication No. WO2010/134350

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SUMMARY OF THE INVENTION

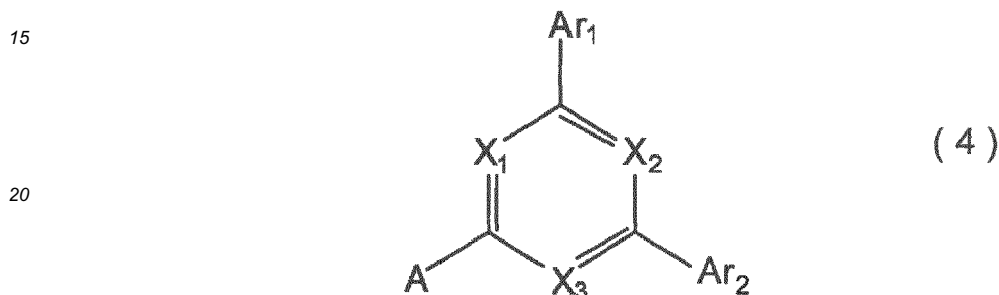
PROBLEMS TO BE SOLVED BY THE INVENTION

5 **[0012]** An object of the invention is to provide an aromatic heterocyclic derivative applicable to an organic EL device, and an organic-EL-device material comprising the aromatic heterocyclic derivative.

MEANS FOR SOLVING THE PROBLEMS

10 **[0013]**

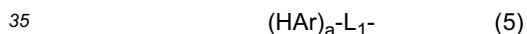
[1] An aromatic heterocyclic derivative according to the invention is represented by a formula (4) below.



25 **[0014]** In the formula (4), X_1 to X_3 are a nitrogen atom or CR_1 , with a proviso that at least one of X_1 to X_3 is a nitrogen atom.

[0015] R_1 independently represents a hydrogen atom, a halogen atom, a cyano group, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, a substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, a substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms.

30 **[0016]** In the formula (4), A is represented by a formula (5) below.



[0017] In the formula (5), HAr is represented by a formula (6) below.

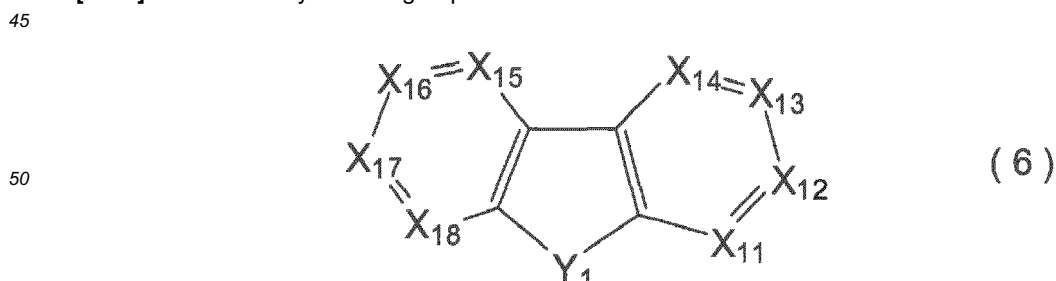
[0018] In the formula (5), a is an integer of 1 to 5.

[0019] When a is 1, L_1 is a divalent linking group.

40 **[0020]** When a is in a range of 2 to 5, L_1 is a trivalent to hexavalent linking group and HAr is the same or different.

[0021] The linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a residue having 2 to 6 valences induced from any one of a group formed by mutually bonding two or three of the substituted or unsubstituted aryl group having 6 to 30 ring carbon.

[0022] The mutually bonded groups are the same or different.



55 **[0023]** In the formula (6), Y_1 is an oxygen atom or a sulfur atom.

[0024] In the formula (6), X_{11} and X_{18} are a nitrogen atom or CR_{13} .

[0025] In the formula (6), one of X_{12} to X_{17} is a carbon atom bonded to L_1 by a single bond and the rest of X_{12} to X_{17} are a nitrogen atom or CR_{13} .

[0026] R_{13} independently represents a hydrogen atom, a halogen atom, a cyano group, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, a substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, a substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms.

A plurality of R_{13} are mutually the same or different. Adjacent R_{13} may bond to each other to form a ring.

[0027] In the above formula (4), Ar_1 and Ar_2 each are independently represented by the formula (5), or represent a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms or a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms.

[0028]

[2] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (6), X_{13} or X_{16} is a carbon atom bonded to L_1 by a single bond.

[3] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (5), a is an integer of 1 to 3.

[4] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (5), a is 1 or 2.

[5] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (5), a is 1 and L_1 is a linking group, and the linking group is a divalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.

[6] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (5), a is 2 and L_1 is a linking group, and the linking group is a trivalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.

[7] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (6), Y_1 is an oxygen atom.

[8] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (6), Y_1 is an oxygen atom, X_{11} and X_{18} are CR_{13} , and one of X_{12} to X_{17} is a carbon atom bonded to L_1 by a single bond and the rest of X_{12} to X_{17} are CR_{13} .

[9] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (4), two or three of X_1 to X_3 are a nitrogen atom.

[10] In the aromatic heterocyclic derivative according to the above aspect of the invention, in the formula (5), L_1 is a divalent or trivalent residue induced from any one of benzene, biphenyl, terphenyl, naphthalene and phenanthrene.

[11] An organic-electroluminescence-device material according to further aspect of the invention comprises the aromatic heterocyclic derivative according to the above aspect of the invention.

[0029] According to the embodiment(s), an aromatic heterocyclic derivative applicable to the organic EL device, and an organic-EL-device material comprising the aromatic heterocyclic derivative can be provided.

BRIEF DESCRIPTION OF DRAWING

[0030]

Fig. 1 is a view showing one example of an organic EL device according to a first exemplary embodiment of the invention.

Fig. 2 is a view showing a relationship of energy gaps between layers of the invention.

Fig. 3 is a view showing an action based on the relationship of the energy gaps between the layers of the invention.

Fig. 4 is an energy band diagram showing a case where an affinity of a host material (A_h) > an affinity of a dopant material (A_d) is satisfied.

Fig. 5 is an energy band diagram showing a case where $A_h < A_d$ is satisfied and a difference between A_h and A_d is less than 0.2 eV.

Fig. 6 is an energy band diagram showing a case where $A_h < A_d$ is satisfied and a difference between A_h and A_d is more than 0.2 eV.

Fig. 7 is an energy band diagram showing a case where a dopant material satisfying $A_h < A_d$ and a dopant material satisfying $A_h > A_d$ coexist.

Fig. 8 is a view showing one example of an organic EL device according to a second exemplary embodiment of the invention.

Fig. 9 is a view showing one example of an organic EL device according to a third exemplary embodiment of the invention.

Fig. 10 is a view showing one example of an organic EL device according to a fourth exemplary embodiment of the invention.

Fig. 11 is a view showing one example of an organic EL device according to a fifth exemplary embodiment of the invention.

5 Fig. 12 is a view showing one example of an organic EL device according to a sixth exemplary embodiment of the invention.

Fig. 13 is a view showing a measurement system of transitional EL waves.

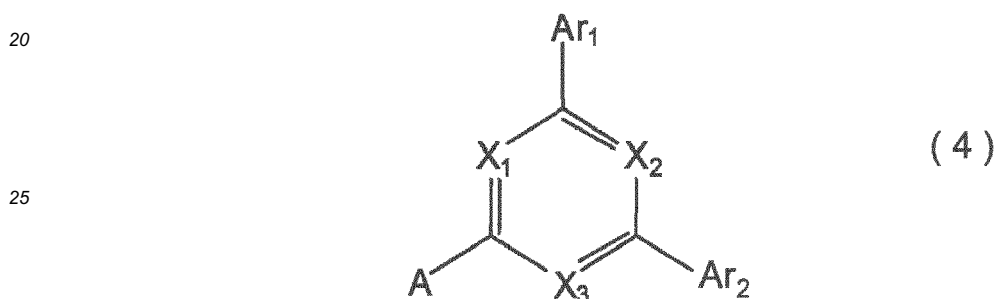
Fig. 14A is a view showing a measurement method of a ratio of luminous intensity derived from TTF and is a graph showing a change over time of luminous intensity of the EL device.

10 Fig. 14B is a view showing a measurement method of a ratio of the luminous intensity derived from TTF and is a graph showing a change over time of a reciprocal square root of the luminous intensity.

DESCRIPTION OF EMBODIMENTS

15 Aromatic Heterocyclic Derivative

[0031] An aromatic heterocyclic derivative according to the invention is represented by a formula (4) below.



30 [0032] In the formula (4), X₁ to X₃ are a nitrogen atom or CR₁.

[0033] However, at least one of X₁ to X₃ is a nitrogen atom.

35 [0034] R₁ independently represents a hydrogen atom, halogen atom, cyano group, substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, or substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms.

40 [0035] In the formula (4), A is represented by a formula (5) below.



[0036] In the formula (5), HAr is represented by a formula (6) below.

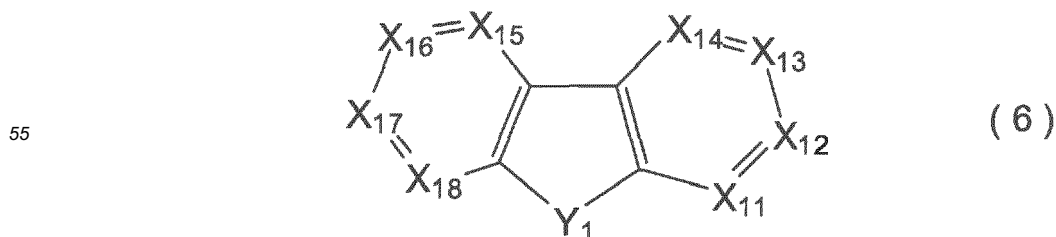
45 [0037] In the formula (5), a is an integer of 1 to 5.

[0038] When a is 1, L₁ is a divalent linking group.

[0039] When a is in a range of 2 to 5, L₁ is a trivalent to hexavalent linking group and HAr is the same or different.

[0040] The linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a divalent to hexavalent residue induced from any one of a group formed by mutually bonding two or three of the above groups.

50 [0041] The mutually bonded groups are the same or different.



[0042] In the formula (6), Y_1 is an oxygen atom or a sulfur atom.

[0043] In the formula (6), X_{11} and X_{18} are a nitrogen atom or CR_{13} .

[0044] In the formula (6), one of X_{12} to X_{17} is a carbon atom bonded to L_1 by a single bond and the rest of X_{12} to X_{17} are a nitrogen atom or CR_{13} .

[0045] R_{13} independently represents a hydrogen atom, halogen atom, cyano group, substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, or substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms. A plurality of R_{13} are mutually the same or different. Adjacent R_{13} may bond to each other to form a ring.

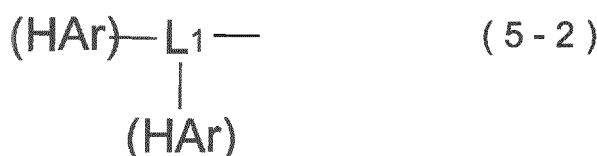
[0046] In the above formula (4), Ar_1 and Ar_2 each are independently represented by the formula (5), or represent a substituted or unsubstituted aryl group having 6 to 30 ring carbon.

[0047] In the formula (6), X_{13} or X_{16} is preferably a carbon atom bonded to L_1 by a single bond.

[0048] In the formula (5), a is an integer in a range of 1 to 5, more preferably of 1 to 3, particularly preferably 1 or 2.

[0049] When a is 1, L_1 is a divalent linking group and the formula (5) is represented by a formula (5-1).

[0050] When a is in a range of 2 to 5, L_1 is a trivalent to hexavalent linking group. When a is 2, L_1 is a trivalent linking group and the formula (5) is represented by a formula (5-2). At this time, HAr is the same or different.



[0051] The linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a divalent or trivalent residue induced from any one of a group formed by bonding two or three of the above groups.

[0052] In L_1 of the formulae (5), (5-1) and (5-2), the group formed by bonding two or three of the above groups means a group formed by bonding, with a single bond, two or three of the divalent or trivalent residue induced from the aryl group having 6 to 30 ring carbon atoms. In the linking group, the mutually bonded groups are the same or different.

[0053] In the formulae (5), (5-1) and (5-2), L_1 is preferably a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.

[0054] Moreover, in the formulae (5), (5-1) and (5-2), L_1 is more preferably a divalent or trivalent residue induced from any one of benzene, biphenyl, terphenyl, naphthalene and phenanthrene.

[0055] In the formula (5), it is more preferable that a is 1 and L_1 is a divalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.

[0056] In the formula (5), it is more preferable that a is 2 and L_1 is a linking group, specifically, a trivalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.

[0057] In the formula (6), X_{13} or X_{16} is preferably a carbon atom bonded to L_1 by a single bond.

[0058] In the formula (6), Y_1 is preferably an oxygen atom.

[0059] Moreover, in the formula (6), it is more preferable that Y_1 is an oxygen atom, X_{11} and X_{18} are CR_{13} , one of X_{12} to X_{17} is a carbon atom bonded to L_1 by a single bond and the rest of X_{12} to X_{17} are CR_{13} .

[0060] In the formula (4), two or three of X_1 to X_3 are preferably a nitrogen atom.

[0061] Ar_1 , Ar_2 , L_1 , R_1 , R_{11} to R_{13} in the formulae (4) to (6) and (5-1) to (5-2) will be described below.

[0062] Examples of the aryl group having 6 to 30 ring carbon atoms in the formulae (4) to (6) and (5-1) to (5-2) are a phenyl group, 1-naphthyl group, 2-naphthyl group, 1-anthryl group, 2-anthryl group, 9-anthryl group, benzanthryl group, 1-phenanthryl group, 2-phenanthryl group, 3-phenanthryl group, 4-phenanthryl group, 9-phenanthryl group, naphthacetyl group, pyrenyl group, 1-chrysenyl group, 2-chrysenyl group, 3-chrysenyl group, 4-chrysenyl group, 5-chrysenyl group, 6-chrysenyl group, benzo[c]phenanthryl group, benzo[g]chrysenyl group, 1-triphenylenyl group, 2-triphenylenyl group, 3-triphenylenyl group, 4-triphenylenyl group, 1-fluorenyl group, 2-fluorenyl group, 3-fluorenyl group, 4-fluorenyl group, 9-fluorenyl group, benzofluorenyl group, dibenzofluorenyl group, 2-biphenyl group, 3-biphenyl group, 4-biphenyl group, o-terphenyl group, m-terphenyl-4-yl group, m-terphenyl-3-yl group, m-terphenyl-2-yl group, p-terphenyl-4-yl group, p-terphenyl-3-yl group, p-terphenyl-2-yl group, m-quarter-phenyl group, 3-fluoranthenyl group, 4-fluoranthenyl group, 8-fluoranthenyl group, 9-fluoranthenyl group, benzofluoranthenyl group, o-tolyl group, m-tolyl group, p-tolyl group, 2,3-

xylyl group, 3,4-xylyl group, 2,5-xylyl group, mesityl group, o-cumenyl group, m-cumenyl group, p-cumenyl group, p-t-butylphenyl group, p-(2-phenylpropyl)phenyl group, 4'-methylbiphenyl group, 4"-t-butyl-p-terphenyl-4-yl group, 9,9-dimethyl-1-fluorenyl group, 9,9-dimethyl-2-fluorenyl group, 9,9-dimethyl-3-fluorenyl group, 9,9-dimethyl-4-fluorenyl group, 9,9-diphenyl-1-fluorenyl group, 9,9-diphenyl-2-fluorenyl group, 9,9-diphenyl-3-fluorenyl group, and 9,9-diphenyl-4-fluorenyl group.

[0063] The aryl group in the formulae (4) to (6) and (5-1) to (5-2) preferably has 6 to 20 ring carbon atoms, more preferably 6 to 12 ring carbon atoms. Among the aryl group, a phenyl group, biphenyl group, naphthyl group, phenanthryl group, terphenyl group, fluorenyl group and triphenylenyl group are particularly preferable. In 1-fluorenyl group, 2-fluorenyl group, 3-fluorenyl group and 4-fluorenyl group, a carbon atom at the ninth position is preferably substituted by the substituted or unsubstituted alkyl group having 1 to 30 carbon atoms in the formula (4).

[0064] The alkyl group having 1 to 30 carbon atoms in the formulae (4) and (6) may be linear, branched or cyclic. Examples of the linear or branched alkyl group are a methyl group, ethyl group, propyl group, isopropyl group, n-butyl group, s-butyl group, isobutyl group, t-butyl group, n-pentyl group, n-hexyl group, n-heptyl group, n-octyl group, n-nonyl group, n-decyl group, n-undecyl group, n-dodecyl group, n-tridecyl group, n-tetradecyl group, n-pentadecyl group, n-hexadecyl group, n-heptadecyl group, n-octadecyl group, neopentyl group, 1-methylpentyl group, 2-methylpentyl group, 1-pentylhexyl group, 1-butylpentyl group, 1-heptyloctyl group, 3-methylpentyl group, hydroxymethyl group, 1-hydroxyethyl group, 2-hydroxyethyl group, 2-hydroxyisobutyl group, 1,2-dihydroxyethyl group, 1,3-dihydroxyisopropyl group, 2,3-dihydroxy-t-butyl group, 1,2,3-trihydroxypropyl group, chloromethyl group, 1-chloroethyl group, 2-chloroethyl group, 2-chloroisobutyl group, 1,2-dichloroethyl group, 1,3-dichloroisopropyl group, 2,3-dichloro-t-butyl group, 1,2,3-trichloropropyl group, bromomethyl group, 1-bromoethyl group, 2-bromoethyl group, 2-bromoisobutyl group, 1,2-dibromoethyl group, 1,3-dibromoisopropyl group, 2,3-dibromo-t-butyl group, 1,2,3-tribromopropyl group, iodomethyl group, 1-iodoethyl group, 2-iodoethyl group, 2-iodoisobutyl group, 1,2-diiodoethyl group, 1,3-diiodoisopropyl group, 2,3-diiodo-t-butyl group, 1,2,3-triiodopropyl group, aminomethyl group, 1-aminoethyl group, 2-aminoethyl group, 2-aminoisobutyl group, 1,2-diaminoethyl group, 1,3-diaminoisopropyl group, 2,3-diamino-t-butyl group, 1,2,3-triaminopropyl group, cyanomethyl group, 1-cyanoethyl group, 2-cyanoethyl group, 2-cyanoisobutyl group, 1,2-dicyanoethyl group, 1,3-dicyanoisopropyl group, 2,3-dicyano-t-butyl group, 1,2,3-tricyanopropyl group, nitromethyl group, 1-nitroethyl group, 2-nitroethyl group, 1,2-dinitroethyl group, 2,3-dinitro-t-butyl group, and 1,2,3-trinitropropyl group.

[0065] Examples of the cyclic alkyl group (cycloalkyl group) are a cyclopropyl group, cyclobutyl group, cyclopentyl group, cyclohexyl group, 4-methylcyclohexyl group, 1-adamantyl group, 2-adamantyl group, 1-norbornyl group, and 2-norbornyl group.

[0066] The linear or branched alkyl group in the formulae (4) and (6) preferably has 1 to 10 carbon atoms, more preferably 1 to 6 carbon atoms. Among the linear or branched alkyl group, a methyl group, ethyl group, propyl group, isopropyl group, n-butyl group, s-butyl group, isobutyl group, t-butyl group, n-pentyl group and n-hexyl group are preferable.

[0067] The cycloalkyl group in the formulae (4) and (6) preferably has 3 to 10 ring carbon atoms, more preferably 5 to 8 ring carbon atoms. Among the cycloalkyl group, a cyclopentyl group and a cyclohexyl group are preferable.

[0068] A halogenated alkyl group provided by substituting the alkyl group with a halogen atom is exemplified by a halogenated alkyl group provided by substituting the alkyl group having 1 to 30 carbon atoms with one or more halogen groups. Specific examples of the halogenated alkyl group are a fluoromethyl group, difluoromethyl group, trifluoromethyl group, fluoroethyl group and trifluoromethylmethyl group.

[0069] The alkenyl group having 2 to 30 carbon atoms in the formulae (4) and (6) may be linear, branched or cyclic. Examples of the alkenyl group are vinyl, propenyl, butenyl, oleyl, eicosapentaenyl, docosahexaenyl, styryl, 2,2-diphenylvinyl, 1,2,2-triphenylvinyl, and 2-phenyl-2-propenyl. Among the alkenyl group, a vinyl group is preferable.

[0070] The alkynyl group having 2 to 30 carbon atoms in the formulae (4) and (6) may be linear, branched or cyclic. Examples of the alkynyl group are ethynyl, propynyl, and 2-phenylethynyl. Among the alkynyl group, an ethynyl group is preferable.

[0071] The alkylsilyl group having 3 to 30 carbon atoms in the formulae (4) and (6) is exemplified by a trialkylsilyl group having the examples of the alkyl group having 1 to 30 carbon atoms. Specific examples of the alkylsilyl group are a trimethylsilyl group, triethylsilyl group, tri-n-butylsilyl group, tri-n-octylsilyl group, triisobutylsilyl group, dimethylethylsilyl group, dimethylisopropylsilyl group, dimethyl-n-propylsilyl group, dimethyl-n-butylsilyl group, dimethyl-t-butylsilyl group, diethylisopropylsilyl group, vinyldimethylsilyl group, propyldimethylsilyl group and triisopropylsilyl group. Three alkyl groups may be the same or different.

[0072] The arylsilyl group having 6 to 30 ring carbon atoms in the formulae (4) and (6) may be a dialkylarylsilyl group, alkylarylsilyl group and triarylsilyl group.

[0073] The dialkylarylsilyl group is exemplified by a dialkylarylsilyl group having two of the examples of the alkyl group having 1 to 30 carbon atoms and one of the aryl group having 6 to 30 ring carbon atoms. The dialkylarylsilyl group preferably has 8 to 30 carbon atoms. Two alkyl groups may be the same or different.

[0074] The alkylarylsilyl group is exemplified by an alkylarylsilyl group having one of the examples of the alkyl

group having 1 to 30 carbon atoms and two of the aryl group having 6 to 30 ring carbon atoms. The dialkylarylsilyl group preferably has 13 to 30 carbon atoms. Two aryl groups may be the same or different.

[0075] The triarylsilyl group is exemplified by a triarylsilyl group having three of the aryl group having 6 to 30 ring carbon atoms. The dialkylarylsilyl group preferably has 18 to 30 carbon atoms. Three aryl groups may be the same or different.

[0076] The alkoxy group having 1 to 30 carbon atoms in the formulae (4) and (6) is represented by -OY. Y is exemplified by the alkyl group having 1 to 30 carbon atoms. Examples of the alkoxy group are a methoxy group, ethoxy group, propoxy group, butoxy group, pentyloxy group and hexyloxy group.

[0077] A halogenated alkoxy group provided by substituting the alkoxy group with a halogen atom is exemplified by a halogenated alkoxy group provided by substituting the alkoxy group having 1 to 30 carbon atoms with one or more halogen groups.

[0078] The aralkyl group having 6 to 30 ring carbon atoms in the formulae (4) and (6) is represented by -Y-Z. Y is exemplified by an alkylene group corresponding to the alkyl group having 1 to 30 carbon atoms. Z is exemplified by the examples of the aryl group having 6 to 30 ring carbon atoms. This aralkyl group is preferably an aralkyl group having 7 to 30 carbon atoms, in which an aryl moiety has 6 to 30 carbon atoms, preferably 6 to 20 carbon atoms, more preferably 6 to 12 carbon atoms, and an alkyl moiety has 1 to 30 carbon atoms, preferably 1 to 20 carbon atoms, more preferably 1 to 10 carbon atoms, further preferably 1 to 6 carbon atoms. Examples of the aralkyl group are a benzyl group, 2-phenylpropane-2-yl group, 1-phenylethyl group, 2-phenylethyl group, 1-phenylisopropyl group, 2-phenylisopropyl group, phenyl-t-butyl group, α -naphthylmethyl group, 1- α -naphthylethyl group, 2- α -naphthylethyl group, 1- α -naphthylisopropyl group, 2- α -naphthylisopropyl group, β -naphthylmethyl group, 1- β -naphthylethyl group, 2- β -naphthylethyl group, 1- β -naphthylisopropyl group, 2- β -naphthylisopropyl group, p-methylbenzyl group, m-methylbenzyl group, o-methylbenzyl group, p-chlorobenzyl group, m-chlorobenzyl group, o-chlorobenzyl group, p-bromobenzyl group, m-bromobenzyl group, o-bromobenzyl group, p-iodobenzyl group, m-iodobenzyl group, o-iodobenzyl group, p-hydroxybenzyl group, m-hydroxybenzyl group, o-hydroxybenzyl group, p-aminobenzyl group, m-aminobenzyl group, o-aminobenzyl group, p-nitrobenzyl group, m-nitrobenzyl group, o-nitrobenzyl group, p-cyanobenzyl group, m-cyanobenzyl group, o-cyanobenzyl group, 1-hydroxy-2-phenylisopropyl group and 1-chloro-2-phenylisopropyl group.

[0079] The aryloxy group having 6 to 30 ring carbon atoms in the formulae (4) and (6) is represented by -OZ. Z is exemplified by the aryl group having 6 to 30 ring carbon atoms or the following monocyclic group and fused cyclic group. The aryloxy group is exemplified by a phenoxy group.

[0080] Examples of the halogen atom in the formulae (4) and (6) are fluorine, chlorine, bromine and iodine, among which fluorine is preferable.

[0081] In the invention, "carbon atoms forming a ring (ring carbon atoms)" mean carbon atoms forming a saturated ring, unsaturated ring, or aromatic ring. "Atoms forming a ring (ring atoms)" mean carbon atoms and hetero atoms forming a hetero ring including a saturated ring, unsaturated ring, or aromatic ring.

[0082] Examples of the substituent meant by "substituted or unsubstituted" are the above-described aryl group, alkyl group (linear or branched alkyl group, cycloalkyl group and halogenated alkyl group), alkenyl group, alkynyl group, alkylsilyl group, arylsilyl group, alkoxy group, halogenated alkoxy group, aralkyl group, aryloxy group, halogen atom, cyano group, hydroxyl group, nitro group and carboxy group. In the above-described substituents, the aryl group, alkyl group, halogen atom, alkylsilyl group, arylsilyl group and cyano group are preferable. The preferable ones of the specific examples of each substituent are further preferable. Moreover, these substituents may be further substituted by the above-described substituents.

[0083] "Unsubstituted" in "substituted or unsubstituted" means that a hydrogen atom is substituted.

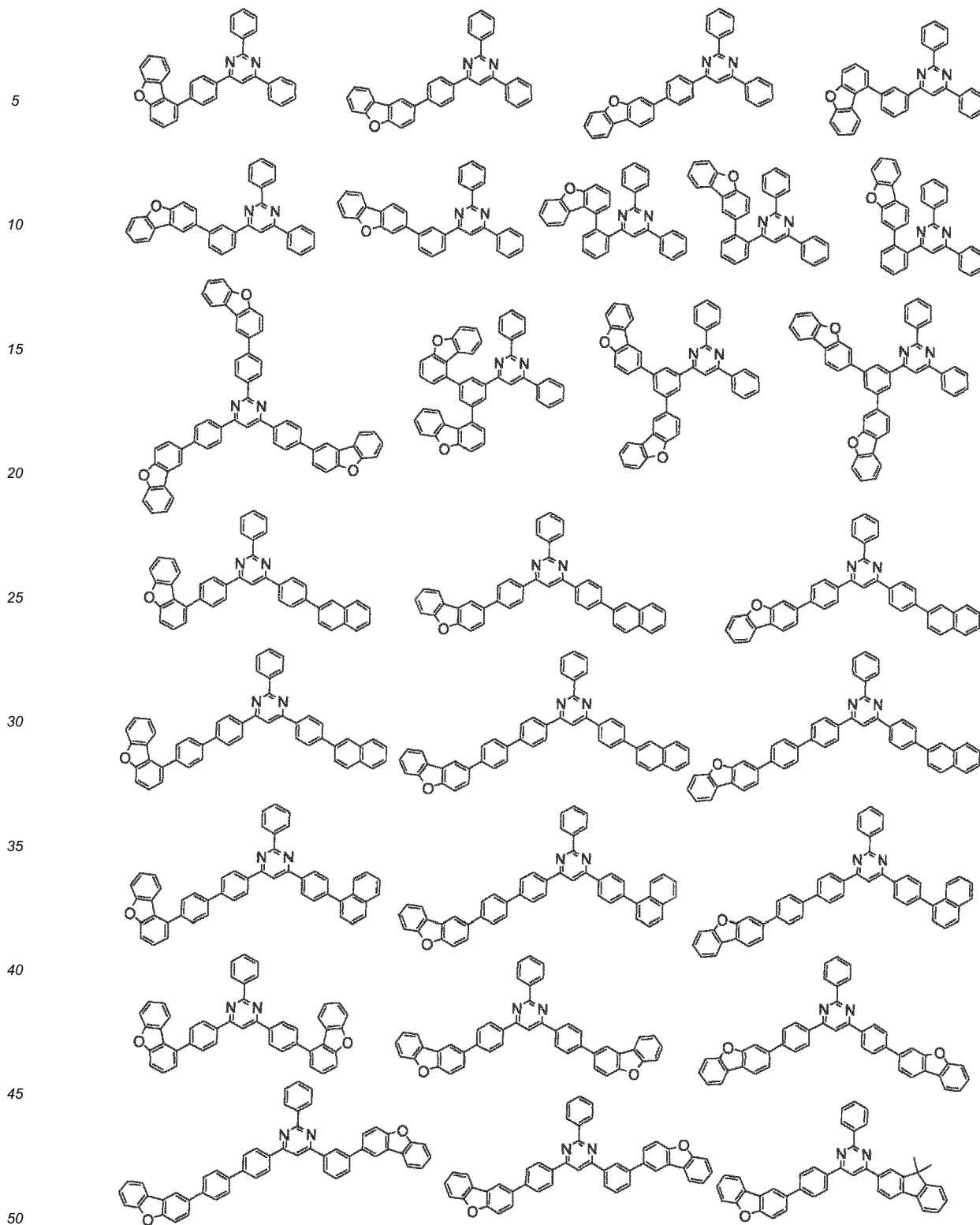
[0084] In a later-described compound or a partial structure thereof, the same applies to the description of "substituted or unsubstituted."

[0085] In the invention, a "hydrogen atom" means isotopes having different neutron numbers and specifically encompasses protium, deuterium and tritium.

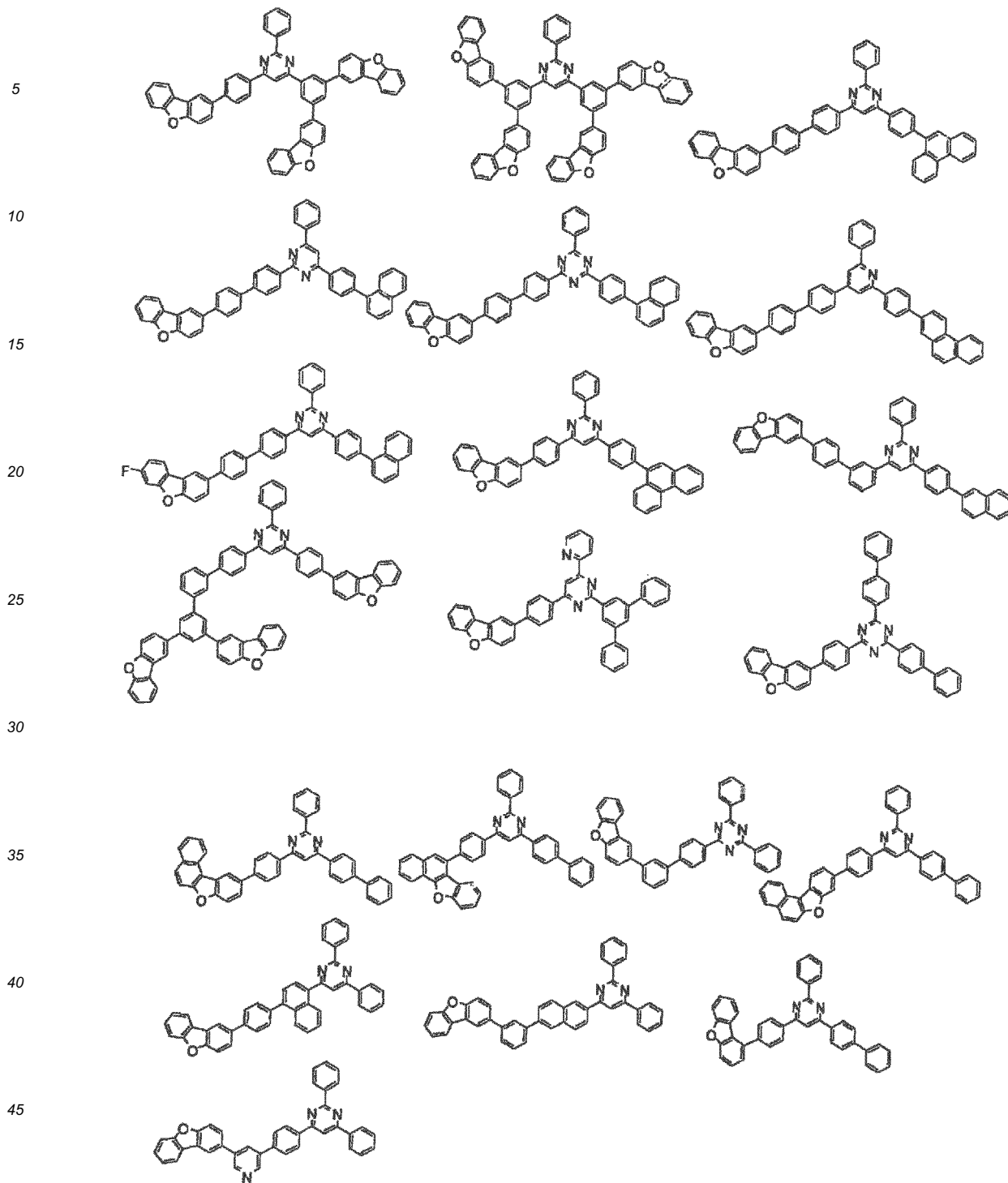
[0086] Examples of specific structures of the aromatic heterocyclic derivative according to the invention represented by the formula (4) are as follows. However, the invention is not limited to the aromatic derivative having these structures.

[Formula 12]

[0087]

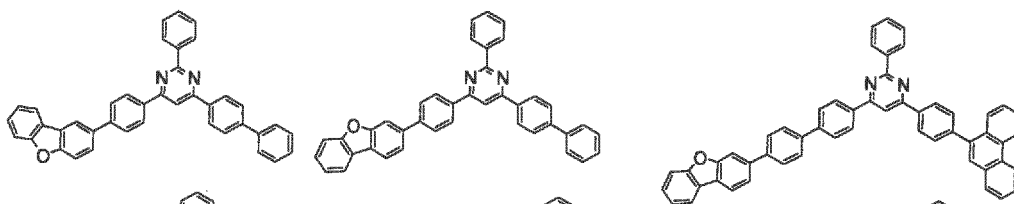


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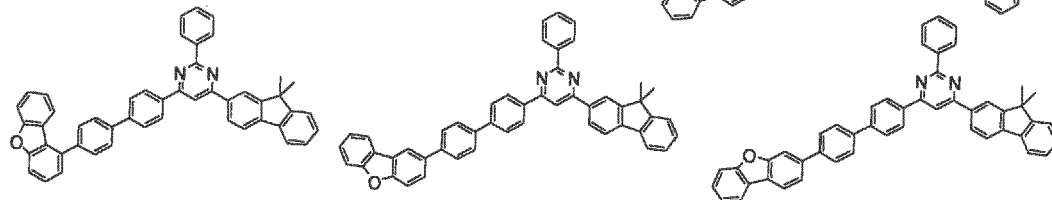


[Formula 15]

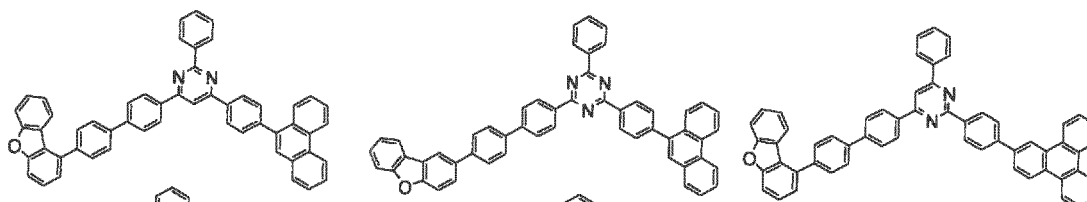
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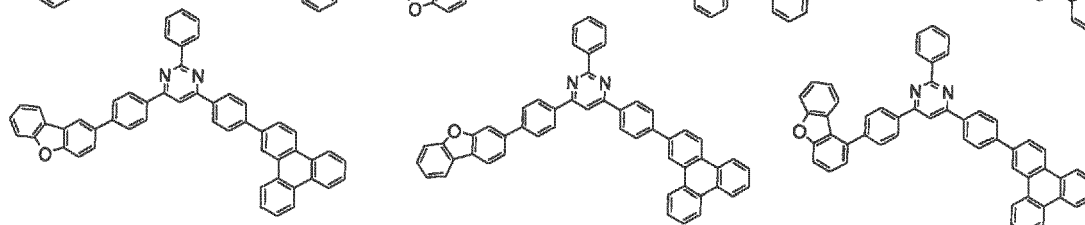
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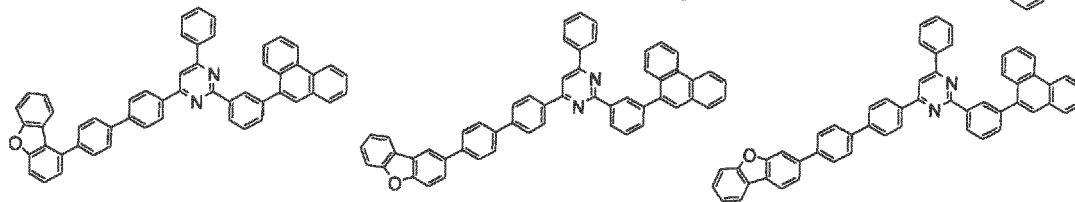
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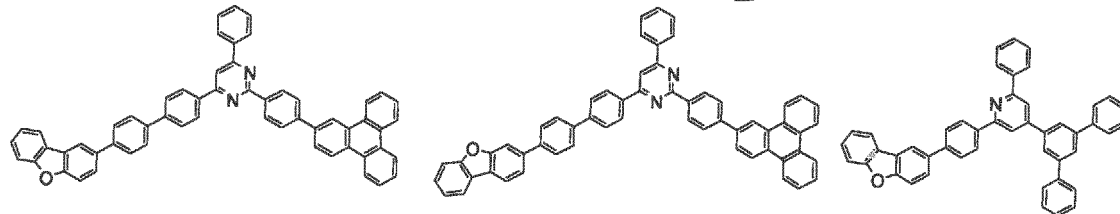
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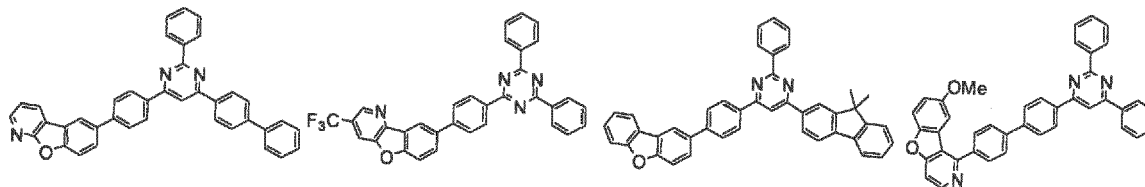


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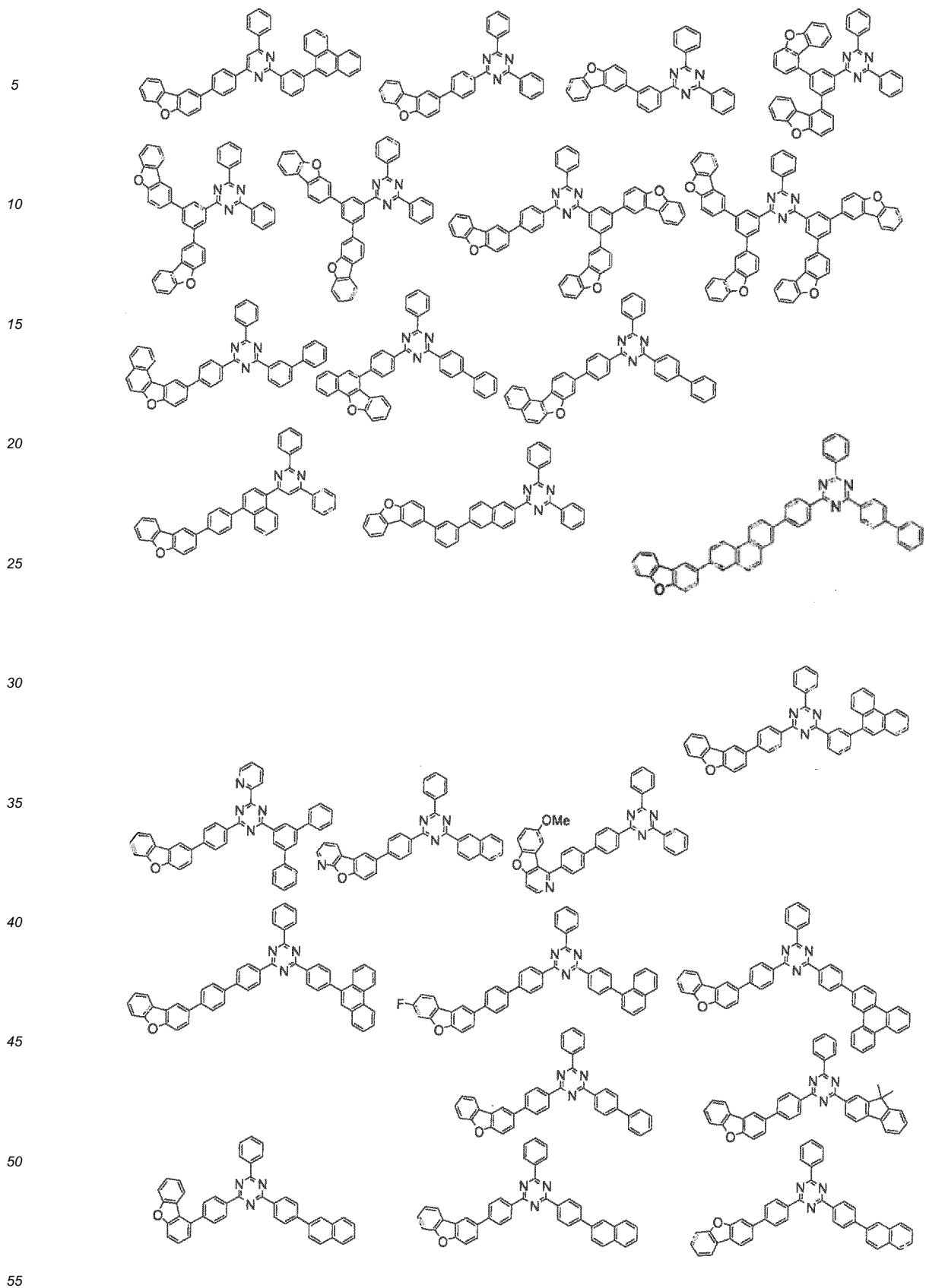
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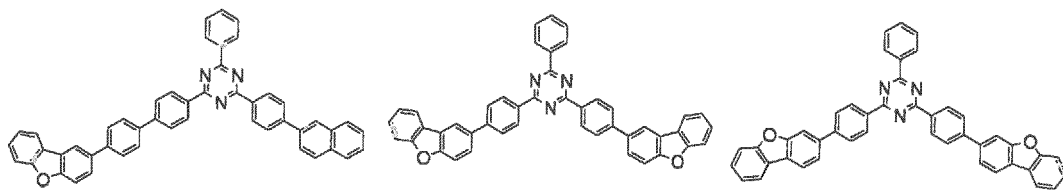
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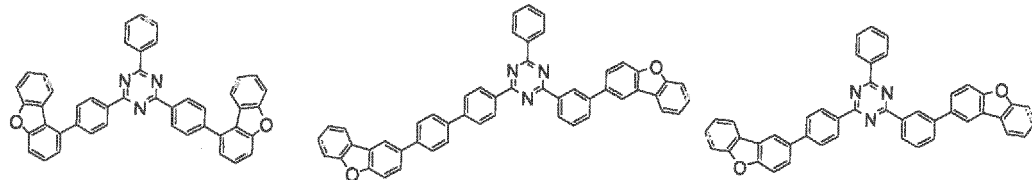
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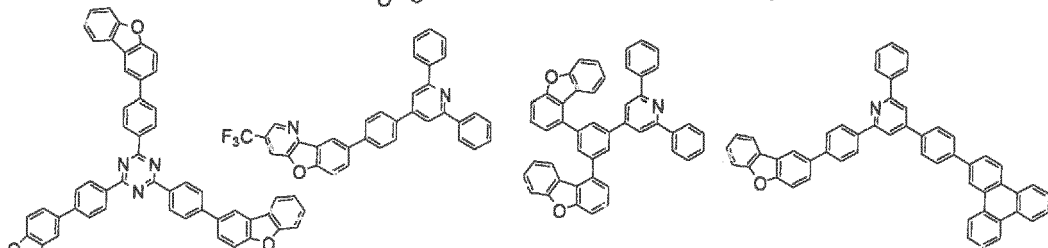
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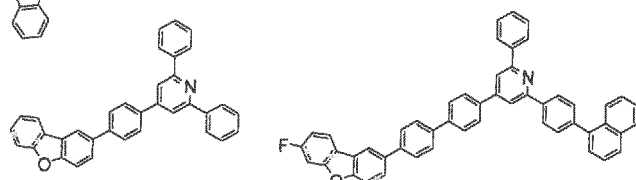
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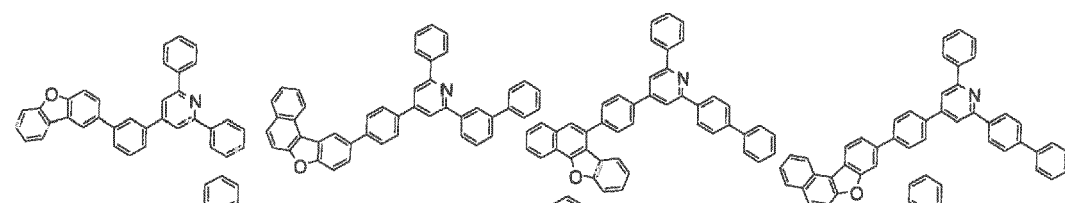


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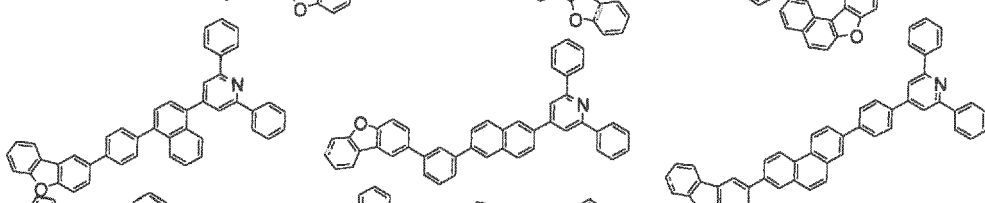


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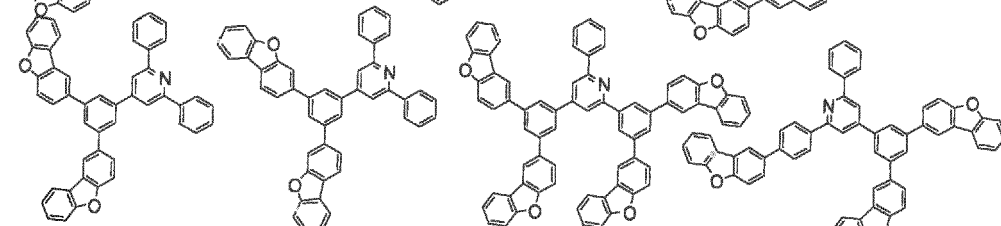
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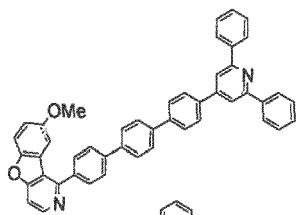


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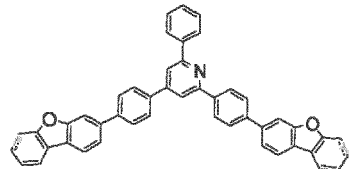
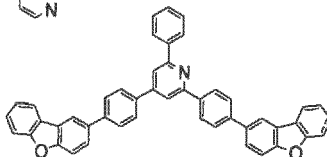
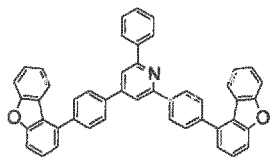
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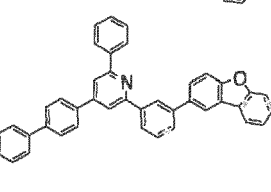
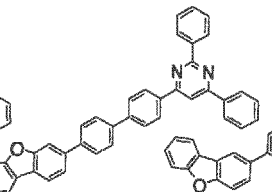
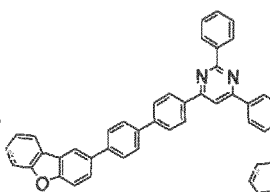
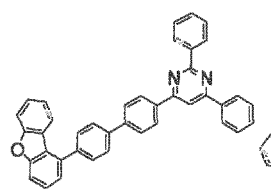
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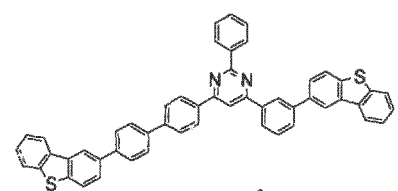
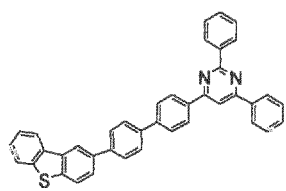
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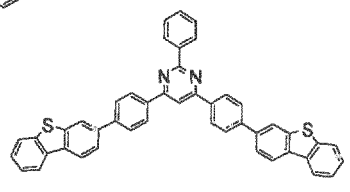
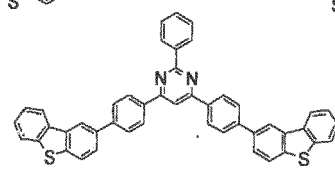
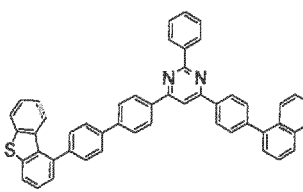
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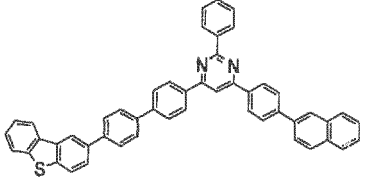
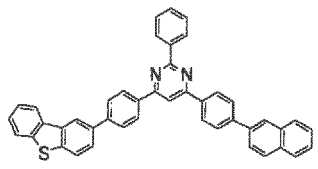
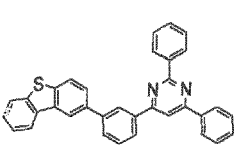
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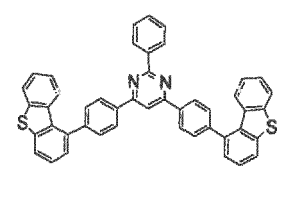
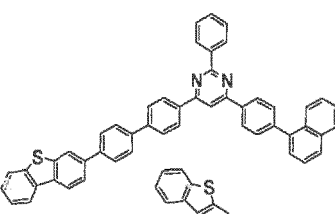
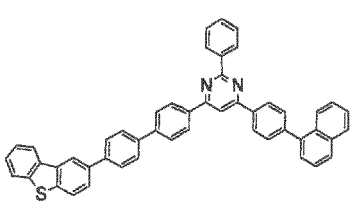
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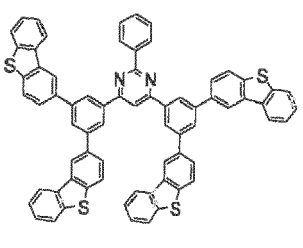
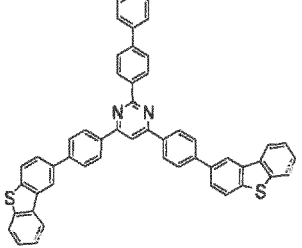
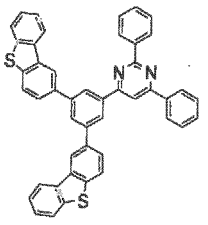
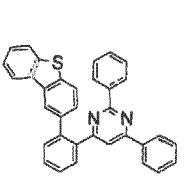
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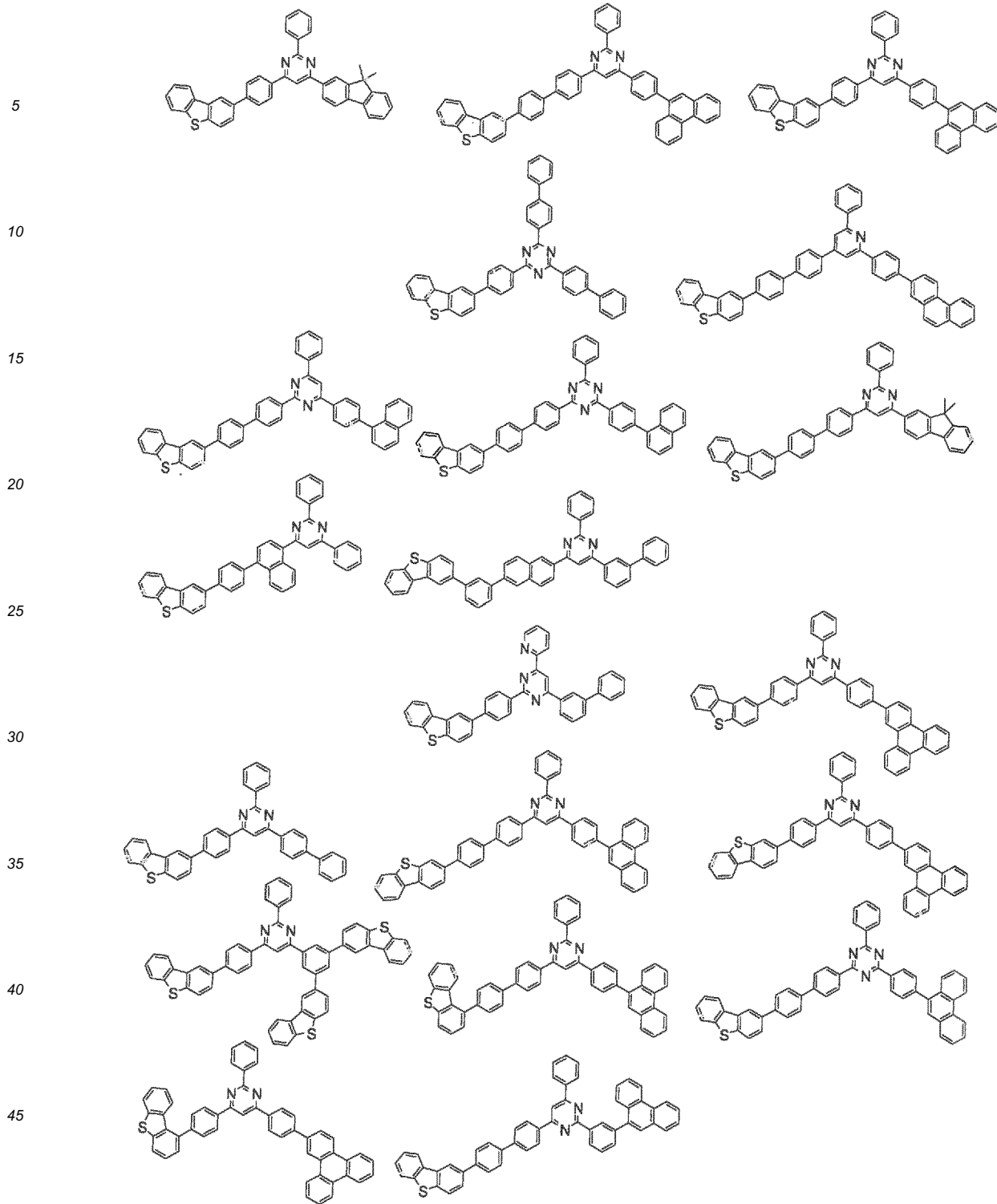
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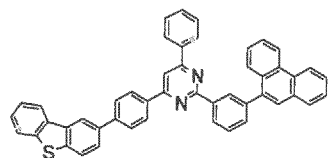
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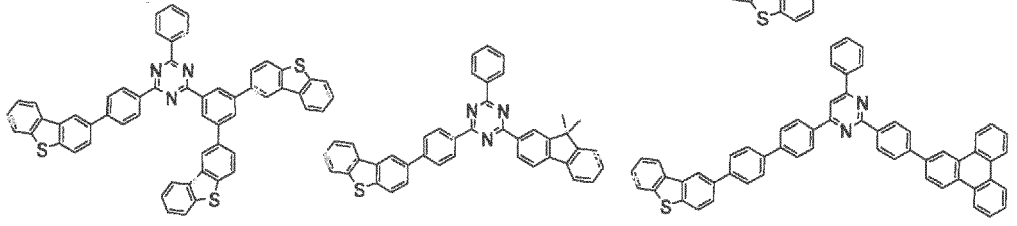
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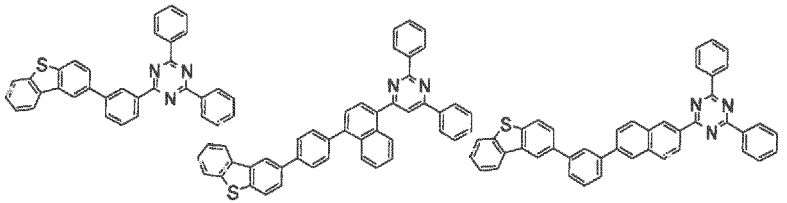
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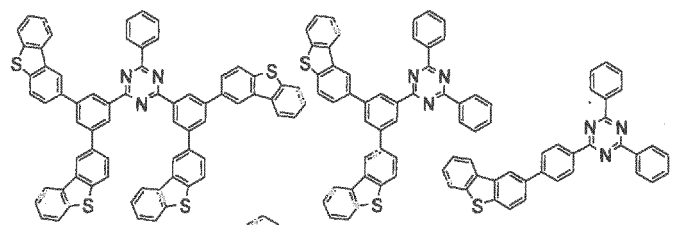
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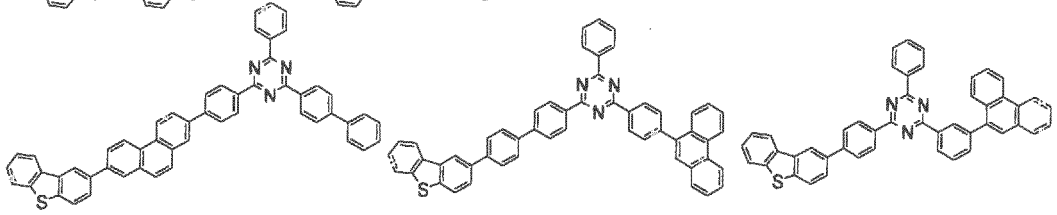
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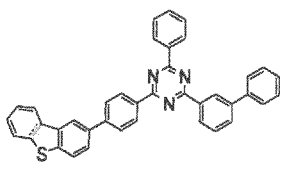
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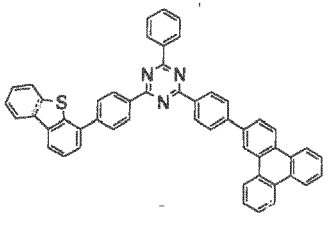
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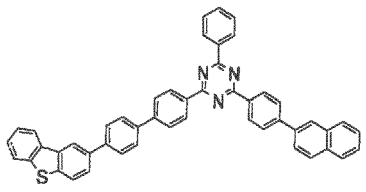
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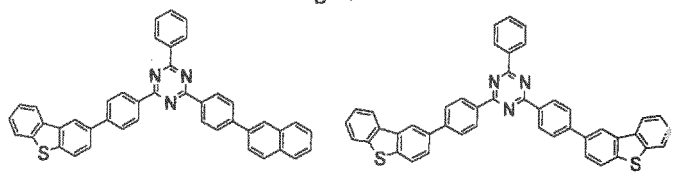
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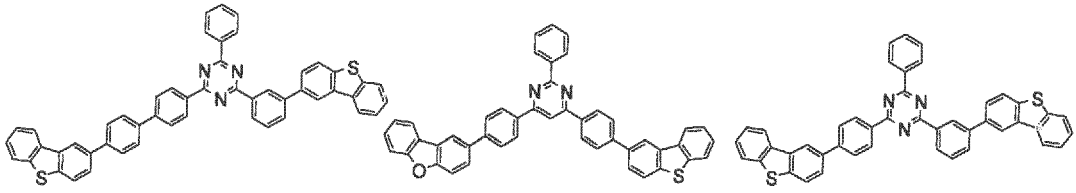
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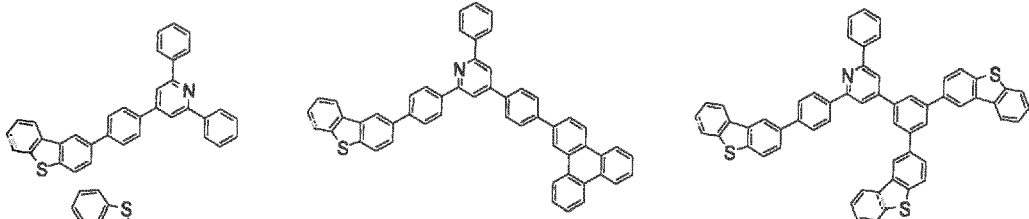
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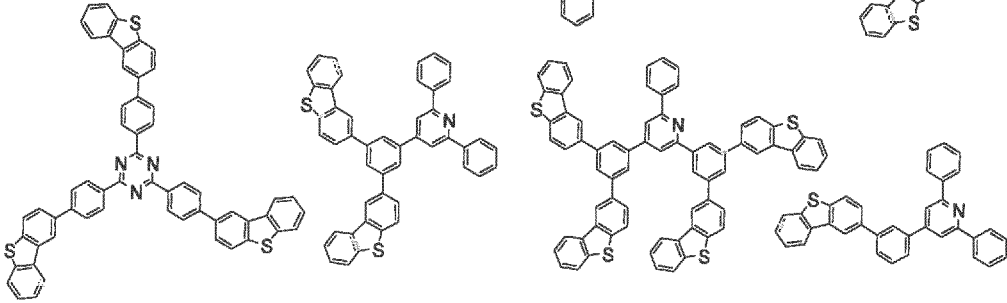
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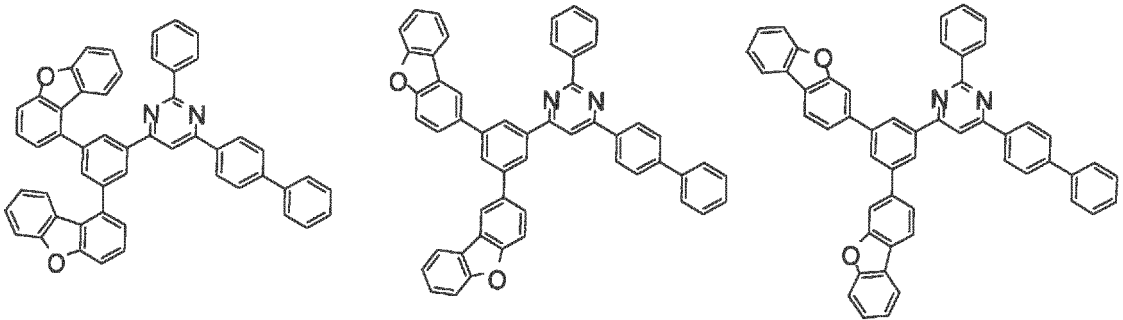


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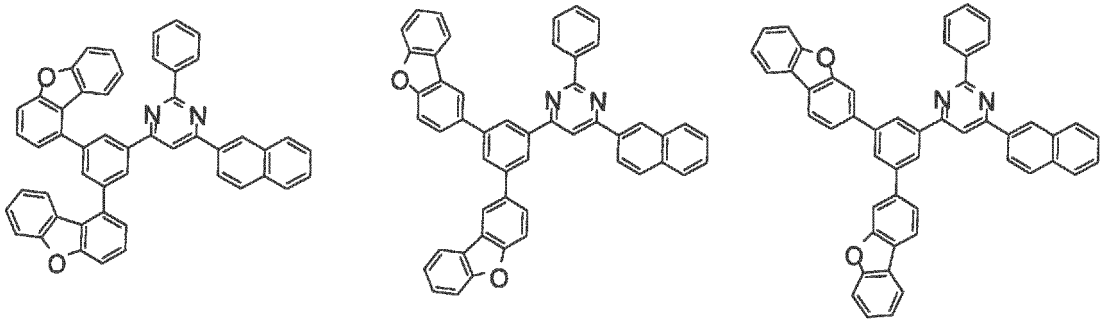
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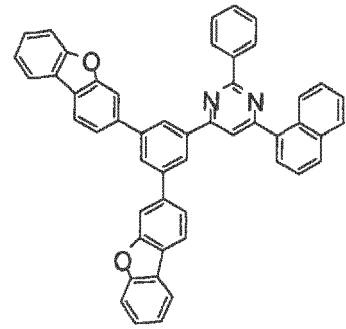
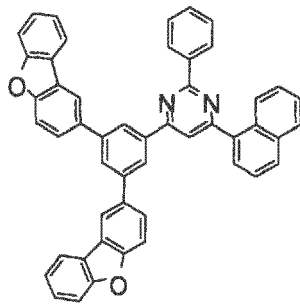
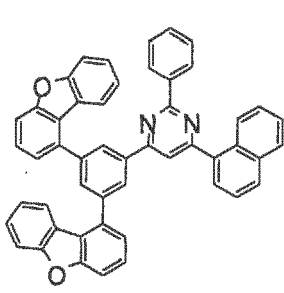


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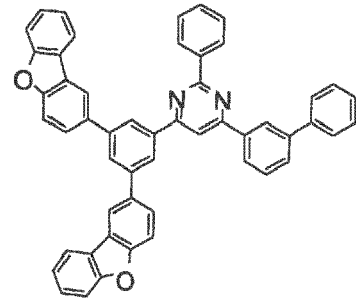
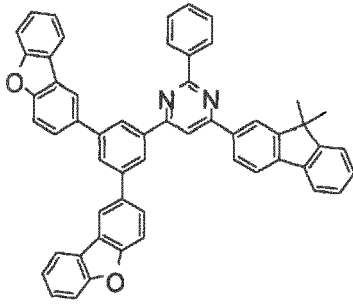
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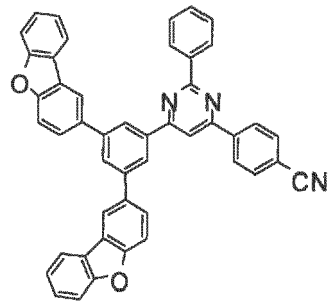
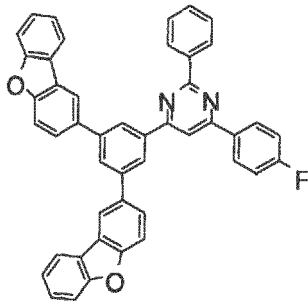
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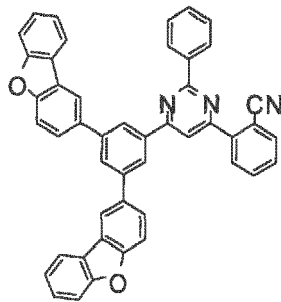
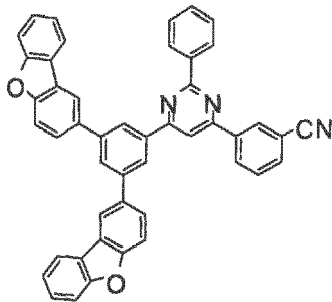
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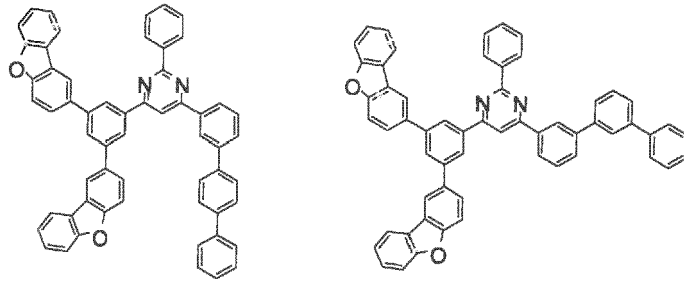
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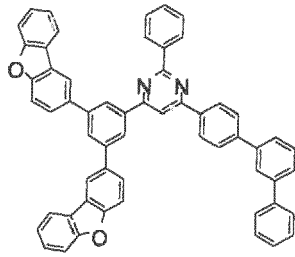
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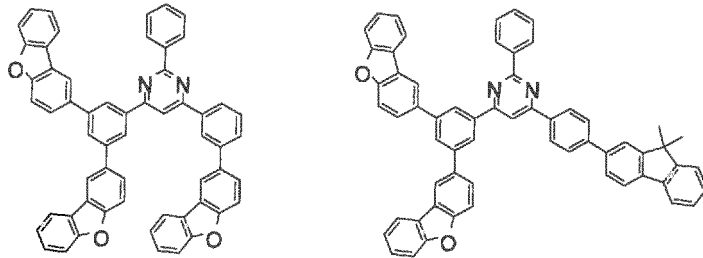
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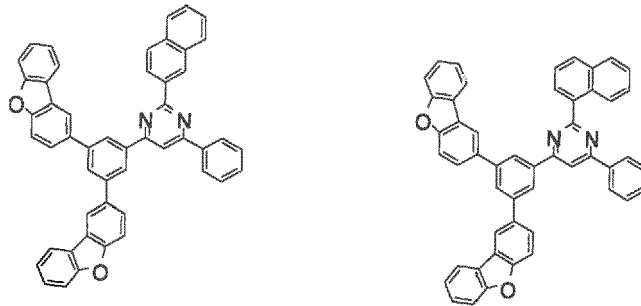
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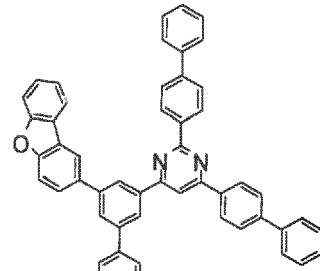
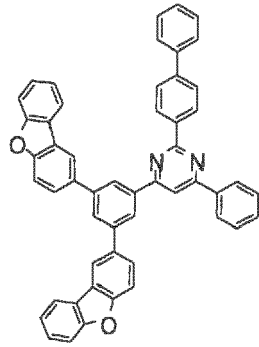
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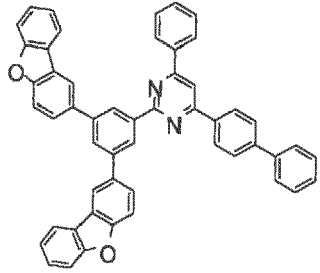
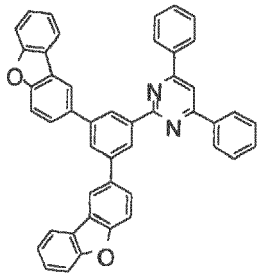
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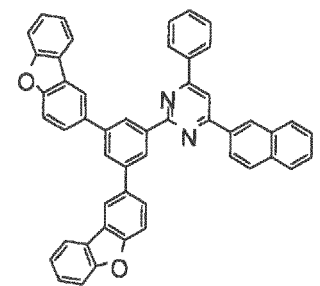
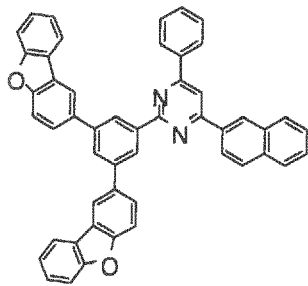
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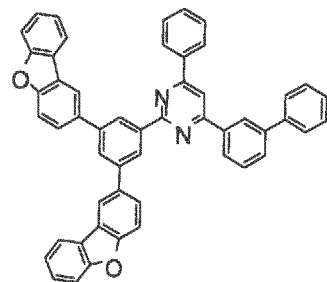
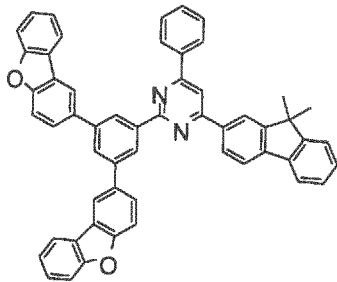
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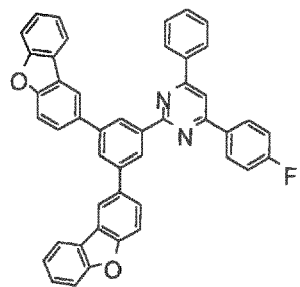
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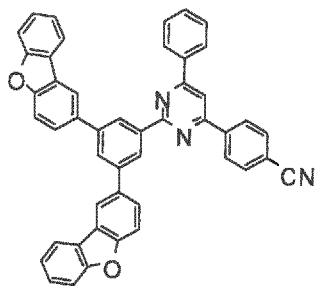
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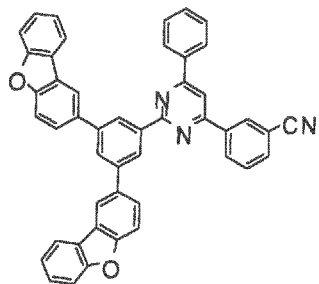
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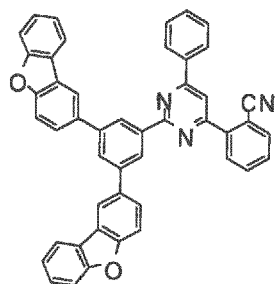
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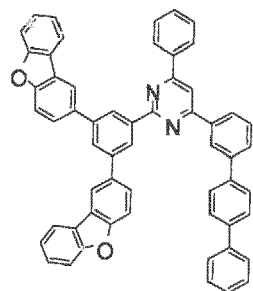
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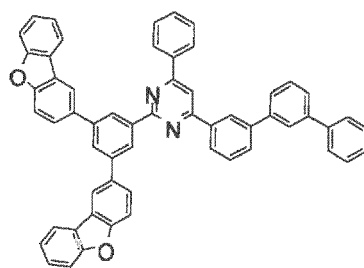
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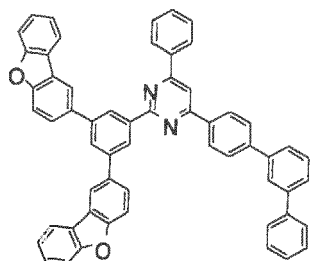
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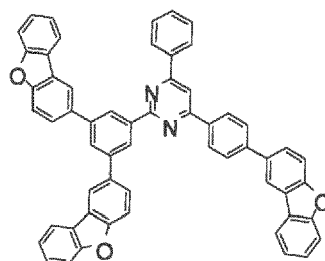
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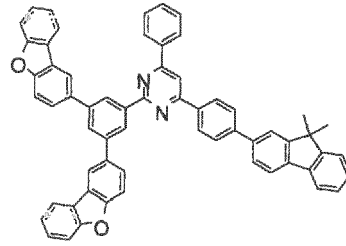
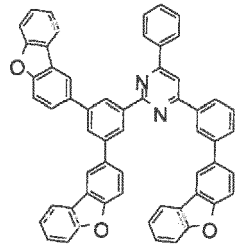


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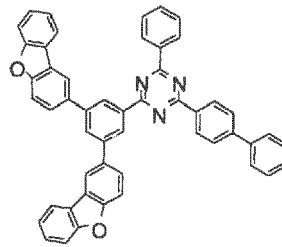
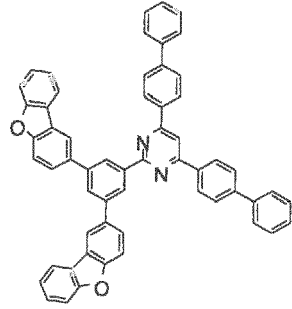
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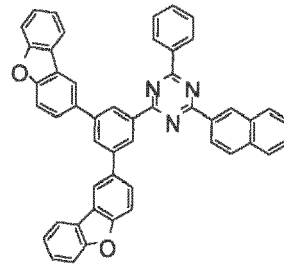
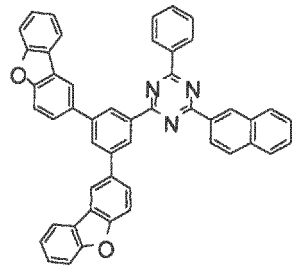
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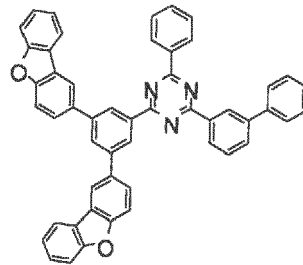
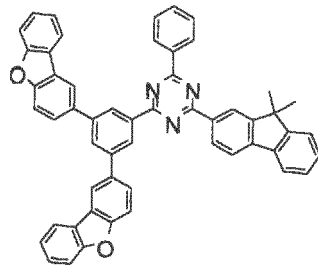
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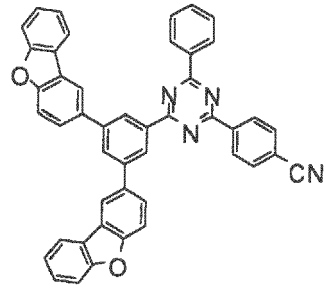
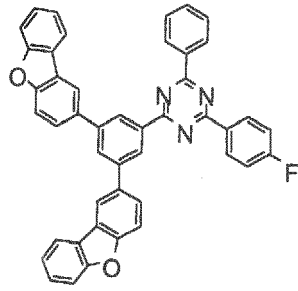
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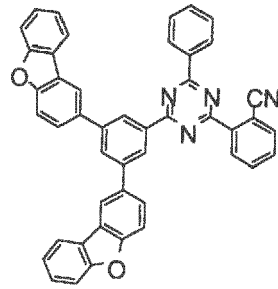
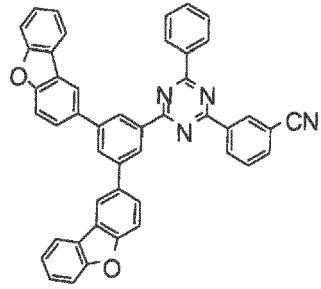
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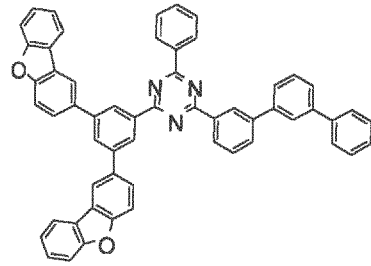
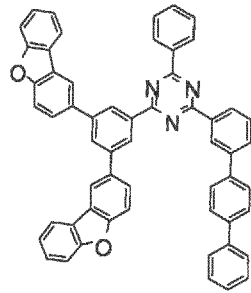
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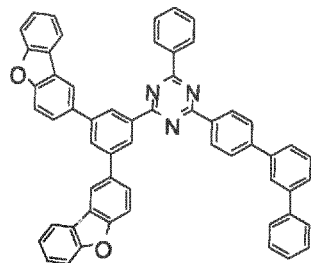
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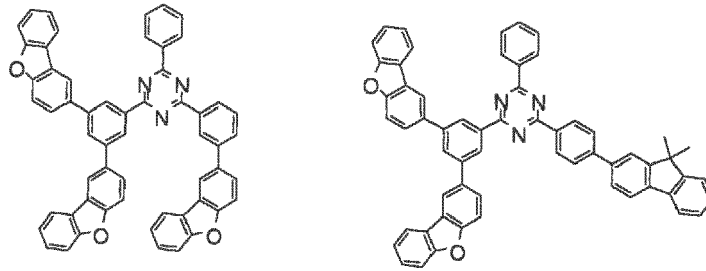
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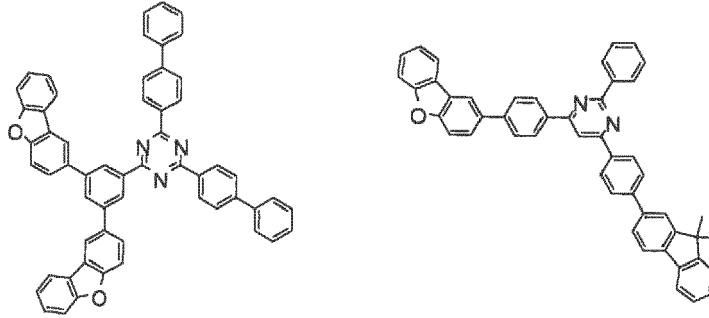
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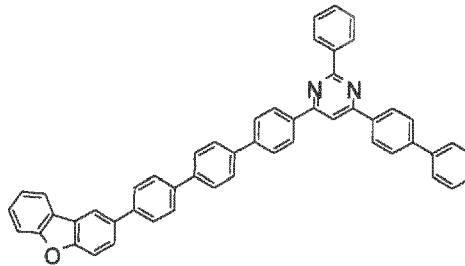
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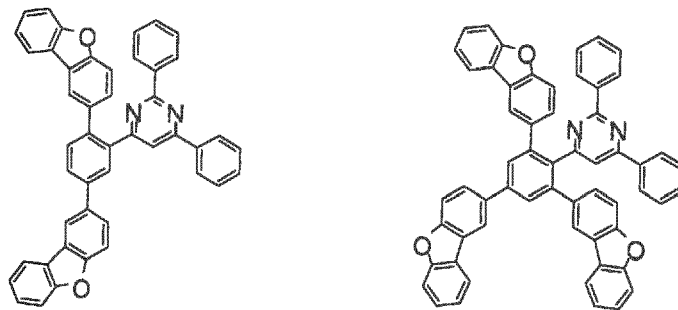
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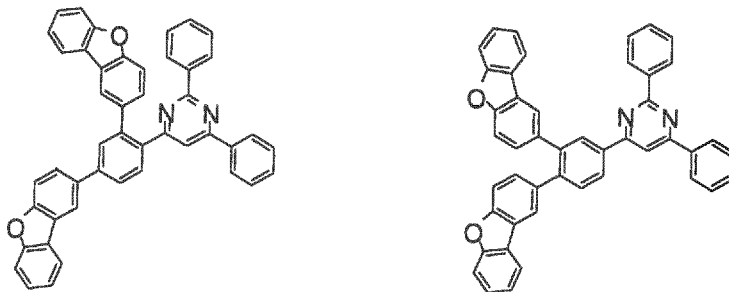
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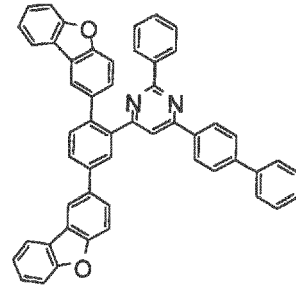
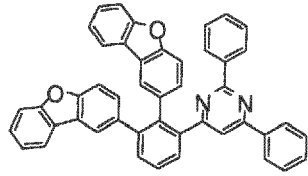
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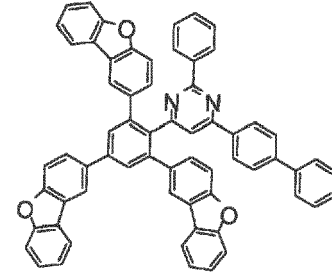
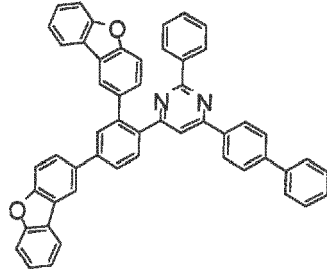
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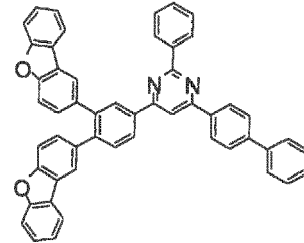
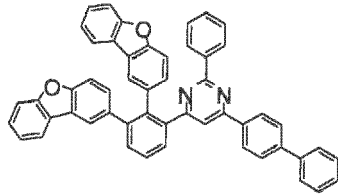
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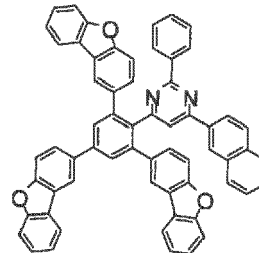
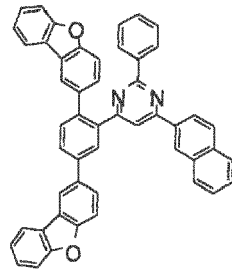
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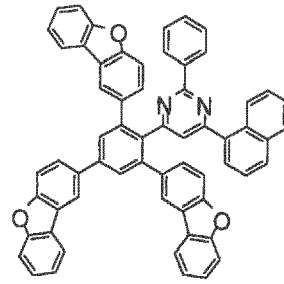
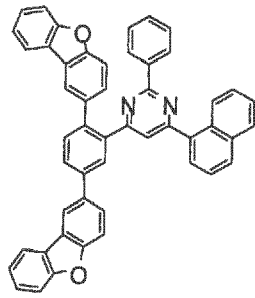
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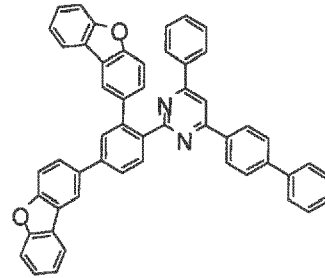
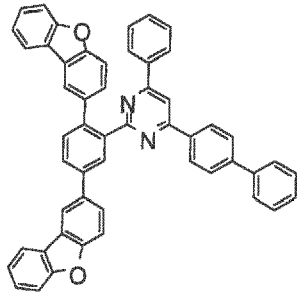
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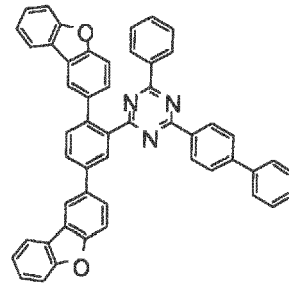
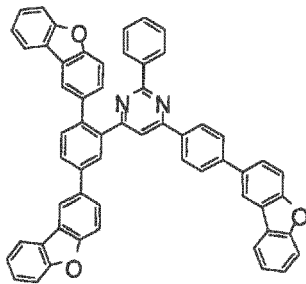
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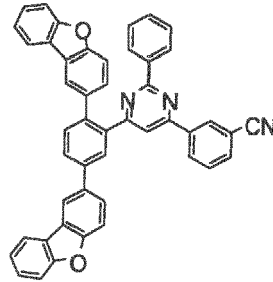
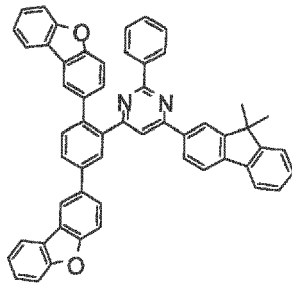
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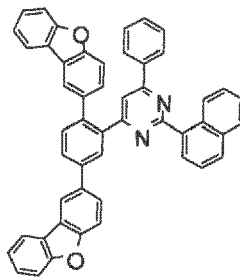
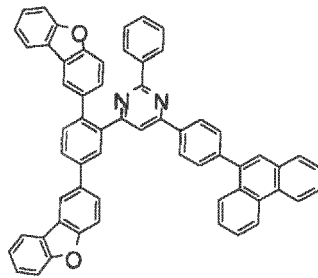
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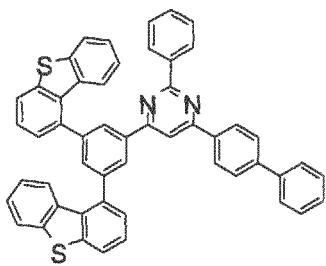
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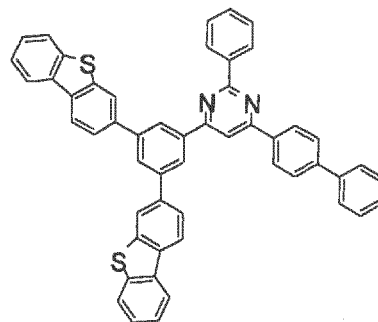
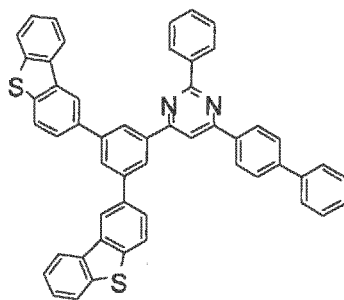


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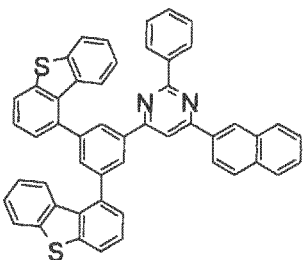
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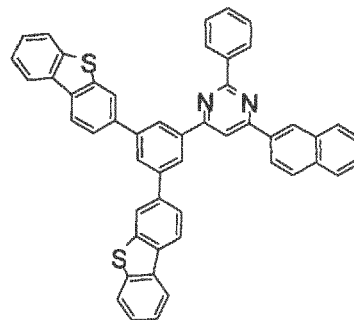
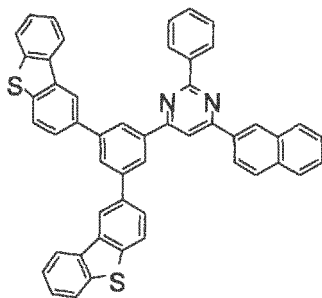
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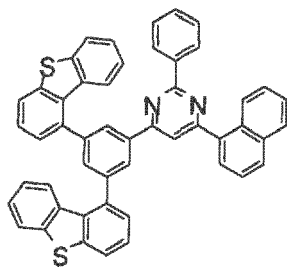
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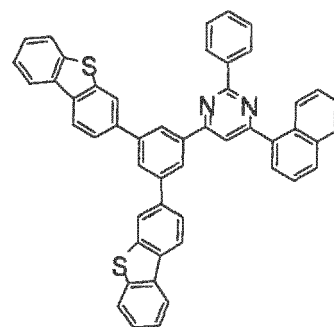
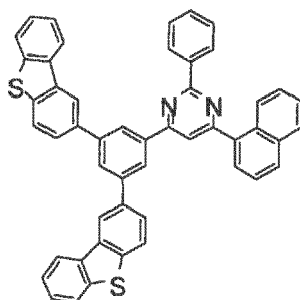
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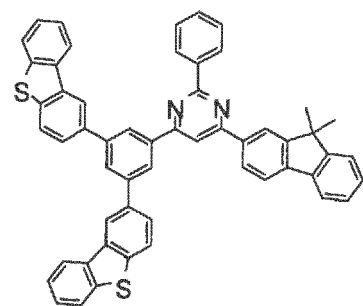
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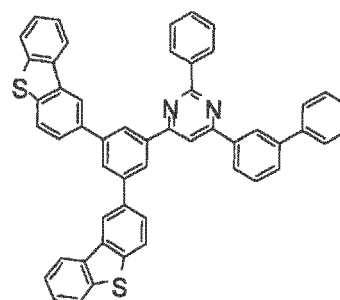
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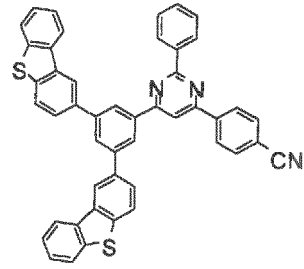
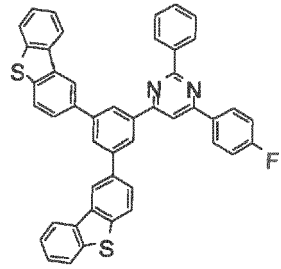


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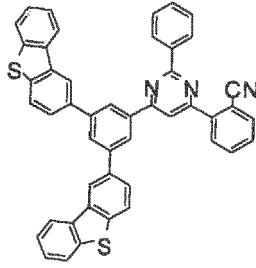
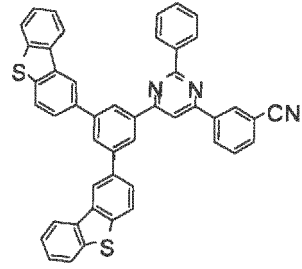
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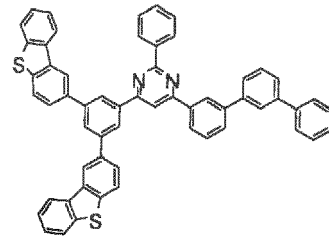
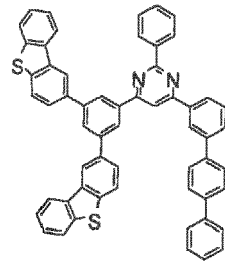
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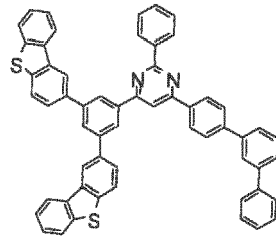
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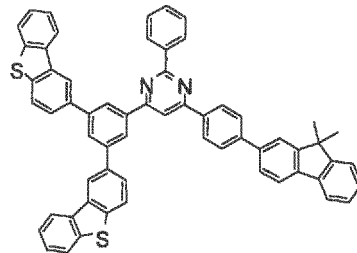
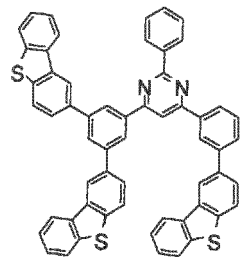
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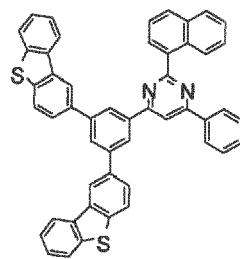
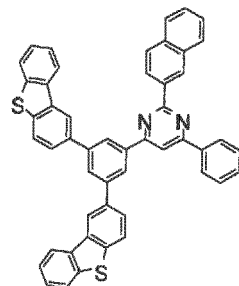
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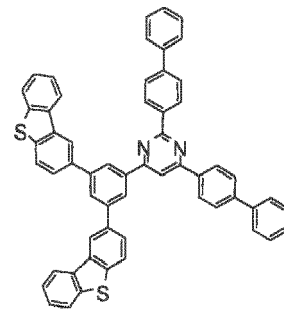
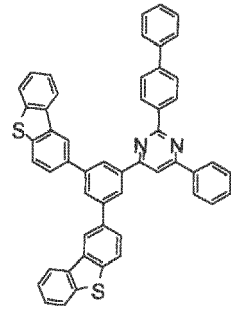


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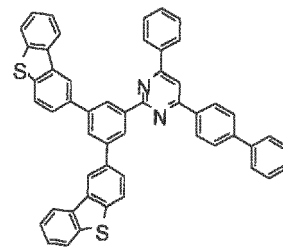
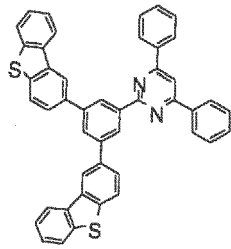
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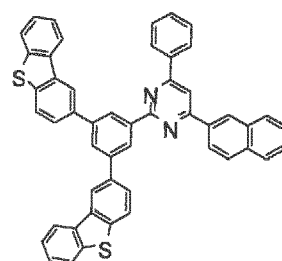
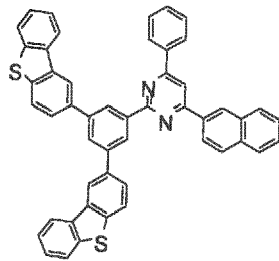


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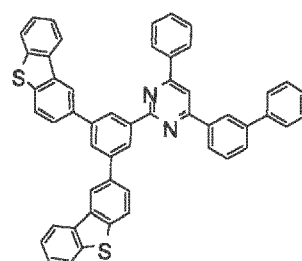
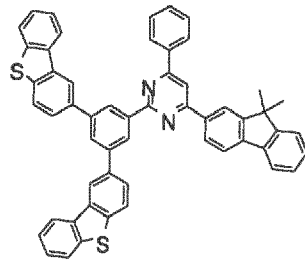
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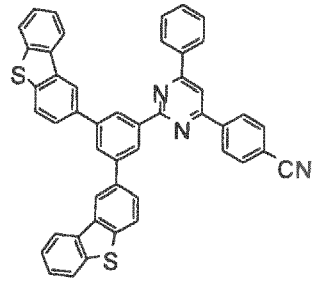
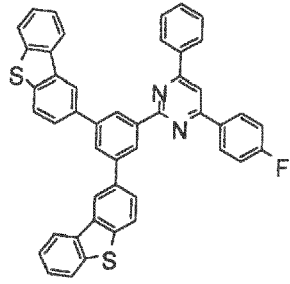
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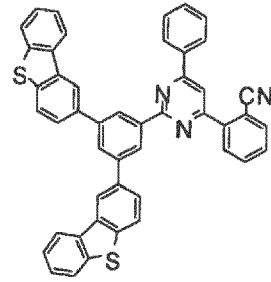
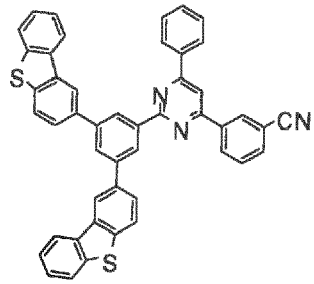
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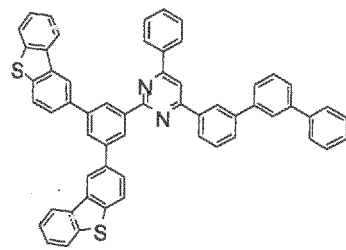
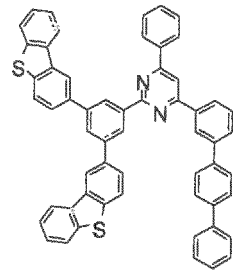
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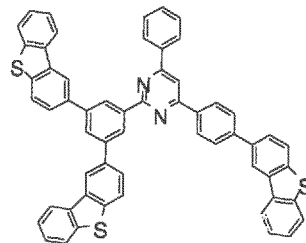
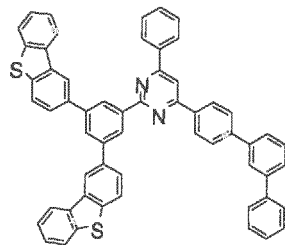
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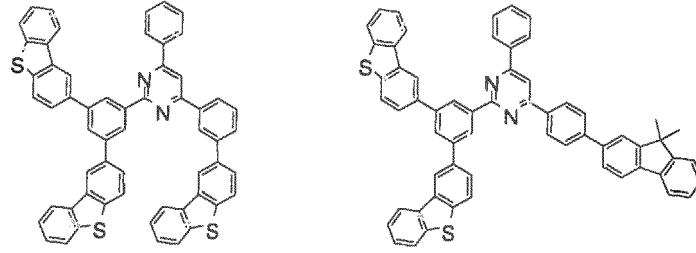
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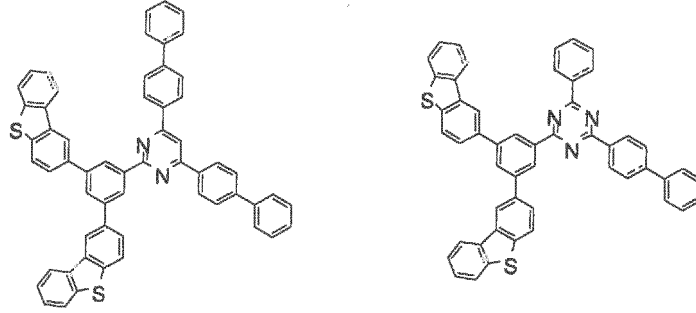
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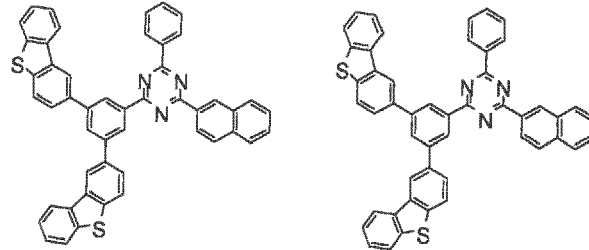
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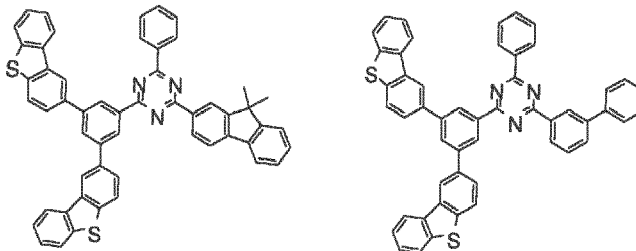
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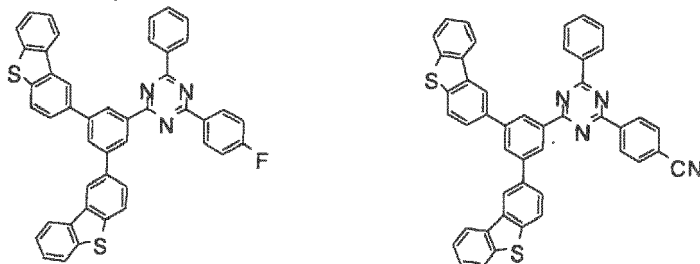
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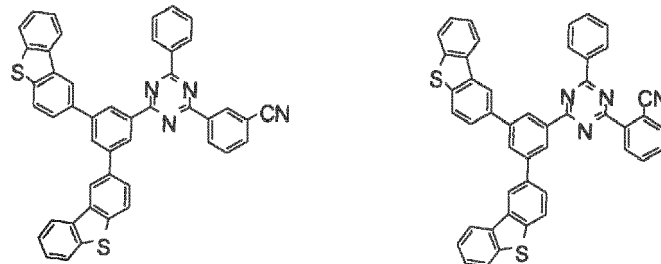
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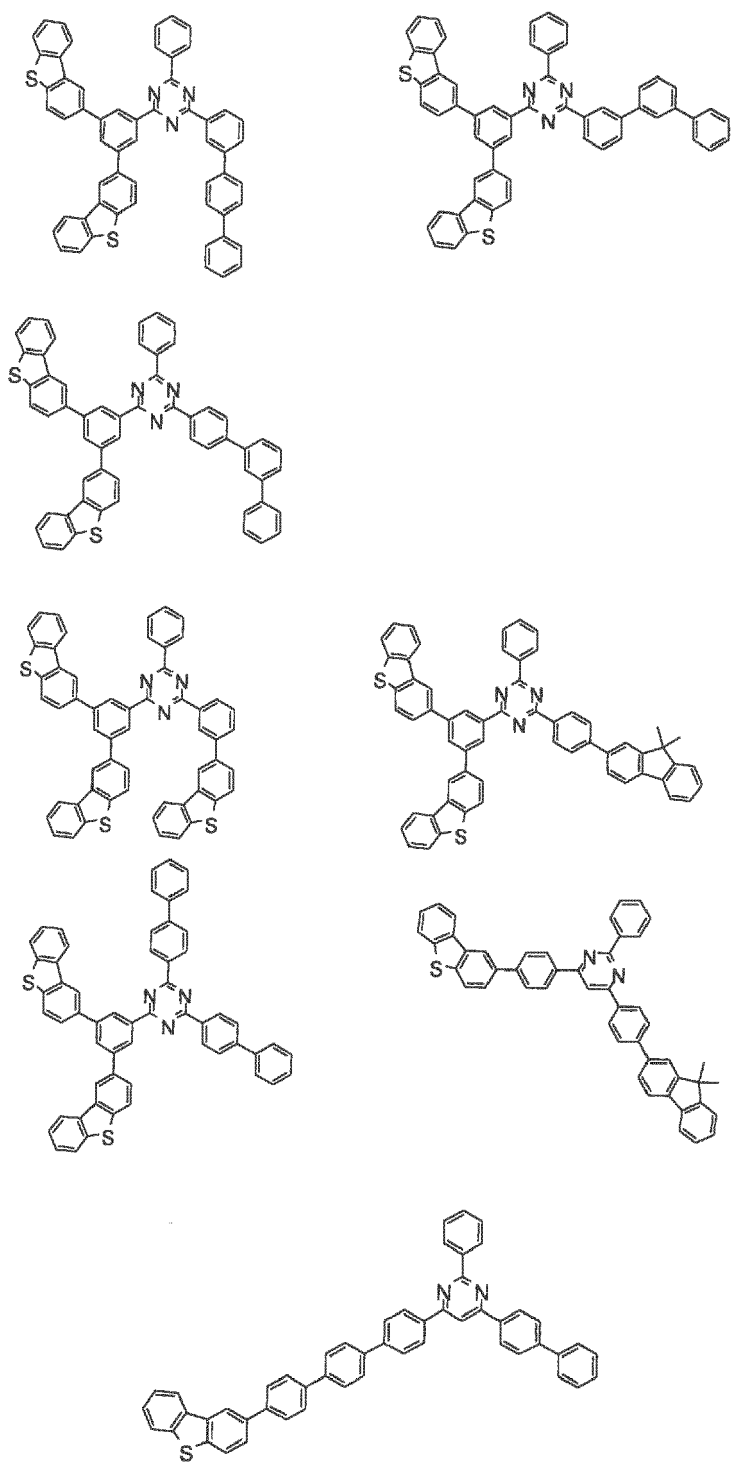


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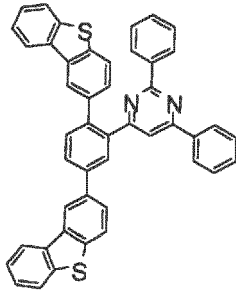
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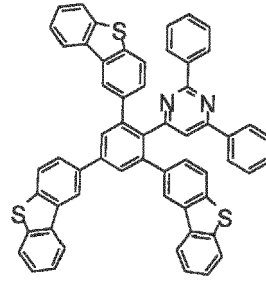
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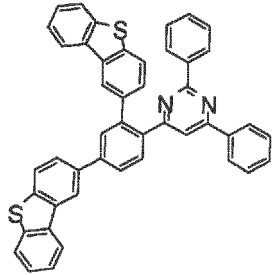
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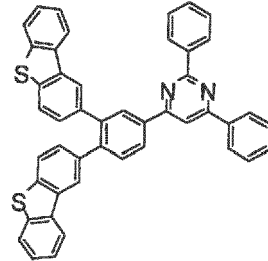
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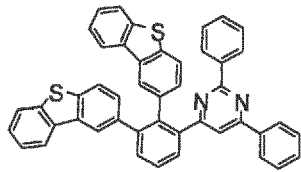
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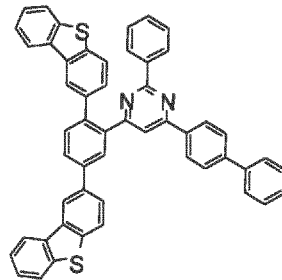
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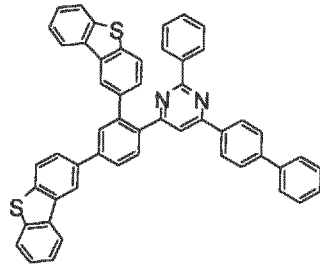
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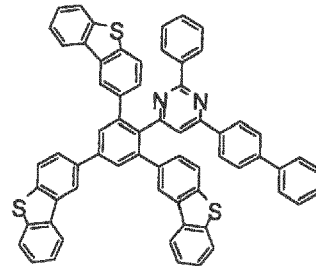
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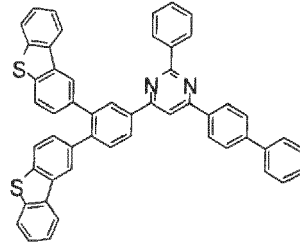
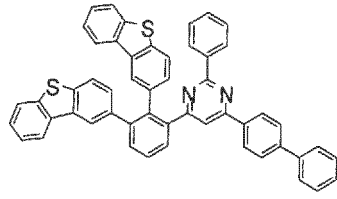


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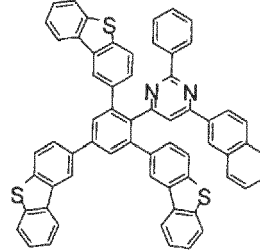
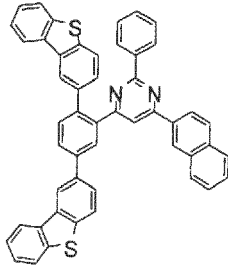
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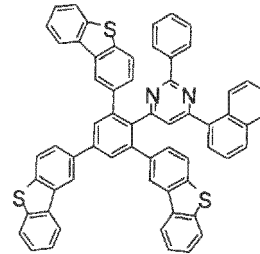
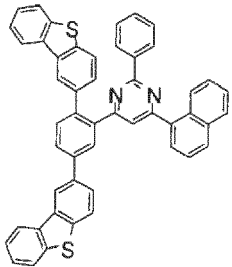


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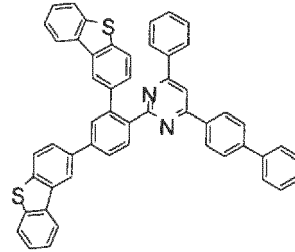
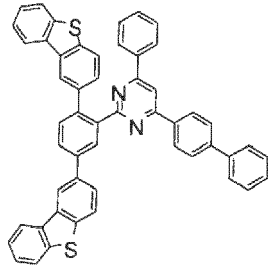
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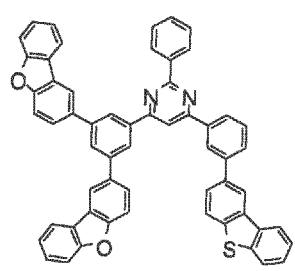
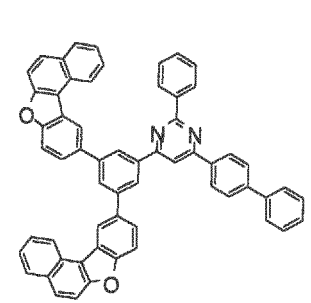
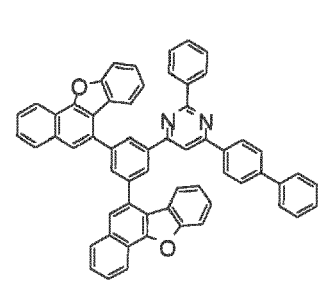
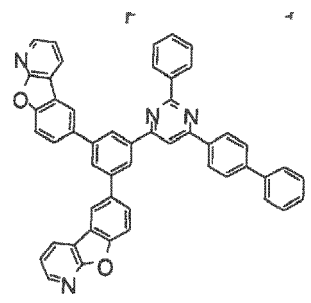
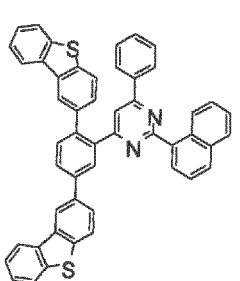
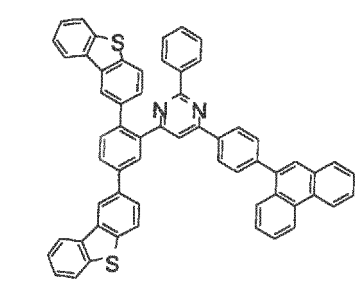
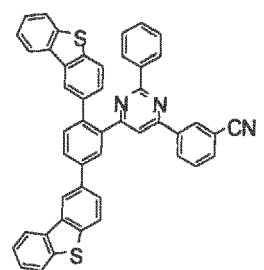
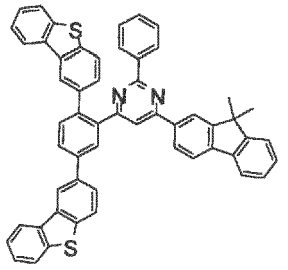
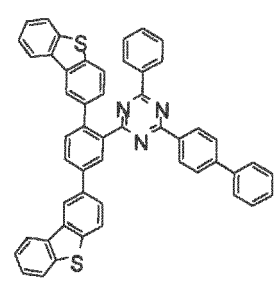
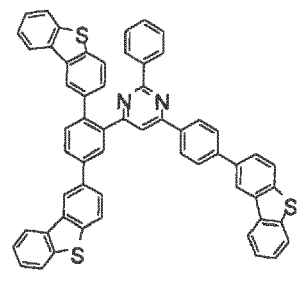
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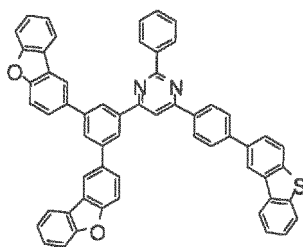
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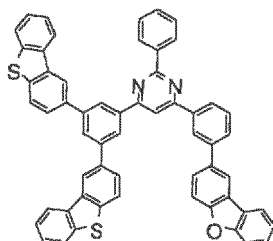
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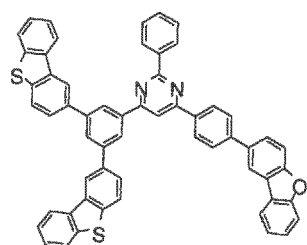


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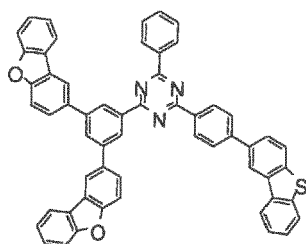
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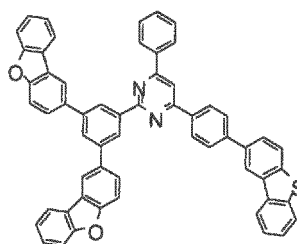


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Organic-EL-Device Material

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[0088] The aromatic heterocyclic derivative of the invention is usable as an organic-EL-device material. The organic-EL-device material may only contain the aromatic heterocyclic derivative of the invention, or alternatively, may further contain another compound. The organic-EL-device material containing the aromatic heterocyclic derivative according to the exemplary embodiment is usable, for instance, as a material for an electron transporting zone, e.g., a material for a blocking layer. It should be noted that, in the specification, the electron transporting zone means any one of an electron transporting layer, an electron injecting layer and a blocking layer, or combination of two or more thereof.

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Organic EL Device

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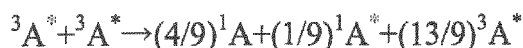
First Exemplary Embodiment

[0089] This exemplary embodiment utilizes TTF phenomenon. The TTF phenomenon will be initially described below.

[0090] Holes and electrons respectively injected from an anode and a cathode are recombined in an emitting layer to generate excitons. As for the spin state, as is conventionally known, singlet excitons account for 25% and triplet excitons account for 75%. In a conventionally known fluorescent device, light is emitted when singlet excitons of 25% are relaxed to the ground state. The remaining triplet excitons of 75% are returned to the ground state without emitting light through a thermal deactivation process. Accordingly, the theoretical limit value of the internal quantum efficiency of a conventional fluorescent device is believed to be 25%.

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[0091] The behavior of triplet excitons generated within an organic substance has been theoretically examined. According to S. M. Bachilo et al. (J. Phys. Chem. A, 104, 7711 (2000)), assuming that high-order excitons such as quintet excitons are quickly returned to triplet excitons, triplet excitons (hereinafter abbreviated as $^3A^*$) collide with one another with an increase in the density thereof, whereby a reaction shown by the following formula occurs. In the formula, 1A represents the ground state and $^1A^*$ represents the lowest singlet excitons.



[0092] In other words, $5^3A^* \rightarrow 4^1A + ^1A^*$, and it is expected that, among triplet excitons initially generated, which account for 75%, one fifth thereof (i.e., 20%) is changed to singlet excitons. Accordingly, the amount of singlet excitons which contribute to emission is 40%, which is a value obtained by adding 15% ($75\% \times (1/5) = 15\%$) to 25%, which is the amount ratio of initially generated singlet excitons. At this time, a ratio of luminous intensity derived from TTF (TTF ratio) relative to the total luminous intensity is 15/40, i.e., 37.5%. Assuming that singlet excitons are generated by collision of initially-generated triplet excitons which account for 75% (i.e., one singlet exciton is generated from two triplet excitons), a significantly high internal quantum efficiency of 62.5% is obtained which is a value obtained by adding 37.5% ($75\% \times (1/2) = 37.5\%$) to 25%, which is the amount ratio of initially generated singlet excitons. At this time, the TTF ratio is 60% ($37.5/62.5$).

[0093] Fig. 1 is schematic view showing one example of an organic EL device according to a first exemplary embodiment of the invention. Fig. 2 is a view showing a relationship between a triplet energy of the emitting layer and a triplet energy of an electron transporting zone in the organic EL device according to the first exemplary embodiment. In the exemplary embodiment, the triplet energy is referred to as a difference between energy in the lowest triplet state and energy in the ground state. The singlet energy (occasionally referred to as energy gap) is referred to as a difference between energy in the lowest singlet state and energy in the ground state.

[0094] An organic EL device 1 shown in Fig. 1 includes an anode 10, a hole transporting zone 60, an emitting layer 20, an electron transporting zone 70, and a cathode 50 in sequential order. These components are adjacent to one another in the organic EL device 1 in the exemplary embodiment. The electron transporting zone 70 in the exemplary embodiment includes a blocking layer 30 and an electron injecting layer 40. It is preferred that the hole transporting zone 60 is interposed between the anode 10 and the emitting layer 20. The hole transporting zone includes at least one of a hole injecting layer and a hole transporting layer.

[0095] In the invention, a simply-called blocking layer means a layer functioning as a barrier against triplet energy. Accordingly, the blocking layer functions differently from a hole blocking layer and a charge blocking layer.

[0096] The emitting layer includes a host material and a dopant material. The dopant material is preferably a dopant material exhibiting fluorescence (hereinafter, also referred to as a fluorescent dopant material). A fluorescent dopant material having a main peak wavelength of 550 nm or less is preferable. A fluorescent dopant material having a main peak wavelength of 500 nm or less is more preferable. A main peak wavelength means a peak wavelength of luminescence spectrum exhibiting a maximum luminous intensity among luminous spectra measured in a toluene solution with a concentration from 10^{-5} mol/liter to 10^{-6} mol/liter. The main peak wavelength of 550 nm is substantially equivalent to a green emission. In this wavelength zone, improvement in luminous efficiency of a fluorescent device utilizing the TTF phenomenon is desired. In a blue-emitting fluorescent device of 480 nm or less, further improvement in luminous efficiency is expectable. In a red-emitting fluorescent device of 550 nm or more, a phosphorescent device exhibiting a high internal quantum efficiency has already been at a practical level. Accordingly, improvement in luminous efficiency as a fluorescent device is not desired.

[0097] In Fig. 2, the holes injected from the anode are injected to the emitting layer via the hole transporting zone. The electrons injected from the cathode are injected to the emitting layer via the electron injecting layer and the blocking layer. Subsequently, the holes and the electrons are recombined in the emitting layer to generate singlet excitons and triplet excitons. There are two manners as for the occurrence of recombination: recombination may occur either on host material molecules or on dopant material molecules.

[0098] In this exemplary embodiment, as shown in Fig. 2, when the triplet energy of the host material and that of the dopant material are respectively taken as E_h^T and E_d^T , a relationship of the following formula (2A) is satisfied.

$$E_h^T < E_d^T \quad \cdot \cdot \cdot \quad (2A)$$

[0099] When this relationship of the formula (2A) is satisfied, triplet excitons generated by recombination on the host material do not transfer to the dopant material which has a higher triplet energy, as shown in Fig. 3. Triplet excitons generated by recombination on dopant material molecules quickly energy-transfer to host material molecules. In other words, triplet excitons on the host material do not transfer to the dopant material but collide with one another efficiently

on the host material to generate singlet excitons by the TTF phenomenon. Moreover, since the singlet energy E_d^S of the dopant material is smaller than the singlet energy E_h^S of the host material, a relationship of the following formula (2B) is satisfied.

$$E_d^S < E_h^S \quad \cdot \cdot \cdot (2B)$$

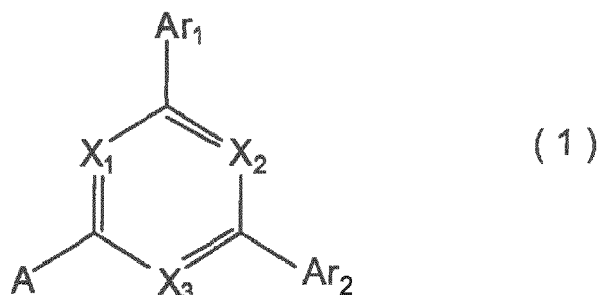
[0100] Since the relationship of the formula (2B) is satisfied, the singlet excitons generated by the TTF phenomenon energy-transfer from the host material to the dopant material, thereby contributing to fluorescence of the dopant material.

In the dopant material which is usually used in a fluorescent device, transition from the triplet state to the ground state should be inhibited. In such a transition, triplet excitons are not optically energy-deactivated, but are thermally energy-deactivated. By causing the triplet energy of a host material and the triplet energy of a dopant material to satisfy the above-mentioned relationship, singlet excitons are generated efficiently due to the collision of triplet excitons before they are thermally deactivated, whereby luminous efficiency is improved. As a consequence, luminous efficiency is improved.

[0101] In the exemplary embodiment, the blocking layer is adjacent to the emitting layer. The blocking layer has a function of preventing triplet excitons generated in the emitting layer to be diffused to the electron transporting zone and confining the triplet excitons within the emitting layer to increase a density of the triplet excitons therein, thereby causing the TTF phenomenon efficiently.

[0102] The blocking layer also serves for efficiently injecting the electrons to the emitting layer. When the electron injecting properties to the emitting layer are deteriorated, the density of the triplet excitons is decreased since the electron-hole recombination in the emitting layer is decreased. When the density of the triplet excitons is decreased, the frequency of collision of the triplet excitons is reduced, whereby the TTF phenomenon does not occur efficiently.

[0103] The blocking layer of the organic EL device according to the exemplary embodiment contains an aromatic heterocyclic derivative represented by a formula (1) below (not according to the invention).

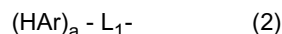


[0104] In the formula (1), X_1 to X_3 are a nitrogen atom or CR_1 .

[0105] However, at least one of X_1 to X_3 is a nitrogen atom.

[0106] R_1 independently represents a hydrogen atom, a halogen atom, a cyano group, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms, a substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, a substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, a substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, and a substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms.

[0107] In the formula (1), A is represented by a formula (2) below.



[0108] In the formula (2), HAr is represented by a formula (3) below.

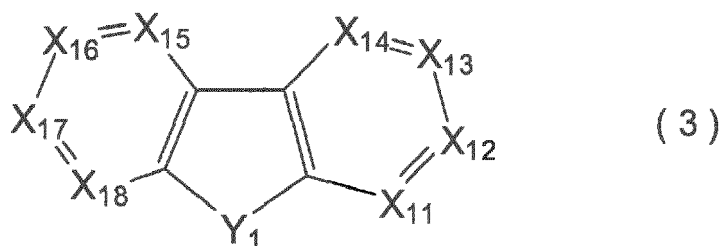
[0109] In the formula (2), a is an integer of 1 to 5.

[0110] When a is 1, L_1 is a single bond or a divalent linking group.

[0111] When a is in a range of 2 to 5, L_1 is a trivalent to hexavalent linking group and HAr is the same or different.

[0112] The linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms, or a residue having 2 to 6 valences induced from any one of a group formed by bonding two or three of the above groups.

[0113] The mutually bonded groups are the same or different.



10 **[0114]** In the formula (3), X_{11} to X_{18} each are independently a nitrogen atom, CR_{13} or a carbon atom bonded to L_1 by a single bond.

[0115] In the formula (3), Y_1 is a nitrogen atom, a sulfur atom, $SiR_{11}R_{12}$ or a silicon atom bonded to each of R_{11} and L_1 by a single bond.

[0116] However, L_1 is bonded by one of a carbon atom at X_{11} to X_{18} and R_{11} to R_{12} and a silicon atom at Y_1 .

15 **[0117]** R_{11} and R_{12} represent the same as R_1 in the formula (1). R_{11} and R_{12} are the same or different.

[0118] R_{13} independently represents a hydrogen atom, a halogen atom, a cyano group, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms, a substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, a substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, a substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms. A plurality of R_{13} are mutually the same or different. Adjacent R_{13} may bond to each other to form a ring.

20 **[0119]** In the above formula (1), Ar_1 and Ar_2 each are independently represented by the formula (2), or represent a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms or a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms.

[0120] In the formula (3), X_{13} or X_{16} is preferably a carbon atom bonded to L_1 by a single bond. Alternatively, in the formula (3), X_{11} or X_{18} is preferably a carbon atom bonded to L_1 by a single bond.

[0121] In the formula (2), a is an integer in a range of 1 to 5, more preferably of 1 to 3, particularly preferably 1 or 2.

30 **[0122]** When a is 1, L_1 is a single bond or a divalent linking group and the formula (2) is represented by a formula (2-1) below.

[0123] When a is in a range of 2 to 5, L_1 is a trivalent to hexavalent linking group. When a is 2, the formula (2) is represented by a formula (2-2) below. At this time, HAr is the same or different.



[0124] The linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms, or a divalent or trivalent residue induced from any one of a group formed by bonding two or three of the above groups.

[0125] The group formed by bonding two or three of the above groups means a group formed by bonding, with a single bond, two or three of the divalent or trivalent residue induced from the aryl group having 6 to 30 ring carbon atoms and the heterocyclic group having 5 to 30 ring atoms. In the linking group, the mutually bonded groups are the same or different.

50 **[0126]** In the formulae (2), (2-1) and (2-2), L_1 is preferably a linking group. The linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms, or a divalent or trivalent residue induced from any one of a group formed by bonding two or three of the above groups. Further, the linking group is preferably a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms.

55 **[0127]** In the formula (2), it is preferable that when a is 1 (see the formula (2-1)) and L_1 is a linking group, the linking group is a divalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a divalent residue of a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms. More specifically, L_1 is preferably a divalent residue induced from any one of benzene, biphenyl, terphenyl, naphthalene and phenanthrene.

[0128] In the formula (2), it is preferable that when a is 2 (see the formula (2-2)) and L_1 is a linking group, the linking group is a trivalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a trivalent residue of a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms. More specifically, L_1 is preferably a divalent residue induced from any one of benzene, biphenyl, terphenyl, naphthalene and phenanthrene.

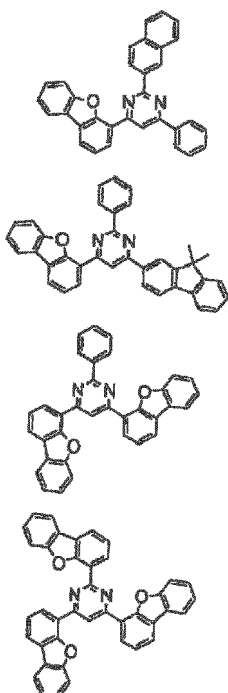
[0129] In the formula (3), Y_1 is preferably an oxygen atom or a sulfur atom. Moreover, in the formula (3), it is preferable that Y_1 is an oxygen atom or a sulfur atom, one of X_{11} to X_{18} is a carbon atom bonded to L_1 by a single bond and the rest of X_{11} to X_{18} are CR_{13} .

[0130] In the formula (1), two or three of X_1 to X_3 are more preferably a nitrogen atom.

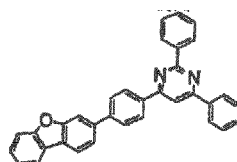
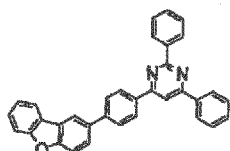
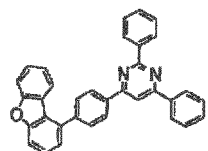
[0131] In the formula (2), L_1 is preferably a divalent or trivalent residue induced from any one of benzene, biphenyl, terphenyl, naphthalene and phenanthrene.

[0132] Specific examples of each group for Ar^1 , Ar^2 , L_1 , R_1 , R_{11} to R_{13} , X_{11} to X_{18} and Y_1 in the formulae (1) to (3) and (2-1) to (2-2) are the groups described in relation to the aromatic heterocyclic derivative represented by the formula (4).

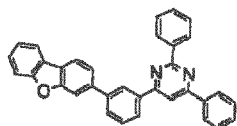
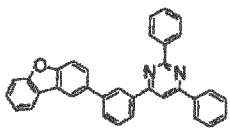
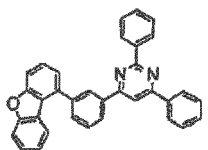
[0133] Examples of specific structures of the aromatic heterocyclic derivative represented by the formula (4) and contained in the blocking layer of the organic EL device according to this embodiment are as follows.



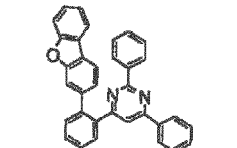
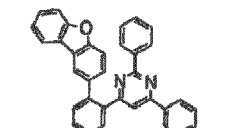
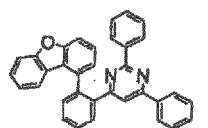
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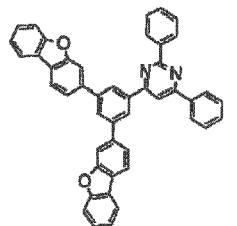
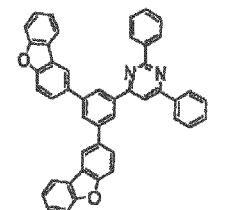
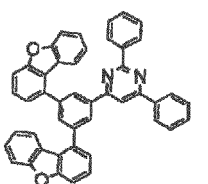
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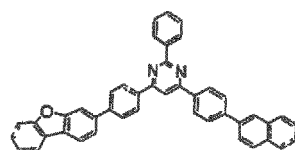
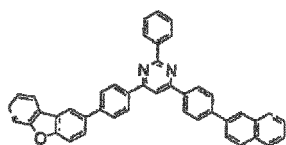
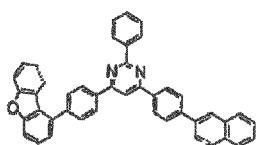
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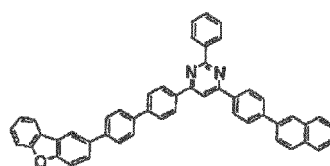
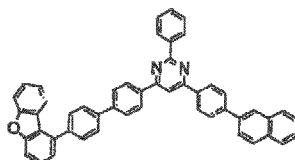
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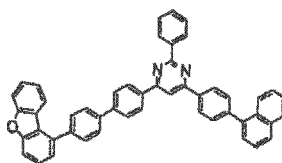
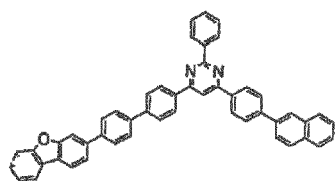
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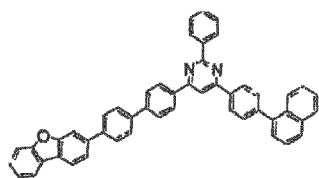
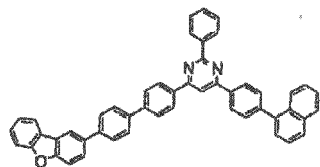
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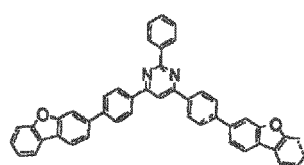
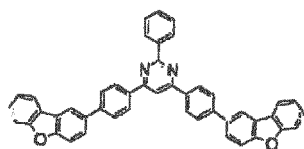
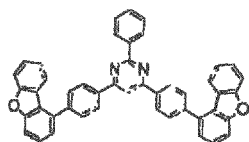
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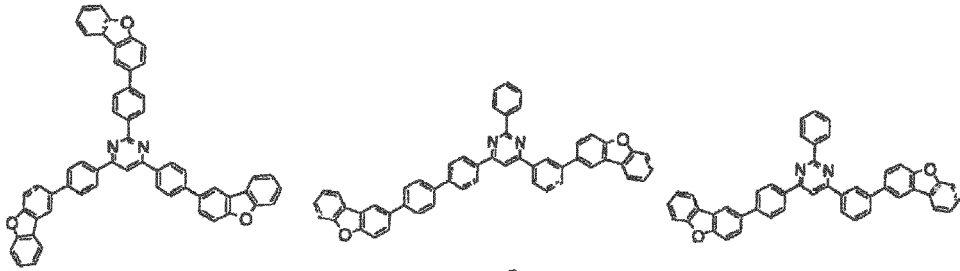


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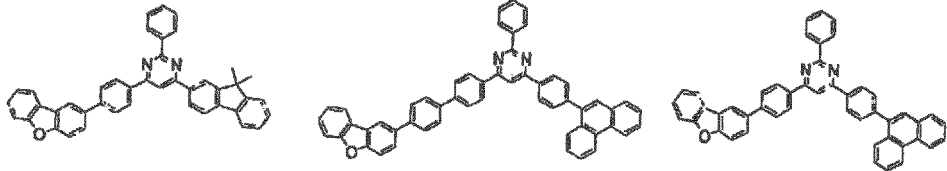


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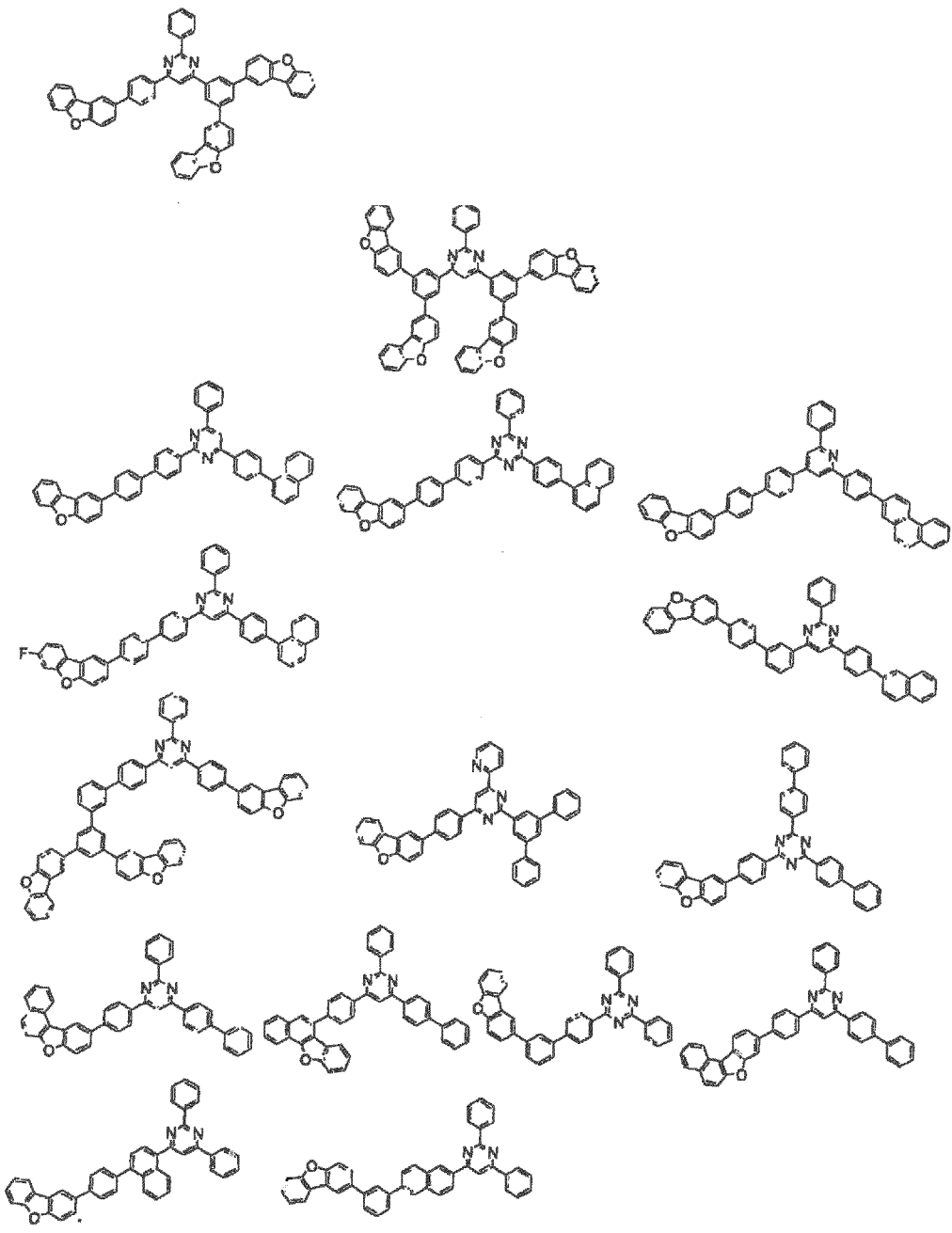
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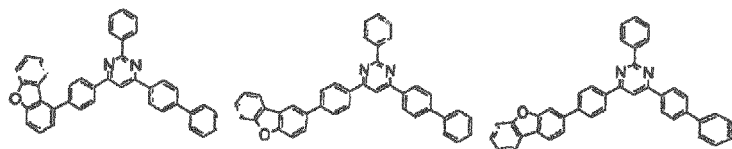
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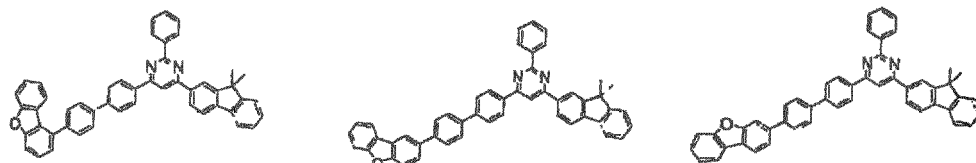
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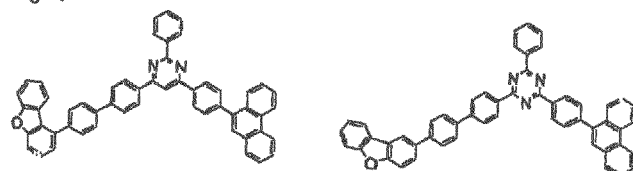
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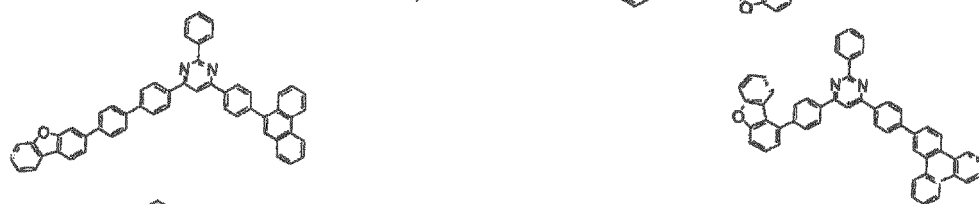
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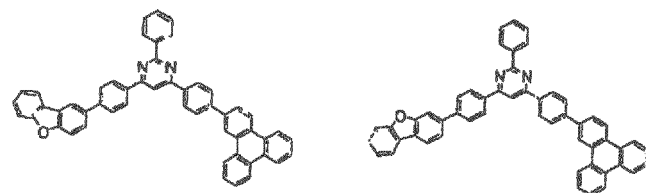
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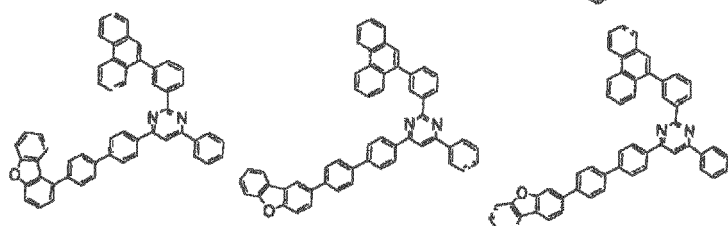
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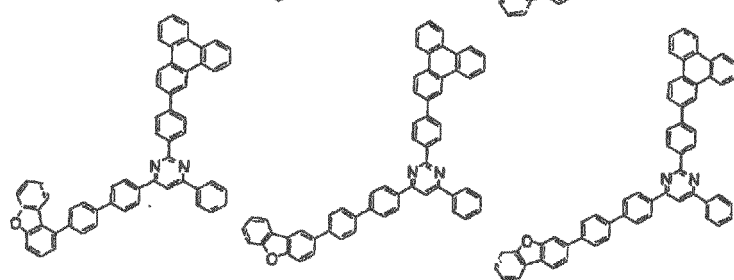


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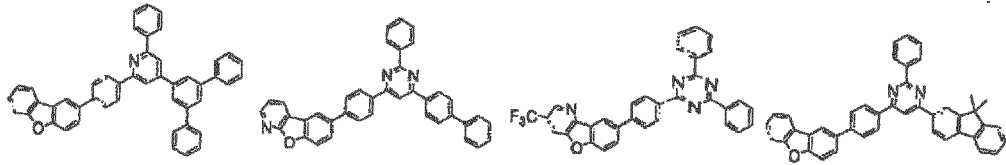


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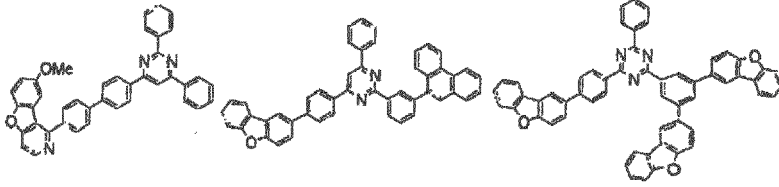
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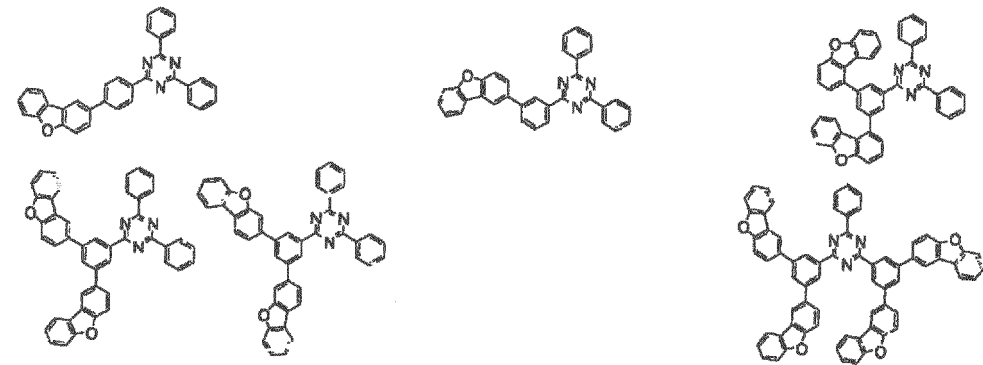
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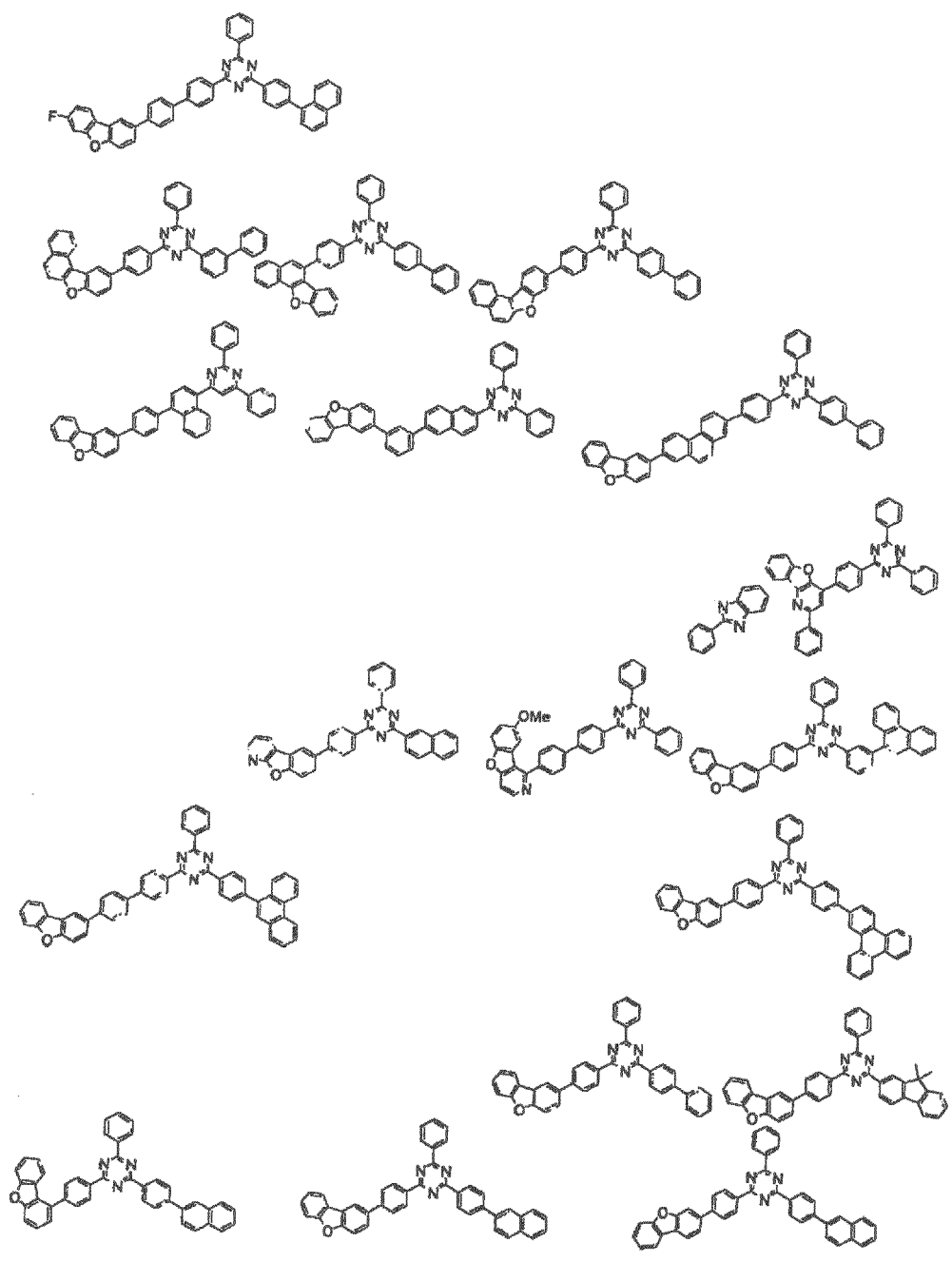
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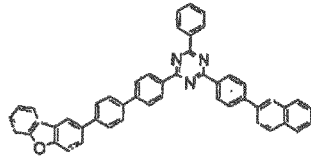
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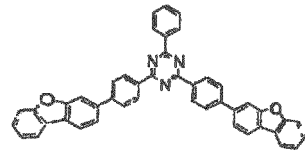
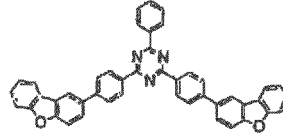
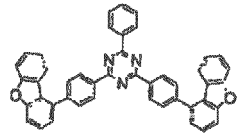
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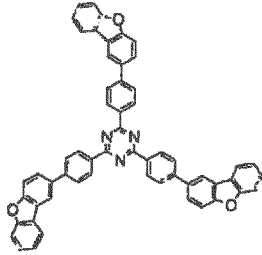
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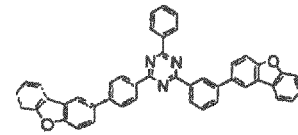
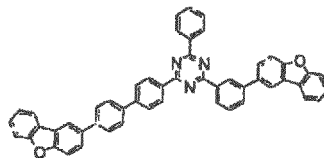
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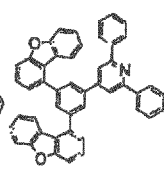
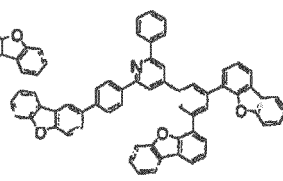
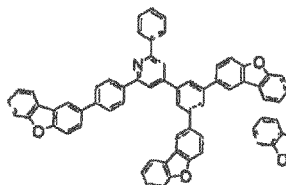
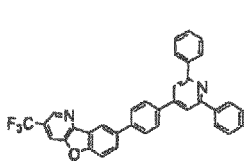
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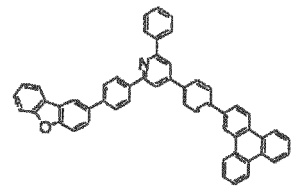
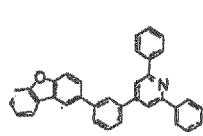
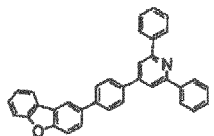
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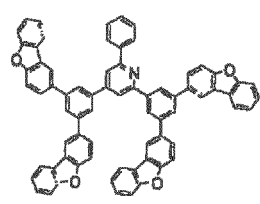
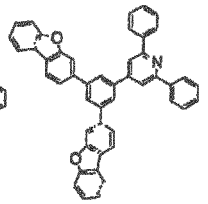
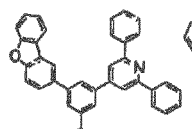
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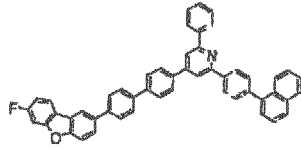
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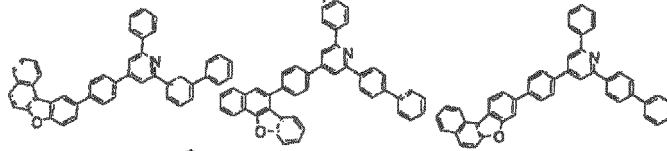
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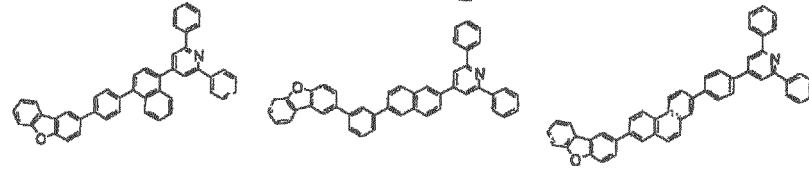
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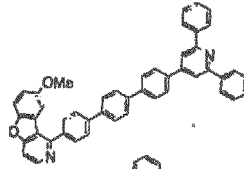
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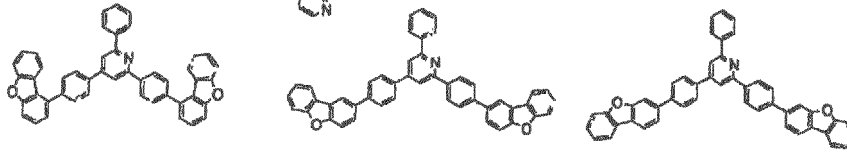
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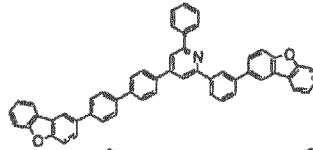
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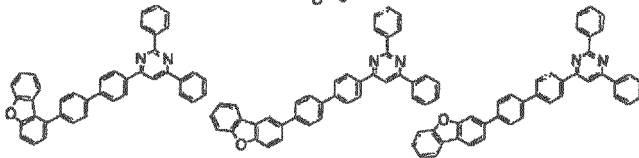
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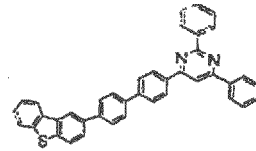
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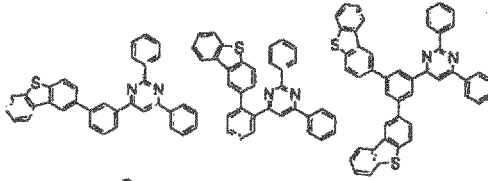
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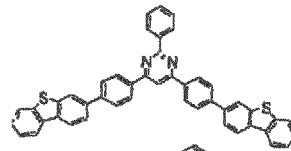
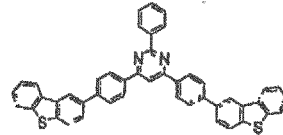
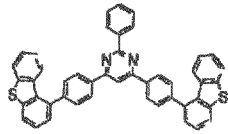
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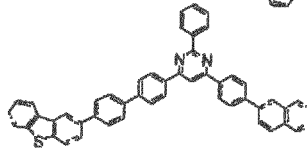
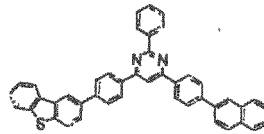
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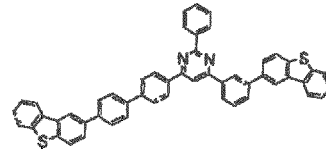
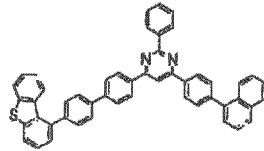
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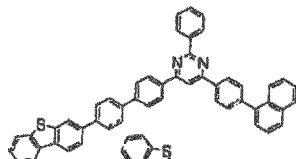
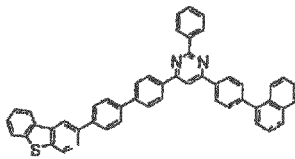
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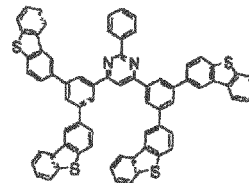
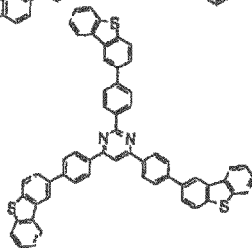
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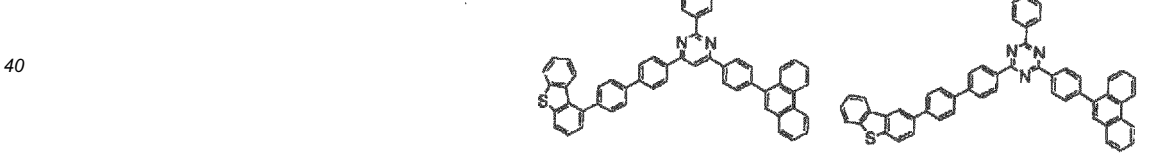
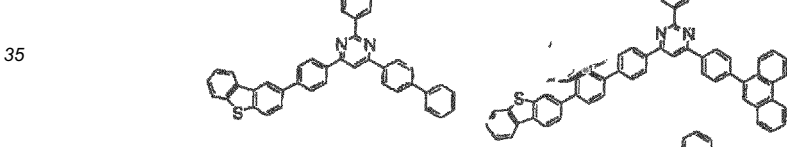
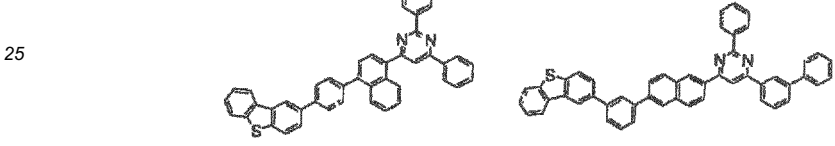
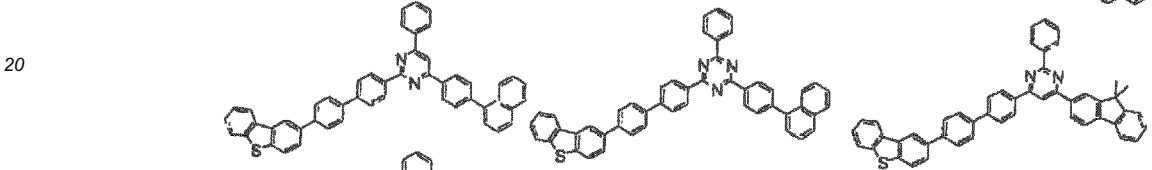
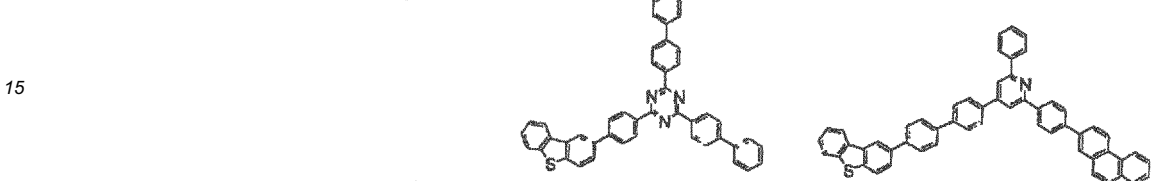
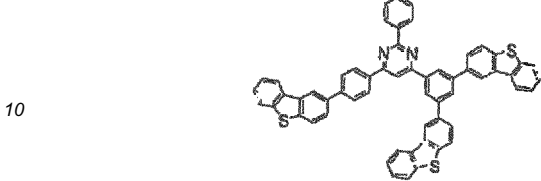
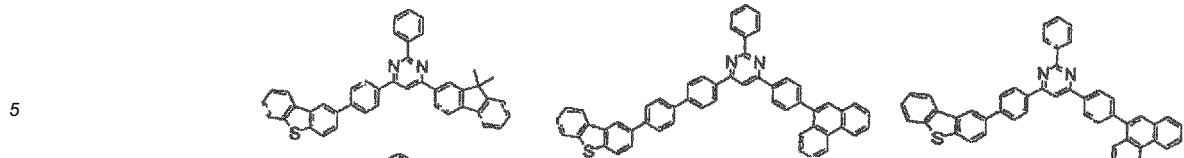


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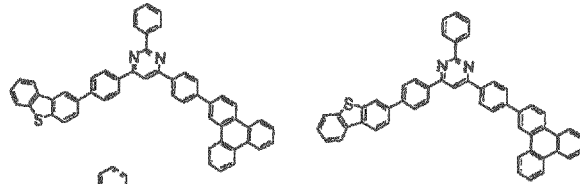
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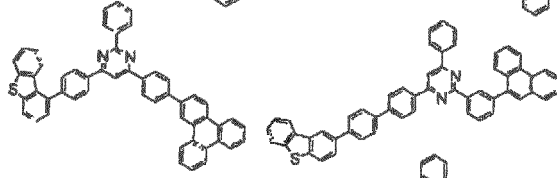
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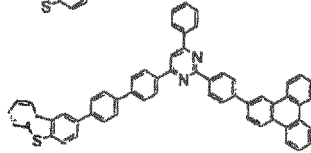
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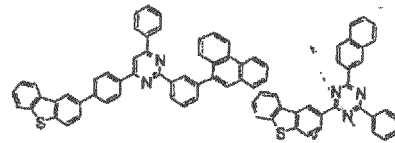
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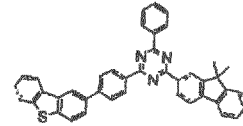
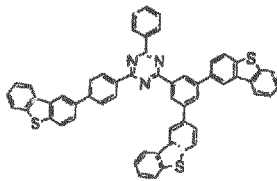
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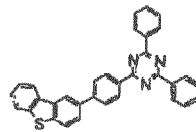
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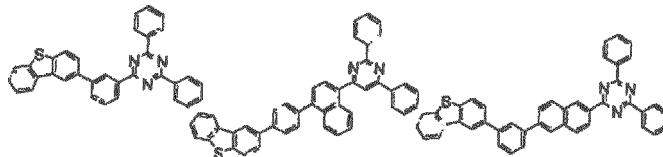
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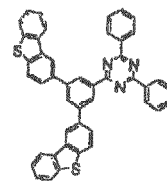
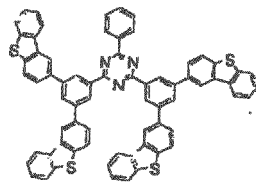
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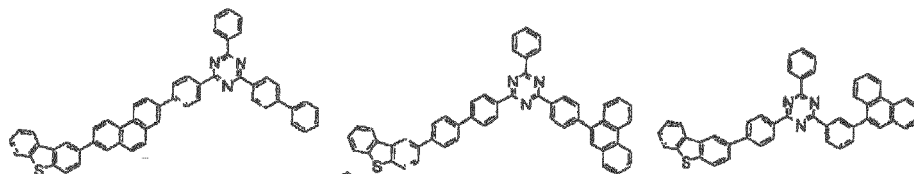


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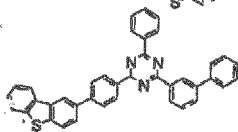
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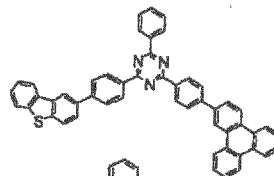
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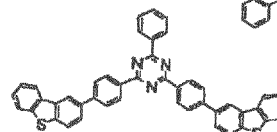
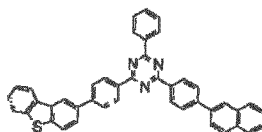
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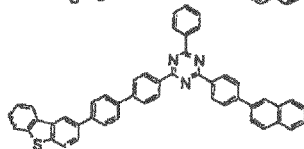
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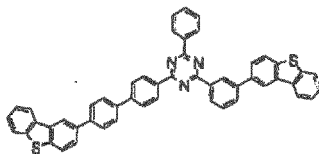
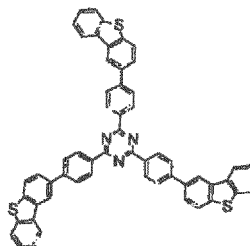
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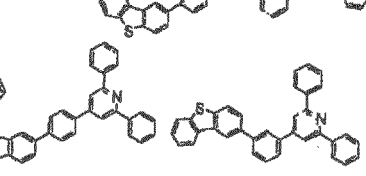
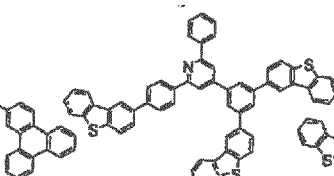
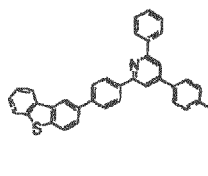
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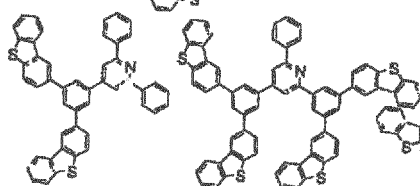
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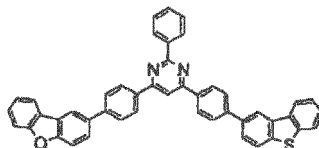


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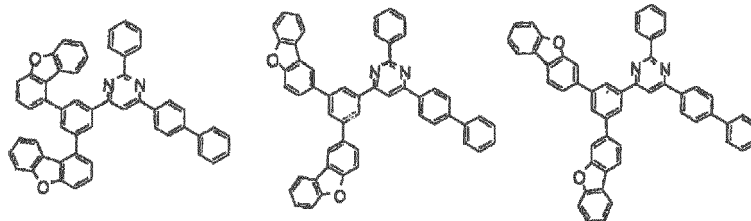
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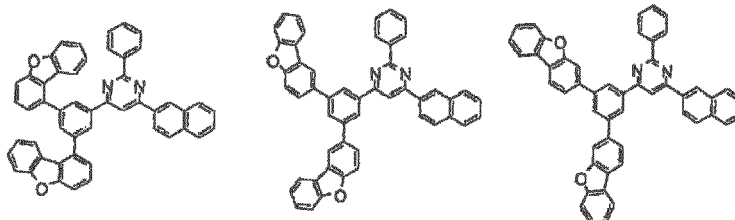


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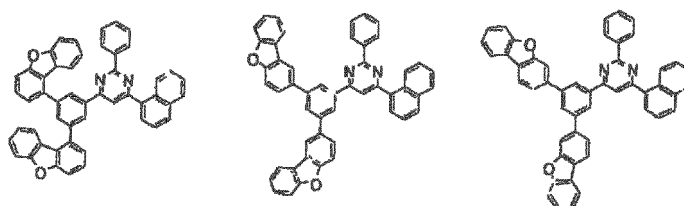


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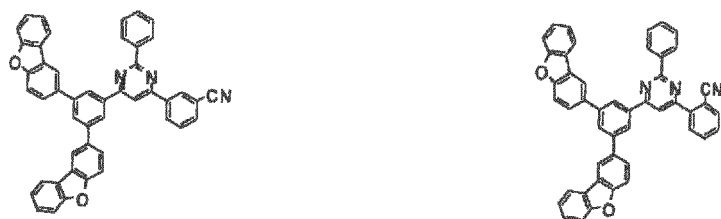
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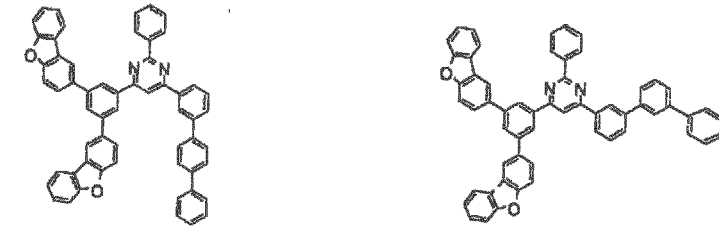
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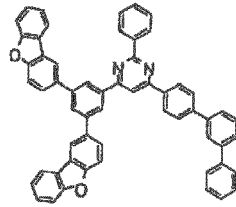
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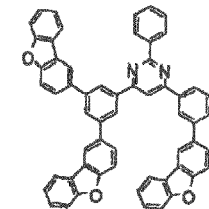


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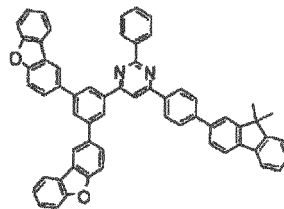


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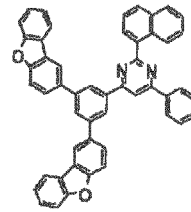
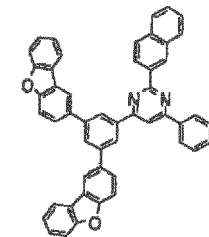
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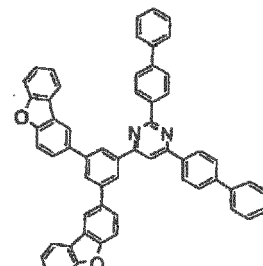
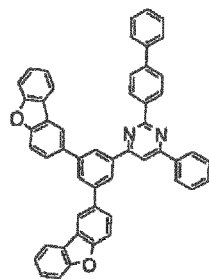


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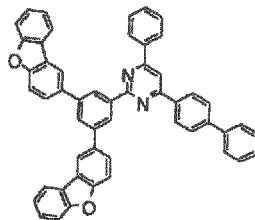
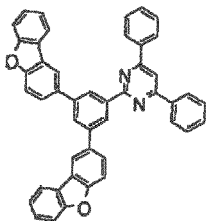
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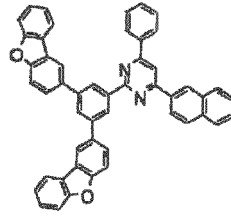
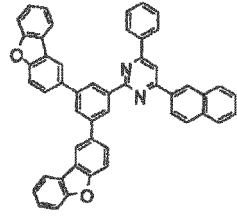
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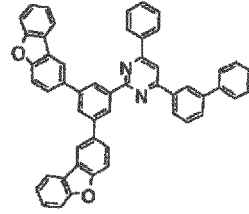
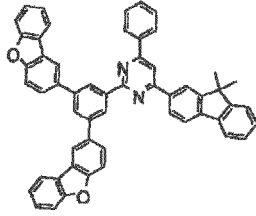


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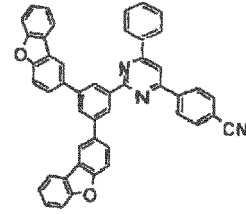
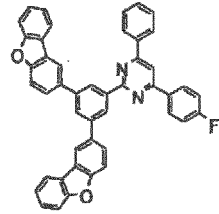


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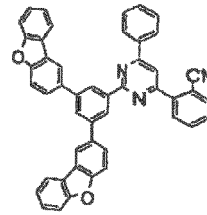
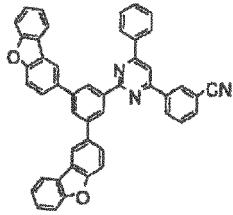
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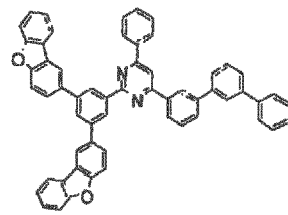
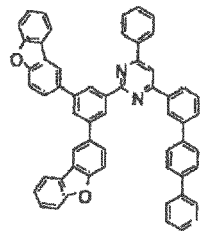
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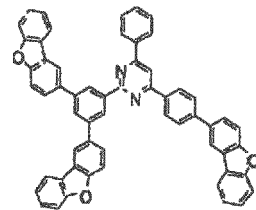
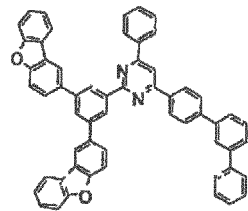
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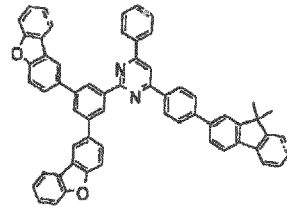
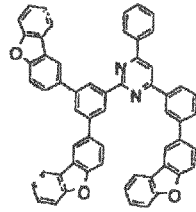
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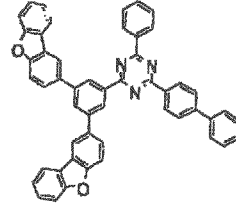
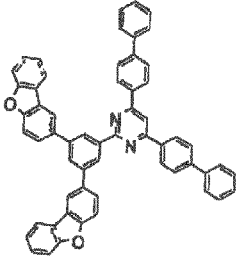


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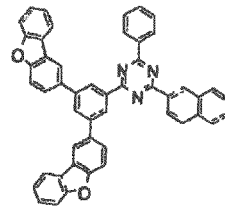
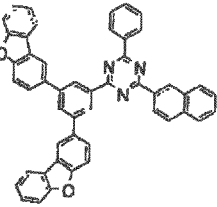


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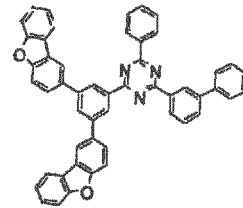
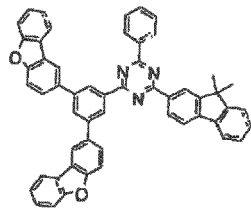


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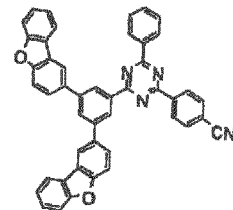
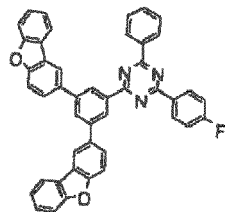


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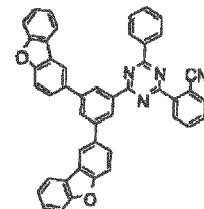
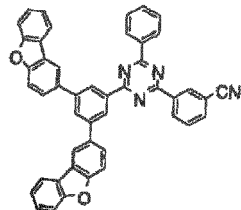
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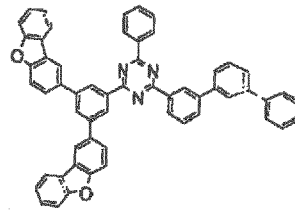
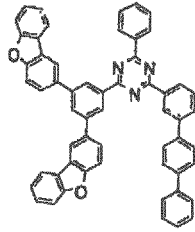
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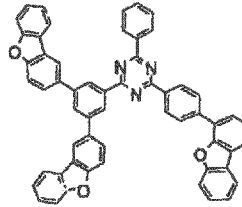
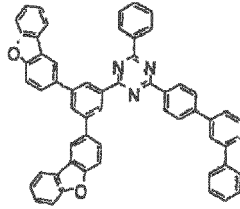
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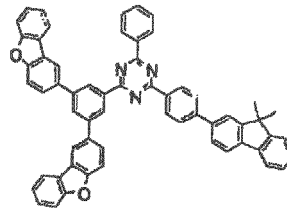
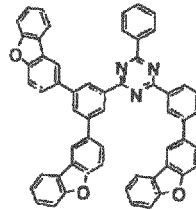


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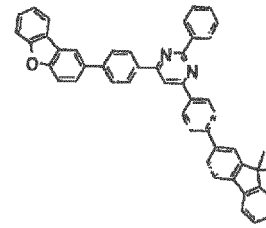
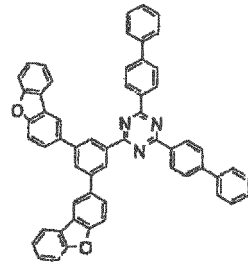
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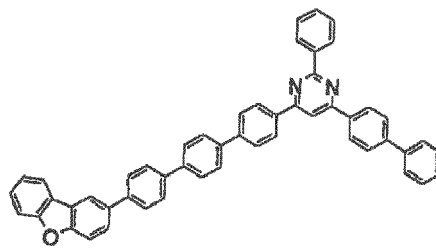
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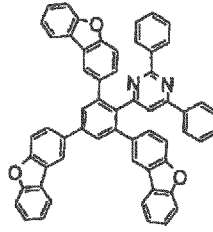
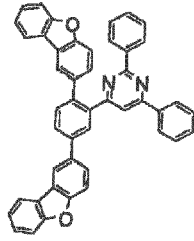


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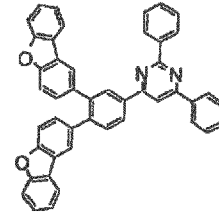
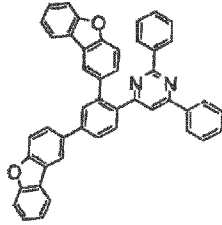
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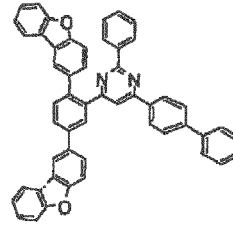
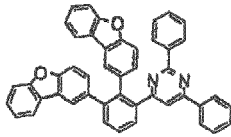


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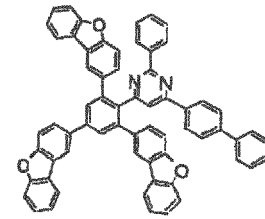
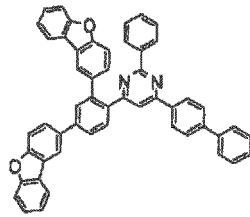
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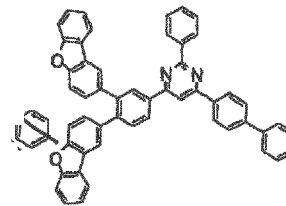
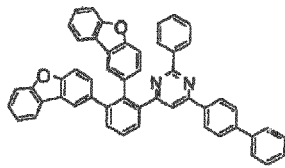
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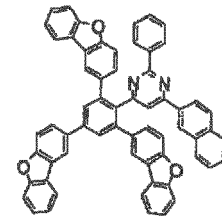
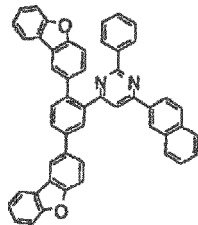
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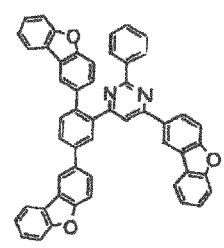
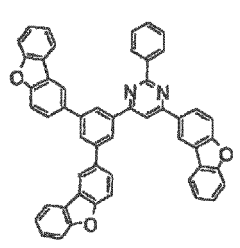
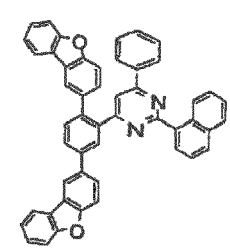
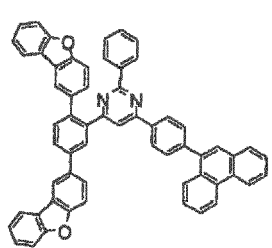
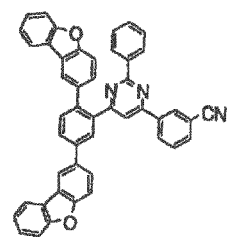
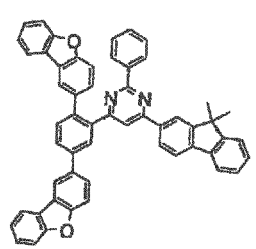
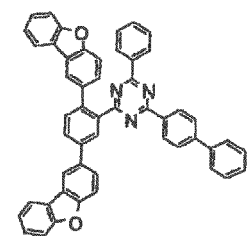
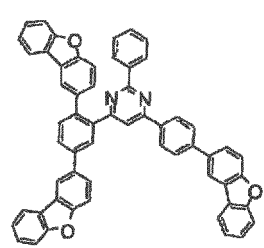
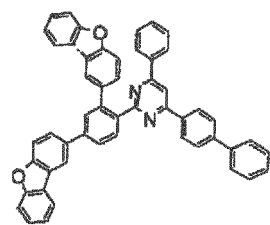
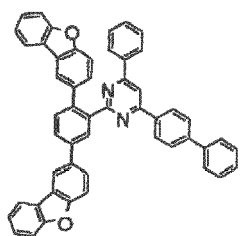
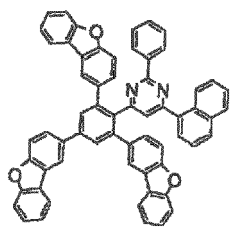
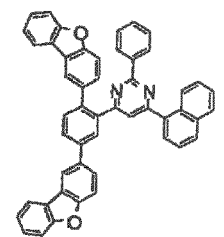
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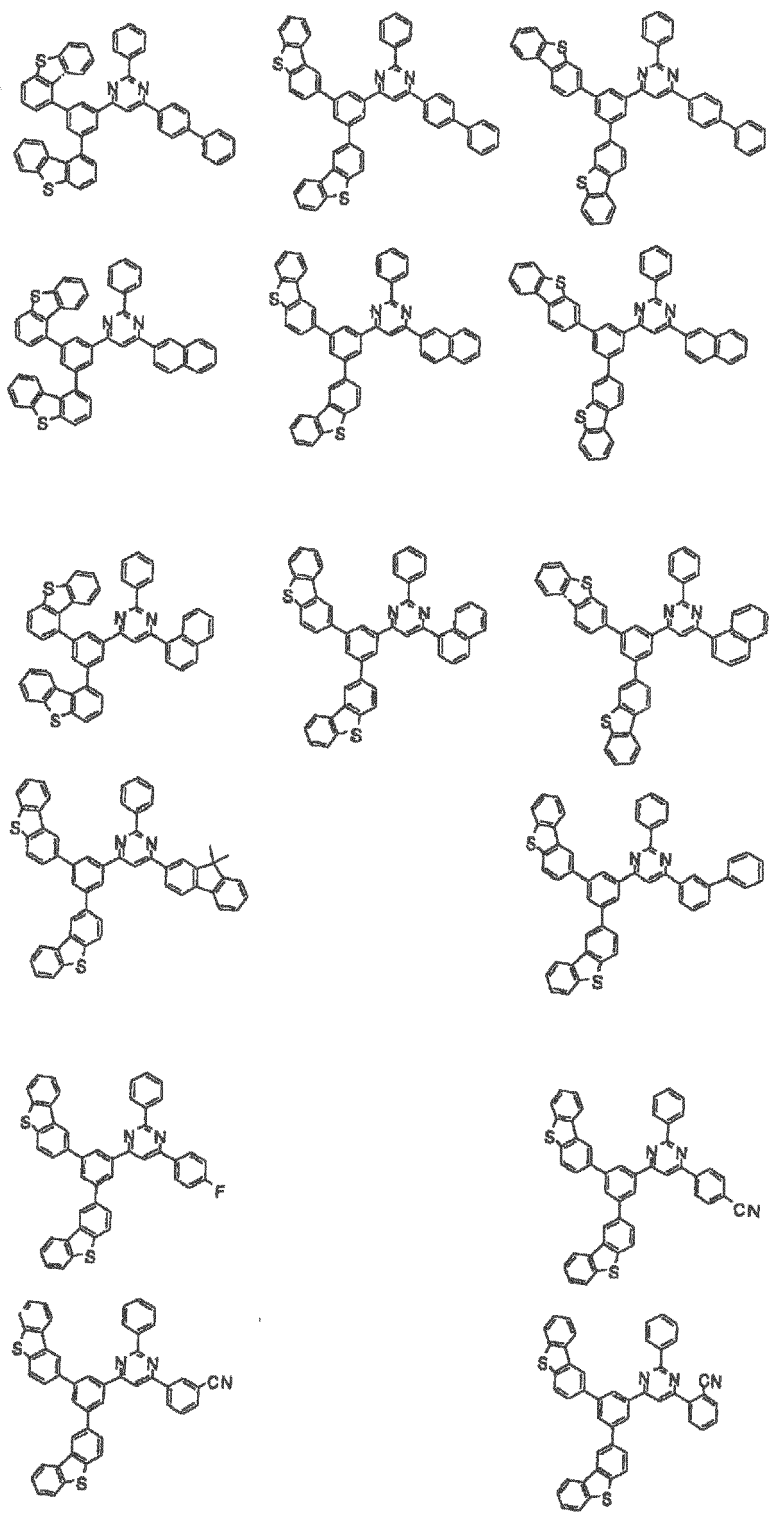


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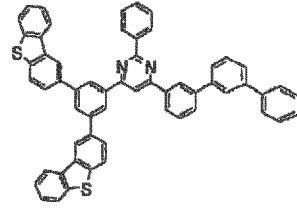
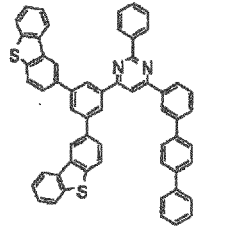
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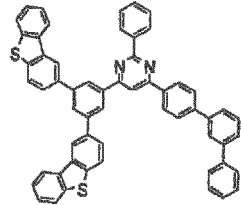
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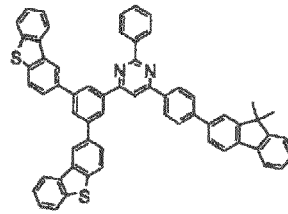
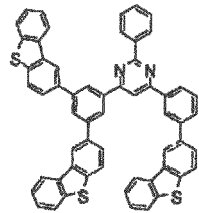


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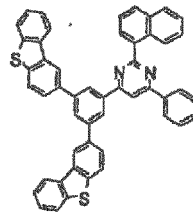
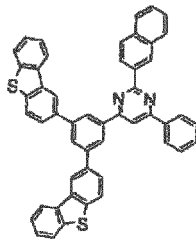
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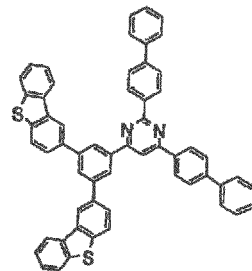
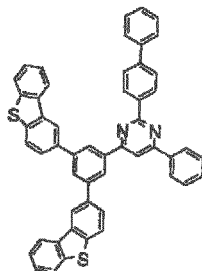
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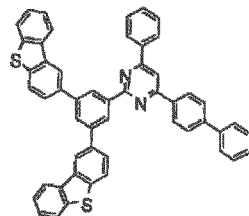
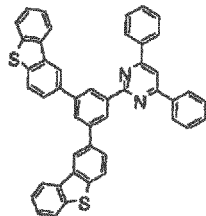
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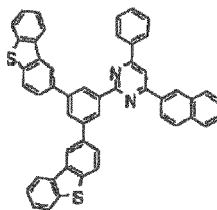
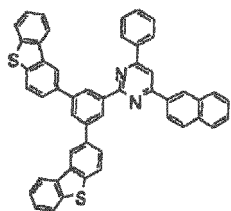
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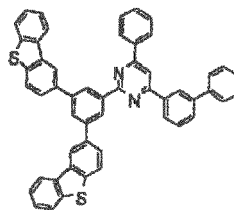
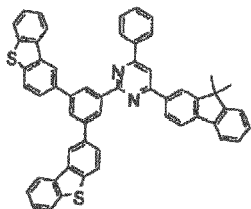


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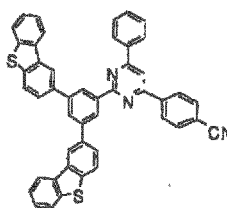
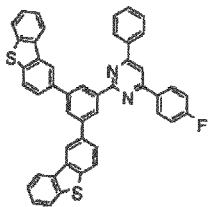


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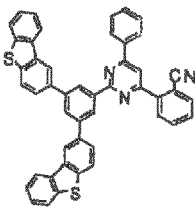
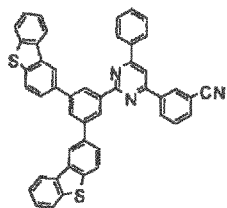
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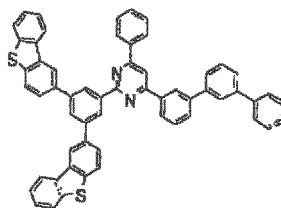
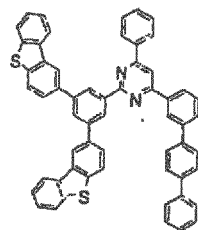
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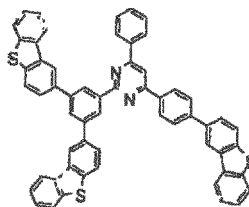
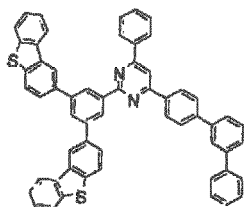
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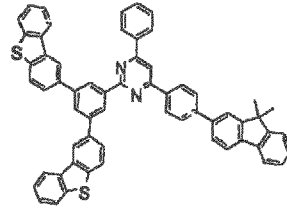
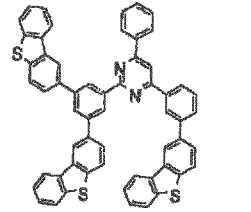
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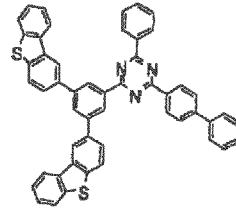
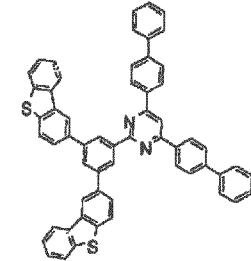


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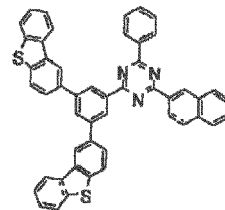
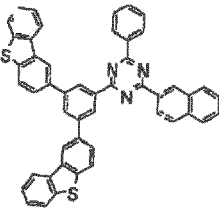


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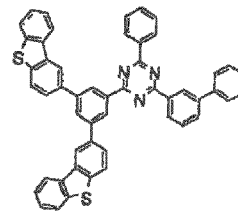
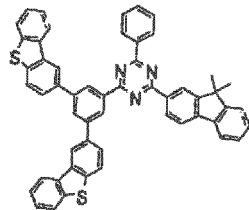


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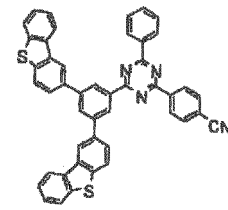
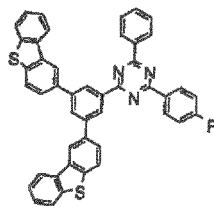


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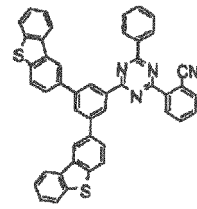
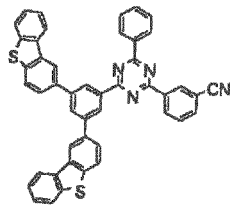
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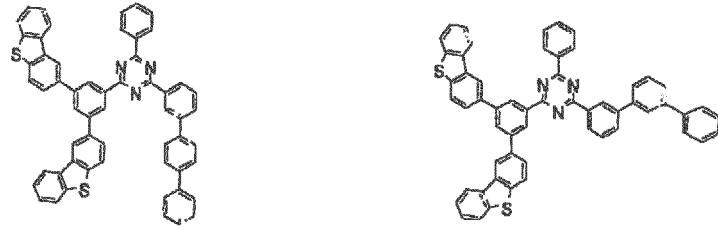
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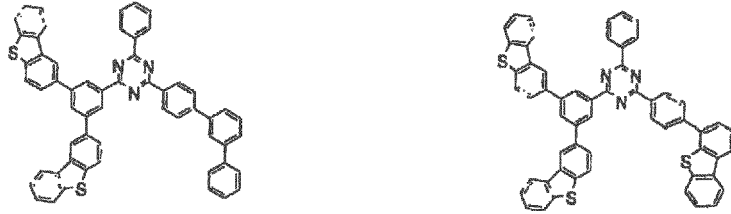
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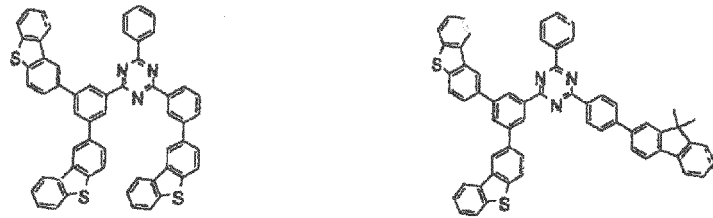


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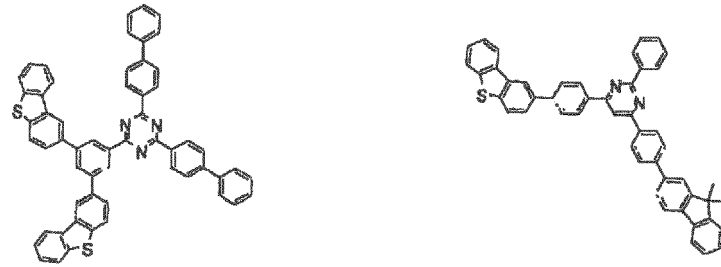
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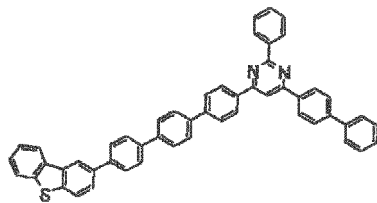
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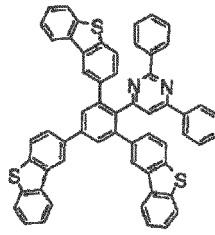
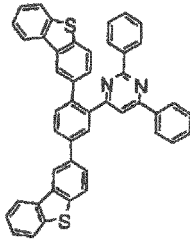


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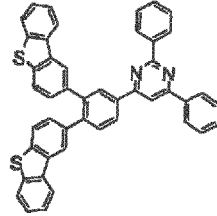
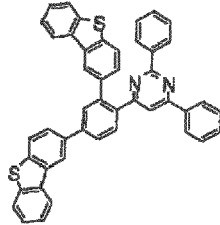
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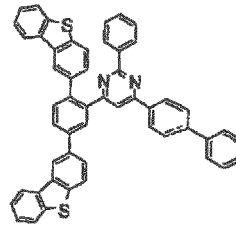
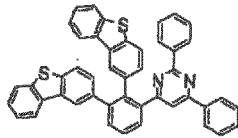


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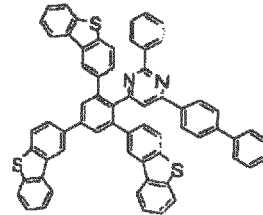
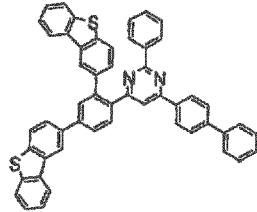
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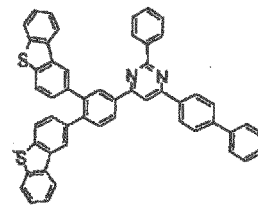
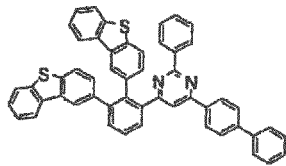
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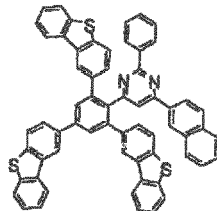
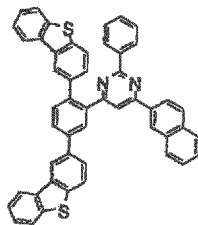
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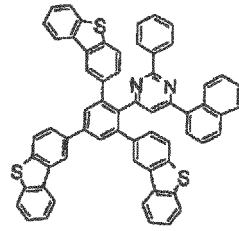
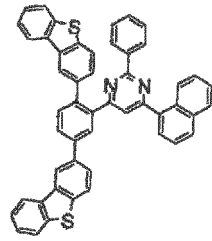
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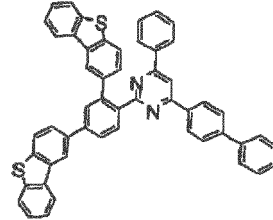
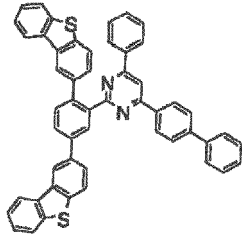


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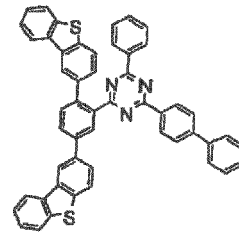
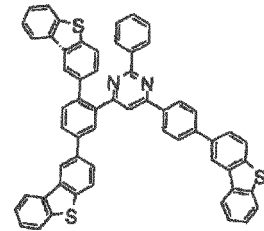


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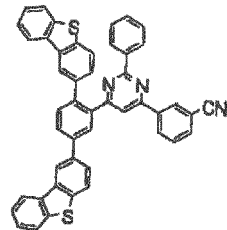
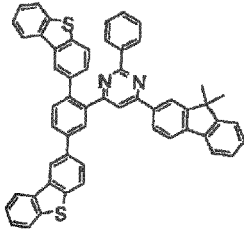
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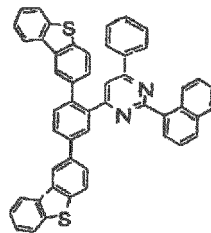
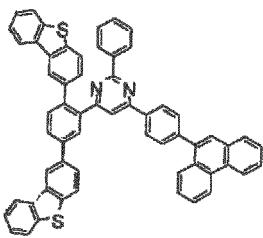
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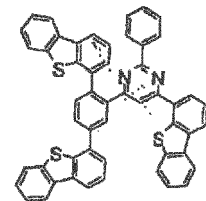
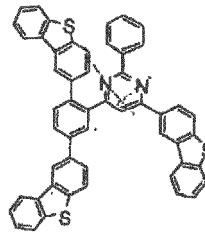
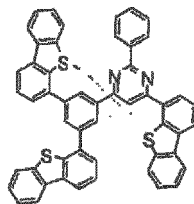
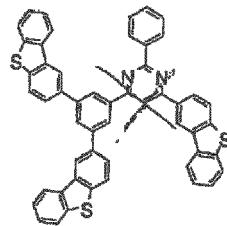
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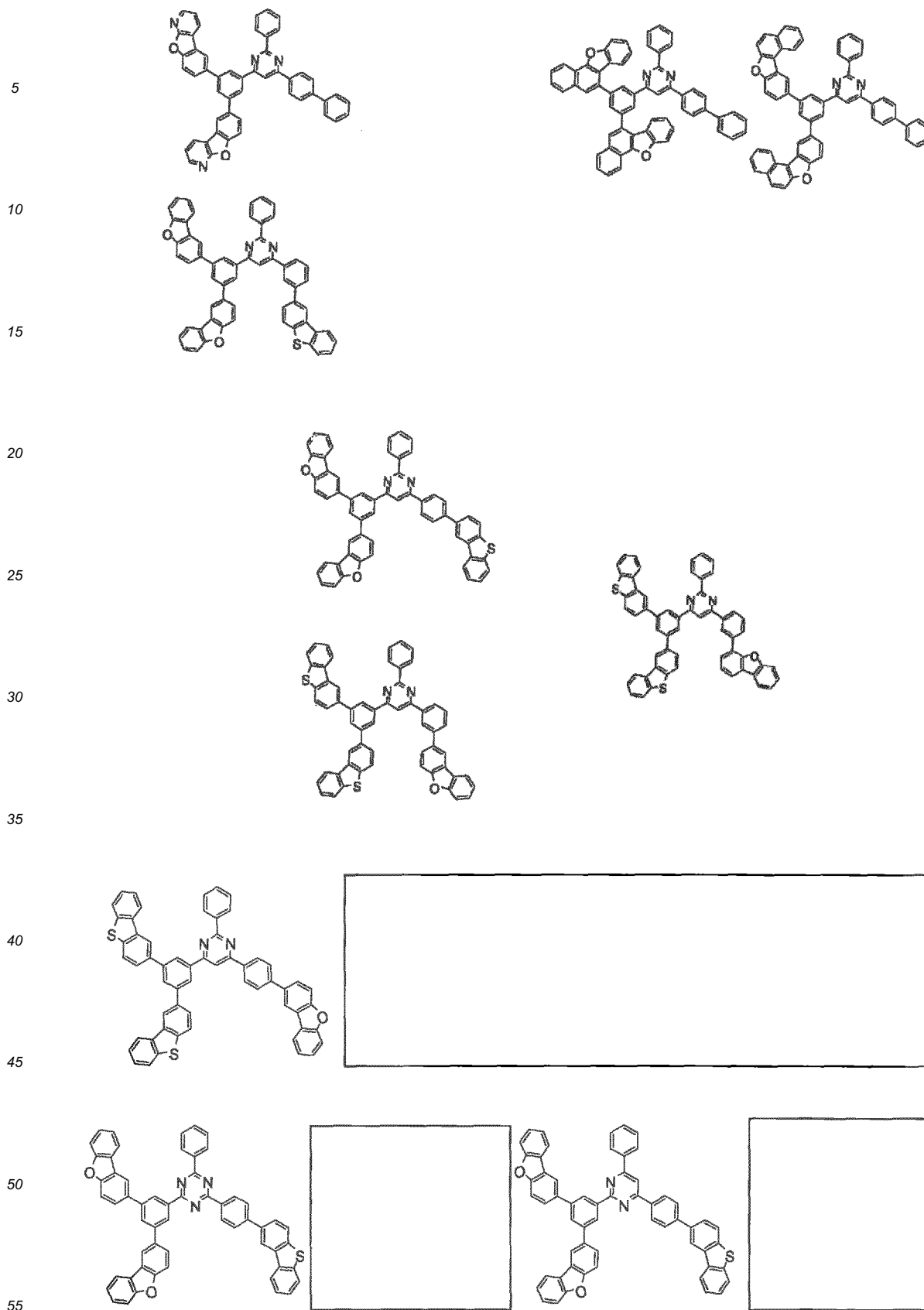


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[0134] The organic EL device 1 in the exemplary embodiment includes the electron injecting layer 40 between the blocking layer 30 and the cathode 50 as described above. The electron injecting layer 40 preferably contains the aromatic

heterocyclic derivative represented by the formula (4). Herein, the aromatic heterocyclic derivative contained in the blocking layer 30 and the aromatic heterocyclic derivative contained in the electron injecting layer 40 may be the same or different.

[0135] The electron injecting layer facilitates electron injection from the cathode. Specifically, for instance, the electron injecting layer may be provided by a laminate of a typical electron transporting material and at least one of an electron-donating dopant material and an organic metal complex. Alternatively, the electron injecting layer may be provided by adding at least one of the electron-donating dopant material and the organic metal complex to a material for forming the blocking layer, specifically, in a vicinity of an interface of the electron injecting layer to the cathode.

[0136] The electron-donating dopant material may be at least one selected from an alkali metal, an alkali metal compound, an alkaline-earth metal, an alkaline-earth metal compound, a rare-earth metal, a rare-earth metal compound and the like.

[0137] The organic metal complex may be at least one selected from an organic metal complex including an alkali metal, an organic metal complex including an alkaline-earth metal, an organic metal complex including a rare-earth metal, and the like.

[0138] Examples of the alkali metal are lithium (Li) (work function: 2.93eV), sodium (Na) (work function: 2.36 eV), potassium (K) (work function: 2.28 eV), rubidium (Rb) (work function: 2.16 eV) and cesium (Cs) (work function: 1.95 eV), which particularly preferably has a work function of 2.9 eV or less. Among the above, the alkali metal is preferably K, Rb or Cs, more preferably Rb or Cs, the most preferably Cs.

[0139] Examples of the alkaline-earth metal are calcium (Ca) (work function: 2.9 eV), strontium (Sr) (work function: 2.0 to 2.5 eV), and barium (Ba) (work function: 2.52 eV), among which a substance having a work function of 2.9 eV or less is particularly preferable.

[0140] Examples of the rare-earth metal are scandium (Sc), yttrium (Y), cerium (Ce), terbium (Tb), and ytterbium (Yb), among which a substance having a work function of 2.9 eV or less is particularly preferable.

[0141] Since the above preferred metals have particularly high reducibility, addition of a relatively small amount of the metals to an electron injecting zone can enhance luminance intensity and lifetime of the organic EL device.

[0142] Examples of the alkali metal compound are an alkali oxide such as lithium oxide (Li_2O) cesium oxide (Cs_2O) and potassium oxide (K_2O) and an alkali halogenide such as sodium fluoride (NaF), cesium fluoride (CsF) and potassium fluoride (KF), among which lithium fluoride (LiF), lithium oxide (Li_2O) and sodium fluoride (NaF) are preferable.

[0143] Examples of the alkaline-earth metal compound are barium oxide (BaO), strontium oxide (SrO), calcium oxide (CaO) and a mixture thereof, i.e., barium strontium oxide ($\text{Ba}_x\text{Sr}_{1-x}\text{O}$) ($0 < x < 1$), barium calcium oxide ($\text{Ba}_x\text{Ca}_{1-x}\text{O}$) ($0 < x < 1$), among which BaO , SrO and CaO are preferable.

[0144] Examples of the rare earth metal compound are ytterbium fluoride (YbF_3), scandium fluoride (ScF_3), scandium oxide (ScO_3), yttrium oxide (Y_2O_3), cerium oxide (Ce_2O_3), gadolinium fluoride (GdF_3) and terbium fluoride (TbF_3), among which YbF_3 , ScF_3 , and TbF_3 are preferable.

[0145] The organic metal complex is not specifically limited as long as containing at least one metal ion of an alkali metal ion, an alkaline-earth metal ion and a rare earth metal ion. A ligand for each of the complexes is preferably quinolinol, benzoquinolinol, acridinol, phenanthridinol, hydroxyphenyl oxazole, hydroxyphenyl thiazole, hydroxydiaryl oxadiazole, hydroxydiaryl thiadiazole, hydroxyphenyl pyridine, hydroxyphenyl benzoimidazole, hydroxybenzo triazole, hydroxy fluorane, bipyridyl, phenanthroline, phthalocyanine, porphyrin, cyclopentadiene, β -diketones, azomethines, or a derivative thereof, but the ligand is not limited thereto.

[0146] The electron-donating dopant material and the organic metal complex are added to preferably form a layer or an island pattern in an interfacial region. A layer or an island pattern of the electron-donating dopant and the organic metal complex is preferably formed by evaporating at least one of the electron-donating dopant material and the organic metal complex by resistance heating evaporation while an emitting material for forming the interfacial region or an organic substance as an electron-injecting material are simultaneously evaporated, so that at least one of the electron-donating dopant material and an organic metal complex reduction-causing dopant is dispersed in the organic substance. Dispersion concentration at which the electron-donating dopant is dispersed in the organic substance is a mole ratio (the organic substance to the electron-donating dopant or the organic metal complex) of 100:1 to 1:100, preferably 5:1 to 1:5.

[0147] When at least one of the electron-donating dopant material and the organic metal complex forms a layer, the emitting material or the electron injecting material for forming the organic layer of the interfacial region is initially layered, and then, at least one of the electron-donating dopant material and the organic metal complex is singularly evaporated thereon by resistance heating evaporation to preferably form a layer having a thickness of 0.1 nm to 15 nm.

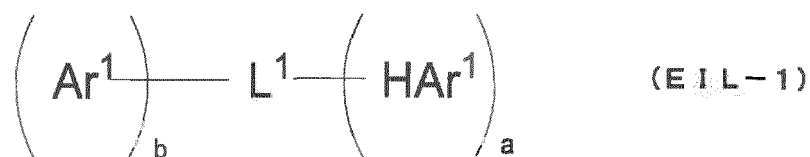
[0148] When at least one of the electron-donating dopant material and the organic metal complex forms an island pattern, the emitting material or the electron injecting material for forming the organic layer of the interfacial region is initially layered, and then, at least one of the electron-donating dopant material is singularly evaporated thereon by resistance heating evaporation to preferably form an island pattern having a thickness of 0.05 nm to 1 nm.

[0149] A ratio of the main component to at least one of the electron-donating dopant material and the organic metal complex in the organic EL device according to the exemplary embodiment is preferably a mole ratio (the main component

to the electron-donating dopant or the organic metal complex) of 5:1 to 1:5, more preferably 2:1 to 1:2.

[0150] A compound other than the electron-donating dopant material and the organic metal complex, which is used in the electron injecting layer, is exemplified by a compound represented by the following formula (EIL-1).

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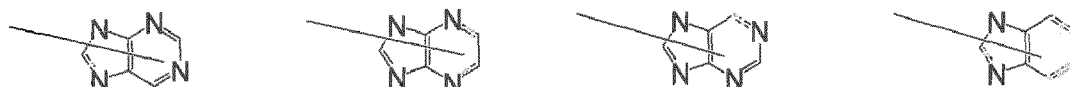
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[0151] In the formula (EIL-1), HAr¹ is a substituted or unsubstituted nitrogen-containing heterocyclic group, preferably having the following structures.

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[0152] Examples of a substituent for HAr¹ in the formula (EIL-1) are a fluorine atom, a cyano group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted cycloalkyl group having 3 to 20 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 20 carbon atoms, a substituted or unsubstituted arylsilyl group having 8 to 30 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms.

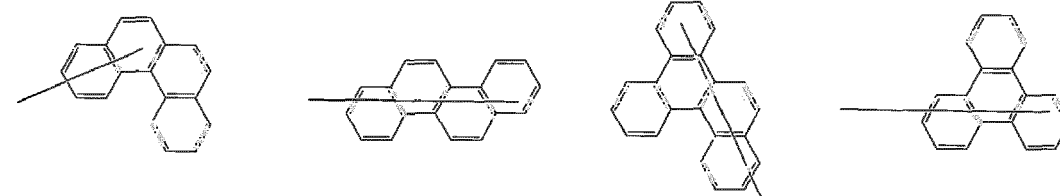
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[0153] In the formula (EIL-1), Ar¹ is a substituted or unsubstituted fused cyclic group having 10 to 30 ring carbon atoms, preferably having the following fused cyclic structures.

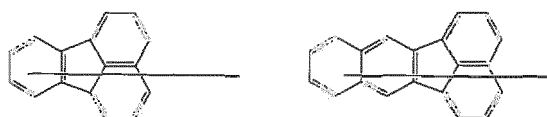
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[0154] Examples of a substituent for Ar¹ in the formula (EIL-1) are a fluorine atom, a cyano group, a substituted or unsubstituted alkyl group having 1 to 20 carbon atoms, a substituted or unsubstituted cycloalkyl group having 3 to 20 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 20 carbon atoms, a substituted or unsubstituted arylsilyl group having 8 to 30 carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 20 carbon atoms, a substituted or unsubstituted aryloxy group having 6 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms.

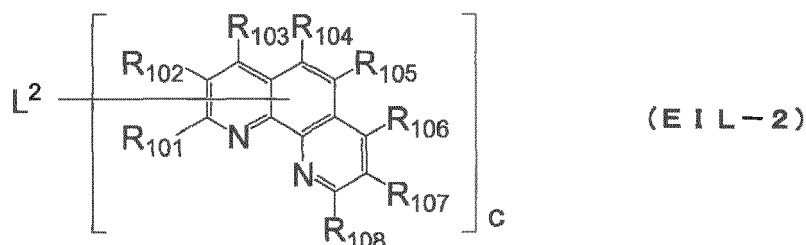
[0155] L¹ in the formula (EIL-1) represents a single bond, a substituted or unsubstituted a+b valent aromatic hydrocarbon ring group having 6 to 30 ring carbon atoms, a substituted or unsubstituted a+b valent heterocyclic group having 5 to 30 ring atoms, a substituted or unsubstituted hydrocarbon ring group having 6 to 30 ring carbon atoms, or an a+b valent group formed by bonding a plurality of substituted or unsubstituted heterocyclic groups having 5 to 30 ring atoms.

[0156] Examples of a substituent for L¹ in the formula (EIL-1) are a fluorine atom, a cyano group, a substituted or unsubstituted alkylsilyl group having 3 to 20 carbon atoms, a substituted or unsubstituted arylsilyl group having 8 to 30 carbon atoms, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted heterocyclic group having 5 to 30 ring atoms.

[0157] In the formula (EIL-1), a is an integer of 1 to 3, preferably a=1.

[0158] In the formula (EIL-1), b is an integer of 1 to 3, preferably b=1.

[0159] A compound used for the electron injecting layer is exemplified by a compound represented by the following formula (EIL-2).



[0160] In the formula (EIL-2), one of R₁₀₁ to R₁₀₈ is bonded to L² by a single bond. The rest of R₁₀₁ to R₁₀₈ are a hydrogen atom or a substituent.

[0161] Examples of the substituent for R₁₀₁ to R₁₀₈ in the formula (EIL-2) are the same as those listed in the formula (EIL-1). A preferred example is an alkyl group having 1 to 20 carbon atoms or an aryl group having 6 to 30 ring carbon atoms.

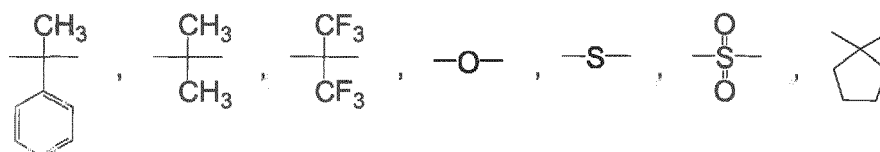
[0162] In the formula (EIL-2), L² represents a single bond or a linking group, in which the linking group is a c-valent aromatic hydrocarbon group or c-valent group having a structure represented by the following formula (EIL-2-1).



[0163] In the formula (EIL-2-1), R₁₀₉ to R₁₁₀ are a hydrogen atom or a substituent.

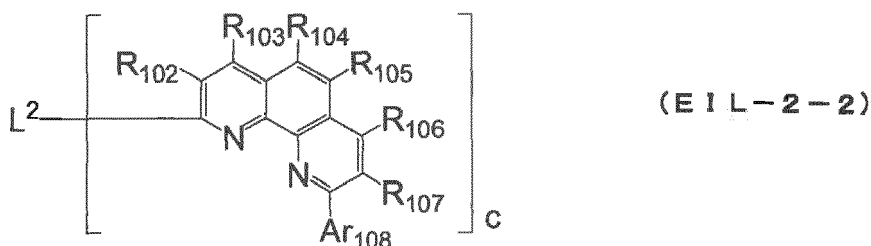
[0164] In the formula (EIL-2-1), d and e are independently an integer of 1 to 5.

[0165] In the formula (EIL-2-1), X is selected from the following structures.



[0166] In the formula (EIL-2), c is an integer of 2 to 4, preferably c is 2.

[0167] Among the compounds represented by the formula (EIL-2), a compound bonded to L² in R₁₀₁ and represented by the following formula (EIL-2-2) is preferable.



10 **[0168]** In the formula (EIL-2-2), R_{102} to R_{107} are a hydrogen atom or a substituent, preferably a hydrogen atom.

[0169] In the formula (EIL-2-2), c and L^2 are the same as those in the formula (EIL-2).

[0170] In the formula (EIL-2-2), c is preferably 2.

[0171] In the formula (EIL-2-2), L^2 is preferably a substituted or unsubstituted phenylene group or a substituted or unsubstituted naphthylene group.

15 **[0172]** In the formula (EIL-2-2), Ar^{108} represents a hydrogen atom, alkyl group having 1 to 20 carbon atoms, or substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, more preferably a methyl group, t-butyl group, substituted or unsubstituted phenyl group, or substituted or unsubstituted naphthyl group.

[0173] In order to efficiently cause The TTF phenomenon, it is preferable to adjust a relationship between an affinity of the host material and an affinity of the dopant material as described below. Hereinafter, the affinity of the host material is described as A_h , the affinity of the dopant material as A_d , ionization potential of the host material as I_h and ionization potential of the dopant material as I_d .

[0174] Now, the following cases will be described.

[1] Case of $A_h > A_d$

[2] Case of $A_h < A_d$

[3] Case where a dopant material satisfying $A_h < A_d$ and a dopant material satisfying $A_h > A_d$ coexist

[1] Case of $A_h > A_d$

30 **[0175]** A case where a relationship of $A_h > A_d$ is satisfied will be described. The dopant material used in this exemplary embodiment is a dopant material emitting fluorescence of a main peak wavelength of 550 nm or less and exhibiting a relatively large energy gap. Accordingly, when the relationship of $A_h > A_d$ is satisfied, a relationship of $I_h > I_d$ is simultaneously satisfied. Consequently, the dopant material easily functions as a hole trap.

[0176] For instance, Fig. 4 shows an I_p (ionization potential)- A_f (affinity) relationship of the host material and the dopant material in the emitting layer in the above case. In Fig. 4, a shaded area in the emitting layer shows an exciton-density distribution. The same applies to Figs. 5 to 7. Fig. 4 shows the relationship in the case of $A_h > A_b > A_e$.

[0177] When a gap in ionization potential between the host material and the dopant material becomes large, the dopant material is likely to have a hole-trapping property, whereby triplet excitons are generated not only on the host material molecule but directly on the dopant material molecule. Consequently, the triplet excitons generated directly on the dopant material are increased. When a relationship of $E_h^T < E_d^T$ is satisfied, triplet exciton energy on the dopant material molecule is transferred onto the host molecule by Dexter energy transfer, resulting in that all the triplet excitons gather on the host material. As a result, The TTF phenomenon occurs efficiently.

[0178] In the exemplary embodiment, it is preferred that the hole transporting layer is adjacent to the emitting layer in the hole transporting zone and a triplet energy E_{ho}^T of the hole transporting layer is larger than a triplet energy E_h^T of the host material.

[0179] When the dopant material has a hole-trapping property, the holes injected from the hole transporting zone to the emitting layer are trapped by the dopant material. Accordingly, recombination often occurs in the emitting layer near the anode. A typically-known hole transporting material used for the hole transporting zone often exhibits a larger triplet energy than the host material. Accordingly, diffusion of the triplet excitons on holes-side has not been a problem.

50 **[0180]** However, even though recombination often occurs near the anode, the triplet exciton density in the interface of the electron transporting zone cannot be ignored. Even under such conditions, highly efficient recombination can be achieved by increasing the triplet energy of the blocking layer.

[0181] Other factors to determine recombination areas are a carrier mobility, ionization potential, affinity and thickness of each of the hole transporting zone and the electron transporting zone. For instance, when the thickness of the hole transporting zone is thicker than that of the electron transporting zone, an amount of the electrons injected to the emitting layer is relatively decreased. As a result, the recombination areas are shifted near the electron transporting zone. In such a case, when the blocking layer having a large triplet energy as in the invention is used, the TTF phenomenon can be efficiently induced.

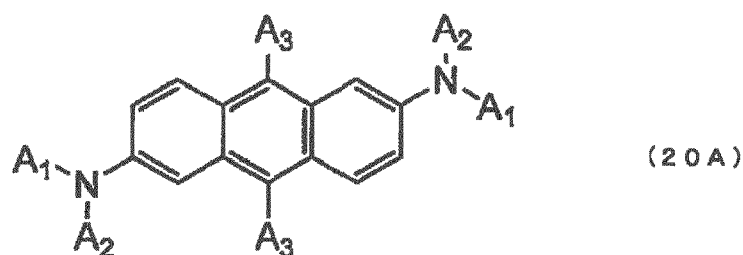
[0182] The host material and the dopant material that satisfy the above relationship in the affinity are selected from, for instance, the following compounds (see JP-A-2010-50227 (Japanese Patent Application No. 2008-212102) and the like).

[0183] The host material is an anthracene derivative and a polycyclic aromatic skeleton-containing compound, preferably the anthracene derivative.

[0184] The dopant material is at least one compound selected from the group consisting of a pyrene derivative, aminoanthracene derivative, aminochrysene derivative and aminopyrene derivative.

[0185] Examples of preferable combinations of the host material and the dopant material are the anthracene derivative as the host material and at least one compound selected from the group consisting of a pyrene derivative, aminoanthracene derivative, aminochrysene derivative and aminopyrene derivative as the dopant material.

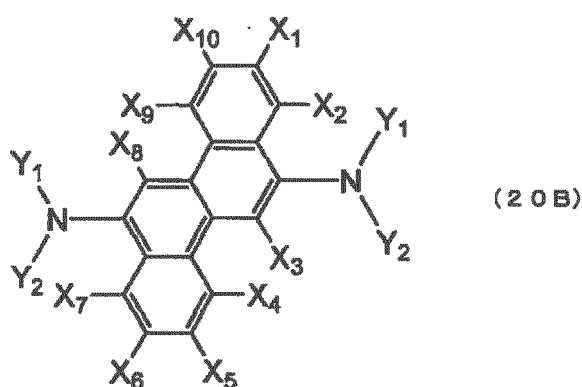
[0186] The aminoanthracene derivative is specifically exemplified by a compound represented by the following formula (20A).



[0187] In the formula (20A), A₁ and A₂ independently represent a substituted or unsubstituted aliphatic hydrocarbon group having 1 to 6 carbon atoms, substituted or unsubstituted aromatic hydrocarbon group having 6 to 20 ring carbon atoms, or substituted or unsubstituted heterocyclic aromatic hydrocarbon group having 5 to 19 ring atoms and containing nitrogen, sulfur or oxygen atom.

[0188] A₃ independently represents a substituted or unsubstituted aliphatic hydrocarbon group having 1 to 6 carbon atoms, substituted or unsubstituted aromatic hydrocarbon group having 6 to 20 ring carbon atoms, substituted or unsubstituted heterocyclic aromatic hydrocarbon group having 5 to 19 ring atoms, or a hydrogen atom. The heterocyclic aromatic hydrocarbon group includes nitrogen, sulfur or oxygen atom.

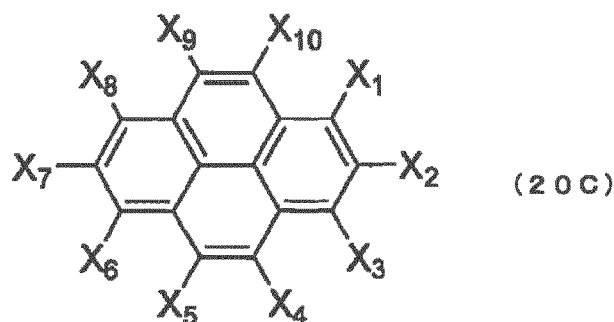
[0189] The aminochrysene derivative is specifically exemplified by a compound represented by the following formula (20B).



[0190] In the formula (20B), X₁ to X₁₀ each represent a hydrogen atom or a substituent. Y₁ and Y₂ each represent a substituent.

[0191] X₁ to X₁₀ are preferably a hydrogen atom. Y₁ and Y₂ are preferably a substituted or unsubstituted aromatic ring having 6 to 30 ring carbon atoms. The substituent of the aromatic ring is preferably an alkyl group having 1 to 6 carbon atoms. The aromatic ring is preferably an aromatic ring having 6 to 10 ring carbon atoms or a phenyl group.

[0192] The aminopyrene derivative is exemplified by a compound represented by the following formula (20C).



[0193] In the formula (20C), X_1 to X_{10} each represent a hydrogen atom or a substituent. X_3 and X_8 or X_2 and X_7 each represent $-NY_1Y_2$ (Y_1 and Y_2 : substituents). When X_3 and X_8 each represent $-NY_1Y_2$, it is preferred that $X_{2,4,5,7,9,10}$ represent a hydrogen atom, X_1 and X_6 represent a hydrogen atom, alkyl group or cycloalkyl group. When X_2 and X_7 each represent $-NY_1Y_2$, it is preferred that $X_{1,3,6,8-10}$ are a hydrogen atom. Y_1 and Y_2 are preferably a substituted or unsubstituted aromatic ring, e.g., a phenyl group and a naphthyl group. The substituent of the aromatic ring is exemplified by an alkyl group having 1 to 6 carbon atoms.

[0194] The anthracene derivative is preferably a compound represented by the formula (20D).

[0195] In the formula (20D), Ar^{11} and Ar^{12} each independently represent a substituted or unsubstituted monocyclic group having 5 to 30 ring atoms, a substituted or unsubstituted fused cyclic group having 10 to 30 ring atoms or a combination of the monocyclic group and the fused cyclic group.

[0196] The monocyclic group in the formula (20D) is a group formed only by ring structure(s) having no fused structure.

[0197] The monocyclic group has 5 to 30 ring atoms, preferably 5 to 20 ring atoms. Examples of the monocyclic group include: an aromatic group such as a phenyl group, biphenyl group, terphenyl group and quarter-phenyl group; and a heterocyclic group such as a pyridyl group, pyrazyl group, pyrimidyl group, triazinyl group, furyl group and thienyl group. Among the monocyclic group, a phenyl group, biphenyl group and terphenyl group are preferable.

[0198] The fused cyclic group in the formula (20D) is a group formed by fusing two or more ring structures.

[0199] The fused cyclic group has 10 to 30 ring atoms, preferably 10 to 20 ring atoms. Examples of the fused cyclic group include: a fused aromatic cyclic group such as a naphthyl group, phenanthryl group, anthryl group, chrysenyl group, benzoanthryl group, benzophenanthryl group, triphenylenyl group, benzochrysenyl group, indenyl group, fluorenyl group, 9,9-dimethylfluorenyl group, benzofluorenyl group, dibenzofluorenyl group, fluoranthenyl group and benzofluoranthenyl group; and a fused heterocyclic group such as a benzofuranyl group, benzothiophenyl group, indolyl group, dibenzofuranyl group, dibenzothiophenyl group, carbazolyl group, quinolyl group and phenanthrolinyl group. Among the fused cyclic group, a naphthyl group, phenanthryl group, anthryl group, 9,9-dimethylfluorenyl group, fluoranthenyl group, benzoanthryl group, dibenzothiophenyl group, dibenzofuranyl group and carbazolyl group are preferable.

[0200] The group formed in a combination of the monocyclic group and the fused cyclic group in the formula (20D) is exemplified by a group combined by sequentially bonding a phenyl group, a naphthyl group and a phenyl group to an anthracene ring.

[0201] Specific examples of the alkyl group, silyl group, alkoxy group, aryloxy group, aralkyl group and the halogen atom for R^{101} to R^{108} in the formula (20D) are the same as those described in relation to R_1 in the formula (1). Examples of the cycloalkyl group are the same as the above examples of the cycloalkyl group. The same applies to the description of "substituted or unsubstituted" in these substituents.

[0202] Preferable examples of the "substituted or unsubstituted" substituent for Ar^{11} and Ar^{12} and R^{101} to R^{108} in the formula (20D) are a monocyclic group, fused cyclic group, alkyl group, cycloalkyl group, silyl group, alkoxy group, cyano group and halogen atom (particularly, fluorine). The monocyclic group and the fused cyclic group are particularly preferable. Preferable specific examples of the substituents are the same as the above examples of the groups in the formula (20D) and the above examples of the groups in the formula (1).

[2] Case of $A_h < A_d$

[0203] Using the combination of the host material and the dopant material which allows $A_h < A_d$, the advantageous effects of the blocking layer provided within the electron transporting zone is exhibited outstandingly, whereby improvement in efficiency due to the TTF phenomenon can be attained. Description will be given in the following cases of [2-1] and [2-2]. In general, an organic material has a broadening of a LUMO level in a range larger than the measured affinity level by approximately 0.2 eV.

[2-1] Difference Between A_d and A_h is Smaller than 0.2 eV

[0204] Fig. 5 shows one example of an energy band diagram in this case. Dotted lines in the emitting layer show an energy level of the dopant material. As shown in Fig. 5, when a difference between A_d and A_h is smaller than 0.2 eV, the LUMO level of the dopant material is included in the range of the broadening of the LUMO level of the host material, so that the electrons carried within the emitting layer is less likely to be trapped by the dopant material. In other words, the dopant material is less likely to exhibit an electron-trapping property. Moreover, the dopant material in the exemplary embodiment is a wide-gap fluorescent dopant material having a main peak wavelength of 550 nm or less. When the relationship of $A_h < A_d$ is satisfied, since the difference between A_h and A_d is approximately 0.2 eV, a difference between the ionization potential of the host material and the ionization potential of the dopant material is reduced. As a result, the dopant material does not tend to exhibit an outstanding hole-trapping property. Fig. 5 shows the relationship in the case of $A_h > A_b > A_e$.

[0205] In other words, the dopant material in this case does not tend to exhibit an outstanding trapping property for both electrons and holes. In this case, as shown by the shaded area of the emitting layer in Fig. 5, the electron-hole recombination occur mainly on the host material molecule in the broad whole area in the emitting layer, thereby generating 25% of singlet excitons and 75% of triplet excitons mainly on the host material molecule. Energy of the singlet excitons generated on the host material is transferred to the dopant material by Forster energy transfer to contribute to a fluorescent emission of the dopant material molecule. On the other hand, the transfer direction of the energy of triplet excitons depends on the triplet energy relationship of the host material and the dopant material. When the relationship is $E_h^T > E_d^T$, the triplet excitons generated on the host material are transferred to a dopant material which exists in the vicinity by the Dexter energy transfer. A concentration of the dopant material in the emitting layer of a fluorescent device is typically as low as at a few mass% to approximately 20 mass%. Accordingly, triplet excitons which have transferred to the dopant material collide with one another less frequently, resulting in a less possibility of occurrence of the TTF phenomenon. However, when the relationship of $E_h^T < E_d^T$ is satisfied as in this exemplary embodiment, since the triplet excitons are present on the host material molecule, the frequency of collision is increased, so that the TTF phenomenon easily and efficiently occur.

[0206] In the exemplary embodiment, the blocking layer is adjacent to the emitting layer. Since the triplet energy E_b^T of the blocking layer is set to be larger than the triplet energy E_h^T of the host material, the triplet excitons is prevented from dispersing in the electron transporting zone, so that the TTF phenomenon can occur efficiently in the emitting layer.

[2-2] Difference Between A_d and A_h is Larger than 0.2 eV

[0207] Fig. 6 shows one example of an energy band diagram in this case. The difference in affinity between the dopant material and the host material is increased, so that a LUMO level of the dopant material is present at a position further higher than the LUMO level broadening of the host material. Accordingly, the dopant material is more likely to exhibit a significant electron-trapping property. Electrons trapped by the dopant material are recombined with holes after the holes are transferred from the host material to the dopant material. In other words, unlike the condition shown in Fig. 5, the electrons and the holes are recombined in a pair not only on the host material molecule but also on the dopant material molecule. As a result, triplet excitons are generated not only on the host material molecule but also directly on the dopant material molecule. Under such conditions, when the relationship of $E_h^T < E_d^T$ is satisfied as in this exemplary embodiment, the triplet excitons generated directly on the dopant material also gather on the host material by Dexter energy transfer, so that the TTF phenomenon occurs efficiently.

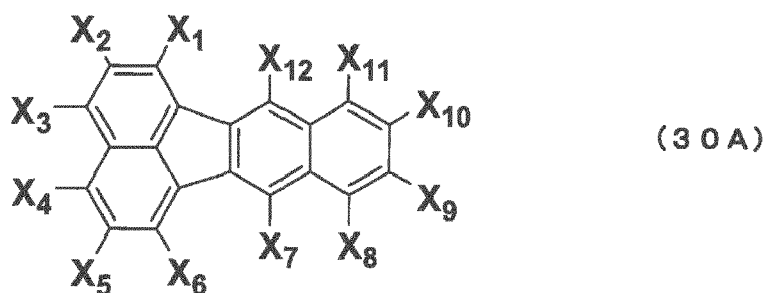
[0208] When the affinities satisfy the above-mentioned relationship, the possibility of trapping of electrons by the dopant material is increased toward the interface between the emitting layer and the blocking layer. As a result, recombination occurs frequently in the vicinity of the interface between the emitting layer and the blocking layer. In this case, the efficiency of confining triplet excitons by the blocking layer is increased as compared with the case mentioned in [2-1], resulting in an increase in density of triplet excitons at the interface between the emitting layer and the blocking layer. Fig. 6 shows the relationship in the case of $A_h > A_b > A_e$.

[0209] The host material and the dopant material that satisfy the above relationship in the $A_h < A_d$ can be selected from, for instance, the following compounds (see JP-A-2010-50227 (Japanese Patent Application No. 2008-212102) and the like).

[0210] Examples of the host material are an anthracene derivative and a polycyclic aromatic skeleton-containing compound, preferably an anthracene derivative.

[0211] Examples of the dopant material are a fluoranthene derivative, pyrene derivative, arylacetylene derivative, fluorene derivative, boron complex, perylene derivative, oxadiazole derivative and anthracene derivatives, preferably fluoranthene derivative, pyrene derivative, and boron complex, more preferably fluoranthene derivative and boron complex. As for the combination of the host material and the dopant material, it is preferred that the host material is an anthracene derivative and the dopant material is a fluoranthene derivative or a boron complex.

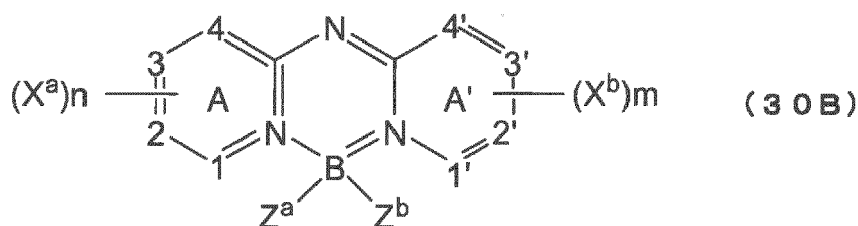
[0212] The fluoranthene derivative is exemplified by the following compound.



[0213] In the formula (30A), X_1 to X_{12} each represent a hydrogen atom or a substituent. Preferably, in the compound, X_1 to X_2 , X_4 to X_6 and X_8 to X_{11} are a hydrogen atom, and X_3 , X_7 and X_{12} are a substituted or unsubstituted aryl group having 5 to 50 ring atoms. Preferably, in the compound, X_1 to X_2 , X_4 to X_6 and X_8 to X_{11} are a hydrogen atom, X_3 , X_7 and X_{12} are a substituted or unsubstituted aryl group having 5 to 50 ring atoms. X_3 is $-Ar_1-Ar_2$, in which Ar_1 is a substituted or unsubstituted arylene group having 5 to 50 ring atoms, and Ar_2 is a substituted or unsubstituted aryl group having 5 to 50 ring atoms.

[0214] More preferably, in the compound, X_1 to X_2 , X_4 to X_6 and X_8 to X_{11} are a hydrogen atom and X_7 and X_{12} are a substituted or unsubstituted aryl group having 5 to 50 ring atoms. X_3 is $-Ar_1-Ar_2-Ar_3$, in which Ar_1 and Ar_3 are each a substituted or unsubstituted arylene group having 5 to 50 ring atoms, and Ar_2 is a substituted or unsubstituted aryl group having 5 to 50 ring atoms.

[0215] The boron complex is exemplified by the following compound.



[0216] In the formula (30B), A and A' represent an independent azine ring system corresponding to a six-membered aromatic ring containing one or more nitrogen. X^a and X^b represent independently-selected substituents, which are bonded together to form a fused ring for the ring A or the ring A'. The fused ring contains an aryl or heteroaryl substituent. m and n independently represent 0 to 4. Z^a and Z^b each represent an independently-selected halide. 1, 2, 3, 4, 1', 2', 3' and 4' each represent an independently-selected carbon atom or nitrogen atom.

[0217] Desirably, the azine ring is preferably a quinolinyl ring or isoquinolinyl ring in which all of 1, 2, 3, 4, 1', 2', 3' and 4' are carbon atoms, m and n each are 2 or more, and X^a and X^b are a substituent having 2 or more carbon atoms that combine with each other to form an aromatic ring. Z^a and Z^b are desirably fluorine atoms.

[0218] The anthracene derivatives as the host material in the case of [2] are the same as those described in the above "[1] Case of $A_h > A_d$."

[3] Case where a dopant material satisfying $A_h < A_d$ and a dopant material satisfying $A_h > A_d$ coexist

[0219] Fig. 7 shows one example of an energy band diagram when a dopant material satisfying $A_h < A_d$ and a dopant material satisfying $A_h > A_d$ are both contained in the emitting layer. In such a case, both electrons and holes are trapped properly, whereby recombination occurs in the entire region of the emitting layer. Accordingly, recombination occurs frequently also on the cathode side. By providing a blocking layer that has a large triplet energy, the TTF phenomenon occurs efficiently. Fig. 7 shows the relationship in the case of $A_h > A_b > A_e$.

[0220] In the exemplary embodiment, the density of excitons is large in the interface between the emitting layer and the blocking layer. In this case, holes which do not contribute to recombination in the emitting layer are injected in the blocking layer with a high probability. Accordingly, among the above-mentioned aromatic heterocyclic derivatives, one having an excellent oxidation resistance is preferable as the material to be used in the blocking layer.

[0221] The blocking layer material desirably exhibits a reversible oxidation process in a cyclic voltammetry measurement.

[0222] The emitting layer may contain two or more fluorescent dopant materials of which the main peak wavelength is 550 nm or less. When the emitting layer contains two or more fluorescent dopant materials, the affinity A_d of at least one dopant material is equal to or larger than the affinity A_h of the host material, and the triplet energy E_d^T of this dopant material is larger than the triplet energy E_h^T of the host material. For instance, the affinity A_d of at least one dopant material of the rest of the dopant materials may be smaller than the affinity A_h of the host material. Containing such two kinds of dopant materials means containing both of a dopant material satisfying $A_h < A_d$ and a dopant material satisfying $A_h > A_d$ as described above. Efficiency can be significantly improved by providing the blocking layer having a large triplet energy.

[0223] Examples of the dopant material having the affinity A_d that is smaller than the affinity A_h of the host material include a pyrene derivative, aminoanthracene derivative, aminochrysene derivative, and aminopyrene derivative.

[0224] In addition to the above-mentioned host materials, dibenzofuran compounds disclosed in WO05/113531 and JP2005-314239, fluorene compounds disclosed in WO02/14244, and benzanthracene compounds disclosed in WO08/145,239 can be used.

[0225] In addition to the above-mentioned dopant materials, pyrene compounds disclosed in JP2004-204238, WO05/108348, WO04/83162, WO09/84512, KR10-2008-79956, KR10-2007-115588 and KR10-2010-24894, chrysene compounds disclosed in WO04/44088, and anthracene compounds disclosed in WO07/21117 can be used.

[0226] Preferably, the host material and the dopant material are each a compound formed by bonding ring structures or single atoms (including bonding of a ring structure and a single atom), in which the bonding is a single bond. A compound having a carbon-carbon double bond in the part other than the ring structure thereof is not preferable. The reason thereof is that the triplet energies generated on the host material and the dopant material are used for the structural change of the double bond, not for a TTF phenomenon.

Formation Method of Each Layer of Organic EL Device

[0227] Each layer of the organic EL device in the exemplary embodiment may be formed by any one of dry film-forming such as vacuum deposition, sputtering, plasma deposition and ion plating, and wet film-forming such as spin coating, dipping, flow coating and ink jet.

[0228] In the wet film-forming, materials for forming each layer are dissolved or dispersed in an appropriate solvent such as ethanol, chloroform, tetrahydrofuran and dioxane to form a thin-film. Any one of the solvents is usable.

[0229] As a solution suitable for the wet film-forming, an organic-EL-material-containing solution, which contains an aromatic amine derivative of the invention as an organic-EL-device material and a solvent, is usable.

[0230] In any one of the organic thin-film layer, a resin or an additive suitable for improving film-forming performance and preventing pin holes on a film may be used.

Thickness of Each Layer of Organic EL Device

[0231] Although a thickness is not limited, an appropriate thickness needs to be set. When the thickness is excessively thick, a large voltage is required to be applied in order to obtain a predetermined emission, so that an efficiency is deteriorated. When the thickness is excessively thin, pin holes or the like generate. Accordingly, even when the electrical field is applied to the film, a sufficient luminescence intensity is not obtained. A thickness of the blocking layer is preferably 20 nm or less. A thickness of each of other layers is typically preferably in a range of 5 nm to 10 μm , more preferably in a range of 10 nm to 0.2 μm .

Second Exemplary Embodiment

[0232] Fig. 8 shows one example of an organic EL device 2 according to a second exemplary embodiment.

[0233] The organic EL device according to the second exemplary embodiment may not include an electron injecting layer. As shown in Fig. 8, the organic EL device 2 according to the second exemplary embodiment includes the anode 10, the hole transporting zone 60, the emitting layer 20, the electron transporting zone (the blocking layer 30 in the exemplary embodiment), and the cathode 50 in this sequence. These layers are adjacent to one another in the organic EL device in the exemplary embodiment.

[0234] The blocking layer 30 of the organic EL device 2 contains the aromatic heterocyclic derivative represented by the formula (4) in the same manner as in the first exemplary embodiment. Other layers forming the organic EL device 2 are also the same as those in the first exemplary embodiment.

Third Exemplary Embodiment

[0235] Fig. 9 shows one example of an organic EL device 3 according to a third exemplary embodiment.

[0236] The organic EL device according to the third exemplary embodiment may include an electron injecting layer on a side of the electron transporting layer near the cathode. As shown in Fig. 9, the organic EL device 3 according to the third exemplary embodiment includes the anode 10, the hole transporting zone 60, the emitting layer 20, the electron transporting zone (the blocking layer 30, the electron transporting layer 41 and the electron injecting layer 40 in the exemplary embodiment), and the cathode 50 in this sequence. These layers are adjacent to one another in the exemplary embodiment.

[0237] In the organic EL device 3, at least one of the electron injecting layer 40 and the electron transporting layer 41 preferably contains the aromatic heterocyclic derivative according to the above exemplary embodiment. A material to be contained in the electron transporting layer can be the material described in relation to the above electron injecting layer and known electron transporting materials. Moreover, the electron injecting layer 40 and the electron transporting layer 41 may further include other material(s) in addition to the aromatic heterocyclic derivative according to the above exemplary embodiment.

[0238] The blocking layer 30 of the organic EL device 3 contains the aromatic heterocyclic derivative represented by the formula (4) in the same manner as in the first exemplary embodiment. Other layers forming the organic EL device 3 are also the same as those in the first exemplary embodiment.

Fourth Exemplary Embodiment

[0239] An organic EL device in the fourth exemplary embodiment may have a tandem device configuration in which at least two organic layer units including emitting layers are provided. An intermediate layer (intermediate conductive layer, charge generation layer or CGL) is interposed between the two emitting layers. An electron transporting zone can be provided in each unit. At least one emitting layer is a fluorescent emitting layer and the unit including the emitting layer satisfies the above-mentioned requirements.

[0240] Fig. 10 shows one example of the organic EL device according to the fourth exemplary embodiment. An organic EL device 4 includes the anode 10, emitting layers 22 and 24 and the cathode 50 in this sequence. An intermediate layer 80 is interposed between the emitting layers 22 and 24. A blocking layer 32 is adjacent to the emitting layer 24. The electron injecting layer 40 is interposed between the blocking layer 32 and the cathode 50. The blocking layer 32, the electron injecting layer 40 and the emitting layer 24 are respectively a blocking layer, an electron injecting layer and a fluorescent emitting layer which satisfy the above-mentioned requirements. The other emitting layer may be either a fluorescent emitting layer or a phosphorescent emitting layer. Another blocking layer and another electron injecting layer are provided adjacent to the emitting layer 22 in sequential order. These blocking layer and electron injecting layer and the emitting layer 22 may be respectively used as the blocking layer, the electron injecting layer, and the fluorescent emitting layer which satisfy the requirements of the invention.

[0241] Note that, in the exemplary embodiment, the blocking layer 32 and the electron injecting layer 40 correspond to the electron transporting zone.

[0242] At least one of an electron transporting zone and hole transporting zone may be interposed between the two emitting layers 22 and 24. Three or more emitting layers may be provided, and two or more intermediate layers may be provided. When three or more emitting layers are present, an intermediate layer may or may not be present between all of the emitting layers.

[0243] The intermediate layer is a layer including at least one of the intermediate conductive layer and the charge generation layer, or at least one of the intermediate conductive layer and the charge generation layer. The intermediate layer serves as a source for supplying electrons or holes to be injected in an emitting unit. In addition to charges injected from a pair of electrodes, charges supplied from the intermediate layer are injected into the emitting unit. Accordingly, by providing the intermediate layer, luminous efficiency (current efficiency) relative to injected current is improved.

[0244] Examples of the intermediate layer include a metal, metal oxide, mixture of metal oxides, composite oxide, and electron-accepting organic compound. Examples of the metal are preferably Mg, Al, and a film formed by co-evaporating Mg and Al. Examples of the metal oxide include ZnO, WO₃, MoO₃ and MoO₂. Examples of the mixture of the metal oxides include ITO, IZO (registered trade mark), and ZnO:Al. Examples of the electron-accepting organic compound include an organic compound having a CN group as a substituent. The organic compound having a CN group is preferably a triphenylene derivative, tetracyanoquinodimethane derivative and indenofluorene derivative. The triphenylene derivative is preferably hexacyanohexaazatriphenylene. The tetracyanoquinodimethane derivative is preferably tetrafluoroquinodimethane and dicyanoquinodimethane. The indenofluorene derivative is preferably a compound disclosed in WO2009/011327, WO2009/069717, or WO2010/064655. The electron accepting substance may be a single substance, or a mixture with other organic compounds.

[0245] In order to easily accept the electrons from the charge generation layer, at least one of the electron-donating dopant represented by an alkali metal and the organic metal complex is added in the vicinity of the interface of the charge generation layer in the electron transporting layer. Examples of the electron-donating dopant and the organic metal complex are those described above in the first exemplary embodiment.

[0246] Specific examples of the compounds usable for the electron-donating dopant and the organic metal complex are compounds disclosed in International Patent Application No. PCT/JP2010/003434 (International Publication No. WO2010/134352).

5 Fifth Exemplary Embodiment

[0247] In the fifth exemplary embodiment, an anode, a plurality of emitting layers, an electron transporting zone that includes a blocking layer adjacent to one of the emitting layers and an electron injecting layer adjacent to the blocking layer, and a cathode are provided in sequential order. A charge blocking layer is provided between two emitting layers of the plurality of the emitting layers. The emitting layers in contact with the charge blocking layer are fluorescent emitting layers. The fluorescent emitting layer, and the blocking layer and the electron injecting layer in the electron transporting zone satisfy the above requirements.

[0248] As a configuration of a suitable organic EL device according to the fifth exemplary embodiment, there can be given a configuration as disclosed in Japanese Patent No. 4134280, US patent publication US2007/0273270A1 and International Publication WO2008/023623A1. Specifically, the configuration in which an anode, a first emitting layer, a charge blocking layer, a second emitting layer and a cathode are sequentially stacked, and an electron-transporting zone having a blocking layer and an electron injecting layer for preventing diffusion of triplet excitons is further provided between the second emitting layer and the cathode. Here, the charge blocking layer means a layer to control the carrier injection to an emitting layer and the carrier balance between electrons and holes injected in the emitting layer by providing an energy barrier of a HOMO level or a LUMO level between adjacent emitting layers

[0249] Specific examples of such a configuration are given below.

anode / first emitting layer / charge blocking layer / second emitting layer / electron transporting zone / cathode

anode / first emitting layer / charge blocking layer / second emitting layer / third emitting layer / electron transporting zone / cathode

[0250] It is preferred that a hole transporting zone is provided between the anode and the first emitting layer in the same manner as in other embodiments

[0251] Fig. 11 shows one example of an organic EL device according to the fifth exemplary embodiment. An upper view in Fig. 11 shows a device configuration, and the HOMO and LUMO energy levels of each layer. A lower view in Fig. 11 shows a relationship between energy gaps of the third emitting layer and the blocking layer. The upper view in Fig. 11 shows the relationship in the case of $A_h > A_b > A_e$.

[0252] The organic EL device includes the anode, first emitting layer, second emitting layer, third emitting layer, electron transporting zone, and cathode in sequential order. A charge blocking layer is interposed between the first and second emitting layers. The electron transporting zone is formed of the blocking layer. This blocking layer and third emitting layer are the blocking layer and the fluorescent emitting layer that satisfy the above-mentioned requirements. The first and second emitting layers may be either a fluorescent emitting layer or a phosphorescent emitting layer.

[0253] The device of this embodiment is suitable as a white emitting device. The device can be a white emitting device by adjusting the emission colors of the first emitting layer, second emitting layer and third emitting layer. Moreover, the device can be a white emitting device by arranging only the first emitting layer and the second emitting layer and adjusting the emission colors of these two emitting layers. In this case, the second emitting layer is a fluorescent emitting layer satisfying the above-mentioned requirements.

[0254] In particular, by using a hole transporting material as the host in the first emitting layer, by adding a fluorescent dopant material of which the main peak wavelength is larger than 550 nm, by using an electron transporting material as the host material in the second emitting layer (and the third emitting layer), and by adding a fluorescent dopant material of which the main peak wavelength is equal to or smaller than 550 nm, it is possible to achieve a white emitting device that exhibits a higher luminous efficiency as compared with conventional white emitting devices, even though all of them are entirely formed of fluorescent materials.

[0255] Reference is made particularly to a hole transporting layer which is adjacent to the emitting layer. In order to allow the TTF phenomenon to occur effectively, it is preferred that the triplet energy of the hole transporting material is larger than the triplet energy of the host material, when the triplet energy of the hole transporting material and that of the host material are compared.

Sixth Exemplary Embodiment

[0256] In a sixth exemplary embodiment, a blue pixel, a green pixel and a red pixel are arranged in parallel on a substrate. Of these three color pixels, at least one of the blue pixel and the green pixel has the configuration of the first exemplary embodiment or second exemplary embodiment.

[0257] Fig. 12 shows one example of an organic EL device according to the fifth exemplary embodiment.

[0258] In a top-emission type organic EL device 5 shown in Fig. 12, a blue pixel B, a green pixel G and a red pixel R

are arranged in parallel on a common substrate 100.

[0259] The blue pixel B includes the anode 10, the hole transporting zone 60, a blue emitting layer 20B, the blocking layer 32, the electron injecting layer 40, the cathode 50, and a protection layer 90 on the substrate 100 in sequential order.

[0260] The green pixel G includes the anode 10, the hole transporting zone 60, a green emitting layer 20G, the blocking layer 32, the electron injecting layer 40, the cathode 50, and the protection layer 90 on the substrate 100 in sequential order.

[0261] The red pixel R includes the anode 10, the hole transporting zone 60, a red emitting layer 20R, the blocking layer 32, the electron injecting layer 40, the cathode 50, and the protection layer 90 on the substrate 100 in sequential order.

[0262] An insulating film 200 is formed between the anodes of adjacent pixels so as to keep the insulation between the pixels. The electron transporting zone is formed of the blocking layer 32 and the electron injecting layer 40.

[0263] In the organic EL device 5, the blocking layer is provided as a common blocking layer for the blue pixel B, the red pixel R and the green pixel G.

[0264] The advantageous effects brought by the blocking layer are outstanding comparing to the luminous efficiency conventionally attained in a blue fluorescent device. In a green fluorescent device and a red fluorescent device, similar advantageous effects, such as confining triplet energies in the emitting layer, can be attained, and improvement in luminous efficiency can also be expected.

[0265] On the other hand, in a phosphorescent emitting layer, it is possible to attain the advantageous effects of confining triplet energies in the emitting layer, and as a result, diffusion of triplet energies is prevented, thereby contributing to improvement in luminous efficiency of a phosphorescent dopant material.

[0266] The hole transporting zone is formed of, for instance, a hole transporting layer, or a combination of a hole transporting layer and a hole injecting layer. A common hole transporting zone may be provided or different hole transporting zones may be provided for the blue pixel B, the red pixel R and the green pixel G. Typically, the hole transporting zones respectively have a configuration suited to the color of emitted light.

[0267] The configuration of the organic layer formed of the emitting layers 20B, G and R and the blocking layer is not limited to that shown in the figure and is changeable appropriately.

[0268] The host material and the dopant material usable in the exemplary embodiment are the same as described above. In particular, emitting layers for each color will be described below.

[0269] A green emitting layer is preferably formed of the following host material and dopant material.

[0270] The host material is preferably a fused aromatic ring derivative. As the fused aromatic ring derivative, an anthracene derivative, pyrene derivative and the like are more preferable in view of luminous efficiency and luminous lifetime.

[0271] The host material is exemplified by a heterocycle-containing compound. Examples of the heterocycle-containing compound are a carbazole derivative, dibenzofuran derivative, ladder-type furan compound and pyrimidine derivative.

[0272] The dopant material is not particularly limited so long as it functions as a dopant, but an aromatic amine derivative is preferable in view of luminous efficiency and the like. As the aromatic amine derivative, a fused aromatic ring derivative having a substituted or unsubstituted arylamino group is preferable. Examples of such a compound are pyrene, anthracene and chrysene having an arylamino group.

[0273] A styrylamine compound is also preferable as the dopant material. Examples of the styrylamine compound are styrylamine, styryldiamine, styryltriamine and styryltetraamine. Here, the styrylamine means a compound in which a substituted or unsubstituted arylamine is substituted with at least one arylvinyl group. The arylvinyl group may be substituted with a substituent such as an aryl group, silyl group, alkyl group, cycloalkyl group, or arylamino group, which may have a further substituent.

[0274] Furthermore, as the dopant material, a boron complex and a fluoranthene compound are preferable. A metal complex is also preferable as the dopant material. The metal complex is exemplified by an iridium complex or platinum complex.

[0275] A red emitting layer is preferably formed of the following host material and dopant material. The host material is preferably a fused aromatic ring derivative. As the fused aromatic ring derivative, a naphthalene derivative, pentacene derivative and the like are more preferable in view of luminous efficiency and luminous lifetime.

[0276] The host material is exemplified by a fused polycyclic aromatic compound. Examples of the fused polycyclic aromatic compound are a naphthalene compound, phenanthrene compound and fluoranthene compound.

[0277] The dopant material is preferably an aromatic amine derivative. As the aromatic amine derivative, a fused aromatic ring derivative having a substituted or unsubstituted arylamino group is preferable. Such a compound is exemplified by perflanthene having an arylamino group.

[0278] A metal complex is also preferable as the dopant material. The metal complex is exemplified by an iridium complex or platinum complex.

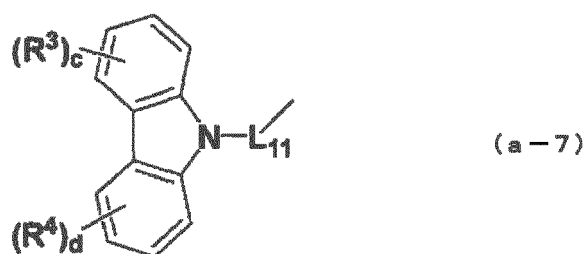
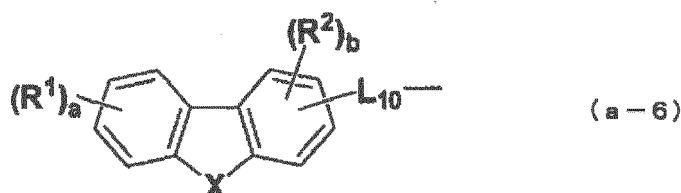
[0279] The organic EL device of the sixth exemplary embodiment is prepared in the following manner.

[0280] On a substrate, an APC (Ag-Pd-Cu) layer as a silver alloy layer (reflective layer) and a transparent conductive layer such as a zinc oxide (IZO) film and a tin oxide film are sequentially formed. Next, by a typical lithographic technology, this conductive material layer is patterned by etching using a mask with a resist pattern, thereby forming an anode. Then,

[0286] L_1 to L_9 are independently a substituted or unsubstituted arylene group having 6 to 50 ring carbon atoms or a substituted or unsubstituted heteroarylene group having 5 to 50 ring atoms.

[0287] Examples of a substituent which Ar_1 to Ar_{24} and L_1 to L_9 may have include a linear or branched alkyl group having 1 to 15 carbon atoms, a cycloalkyl group having 3 to 15 ring carbon atoms, a trialkylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms, a triarylsilyl group having an aryl group having 6 to 14 ring carbon atoms, an alkylarylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms and an aryl group having 6 to 14 ring carbon atoms, an aryl group having 6 to 50 ring carbon atoms, a heteroaryl group having 5 to 50 ring atoms, a halogen atom, and a cyano group. Adjacent substituents may bond to each other to form a saturated or unsaturated divalent group forming a ring.

[0288] At least one of the above Ar_1 to Ar_{24} is preferably a substituent represented by the following formula (a-6) or (a-7).



[0289] In the formula (a-6), X is an oxygen atom, sulfur atom or N-Ra. Ra is a linear or branched alkyl group having 1 to 15 carbon atoms, a cycloalkyl group having 3 to 15 ring carbon atoms, an aryl group having 6 to 50 ring carbon atoms or a heteroaryl group having 5 to 50 ring atoms.

[0290] L_{10} is a single bond, a substituted or unsubstituted arylene group having 6 to 50 ring carbon atoms, or a substituted or unsubstituted heteroarylene group having 5 to 50 ring atoms.

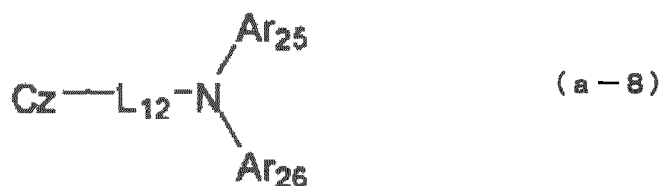
[0291] In the formula (a-7), L_{11} is a substituted or unsubstituted arylene group having 6 to 50 ring carbon atoms, or a substituted or unsubstituted heteroarylene group having 5 to 50 ring atoms.

[0292] In the formulae (a-6) and (a-7), R^1 to R^4 are independently a linear or branched alkyl group having 1 to 15 carbon atoms, a cycloalkyl group having 3 to 15 ring carbon atoms, a trialkylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms, a triarylsilyl group having an aryl group having 6 to 14 ring carbon atoms, an alkylarylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms and an aryl group having 6 to 14 ring carbon atoms, an aryl group having 6 to 14 ring carbon atoms, a heteroaryl group having 5 to 50 ring atoms, a halogen atom, or a cyano group. Adjacent groups of R^1 s to R^4 s may bond to each other to form a ring.

a, c and d are each an integer of 0 to 4.

b is an integer of 0 to 3.

[0293] The compound represented by the formula (a-1) is preferably a compound represented by the following formula (a-8).



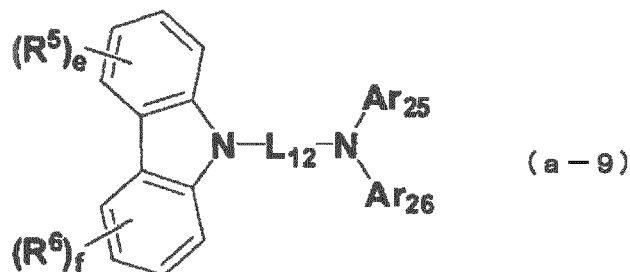
[0294] In the formula (a-8), Cz is a substituted or unsubstituted carbazolyl group.

[0295] L_{12} is a substituted or unsubstituted arylene group having 6 to 50 ring carbon atoms, or a substituted or

unsubstituted heteroarylene group having 5 to 50 ring atoms.

[0296] Ar_{25} and Ar_{26} are independently a substituted or unsubstituted aryl group having 6 to 50 ring carbon atoms or a substituted or unsubstituted heteroaryl group having 5 to 50 ring atoms.

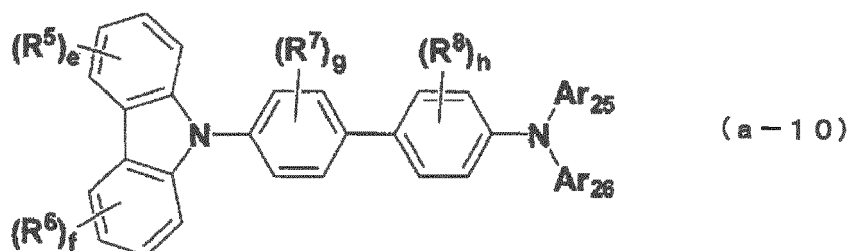
[0297] The compound represented by the formula (a-8) is preferably a compound represented by the following formula (a-9).



[0298] In the formula (a-9), R^5 and R^6 are independently a linear or branched alkyl group having 1 to 15 carbon atoms, a cycloalkyl group having 3 to 15 ring carbon atoms, a trialkylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms, a triarylsilyl group having an aryl group having 6 to 14 ring carbon atoms, an alkylarylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms and an aryl group having 6 to 14 ring carbon atoms, an aryl group having 6 to 14 ring carbon atoms, a heteroaryl group having 5 to 50 ring atoms, a halogen atom, or a cyano group. Adjacent groups of R^5 s to R^6 s may bond to each other to form a ring. e and f are each an integer of 0 to 4.

[0299] L_{12} , Ar_{25} and Ar_{26} are the same as L_{12} , Ar_{25} and Ar_{26} in the formula (a-8).

[0300] The compound represented by the formula (a-9) is preferably a compound represented by the following formula (a-10).



[0301] In the formula (a-10), R^7 and R^8 are independently a linear or branched alkyl group having 1 to 15 carbon atoms, a cycloalkyl group having 3 to 15 ring carbon atoms, a trialkylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms, a triarylsilyl group having an aryl group having 6 to 14 ring carbon atoms, an alkylarylsilyl group having a linear or branched alkyl group having 1 to 15 carbon atoms and an aryl group having 6 to 14 ring carbon atoms, an aryl group having 6 to 14 ring carbon atoms, a heteroaryl group having 5 to 50 ring atoms, a halogen atom, or a cyano group. Adjacent groups of R^5 s to R^6 s may bond to each other to form a ring.

[0302] g and h are each an integer of 0 to 4.

[0303] R^5 , R^6 , e, f, Ar_{25} and Ar_{26} are the same as R^5 , R^6 , e, f, Ar_{25} and Ar_{26} in the formula (a-9).

Seventh Exemplary Embodiment

[0304] An organic EL device according to the seventh exemplary embodiment may include the electron transporting layer in place of the blocking layer 30 as the electron transporting zone in the organic EL device 2 according to the second exemplary embodiment in Fig. 8. Specifically, the organic EL device according to the seventh exemplary embodiment (not shown) include the anode 10, the hole transporting zone 60, the emitting layer 20, the electron transporting layer and the cathode 50 in this sequence. The electron transporting layer includes the aromatic heterocyclic derivative of formula (4) of the invention and may further include other material(s). In the seventh exemplary embodiment, the emitting layer 20 preferably contains a dopant material exhibiting phosphorescence as the dopant material.

[0305] Note that other layers forming the organic EL device in the seventh exemplary embodiment are the same as those in the first and second exemplary embodiments.

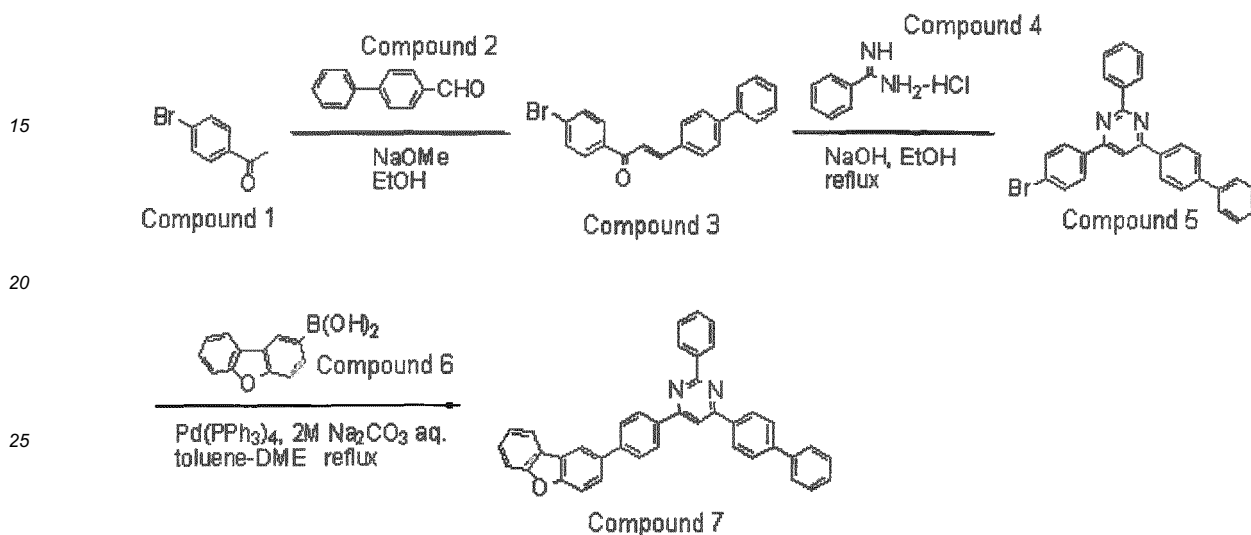
Examples

[0306] Examples of the invention will be described below. However, the invention is not limited by these Examples.

5 Synthesis of Compound(s)

Synthesis Example 1: Synthesis of Compound 7

[0307] A synthesis scheme of a compound 7 is shown below.



(1-1) Synthesis of Compound 3

[0308] 4'-bromoacetophenone (a compound 1) (22g, 120 mmol), 4-phenylbenzaldehyde (a compound 2) (25 g, 126 mmol), sodium methoxide (8.4 g, 156 mmol) and ethanol (200mL) were mixed and stirred for 12 hours at the room temperature. A precipitated solid was separated by filtration and suspended in and washed with ethanol. Subsequently, the obtained solid was dried under reduced pressure, so that a compound 3 (42 g, a yield of 96%) in a form of a white solid was obtained.

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(1-2) Synthesis of Compound 5

[0309] Ethanol (450 mL) was added with the compound 3 (40 g, 111 mmol), benzamidinium hydrochloride (a compound 4) (26 g, 166 mmol) and sodium hydroxide (12 g, 299 mmol), and was heated to reflux for nine hours. After the reaction, a precipitated solid was separated by filtration and refined by silica-gel column chromatography (eluent: toluene). The reactant was recrystallized using toluene, so that a compound 5 in a form of a white solid (24 g, a yield of 47%) was obtained.

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(1-3) Synthesis of Compound 7

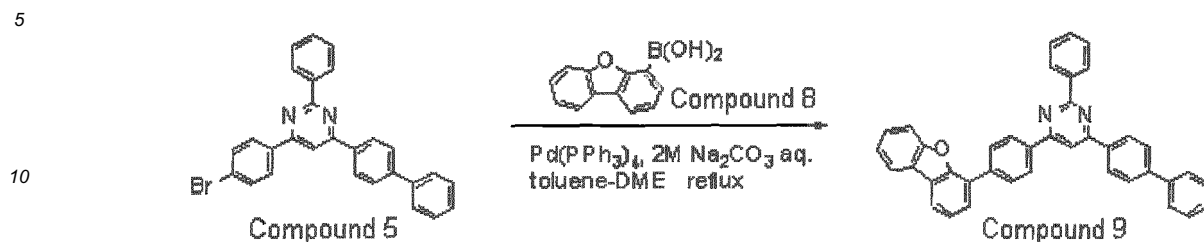
[0310] The compound 5 (6.0 g, 13 mmol) and a compound 6 (3.3 g, 16 mmol) were dissolved in toluene (200 mL) and 1,2-dimethoxyethane (200 mL), to which tetrakis(triphenylphosphine)palladium (0) (0.75 g, 0.65 mmol) and an aqueous solution of 2M sodium carbonate (26 mL) were added. The obtained solution was heated to reflux for 15 hours. After the reaction, the reactant solution was cooled to the room temperature and extracted with toluene. The obtained organic layer was washed with water and saturated saline in sequential order and dried with sodium sulfate. The solvent was distilled away under reduced pressure. The residue was added with toluene and heated to reflux to be dissolved. The reactant solution was cooled and precipitated in a crystal form. The crystal was separated by filtration and washed with toluene. Subsequently, the obtained solid was dried under reduced pressure, so that a compound 7 (5.5 g, a yield of 77%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 7.

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Synthesis Example 2: Synthesis of Compound 9 (not according to the invention)

[0311] A synthesis scheme of a compound 9 is shown below.



(2-1) Synthesis of Compound 9

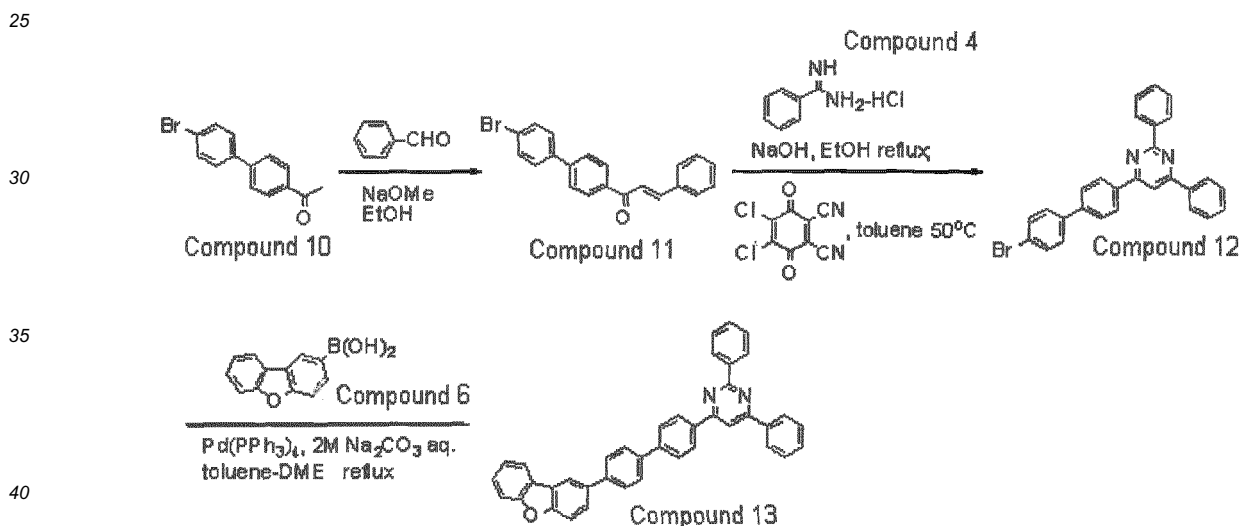
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[0312] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using a compound 8 (2.7 g, 13 mmol) in place of the compound 6, so that a compound 9 (5.5 g, a yield of 92%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 9.

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Synthesis Example 3: Synthesis of Compound 13

[0313] A synthesis scheme of a compound 13 is shown below.



(3-1) Synthesis of Compound 12

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[0314] 4'-(p-bromophenyl)acetophenone (a compound 10) (30 g, 110 mmol) and benzaldehyde (12 g, 110 mmol) were dissolved in ethanol (300 mL), into which sodium methoxide (about-5M methanol solution) (80 mL) was dropped. The obtained solution was stirred for three hours at the room temperature to synthesize a compound 11. Next, benzamidine hydrochloride (the compound 4) (17 g, 110 mmol) and sodium hydroxide (5.3 g, 132 mmol) were added to the above solution and was heated to reflux for 18 hours. After the reaction, the reactant solution was cooled to the room temperature and a precipitated solid was separated by filtration. Then, the obtained solid was washed with methanol. A crude product, 2,3-dichloro-5,6-dicyano-p-benzoquinone (19 g, 84 mmol) and toluene (400 mL) were mixed and stirred for one hour at 50 degrees C. After the reaction, methanol was added to the mixture and a precipitated solid was separated by filtration and refined by silica-gel column chromatography (eluent: toluene), so that a compound 12 (26 g, a yield of 51 %) in a form of a white solid was obtained.

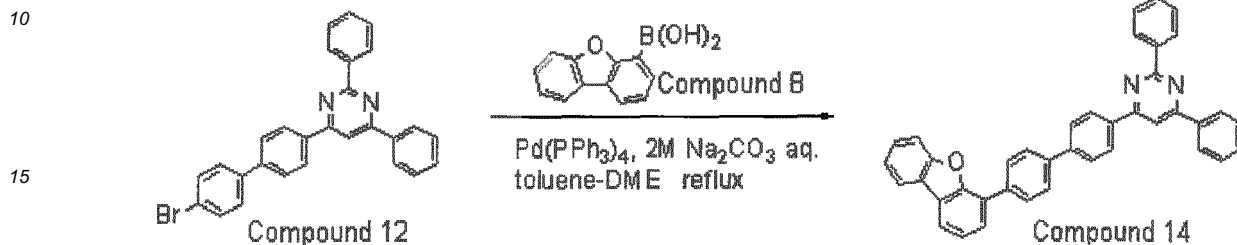
(3-2) Synthesis of Compound 13

[0315] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the

compound 12 (6.0 g, 13 mmol) in place of the compound 5, so that a compound 13 (4.9g, a yield of 69%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 13.

5 Synthesis Example 4: Synthesis of Compound 14 (not according to the invention)

[0316] A synthesis scheme of a compound 14 is shown below,



(4-1) Synthesis of Compound 14

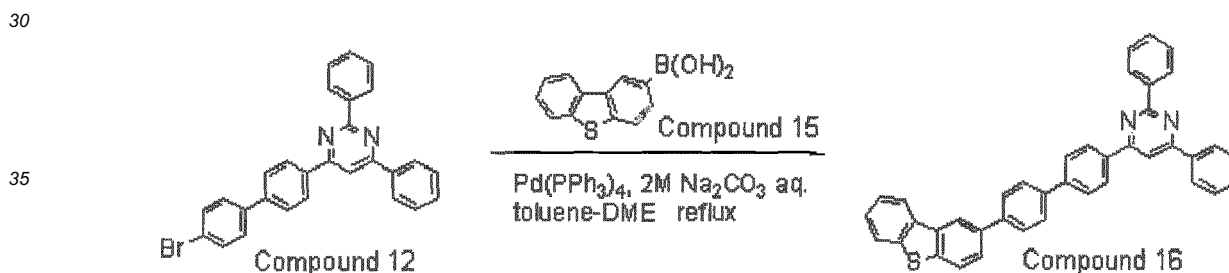
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[0317] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 12 (5.0 g, 11 mmol) in place of the compound 5 and the compound 8 (2.7 g, 13 mmol) in place of the compound 6, so that a compound 14 (5.5 g, a yield of 93%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 14.

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Synthesis Example 5: Synthesis of Compound 16

[0318] A synthesis scheme of a compound 16 is shown below.



40 (5-1) Synthesis of Compound 16

[0319] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 12 (5.0 g, 11 mmol) in place of the compound 5 and the compound 15 (3.0 g, 13 mmol) in place of the compound 6, so that a compound 16 (4.5 g, a yield of 74%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 16.

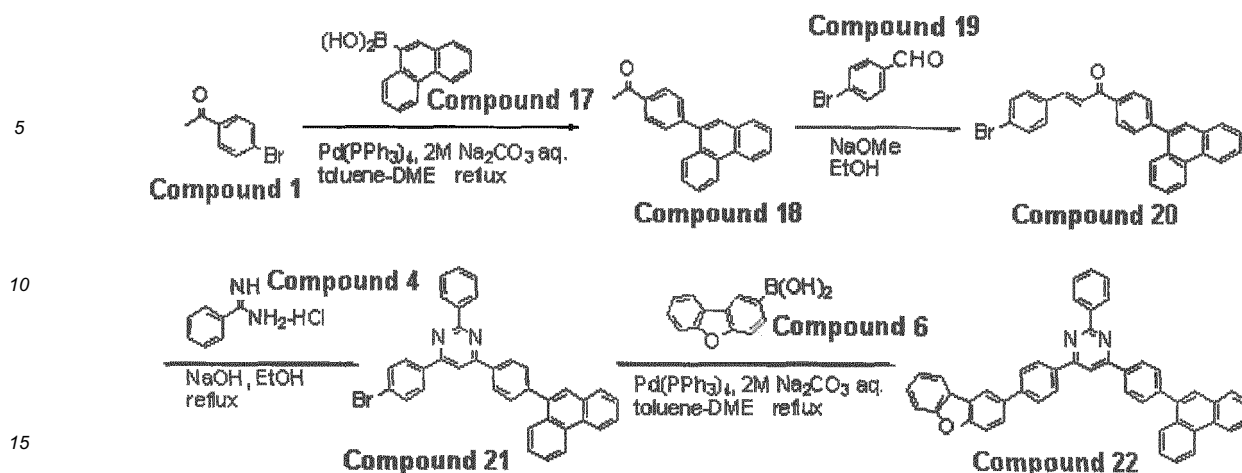
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Synthesis Example 6: Synthesis of Compound 22

[0320] A synthesis scheme of a compound 22 is shown below.

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(6-1) Synthesis of Compound 18

20 **[0321]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 1 (20 g, 100 mmol) in place of the compound 5 and a compound 17 (22 g, 100 mmol) in place of the compound 6, so that a compound 18 (25 g, a yield of 85%) in a form of a white solid was obtained.

(6-2) Synthesis of Compound 20

25 **[0322]** Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using 4-bromobenzaldehyde (a compound 19) (17 g, 89 mmol) in place of the compound 2 and the compound 18 (25 g, 85 mmol) in place of the compound 1, so that a compound 20 (38 g, a yield of 97%) in a form of a white solid was obtained.

(6-3) Synthesis of Compound 21

30 **[0323]** Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 20 (38 g, 82 mmol) in place of the compound 3, so that a compound 21 (16g, a yield of 34%) in a form of a white solid was obtained.

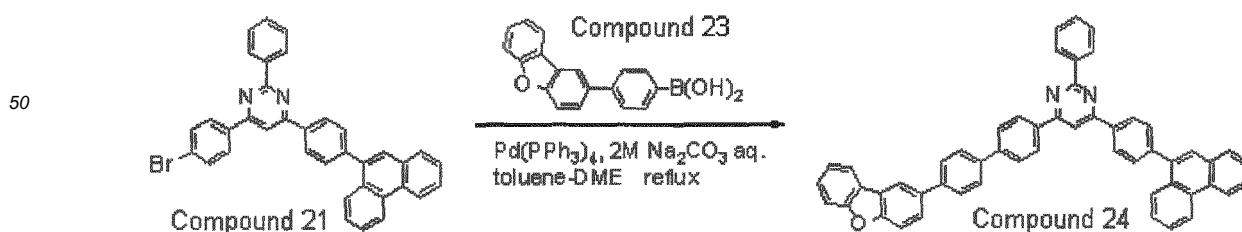
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(6-4) Synthesis of Compound 22

40 **[0324]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 21 (5.0 g, 8.9 mmol) in place of the compound 5, so that a compound 22 (3.9 g, a yield of 68%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 22.

Synthesis Example 7: Synthesis of Compound 24

45 **[0325]** A synthesis scheme of a compound 22 is shown below.



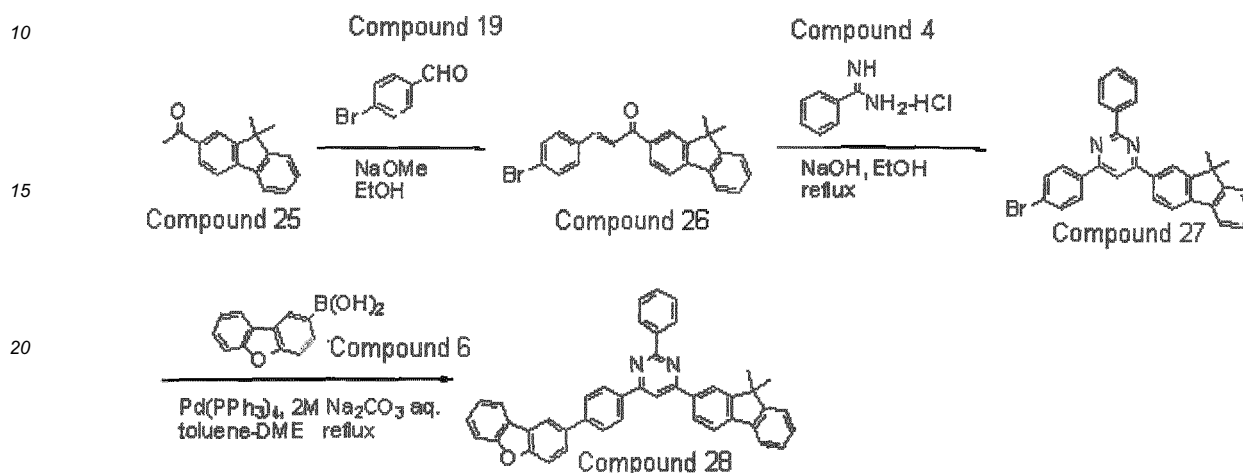
(7-1) Synthesis of Compound 24

[0326] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the

compound 21 (5.0 g, 8.9 mmol) in place of the compound 5 and a compound 23 (2.8 g, 9.8 mmol) in place of the compound 6, so that a compound 24 (25 g, a yield of 78%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 24.

5 Synthesis Example 8: Synthesis of Compound 28

[0327] A synthesis scheme of a compound 28 is shown below.



(8-1) Synthesis of Compound 26

[0328] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 19 (20 g, 109 mmol) in place of the compound 2 and a compound 25 (25 g, 106 mmol) in place of the compound 1, so that a compound 26 (27 g, a yield of 64%) in a form of a white solid was obtained.

(8-2) Synthesis of Compound 27

[0329] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 26 (27 g, 68 mmol) in place of the compound 3, so that a compound 27 (6.7 g, a yield of 20%) in a form of a white solid was obtained.

(8-3) Synthesis of Compound 28

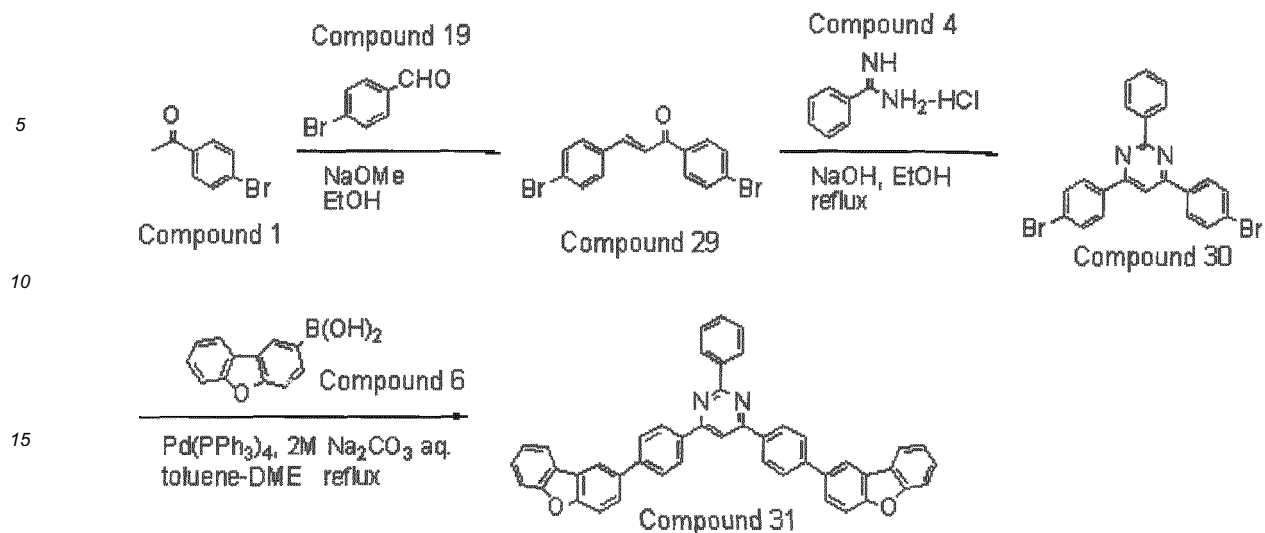
[0330] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 27 (3.4 g, 6.8 mmol) in place of the compound 5, so that a compound 28 (3.3 g, a yield of 82%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 28.

45 Synthesis Example 9: Synthesis of Compound 31

[0331] A synthesis scheme of a compound 31 is shown below.

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(9-1) Synthesis of Compound 29

[0332] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 19 (23 g, 125 mmol) in place of the compound 2, so that a compound 29 (41 g, a yield of 95%) in a form of a white solid was obtained.

25

(9-2) Synthesis of Compound 30

[0333] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 29 (20 g, 55 mmol) in place of the compound 3, so that a compound 30 (15 g, a yield of 61%) in a form of a white solid was obtained.

30

(9-3) Synthesis of Compound 31

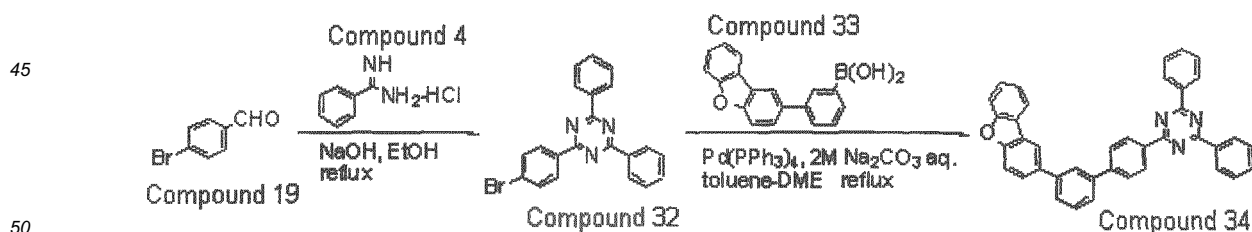
[0334] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using a compound 30 (6.9 g, 15 mmol) in place of the compound 5 and using 2.1 mol equivalent weight of the compound 6 relative to the compound 30, so that a compound 31 (5.3 g, a yield of 56%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 31.

35

Synthesis Example 10: Synthesis of Compound 34

40

[0335] A synthesis scheme of a compound 34 is shown below.



50

(10-1) Synthesis of Compound 32

[0336] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 19 (20 g, 108 mmol) in place of the compound 3 and using 2 mol equivalent weight of the compound 4 relative to the compound 19, so that a compound 32 (11 g, a yield of 26%) in a form of a white solid was obtained.

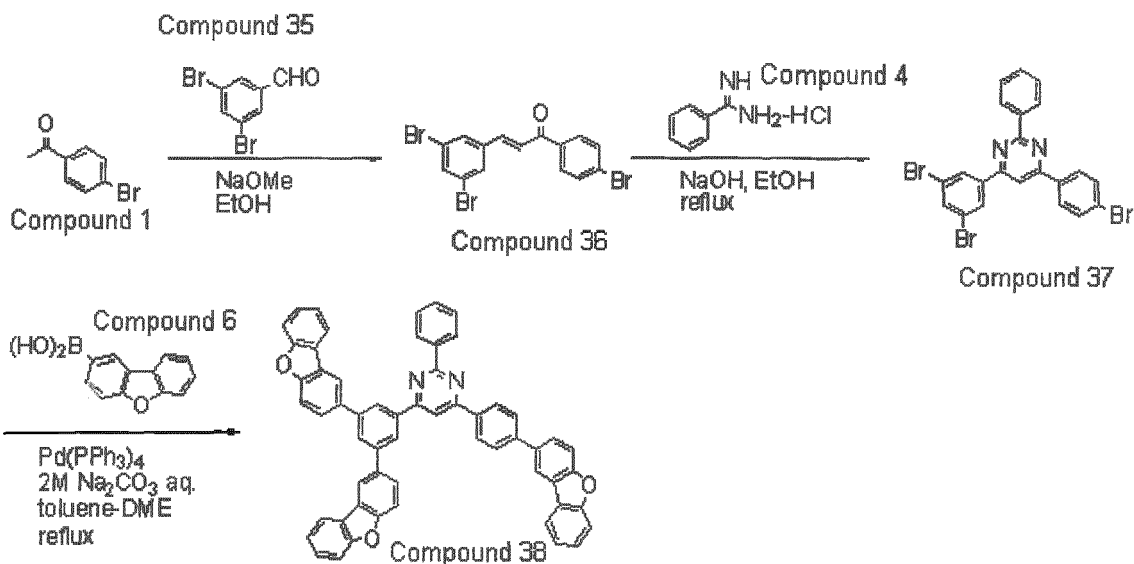
55

(10-2) Synthesis of Compound 34

[0337] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 32 (5.5 g, 14 mmol) in place of the compound 5 and using a compound 33 (4.9 g, 17 mmol) in place of the compound 6, so that a compound 34 (6.1 g, a yield of 78%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 34.

Synthesis Example 11: Synthesis of Compound 38

[0338] A synthesis scheme of a compound 38 is shown below.



(11-1) Synthesis of Compound 36

[0339] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using a compound 35 (28 g, 105 mmol) in place of the compound 2, so that a compound 36 (44 g, a yield of 98%) in a form of a white solid was obtained.

(11-2) Synthesis of Compound 37

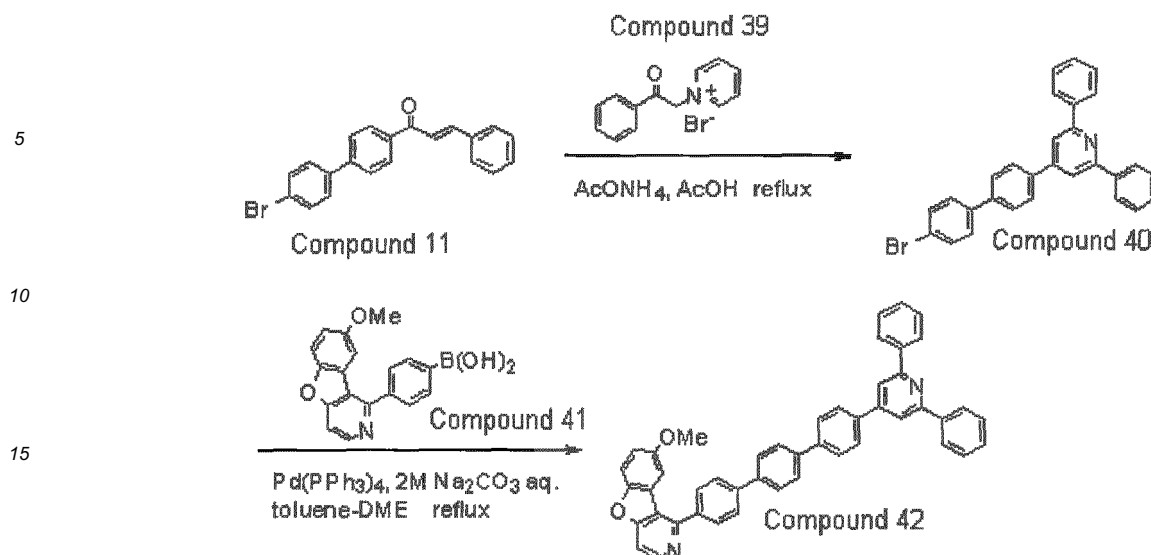
[0340] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 36 (44 g, 98 mmol) in place of the compound 3, so that a compound 37 (17 g, a yield of 23%) in a form of a white solid was obtained.

(11-3) Synthesis of Compound 38

[0341] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 37 (5.0 g, 9.2 mmol) in place of the compound 5 and using 3.1 mol equivalent weight of the compound 6 relative to the compound 37, so that a compound 38 (4.7 g, a yield of 64%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 38.

Synthesis Example 12: Synthesis of Compound 42

[0342] A synthesis scheme of a compound 42 is shown below.



20 (12-1) Synthesis of Compound 40

[0343] The compound 11 (11 g, 31 mmol), 1-phenylpyridinium bromide (a compound 39) (8.7 g, 31 mmol) and ammonium acetate (19 g, 250 mmol) were suspended in acetic acid (27 mL) and heated to reflux for 12 hours. After the reaction, the reactant solution was cooled to the room temperature, added with water and extracted with toluene. The obtained organic layer was washed with a 10-mass% aqueous sodium hydroxide and saturated saline in sequential order and dried with sodium sulfate. The solvent was distilled away under reduced pressure. The residue was added with ethanol and heated to reflux to be dissolved. The reactant solution was cooled and precipitated in a crystal form. The crystal was separated by filtration and washed with ethanol. Subsequently, the obtained solid was dried under reduced pressure, so that a compound 40 (13 g, a yield of 88%) in a form of a light-yellow solid was obtained.

30

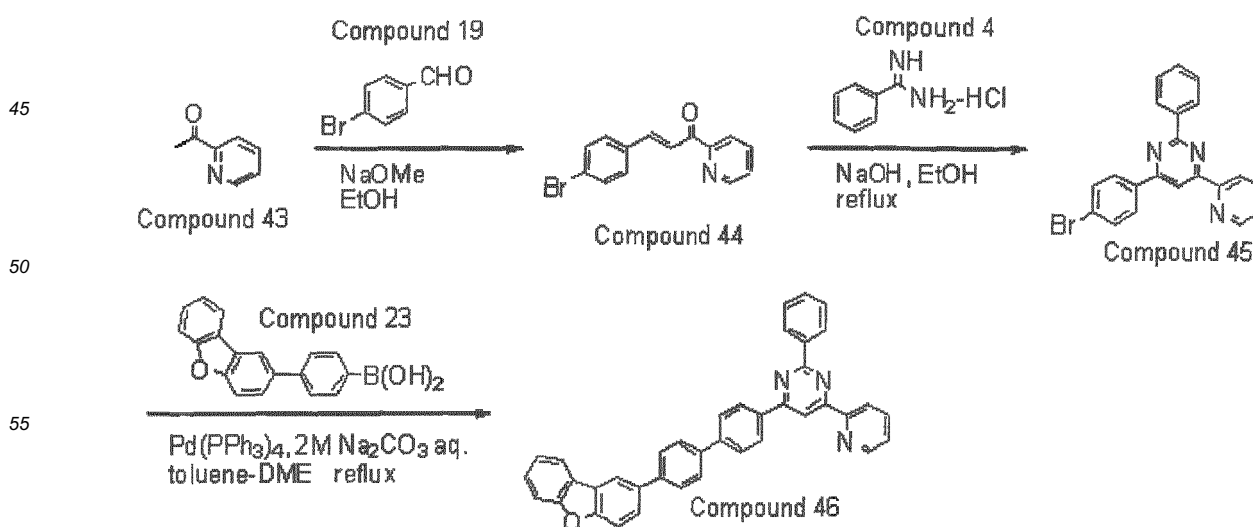
(12-2) Synthesis of Compound 42

[0344] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 40 (5.0 g, 11 mmol) in place of the compound 5 and a compound 41 (3.8 g, 12 mmol) in place of the compound 6, so that a compound 42 (4.4 g, a yield of 62%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 42.

35

Synthesis Example 13: Synthesis of Compound 46 (not according to the invention)

40 [0345] A synthesis scheme of a compound 46 is shown below.



(13-1) Synthesis of Compound 44

[0346] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using a compound 43 (9.8 g, 81 mmol) in place of the compound 1 and the compound 19 (16 g, 85 mmol) in place of the compound 2, so that a compound 44 (9.5g, a yield of 41%) in a form of a light-yellow solid was obtained.

(13-2) Synthesis of Compound 45

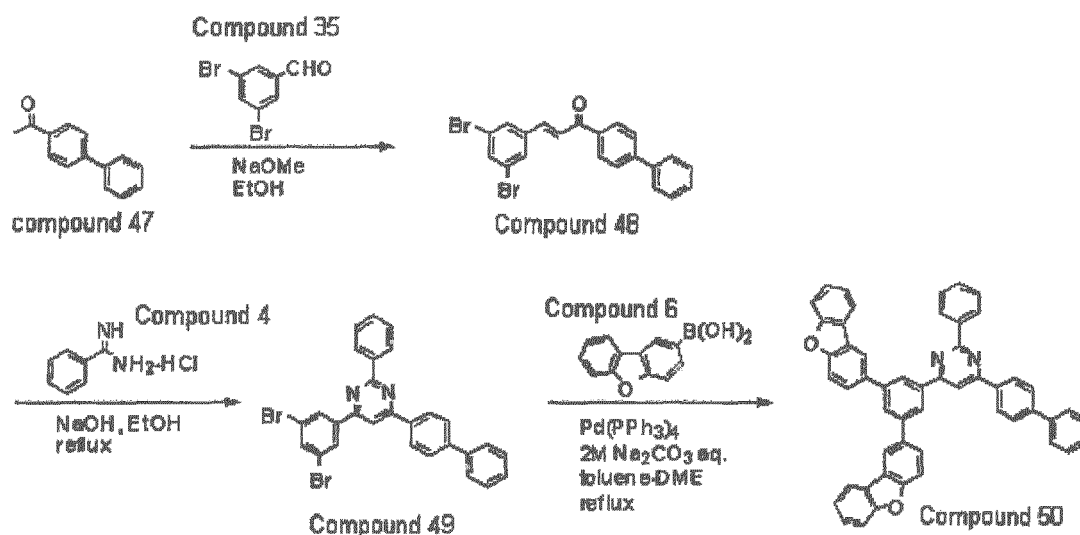
[0347] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 44 (9.5 g, 33 mmol) in place of the compound 3, so that a compound 45 (3.5 g, a yield of 27%) in a form of a white solid was obtained.

(13-3) Synthesis of Compound 46

[0348] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 45 (3.5 g, 9.0 mmol) in place of the compound 5 and the compound 23 (2.9 g, 9.9 mmol) in place of the compound 6, so that a compound 46 (2.9 g, a yield of 58%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 46.

Synthesis Example 14: Synthesis of Compound 50

[0349] A synthesis scheme of a compound 50 is shown below.



(14-1) Synthesis of Compound 48

[0350] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using a compound 47 (37 g, 191 mmol) in place of the compound 1 and using the compound 35 (50 g, 191 mmol) in place of the compound 2, so that a compound 48 (82 g, a yield of 98%) in a form of a yellow solid was obtained.

(14-2) Synthesis of Compound 49

[0351] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 48 (82 g, 186 mmol) in place of the compound 3, so that a compound 49 (40 g, a yield of 40%) in a form of a white solid was obtained.

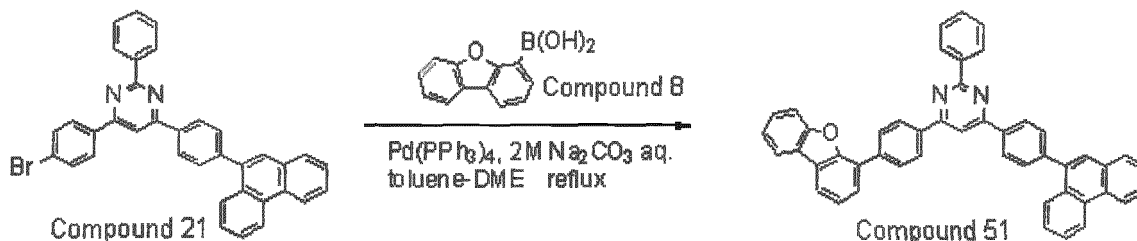
(14-3) Synthesis of Compound 50

[0352] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 49 (6.0 g, 11 mmol) in place of the compound 5 and using 2.2 mol equivalent weight of the compound 6

relative to the compound 49, so that a compound 50 (4.2 g, a yield of 53%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 50.

Synthesis Example 15: Synthesis of Compound 51

[0353] A synthesis scheme of a compound 51 is shown below.

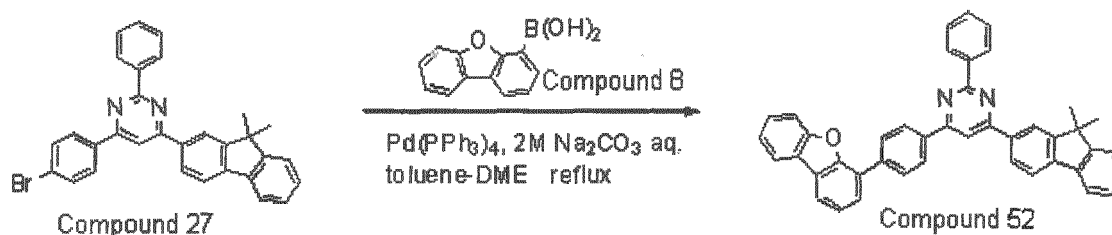


(15-1) Synthesis of Compound 51

[0354] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 21 (6.0 g, 11 mmol) in place of the compound 5 and using the compound 8 (2.7 g, 13 mmol) in place of the compound 6, so that a compound 51 (2.7g, a yield of 39%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 51.

Synthesis Example 16: Synthesis of Compound 52 (not according to the invention)

[0355] A synthesis scheme of a compound 52 is shown below.

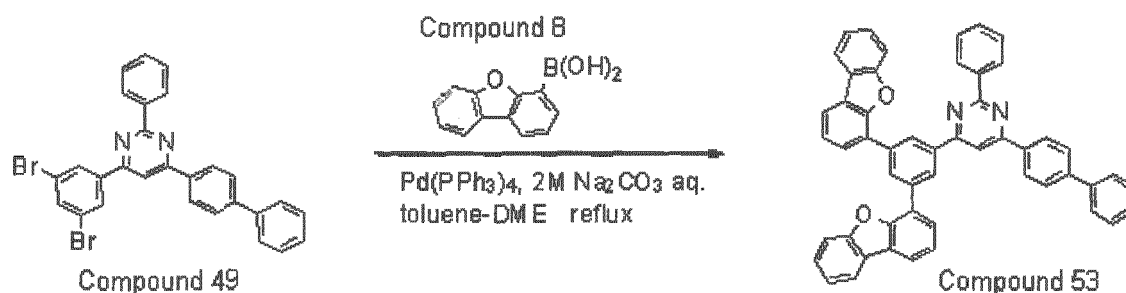


(16-1) Synthesis of Compound 52

[0356] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 27 (6.0 g, 12 mmol) in place of the compound 5 and the compound 8 (3.0 g, 14 mmol) in place of the compound 6, so that a compound 52 (5.6 g, a yield of 79%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 52.

Synthesis Example 17: Synthesis of Compound 53 (not according to the invention)

[0357] A synthesis scheme of a compound 53 is shown below.

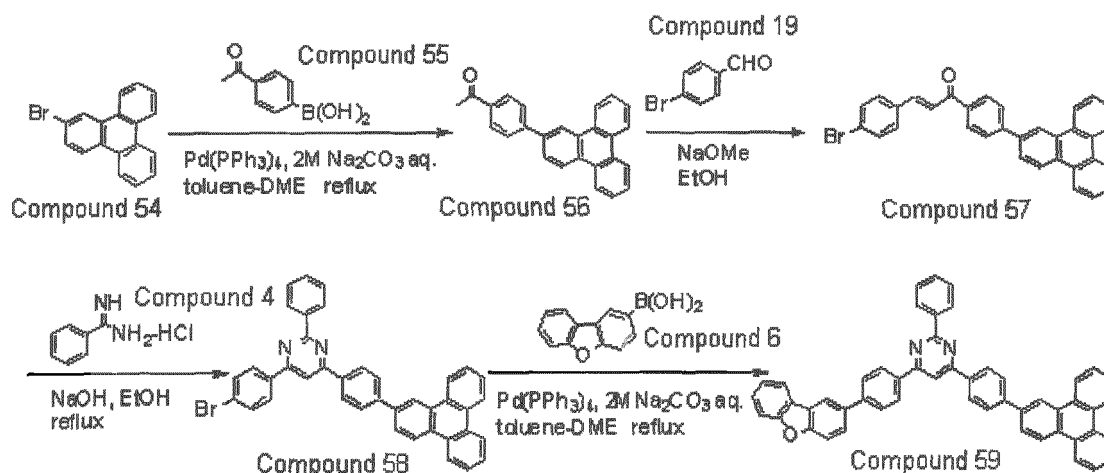


(17-1) Synthesis of Compound 53

[0358] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 49 (6.0 g, 11 mmol) in place of the compound 5 and using the compound 8 (5.2 g, 24 mmol) in place of the compound 6, so that a compound 53 (4.9 g, a yield of 61 %) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 53.

Synthesis Example 18: Synthesis of Compound 59

[0359] A synthesis scheme of a compound 59 is shown below.



(18-1) Synthesis of Compound 56

[0360] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 54 (54 g, 176 mmol) in place of the compound 5 and using the compound 55 (32 g, 194 mmol) in place of the compound 6, so that a compound 56 (58 g, a yield of 95%) in a form of a white solid was obtained.

(18-2) Synthesis of Compound 57

[0361] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 56 (42 g, 122 mmol) in place of the compound 1 and using 4-bromobenzaldehyde (the compound 19) (24 g, 128 mmol) in place of the compound 2, so that a partially refined product (78 g) of a compound 57 in a form of a yellow solid was obtained. Without further refining the product, the subsequent reaction was performed.

(18-3) Synthesis of Compound 58

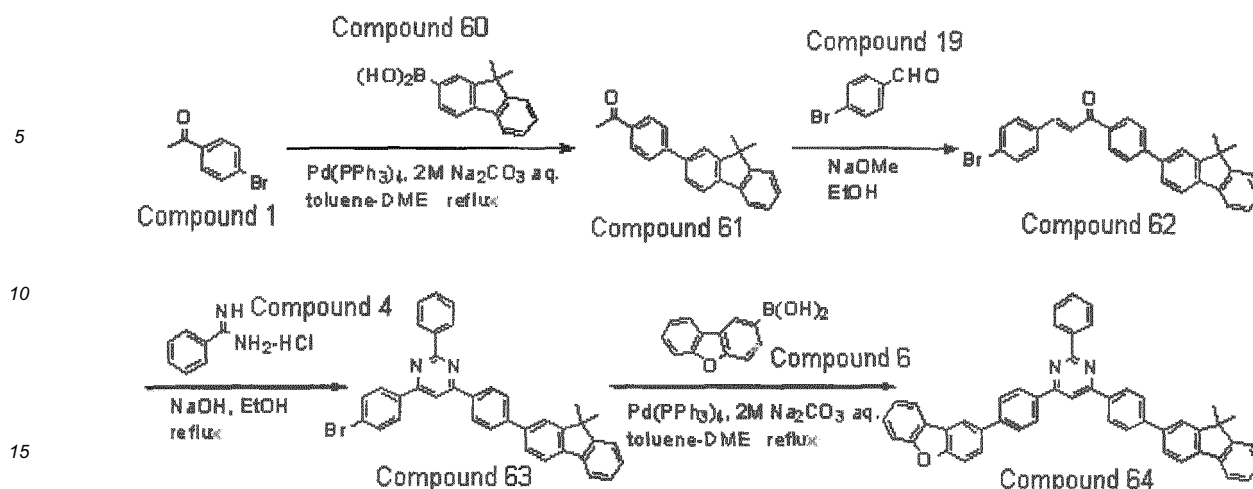
[0362] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the partially refined product (78 g) of the compound 57 in place of the compound 3, so that a compound 58 (27 g, a yield of 36%) in a form of a white solid was obtained.

(18-4) Synthesis of Compound 59

[0363] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 58 (6.0 g, 9.8 mmol) in place of the compound 5, so that a compound 59 (3.2 g, a yield of 47%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 59.

Synthesis Example 19: Synthesis of Compound 64

[0364] A synthesis scheme of a compound 64 is shown below.



(19-1) Synthesis of Compound 61

20 **[0365]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 1 (50 g, 251 mmol) in place of the compound 5 and using the compound 60 (66 g, 276 mmol) in place of the compound 6, so that a compound 61 (72 g, a yield of 92%) in a form of a white solid was obtained.

(19-2) Synthesis of Compound 62

25 **[0366]** Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 61 (72 g, 230 mmol) in place of the compound 1 and using 4-bromobenzaldehyde (the compound 19) (43 g, 230 mmol) in place of the compound 2, so that a compound 62 (107 g, a yield of 97%) in a form of a white solid was obtained.

(19-3) Synthesis of Compound 63

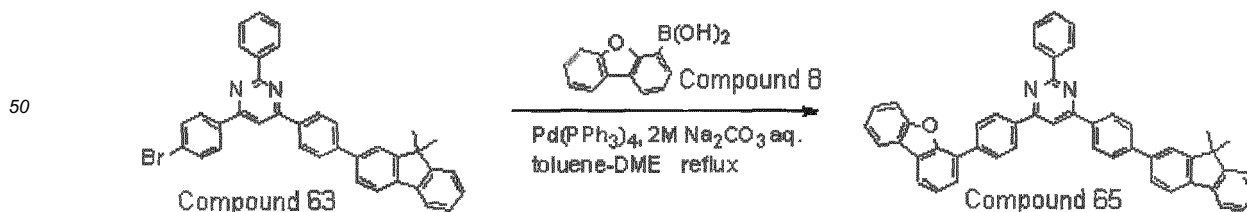
30 **[0367]** Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 62 (107 g, 223 mmol) in place of the compound 3, so that a compound 63 (47 g, a yield of 36%) in a form of a white solid was obtained.

(19-4) Synthesis of Compound 64

35 **[0368]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 63 (7.0 g, 12 mmol) in place of the compound 5, so that a compound 64 (4.9 g, a yield of 61 %) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 64.

Synthesis Example 20: Synthesis of Compound 65 (not according to the invention)

45 **[0369]** A synthesis scheme of a compound 65 is shown below.



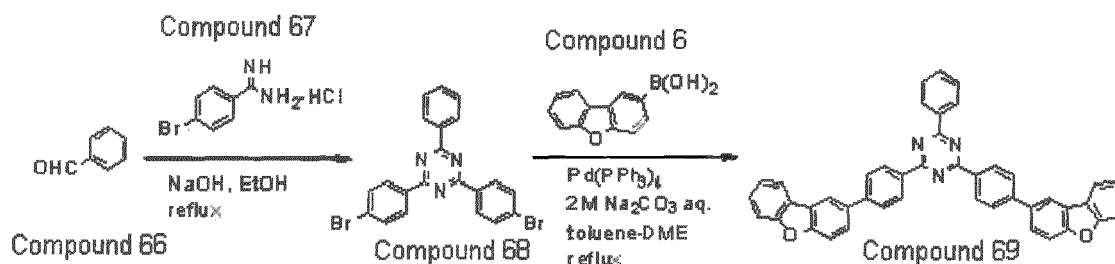
(20-1) Synthesis of Compound 65

[0370] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 63 (7.0 g, 12 mmol) in place of the compound 5 and the compound 8 (2.8 g, 13 mmol) in place of the compound

6, so that a compound 65 (4.3 g, a yield of 53%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 65.

Synthesis Example 21: Synthesis of Compound 69

[0371] A synthesis scheme of a compound 69 is shown below.



(21-1) Synthesis of Compound 68

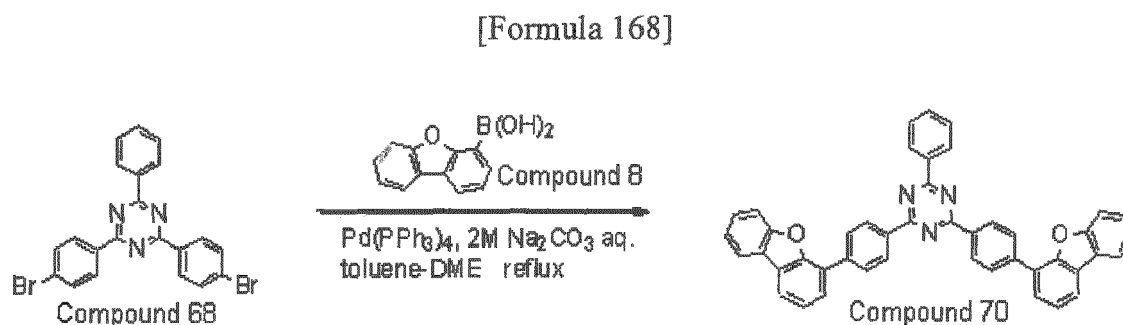
[0372] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using a compound 66 (7.7 g, 72 mmol) in place of the compound 3 and using the compound 67 (34 g, 145 mmol) in place of the compound 4, so that a compound 68 (9.4 g, a yield of 32%) in a form of a white solid was obtained.

(21-2) Synthesis of Compound 69

[0373] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 68 (4.5 g, 9.6 mmol) in place of the compound 5 and using 2.2 mol equivalent weight of the compound 6 relative to the compound 68, so that a compound 69 (4.4 g, a yield of 71%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 69.

Synthesis Example 22: Synthesis of Compound 70 (not according to the invention)

[0374] A synthesis scheme of a compound 70 is shown below.

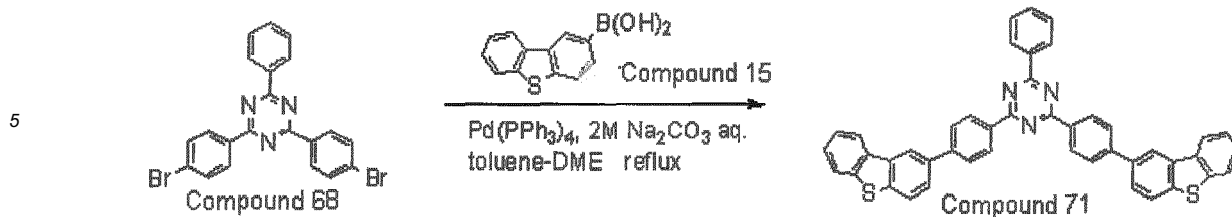


(22-1) Synthesis of Compound 70

[0375] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 68 (4.9 g, 10 mmol) in place of the compound 5 and using the compound 8 (4.9 g, 23 mmol) in place of the compound 6, so that a compound 70 (3.1 g, a yield of 46%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 70.

Synthesis Example 23: Synthesis of Compound 71

[0376] A synthesis scheme of a compound 71 is shown below.

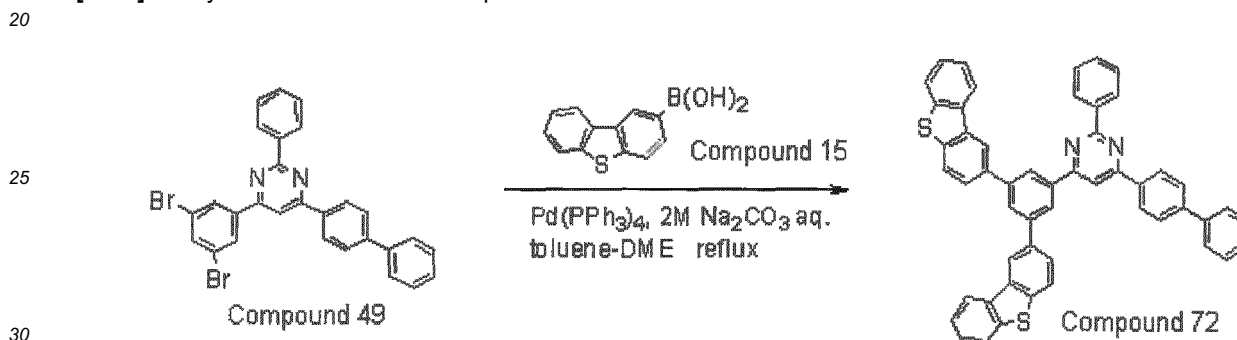


10 (23-1) Synthesis of Compound 71

[0377] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 68 (4.5 g, 9.6 mmol) in place of the compound 5 and using the compound 15 (4.8 g, 21 mmol) in place of the compound 6, so that a compound 71 (3.4 g, a yield of 53%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 71.

15 Synthesis Example 24: Synthesis of Compound 72

[0378] A synthesis scheme of a compound 72 is shown below.

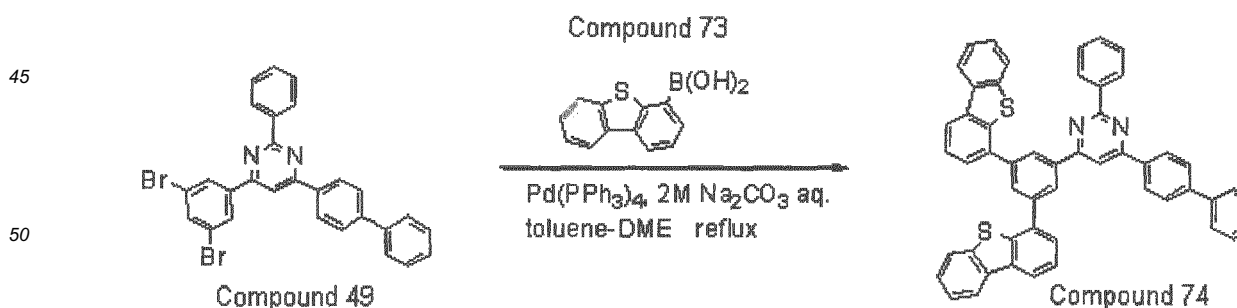


(24-1) Synthesis of Compound 72

[0379] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 49 (10 g, 18 mmol) in place of the compound 5 and using the compound 15 (9.3 g, 41 mmol) in place of the compound 6, so that a compound 72 (9.7 g, a yield of 70%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 72.

35 Synthesis Example 25: Synthesis of Compound 74 (not according to the invention)

[0380] A synthesis scheme of a compound 74 is shown below.



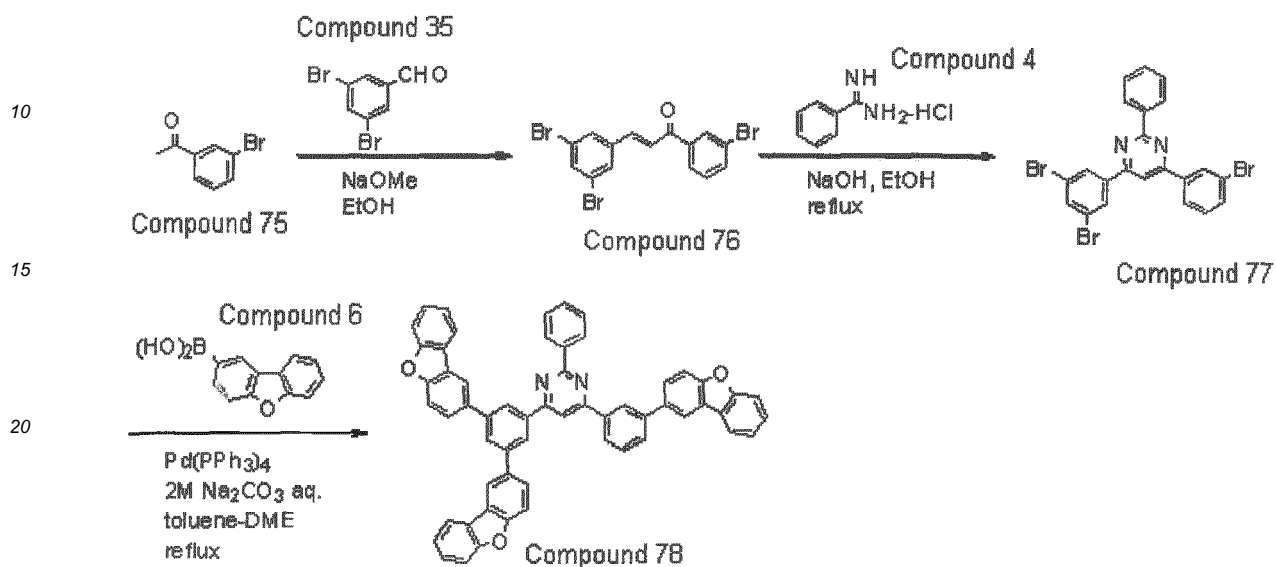
(25-1) Synthesis of Compound 74

[0381] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 49 (10 g, 18 mmol) in place of the compound 5 and using the compound 73 (9.3 g, 41 mmol) in place of the compound 6, so that a compound 74 (9.5 g, a yield of 68%) in a form of a white solid was obtained. As a result of FD-

MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 74.

Synthesis Example 26: Synthesis of Compound 78

5 **[0382]** A synthesis scheme of a compound 78 is shown below.



(26-1) Synthesis of Compound 76

30 **[0383]** Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 75 (38 g, 193 mmol) in place of the compound 1 and the compound 35 (50 g, 189 mmol) in place of the compound 2, so that a compound 76 (75 g, a yield of 89%) in a form of a white solid was obtained.

(26-2) Synthesis of Compound 77

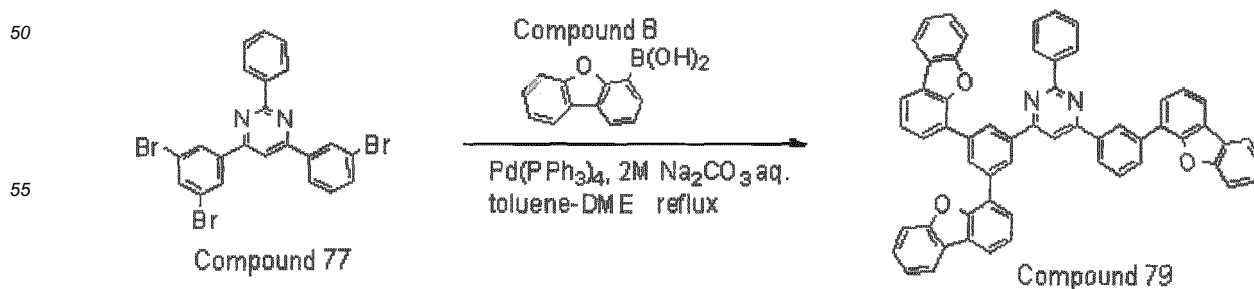
35 **[0384]** Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 76 (71 g, 158 mmol) in place of the compound 3, so that a compound 77 (31 g, a yield of 33%) in a form of a white solid was obtained.

(26-3) Synthesis of Compound 78

40 **[0385]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 77 (15 g, 28 mmol) in place of the compound 5 and using 3.5 equivalent weight of the compound 6 relative to the compound 77, so that a compound 78 (5.0 g, a yield of 23%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 78.

45 Synthesis Example 27: Synthesis of Compound 79 (not according to the invention)

[0386] A synthesis scheme of a compound 79 is shown below.

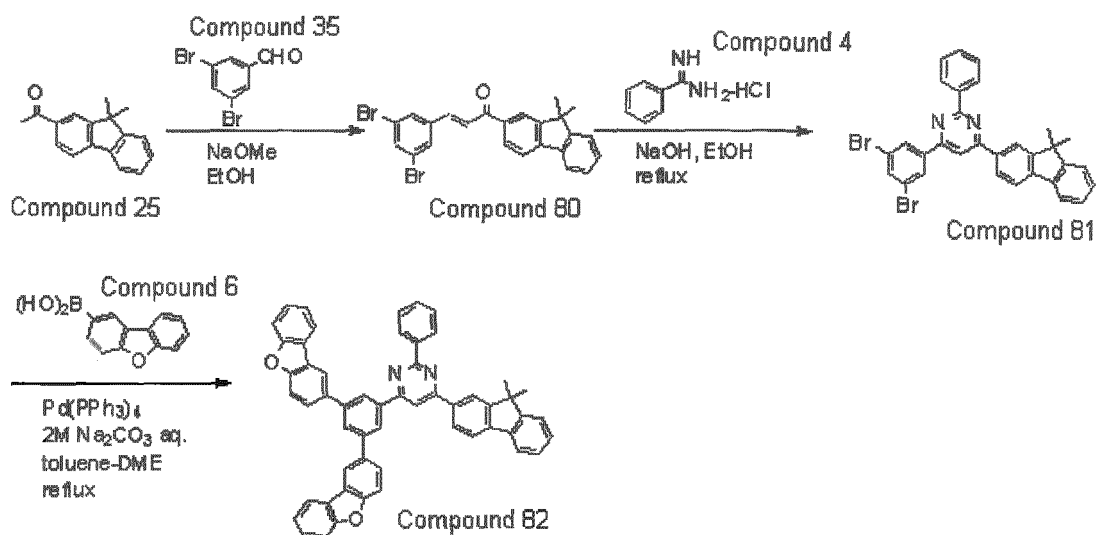


(27-1) Synthesis of Compound 79

[0387] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 77 (15 g, 28 mmol) in place of the compound 5 and using the compound 8 (20 g, 96 mmol) in place of the compound 6, so that a compound 79 (12 g, a yield of 52%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 79.

Synthesis Example 28: Synthesis of Compound 82

[0388] A synthesis scheme of a compound 82 is shown below.



(28-1) Synthesis of Compound 80

[0389] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 25 (34 g, 144 mmol) in place of the compound 1 and using the compound 35 (42 g, 158 mmol) in place of the compound 2, so that a compound 80 (69 g, a yield of 100%) in a form of a light-brown solid was obtained.

(28-2) Synthesis of Compound 81

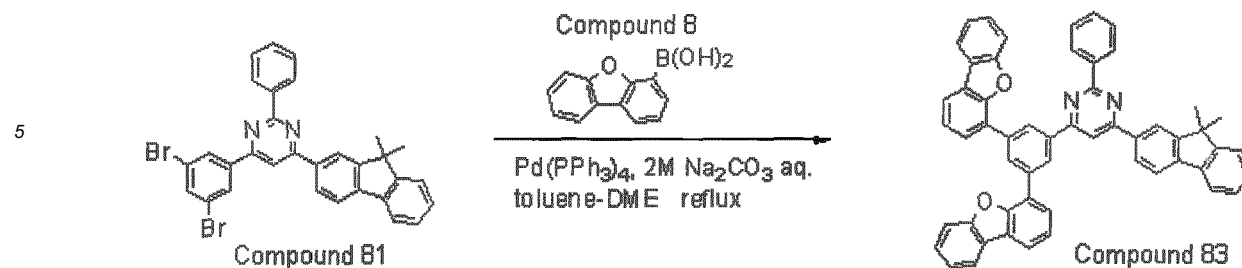
[0390] Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 80 (69 g, 144 mmol) in place of the compound 3, so that a compound 81 (18 g, a yield of 22%) in a form of a light-yellow solid was obtained.

(28-3) Synthesis of Compound 82

[0391] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 81 (9.1g, 16 mmol) in place of the compound 5 and using 2.2 mol equivalent weight of the compound 6 relative to the compound 81, so that a compound 82 (9.4 g, a yield of 79%) in a form of a light-yellow solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 82.

Synthesis Example 29: Synthesis of Compound 83 (not according to the invention)

[0392] A synthesis scheme of a compound 83 is shown below.

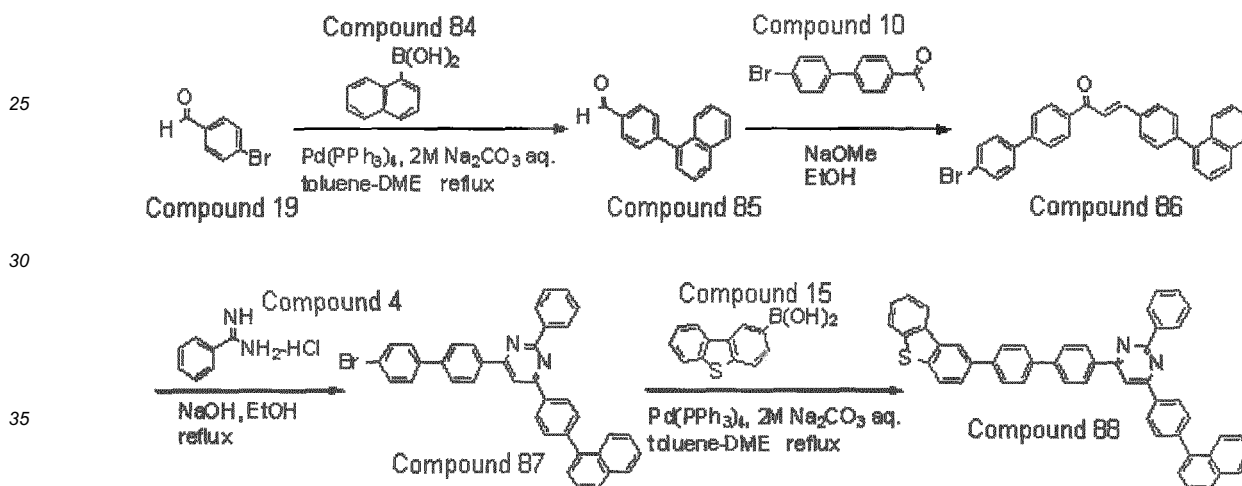


(29-1) Synthesis of Compound 83

15 **[0393]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 81 (9.0 g, 15 mmol) in place of the compound 5 and using the compound 8 (7.2 g, 34 mmol) in place of the compound 6, so that a compound 83 (8.1 g, a yield of 69%) in a form of a light-yellow solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 83.

Synthesis Example 30: Synthesis of Compound 88

20 **[0394]** A synthesis scheme of a compound 88 is shown below.



(30-1) Synthesis of Compound 85

40 **[0395]** Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 19 (50 g, 270 mmol) in place of the compound 5 and using the compound 84 (56 g, 325 mmol) in place of the compound 6, so that a compound 85 (53 g, a yield of 85%) in a form of a white solid was obtained.

45 (30-2) Synthesis of Compound 86

[0396] Synthesis was made by the same method as in (1-1) synthesis of the compound 7, except for using the compound 10 (65 g, 235 mmol) in place of the compound 1 and using the compound 85 (53 g, 229 mmol) in place of the compound 2, so that a compound 86 (109g) in a form of a light-yellow solid was obtained.

50 (30-3) Synthesis of Compound 87

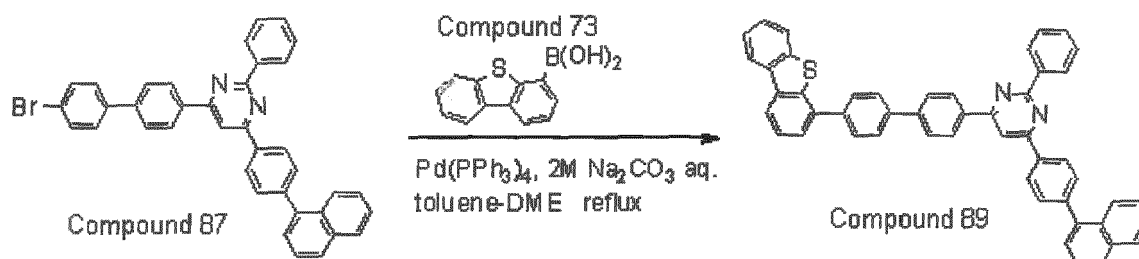
55 **[0397]** Synthesis was made by the same method as in (1-2) synthesis of the compound 7, except for using the compound 86 (108 g, 221 mmol) in place of the compound 3, so that a compound 87 (44 g, a yield of 34%) in a form of a white solid was obtained.

(30-4) Synthesis of Compound 88

[0398] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 87 (6.0 g, 10 mmol) in place of the compound 5 and using the compound 15 (2.8 g, 12 mmol) in place of the compound 6, so that a compound 88 (5.9 g, a yield of 84%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 88.

Synthesis Example 31: Synthesis of Compound 89 (not according to the invention)

[0399] A synthesis scheme of a compound 89 is shown below.

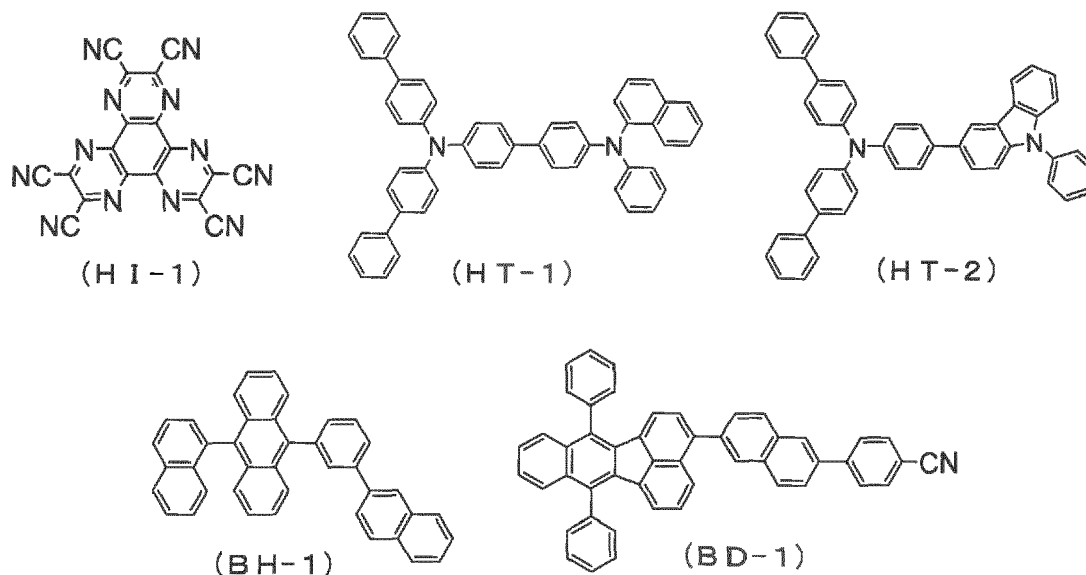


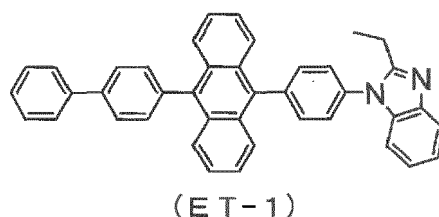
(31-1) Synthesis of Compound 89

[0400] Synthesis was made by the same method as in (1-3) synthesis of the compound 7, except for using the compound 87 (6.0 g, 10 mmol) in place of the compound 5 and using the compound 73 (2.8 g, 12 mmol) in place of the compound 6, so that a compound 89 (6.4 g, a yield of 91%) in a form of a white solid was obtained. As a result of FD-MS (Field Desorption Mass Spectrometry) analysis, the reactant was identified as the compound 89.

Manufacturing of Organic EL Device

[0401] For manufacturing the organic EL device, the following compounds were used in addition to the compounds synthesized in the above Synthesis Examples.





Example 1

[0402] A glass substrate (size: 25 mm × 75 mm × 0.7mm thick, manufactured by Geomatec Co., Ltd.) having an ITO transparent electrode (anode) was ultrasonic-cleaned in isopropyl alcohol for five minutes, and then UV/ozone-cleaned for 30 minutes. After the glass substrate having the transparent electrode line was cleaned, the glass substrate was mounted on a substrate holder of a vacuum evaporation apparatus. Initially, a compound HI-1 was evaporated on a surface of the glass substrate where the transparent electrode line was provided in a manner to cover the transparent electrode, thereby forming a 5-nm thick HI-1 film. The HI-1 film serves as a hole injecting layer.

[0403] After the formation of the HT-1 film, a compound HT-1 was evaporated on the HT-1 film to form an 80-nm thick HT-1 film. The HT-1 film serves as a first hole transporting layer.

[0404] After the film formation of the HT-1 film, a compound HT-2 was evaporated on the HT-1 film to form a 15-nm thick HT-2 film on the HT-1 film. The HT-2 film serves as a second hole transporting layer.

[0405] A compound BH-1 (host material) and a compound BD-1 (dopant material) (mass ratio of BH-1 to BD-1 was 20: 1) were co-evaporated on the HT-2 film to form a 25-nm thick emitting layer.

[0406] The compound 7 was evaporated on this emitting layer to form a 20-nm thick blocking layer.

[0407] A compound ET-1 (electron transporting material) was evaporated on the blocking layer to form a 5-nm thick electron injecting layer.

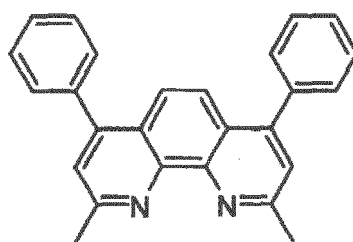
[0408] LiF was evaporated on the electron injecting layer to form a 1-nm thick LiF film.

[0409] A metal A1 was evaporated on the LiF film to form an 80-nm thick metal cathode.

[0410] Thus, the organic EL device of Example 1 was manufactured.

Examples 2 to 19 and Comparative 1

[0411] Organic EL devices of Examples 2 to 19 and Comparative 1 were manufactured in the same manner as the organic EL device in the Example 1, except for using materials shown in Table 1 as a material for the blocking layer. BCP shown below was used as the material for the blocking layer of the organic EL device in Comparative 1.



BCP

Evaluation of Device

[0412] The manufactured organic EL devices were evaluated as below. The results are shown in Table 1.

Initial Performance

[0413] Voltage was applied to the organic EL device so that a current density becomes 10 mA/cm², and a voltage value (V) at that time was measured. EL spectra were measured with a spectral radiance meter (CS-1000, manufactured by KONICA MINOLTA). Chromaticity CIE_x, CIE_y, current efficiency L/J(cd/A) and external quantum efficiency EQE (%) were calculated from the obtained spectral-radiance spectra.

Measurement of TTF Ratio

[0414] When the triplet energy of the host material, the dopant material and the blocking layer material satisfies a predetermined relation, the ratio of luminous intensity derived from TTF relative to the entire emission can be high, so that a fluorescent device can be highly efficient to the level unachievable by a typically known fluorescent device.

[0415] The ratio of luminous intensity derived from TTF is measurable by a transitional EL method. The transitional EL method is a method for measuring reduction behavior (transitional property) of EL emission after DC voltage applied on the device is removed. EL luminous intensity are classified into a luminescence component from singlet excitons generated in first recombination and a luminescence component from singlet excitons generated through TTF phenomenon. Since lifetime of the singlet excitons is very short at nano-second order, EL emission is rapidly reduced after removal of DC voltage.

[0416] On the other hand, since the TTF phenomenon provides emission from singlet excitons generated through long-life triplet excitons, EL emission is gradually reduced. Thus, since there is a large difference in time between emission from the singlet excitons and emission from the triplet excitons, luminous intensity derived from TTF is obtainable. Specifically, the luminous intensity can be determined by the following method.

[0417] Transitional EL waveform is measured as follows (see Fig. 13). Pulse voltage waveform output from a voltage pulse generator (PG) is applied on an EL device. The applied voltage waveform is loaded in an oscilloscope (OSC). When a pulse voltage is applied on the EL device, the EL device generates pulse emission. This emission is loaded in the oscilloscope (OSC) through a photomultiplier (PMT). The voltage waveform and the pulse emission are synchronized and loaded in a personal computer (PC).

[0418] The ratio of luminous intensity derived from TTF is determined as follows based on analysis of the transitional EL waveform.

[0419] A rate equation for reduction behavior of triplet excitons is resolved and the reduction behavior of luminous intensity based on TTF phenomenon is brought into modeling. Time-varying reduction of a density n_T of triplet excitons within the emitting layer is represented by the following rate equation using a reduction speed α due to lifetime of the triplet excitons and a reduction speed γ due to collision of the triplet excitons.

[Formula 1]

$$\frac{dn_T}{dt} = -\alpha \cdot n_T - \gamma \cdot n_T^2$$

[0420] When this differential formula is approximately resolved, the following formula is obtained. Herein, I_{TTF} represents luminous intensity derived from TTF. A is a constant. Thus, when the transitional EL emission is based on TTF, a reciprocal number of the square root of intensity of the transitional EL emission is shown approximately in a linear line. The measured transitional EL waveform data is fit in the following approximate expression to obtain the constant A. At this time, luminous intensity $1/A^2$ at the time $t=0$ when DC voltage is removed is defined as the ratio of luminous intensity derived from TTF.

[Formula 2]

$$\frac{1}{\sqrt{I_{TTF}}} \propto A + \gamma \cdot t$$

[0421] A graph of Fig. 14(A) shows a measurement example where a predetermined DC voltage is applied on the EL device and then the DC voltage is removed and shows time-varying luminous intensity of the EL device. The DC voltage was removed at the time of about 3×10^{-8} seconds in the graph of Fig. 14(A). In the graph, the luminous intensity when voltage is removed is defined as 1. After rapid reduction before the elapse of about 2×10^{-7} , a gradual reduction component

appears. In the graph of Fig. 14(B), the voltage removal time is a starting point and the reciprocal numbers of the square root of luminous intensity before the elapse of 10^{-5} seconds after voltage removal are plotted in an approximately linear line. A value at an intersection A of the ordinate axis and the linear line extended to the starting point is 2.41. Accordingly, the ratio of luminous intensity derived from TTF obtained from the transitional EL waveform is $1/2.41^2=0.17$, which means 17% of the entire luminous intensity is derived from TTF.

[0422] The luminous intensity is preferably fitted in a linear line by the method of least squares. In this case, the luminous intensity before the elapse of 10^{-5} seconds is preferably fitted.

[0423] Voltage pulse waveform (pulse width: 500 micro second, frequency: 20 Hz, voltage: equivalent to 0.1 to 100 mA/cm²) output from a pulse generator (8114A: manufactured by Agilent Technologies) was applied. EL emission was input in a photomultiplier (R928: manufactured by Hamamatsu Photonics K.K.). The pulse voltage waveform and the EL emission were synchronized and loaded in an oscilloscope (2440: Tektronix) to obtain a transitional EL waveform. The transitional EL waveform was analyzed to determine a TTF ratio.

[0424] Voltage was applied on the organic EL device of Example 1 at the room temperature. The pulse voltage was removed at the time of about 3×10^{-8} seconds.

[0425] Based on the graph, where the voltage removal time was a starting point and the reciprocal numbers of the square root of luminous intensity before the elapse of 1.5×10^{-5} seconds after voltage removal were plotted, the TTF ratio was obtained. The same measurement was performed in other Examples and Comparatives. The results are shown in Table 1.

	Blocking layer material	Voltage	Chromaticity CIE		L/J	EQE	TTF Ratio
		(V)	x	y	(cd/A)	(%)	(%)
Example 1	Compound 7	3.38	0.144	0.131	10.32	9.21	28
Example 2*	Compound 9	3.51	0.142	0.137	10.38	9.06	32
Example 3	Compound 13	3.65	0.142	0.151	10.63	8.64	33
Example 4*	Compound 14	3.52	0.145	0.131	8.84	8.76	29
Example 5	Compound 16	3.55	0.145	0.127	9.76	8.81	32
Example 6	Compound 22	3.61	0.143	0.137	11.09	8.85	32
Example 7	Compound 31	3.75	0.143	0.134	10.05	8.87	35
Example 8	Compound 50	4.08	0.144	0.128	9.78	8.87	29
Example 9*	Compound 51	3.59	0.143	0.137	11.10	9.64	31
Example 10*	Compound 52	3.81	0.143	0.132	11.17	9.96	33
Example 11*	Compound 53	4.15	0.143	0.134	11.03	9.72	33
Example 12	Compound 24	3.71	0.143	0.131	10.53	9.42	30
Example 13	Compound 72	3.81	0.141	0.138	10.03	8.74	32
Example 14*	Compound 74	3.64	0.143	0.135	11.34	9.96	30
Example 15	Compound 78	3.75	0.142	0.130	11.04	9.96	31
Example 16*	Compound 79	4.02	0.142	0.132	10.50	9.40	34
Example 17*	Compound 83	3.78	0.143	0.134	11.21	9.87	36
Example 18	Compound 88	3.58	0.142	0.136	11.05	9.69	31
Example 19*	Compound 89	3.56	0.143	0.135	10.78	9.43	28
Comp.1	BCP	4.30	0.144	0.128	8.43	7.65	25
(*: not according to the invention)							

[0426] Since the aromatic heterocyclic compound of the invention was used for the blocking layer of the organic EL devices of Examples 1 to 19, the organic EL devices of Examples 1 to 19 exhibited a higher TTF ratio, a higher current efficiency and a higher external quantum efficiency than the organic EL device of Comparative 1. Moreover, a drive voltage of the organic EL devices of Examples 1 to 19 was also lower than that of the organic EL device of Comparative 1.

INDUSTRIAL APPLICABILITY

[0427] An organic EL device comprising the aromatic heterocyclic compound of the invention is applicable to a display and an illuminator.

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EXPLANATION OF CODES

[0428]

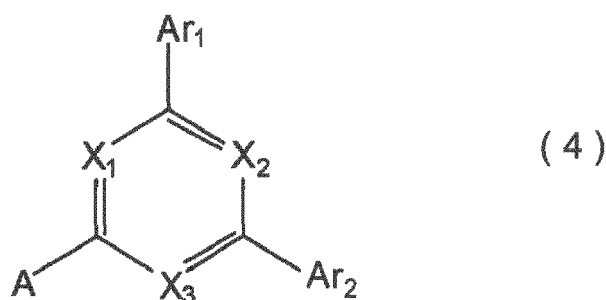
10	1,2,3,4,5	organic EL device
	10	anode
	20, 22, 24	emitting layer
	30, 32	blocking layer
	40	electron injecting layer
15	41	electron transporting layer
	50	cathode
	60	hole transporting zone
	70	electron transporting zone

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Claims

1. An aromatic heterocyclic derivative represented by a formula (4) below,

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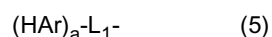


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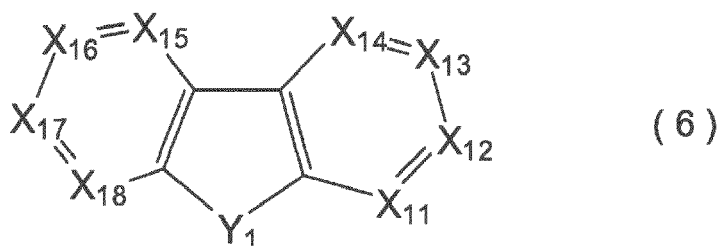
wherein: X_1 to X_3 are a nitrogen atom or CR_1 , with a proviso that at least one of X_1 to X_3 is a nitrogen atom; R_1 independently represents a hydrogen atom, a halogen atom, a cyano group, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, a substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, a substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms; A is represented by a formula (5) below,

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wherein: HAr is represented by a formula (6) below;
 a is an integer of 1 to 5;
 when a is 1, L_1 is a divalent linking group;
 when a is in a range of 2 to 5, L_1 is a trivalent to hexavalent linking group and HAr is the same or different;
 the linking group is a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, or a divalent to hexavalent residue induced from any one of a group formed by mutually bonding two or three of the
 55 substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms; and
 the mutually bonded groups are the same or different,



wherein: Y_1 is an oxygen atom or a sulfur atom;

X_{11} and X_{18} are a nitrogen atom or CR_{13} ;

one of X_{12} to X_{17} is a carbon atom bonded to L_1 by a single bond and the rest of X_{12} to X_{17} are a nitrogen atom or CR_{13} ;

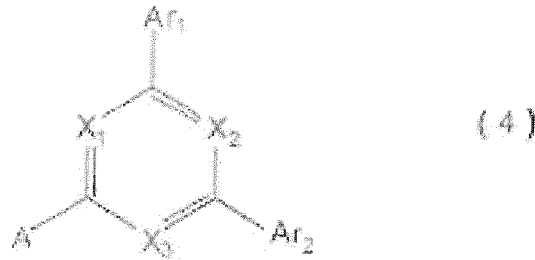
R_{13} independently represents a hydrogen atom, a halogen atom, a cyano group, a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkyl group having 1 to 30 carbon atoms, a substituted or unsubstituted alkenyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkynyl group having 2 to 30 carbon atoms, a substituted or unsubstituted alkylsilyl group having 3 to 30 carbon atoms, a substituted or unsubstituted arylsilyl group having 6 to 30 ring carbon atoms, a substituted or unsubstituted alkoxy group having 1 to 30 carbon atoms, a substituted or unsubstituted aralkyl group having 6 to 30 ring carbon atoms, or a substituted or unsubstituted aryloxy group having 6 to 30 ring carbon atoms; a plurality of R_{13} are mutually the same or different; adjacent R_{13} optionally bond to each other to form a ring; and in the above formula (4), Ar_1 and Ar_2 each are independently represented by the formula (5), or represent a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms, when at least one of R_1 , L_1 , R_{13} , Ar_1 and Ar_2 is substituted, the substituent of R_1 , L_1 , R_{13} , Ar_1 and Ar_2 each being independently selected from the group consisting of an aryl group, linear or branched alkyl group, cycloalkyl group, halogenated alkyl group, alkenyl group, alkynyl group, alkylsilyl group, arylsilyl group, alkoxy group, halogenated alkoxy group, aralkyl group, aryloxy group, halogen atom, cyano group, hydroxyl group, nitro group and carboxy group.

2. The aromatic heterocyclic derivative according to claim 1, wherein in the formula (6), X_{13} or X_{16} is a carbon atom bonded to L_1 by a single bond.
3. The aromatic heterocyclic derivative according to claim 1 or 2, wherein in the formula (5), a is an integer of 1 to 3.
4. The aromatic heterocyclic derivative according to claim 1 or 2, wherein in the formula (5), a is 1 or 2.
5. The aromatic heterocyclic derivative according to claim 1 or 2, wherein in the formula (5), a is 1, L_1 is a linking group, and the linking group is a divalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.
6. The aromatic heterocyclic derivative according to claim 1 or 2, wherein in the formula (5), a is 2 and L_1 is a linking group, and the linking group is a trivalent residue of a substituted or unsubstituted aryl group having 6 to 30 ring carbon atoms.
7. The aromatic heterocyclic derivative according to any one of claims 1 to 6, wherein in the formula (6), Y_1 is an oxygen atom.
8. The aromatic heterocyclic derivative according to any one of claims 1 to 7, wherein in the formula (6), Y_1 is an oxygen atom, X_{11} and X_{18} are CR_{13} , and one of X_{12} to X_{17} is a carbon atom bonded to L_1 by a single bond and the rest of X_{12} to X_{17} are CR_{13} .
9. The aromatic heterocyclic derivative according to any one of claims 1 to 8, wherein in the formula (4), two or three of X_1 to X_3 are a nitrogen atom.

10. The aromatic heterocyclic derivative according to any one of claims 1 to 9, wherein in the formula (5), L_1 is a divalent or trivalent residue induced from any one of benzene, biphenyl, terphenyl, naphthalene and phenanthrene.
11. An organic-electroluminescence-device material comprising the aromatic heterocyclic derivative according to any one of claims 1 to 10.

Patentansprüche

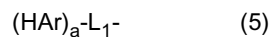
1. Aromatisches heterocyclisches Derivat nach der untenstehenden Formel (4),



worin: X_1 bis X_3 ein Stickstoffatom oder CR_1 sind, unter der Voraussetzung, dass mindestens eines von X_1 bis X_3 ein Stickstoffatom ist;

R_1 jeweils unabhängig voneinander ein Wasserstoffatom, ein Halogenatom, eine Cyanogruppe, eine substituierte oder unsubstituierte Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen, eine substituierte oder unsubstituierte Alkylgruppe mit 1 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Alkenylgruppe mit 2 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Alkynylgruppe mit 2 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Alkylsilylgruppe mit 3 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Arylsilylgruppe mit 6 bis 30 Ringkohlenstoffatomen, eine substituierte oder unsubstituierte Alkoxygruppe mit 1 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Aralkylgruppe mit 6 bis 30 Ringkohlenstoffatomen oder eine substituierte oder unsubstituierte Aryloxygruppe mit 6 bis 30 Ringkohlenstoffatomen ist;

A der untenstehenden Formel (5) entspricht,



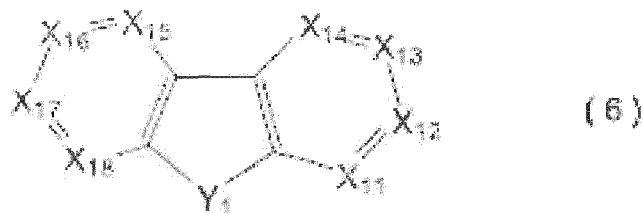
worin: HAr der untenstehenden Formel (6) entspricht;

a eine ganze Zahl von 1 bis 5 ist;

wenn a 1 ist, L_1 eine divalente Verbindungsgruppe ist;

wenn a im Bereich von 2 bis 5 ist, L_1 eine trivalente bis hexavalente Verbindungsgruppe ist und die HAr gleich oder verschieden voneinander sind;

die Verbindungsgruppe eine substituierte oder unsubstituierte Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen oder ein divalenter bis hexavalenter Rest ist, welcher aus einer der Gruppen erhalten wird, die sich durch miteinander Verbinden von zwei oder drei der substituierten oder unsubstituierten Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen bilden, und die miteinander verbundenen Gruppen gleich oder verschieden voneinander sind,



worin: Y_1 ein Sauerstoffatom oder ein Schwefelatom ist;

X_{11} und X_{18} ein Stickstoffatom oder CR_{13} sind;
 einer von X_{12} bis X_{17} ein Kohlenstoffatom ist, welches über eine Einfachbindung an L_1 gebunden ist, und der Rest von X_{12} bis X_{17} ein Stickstoffatom oder CR_{13} ist;
 R_{13} jeweils unabhängig ein Wasserstoffatom, ein Halogenatom, eine Cyanogruppe, eine substituierte oder unsubstituierte Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen, eine substituierte oder unsubstituierte Alkylgruppe mit 1 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Alkenylgruppe mit 2 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Alkynylgruppe mit 2 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Alkylsilylgruppe mit 3 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Arylsilylgruppe mit 6 bis 30 Ringkohlenstoffatomen, eine substituierte oder unsubstituierte Alkoxygruppe mit 1 bis 30 Kohlenstoffatomen, eine substituierte oder unsubstituierte Aralkylgruppe mit 6 bis 30 Ringkohlenstoffatomen oder eine substituierte oder unsubstituierte Aryloxygruppe mit 6 bis 30 Ringkohlenstoffatomen ist; eine Vielzahl von R_{13} gleich oder verschieden voneinander sind; benachbarte R_{13} sich optional miteinander verbinden, um einen Ring zu bilden; und
 in der oben aufgeführten Formel (4), Ar_1 und Ar_2 jeweils unabhängig voneinander der Formel (5) entsprechen oder eine substituierte oder unsubstituierte Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen sind, wenn mindestens einer von R_1 , L_1 , R_{13} , Ar_1 und Ar_2 substituiert ist, der Substituent von R_1 , L_1 , R_{13} , Ar_1 und Ar_2 jeweils unabhängig voneinander aus der Gruppe, bestehend aus einer Arylgruppe, einer geradkettigen oder verzweigten Alkylgruppe, einer Cycloalkylgruppe, einer halogenierten Alkylgruppe, einer Alkenylgruppe, einer Alkynylgruppe, einer Alkylsilylgruppe, einer Arylsilylgruppe, einer Alkoxygruppe, einer halogenierten Alkoxygruppe, einer Aralkylgruppe, einer Aryloxygruppe, einem Halogenatom, einer Cyanogruppe, einer Hydroxylgruppe, einer Nitrogruppe und einer Carboxygruppe, ausgewählt ist.

2. Das aromatische heterocyclische Derivat nach Anspruch 1, wobei, in der Formel (6), X_{13} oder X_{16} ein Kohlenstoffatom ist, welches über eine Einfachbindung an L_1 gebunden ist.
3. Das aromatische heterocyclische Derivat nach Anspruch 1 oder 2, wobei in der Formel (5), a eine ganze Zahl von 1 bis 3 ist.
4. Das aromatische heterocyclische Derivat nach Anspruch 1 oder 2, wobei, in der Formel (5) a eine ganze Zahl von 1 oder 2 ist.
5. Das aromatische heterocyclische Derivat nach Anspruch 1 oder 2, wobei, in der Formel (5) a 1 ist, L_1 eine Verbindungsgruppe ist, und die Verbindungsgruppe ein divalenter Rest aus einer substituierten oder unsubstituierten Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen ist.
6. Das aromatische heterocyclische Derivat nach Anspruch 1 oder 2, wobei, in der Formel (5) a 2 ist und L_1 eine Verbindungsgruppe ist, und die Verbindungsgruppe ein trivalenter Rest aus einer substituierten oder unsubstituierten Arylgruppe mit 6 bis 30 Ringkohlenstoffatomen ist.
7. Das aromatische heterocyclische Derivat nach einem der Ansprüche 1 bis 6, wobei, in der Formel (6) Y_1 ein Sauerstoffatom ist.
8. Das aromatische heterocyclische Derivat nach einem der Ansprüche 1 bis 7, wobei, in der Formel (6) Y_1 ein Sauerstoffatom ist, X_{11} und X_{18} CR_{13} sind, und einer von X_{12} bis X_{17} ein Kohlenstoffatom ist, welches über eine Einfachbindung an L_1 gebunden ist, und der Rest von X_{12} bis X_{17} CR_{13} ist.
9. Das aromatische heterocyclische Derivat nach einem der Ansprüche 1 bis 8, wobei, in der Formel (4) zwei oder drei von X_1 bis X_3 Stickstoffatome sind.
10. Das aromatische heterocyclische Derivat nach einem der Ansprüche 1 bis 9, wobei, in der Formel (5) L_1 ein divalenter oder trivalenter Rest, der sich aus einem von Benzol, Biphenyl, Terphenyl, Naphthalin und Phenanthren ergibt.
11. Material für eine organische Elektrolumineszenzvorrichtung, welches das aromatische heterocyclische Derivat nach einem der Ansprüche 1 bis 10 umfasst.

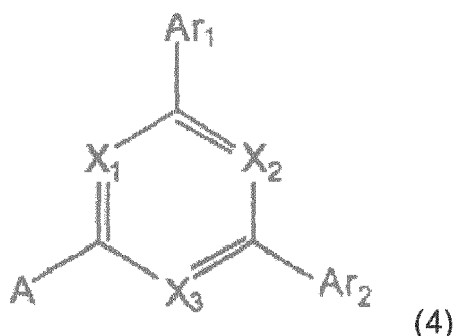
Revendications

1. Dérivé hétérocyclique aromatique représenté par la formule (4) ci-dessous,

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dans laquelle : X_1 à X_3 représentent un atome d'azote ou CR_1 , à condition qu'au moins l'un de X_1 à X_3 représente un atome d'azote ;

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R_1 représente indépendamment un atome d'hydrogène, un atome d'halogène, un groupe cyano, un groupe aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, un groupe alkyle substitué ou non substitué ayant 1 à 30 atome(s) de carbone, un groupe alcényle substitué ou non substitué ayant 2 à 30 atomes de carbone, un groupe alcynyle substitué ou non substitué ayant 2 à 30 atomes de carbone, un groupe alkylsilyle substitué ou non substitué ayant 3 à 30 atomes de carbone, un groupe arylsilyle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, un groupe alcoxy substitué ou non substitué ayant 1 à 30 atome(s) de carbone, un groupe aralkyle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, ou un groupe aryloxy substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle ;

A est représenté par la formule (5) ci-dessous,

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dans laquelle : HAr est représenté par la formule (6) ci-dessous ;

a est un nombre entier allant de 1 à 5 ;

lorsque a vaut 1, L_1 représente un groupe de liaison divalent ;

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lorsque a se trouve dans une plage allant de 2 à 5, L_1 représente un groupe de liaison trivalent à hexavalent et HAr est identique ou différent ;

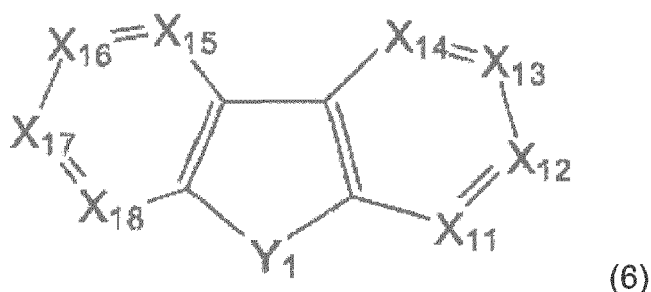
le groupe de liaison représente un groupe aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, ou un résidu divalent à hexavalent engendré à partir de l'un quelconque d'un groupe formé par liaison mutuelle de deux ou trois du groupe aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle ; et

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les groupes liés mutuellement sont identiques ou différents,

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dans laquelle : Y_1 représente un atome d'oxygène ou un atome de soufre ;

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X_{11} et X_{18} représentent un atome d'azote ou CR_{13} ;

l'un de X_{12} à X_{17} représente un atome de carbone lié à L_1 par une liaison simple et le reste de X_{12} à X_{17} représentent un atome d'azote ou CR_{13} ;

R_{13} représente indépendamment un atome d'hydrogène, un atome d'halogène, un groupe cyano, un groupe

- aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, un groupe alkyle substitué ou non substitué ayant 1 à 30 atome(s) de carbone, un groupe alcényle substitué ou non substitué ayant 2 à 30 atomes de carbone, un groupe alcynyle substitué ou non substitué ayant 2 à 30 atomes de carbone, un groupe alkylsilyle substitué ou non substitué ayant 3 à 30 atomes de carbone, un groupe arylsilyle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, un groupe alcoxy substitué ou non substitué ayant 1 à 30 atome(s) de carbone, un groupe aralkyle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, ou un groupe aryloxy substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle ; une pluralité de R_{13} sont mutuellement identiques ou différents ; des groupes R_{13} adjacents se lient facultativement les uns aux autres pour former un cycle ; et
- dans la formule (4) ci-dessus, Ar_1 et Ar_2 sont chacun indépendamment représentés par la formule (5), ou représentent un groupe aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle, lorsque au moins l'un de R_1 , L_1 , R_{13} , Ar_1 et Ar_2 est substitué, le substituant de R_1 , L_1 , R_{13} , Ar_1 et Ar_2 étant chacun indépendamment choisi dans le groupe constitué d'un groupe aryle, d'un groupe alkyle linéaire ou ramifié, d'un groupe cycloalkyle, d'un groupe alkyle halogéné, d'un groupe alcényle, d'un groupe alcynyle, d'un groupe alkylsilyle, d'un groupe arylsilyle, d'un groupe alcoxy, d'un groupe alcoxy halogéné, d'un groupe aralkyle, d'un groupe aryloxy, d'un atome d'halogène, d'un groupe cyano, d'un groupe hydroxyle, d'un groupe nitro et d'un groupe carboxy.
2. Dérivé hétérocyclique aromatique selon la revendication 1, dans lequel dans la formule (6), X_{13} ou X_{16} représente un atome de carbone lié à L_1 par une liaison simple.
 3. Dérivé hétérocyclique aromatique selon la revendication 1 ou 2, dans lequel dans la formule (5), a est un nombre entier allant de 1 à 3.
 4. Dérivé hétérocyclique aromatique selon la revendication 1 ou 2, dans lequel dans la formule (5), a vaut 1 ou 2.
 5. Dérivé hétérocyclique aromatique selon la revendication 1 ou 2, dans lequel, dans la formule (5), a vaut 1, L_1 représente un groupe de liaison, et le groupe de liaison est un résidu divalent d'un groupe aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle.
 6. Dérivé hétérocyclique aromatique selon la revendication 1 ou 2, dans lequel dans la formule (5), a vaut 2 et L_1 représente un groupe de liaison, et le groupe de liaison est un résidu trivalent d'un groupe aryle substitué ou non substitué ayant 6 à 30 atomes de carbone dans le cycle.
 7. Dérivé hétérocyclique aromatique selon l'une quelconque des revendications 1 à 6, dans lequel dans la formule (6), Y_1 représente un atome d'oxygène.
 8. Dérivé hétérocyclique aromatique selon l'une quelconque des revendications 1 à 7, dans lequel dans la formule (6), Y_1 représente un atome d'oxygène, X_{11} et X_{18} représentent CR_{13} , et l'un de X_{12} à X_{17} représente un atome de carbone lié à L_1 par une liaison simple et le reste de X_{12} à X_{17} représentent CR_{13} .
 9. Dérivé hétérocyclique aromatique selon l'une quelconque des revendications 1 à 8, dans lequel dans la formule (4), deux ou trois de X_1 à X_3 représentent un atome d'azote.
 10. Dérivé hétérocyclique aromatique selon l'une quelconque des revendications 1 à 9, dans lequel dans la formule (5), L_1 est un résidu divalent ou trivalent engendré à partir de l'un quelconque parmi le benzène, le biphényle, le terphényle, le naphthalène et le phénanthrène.
 11. Matériau pour dispositif électroluminescent organique comprenant le dérivé hétérocyclique aromatique selon l'une quelconque des revendications 1 à 10.

FIG. 1

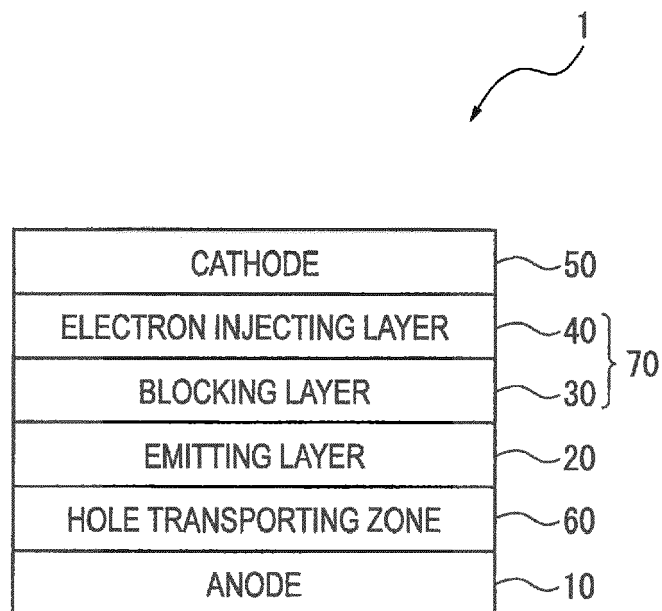


FIG. 2

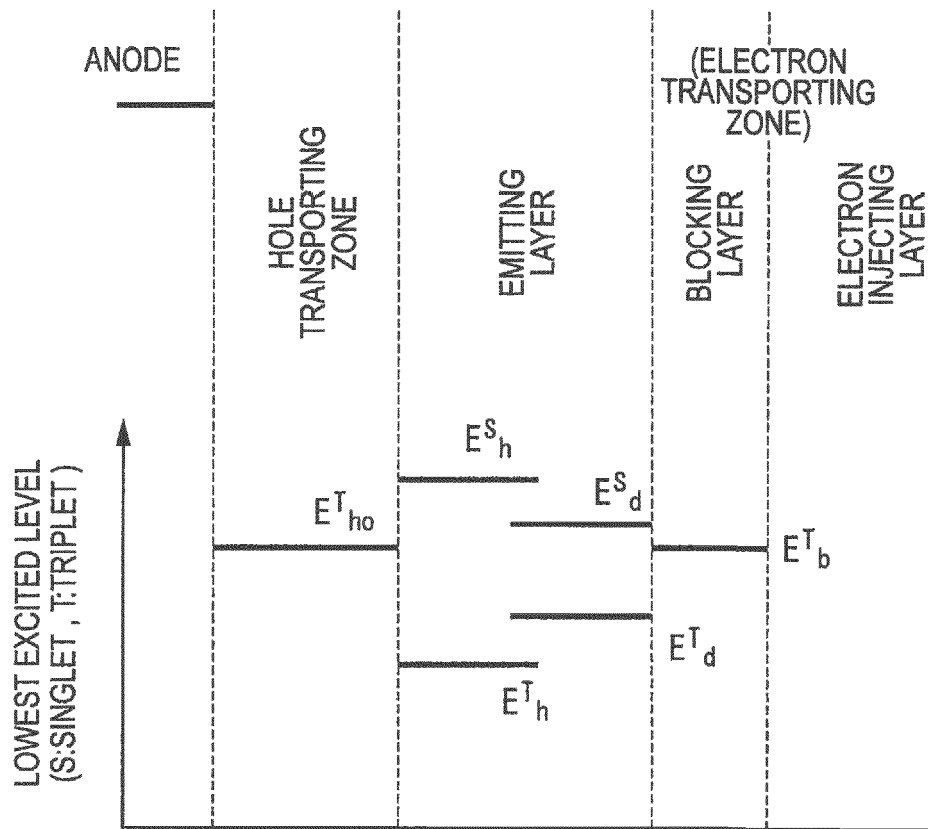


FIG. 3

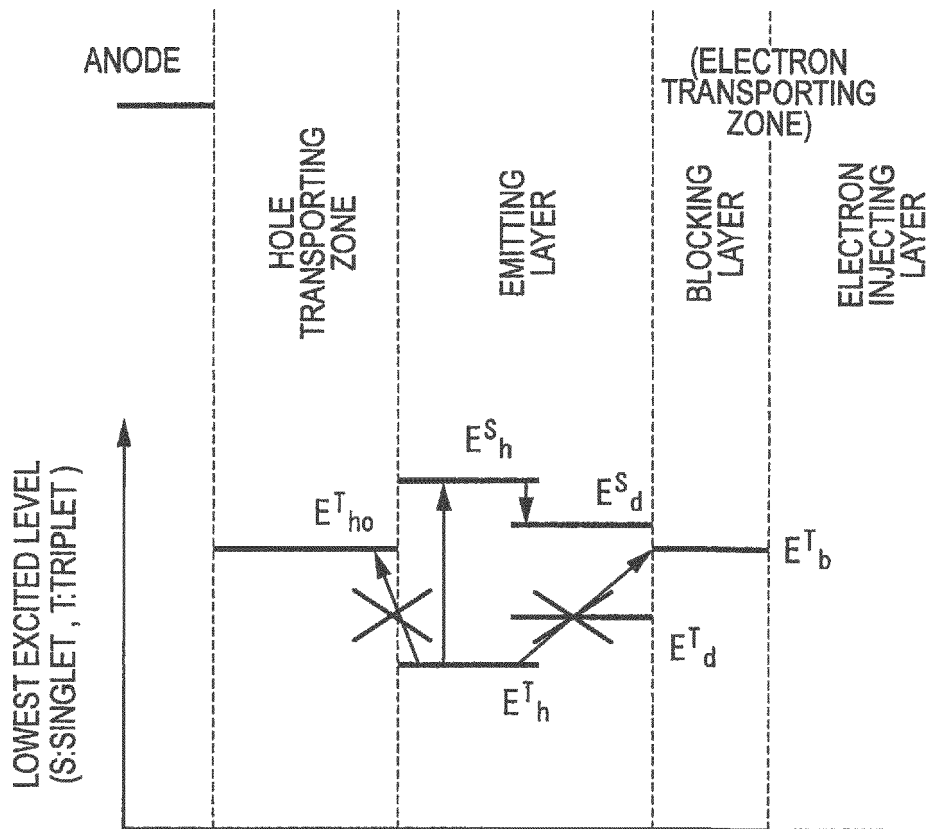


FIG. 4

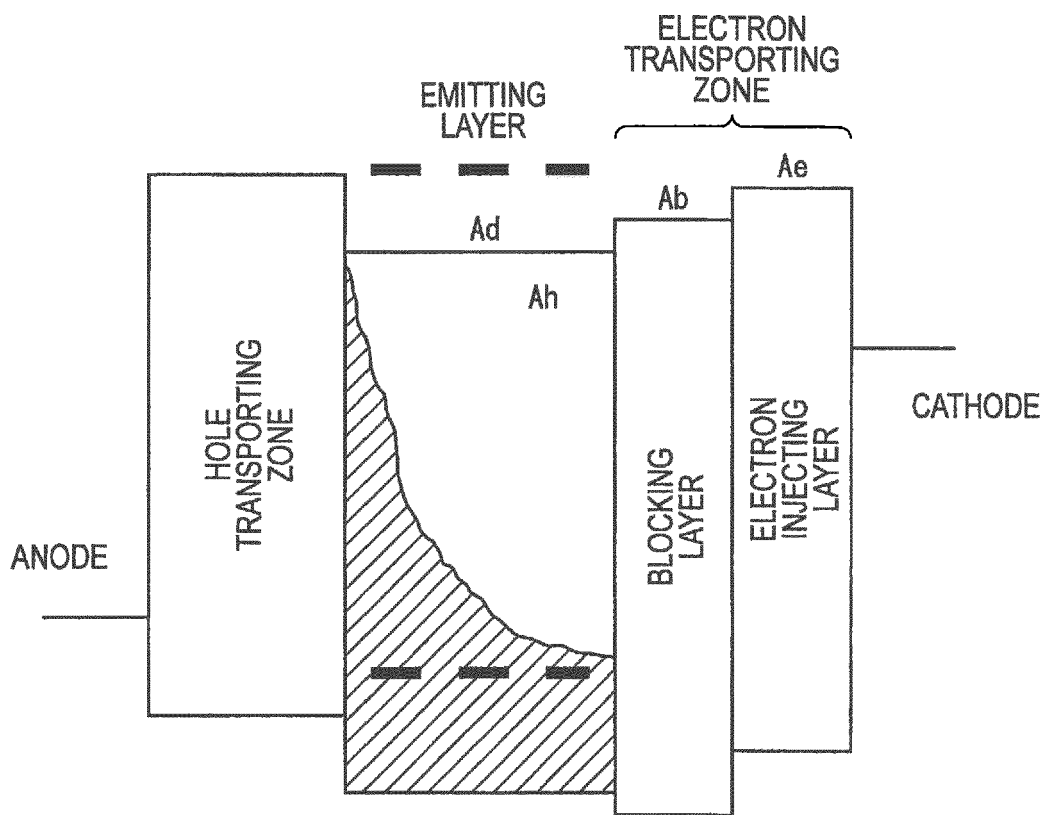


FIG. 5

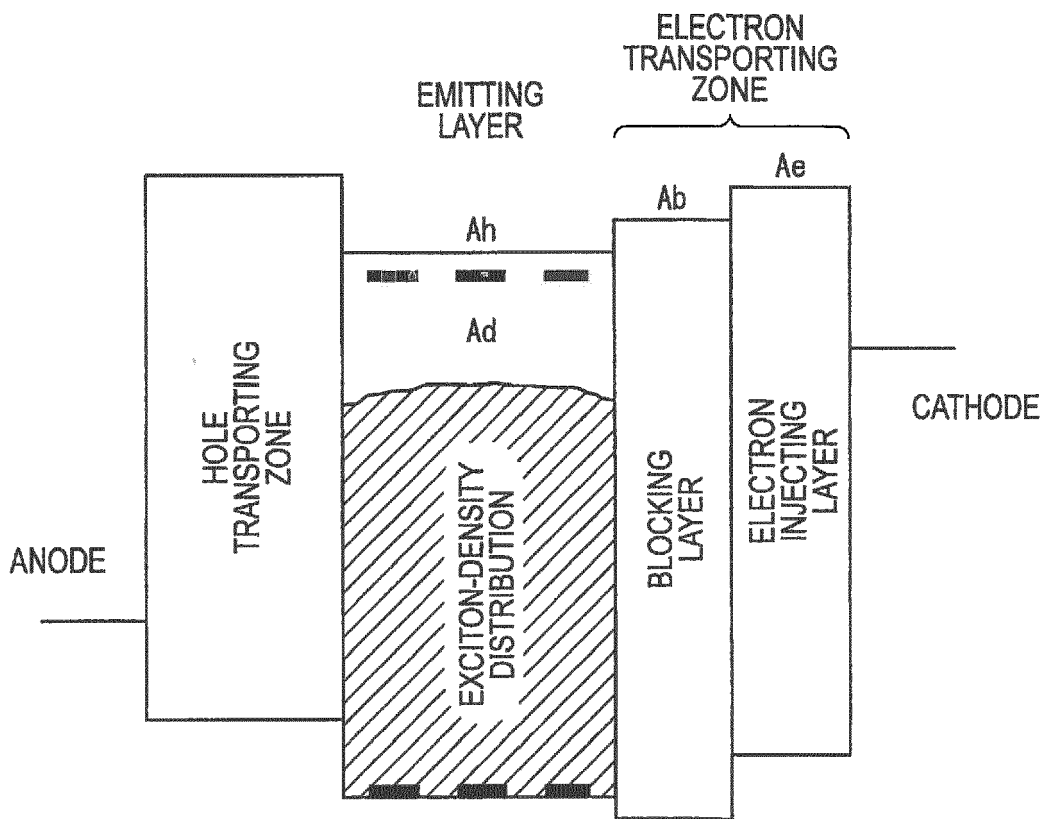


FIG. 6

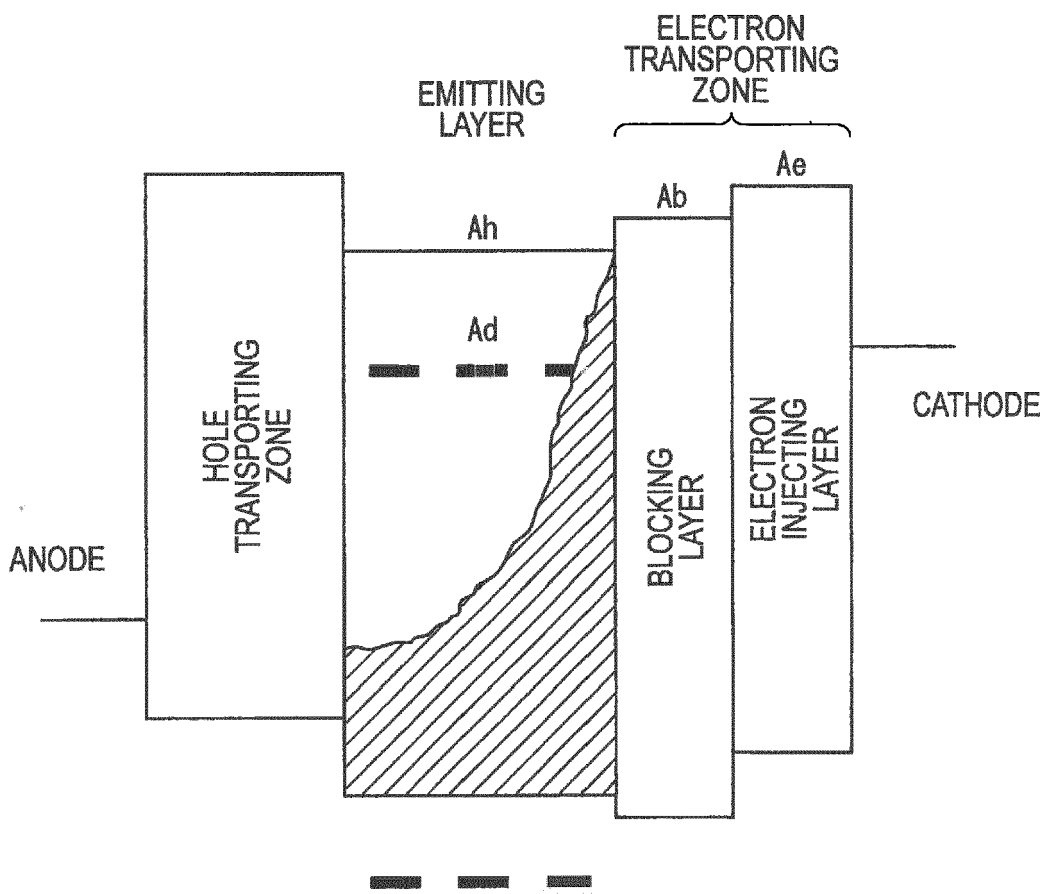


FIG. 7

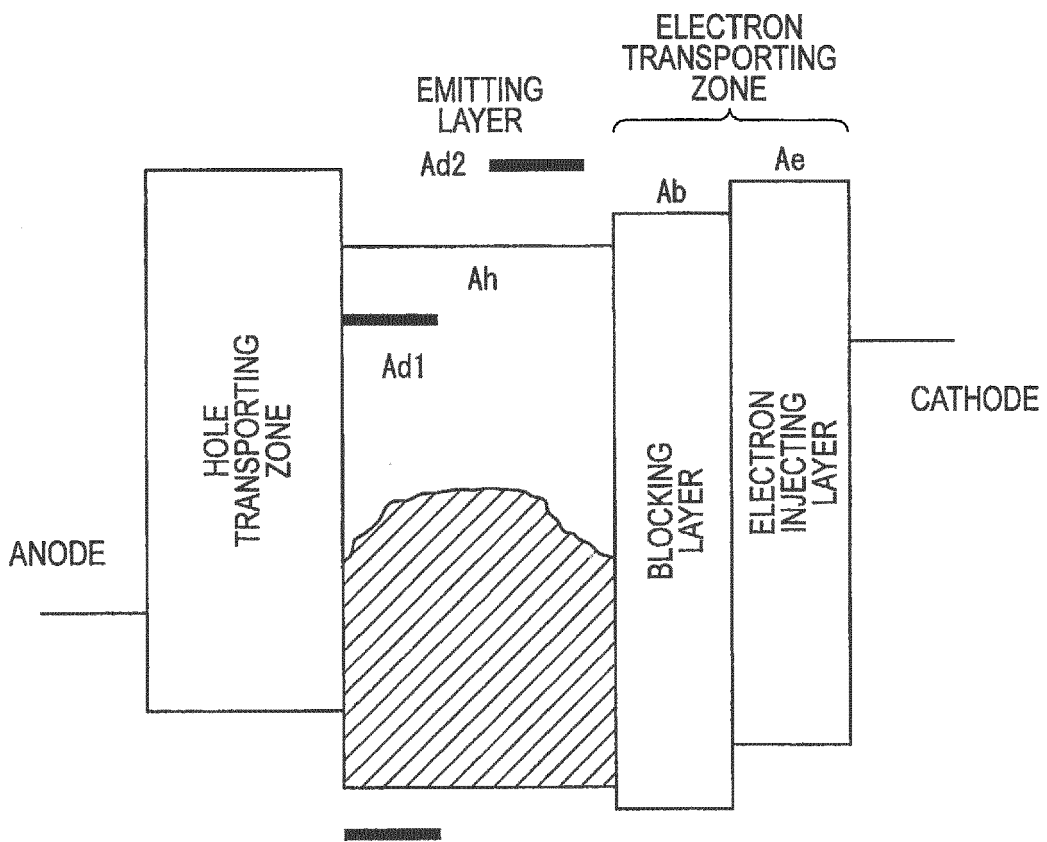


FIG. 8

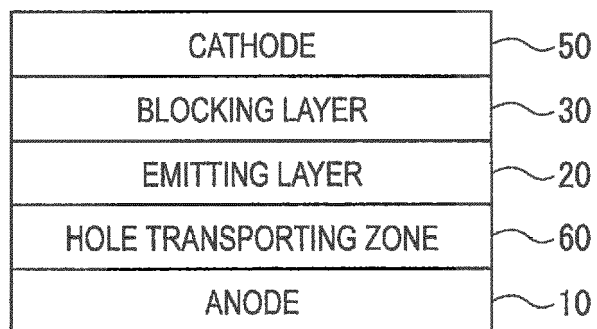


FIG. 9

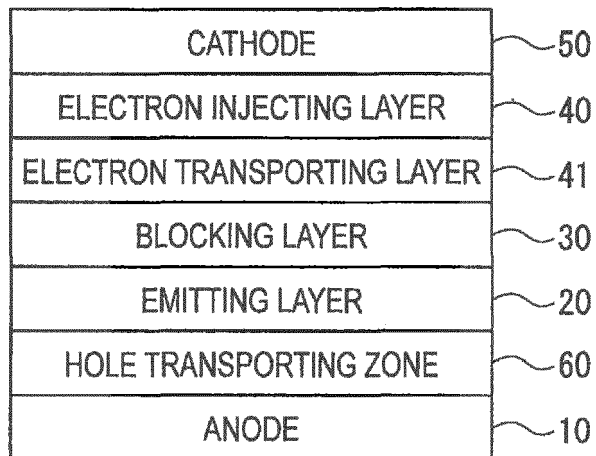


FIG. 10

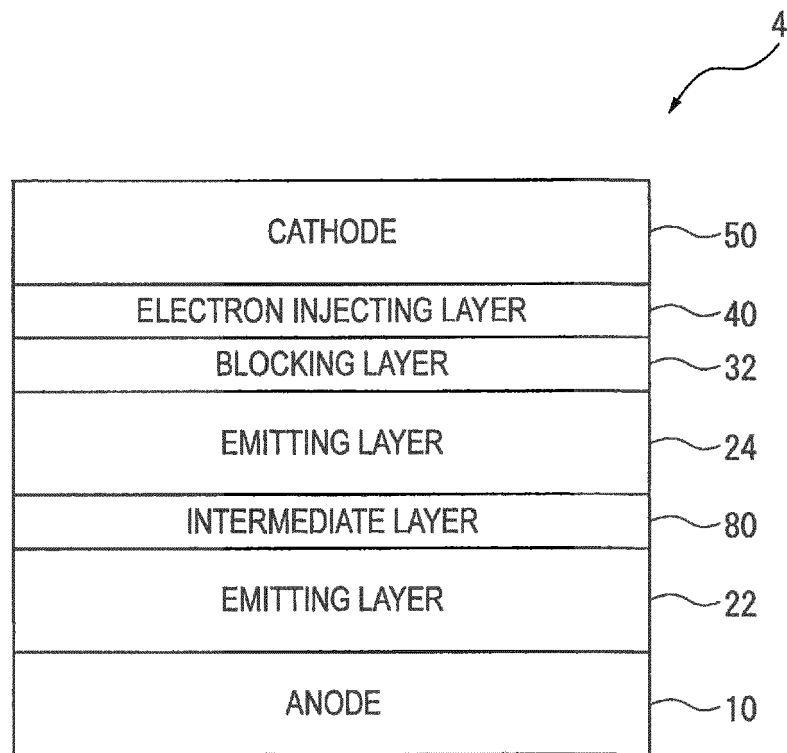


FIG. 11

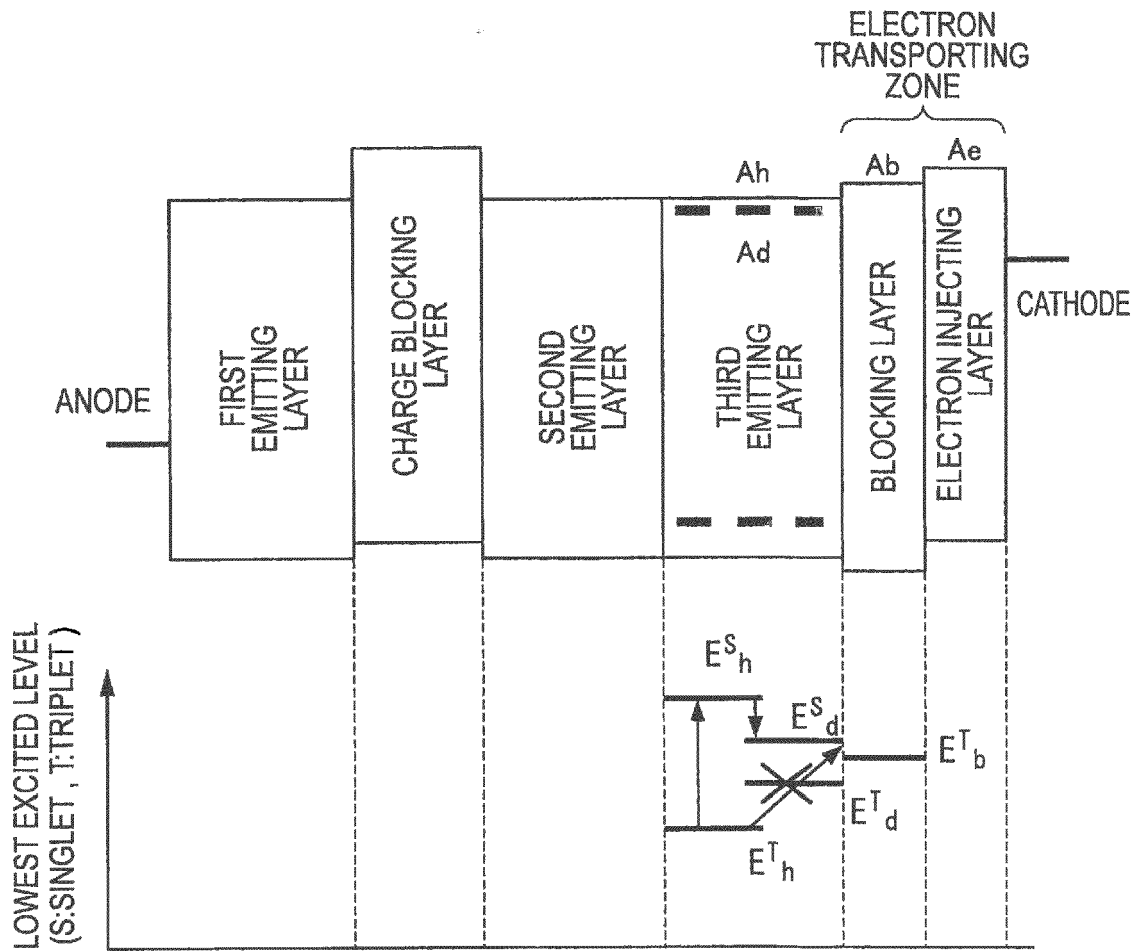


FIG. 12

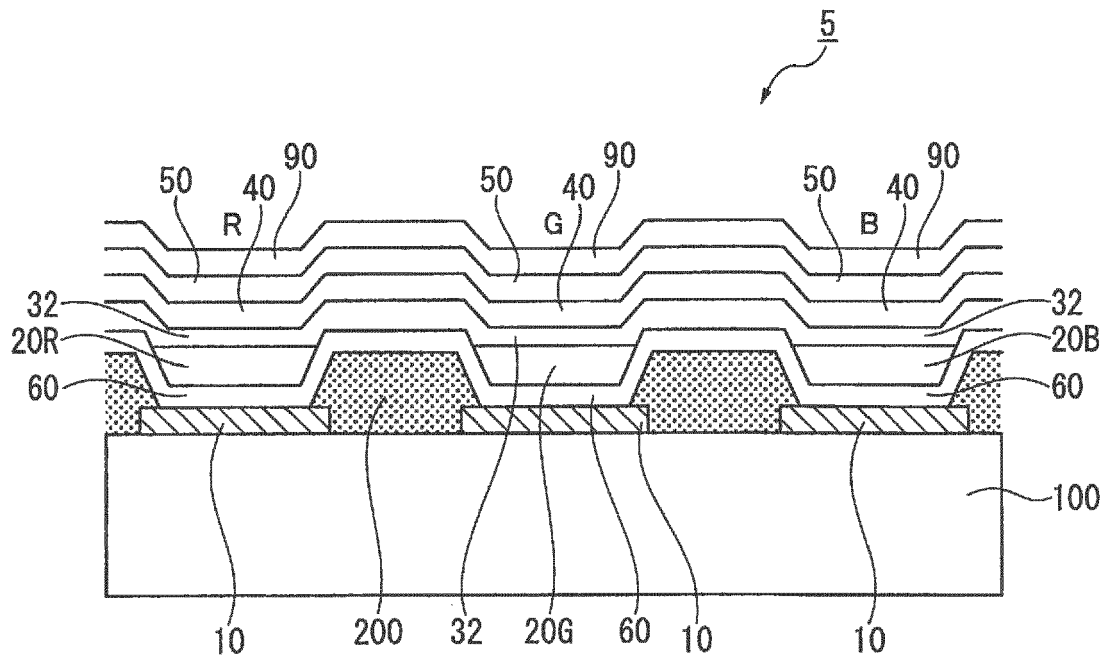


FIG. 13

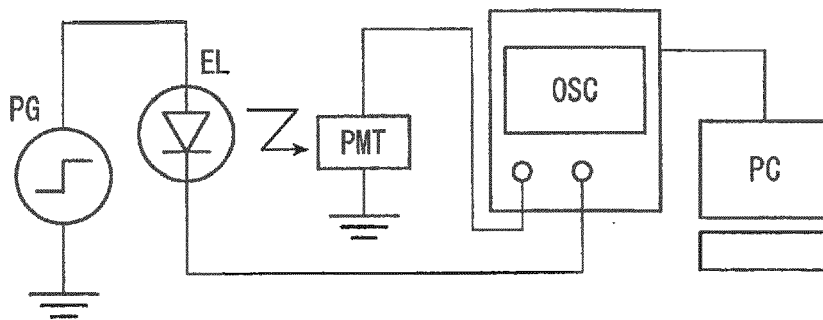


FIG. 14A

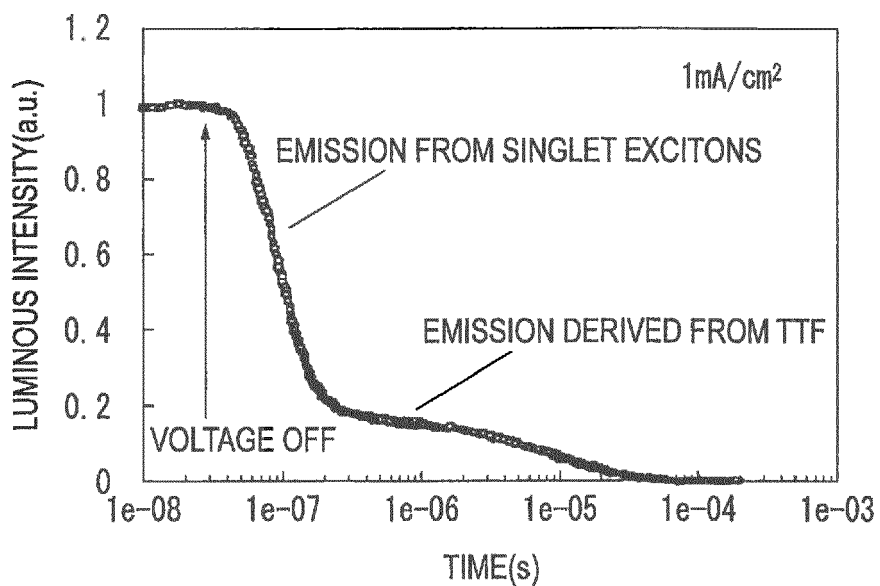
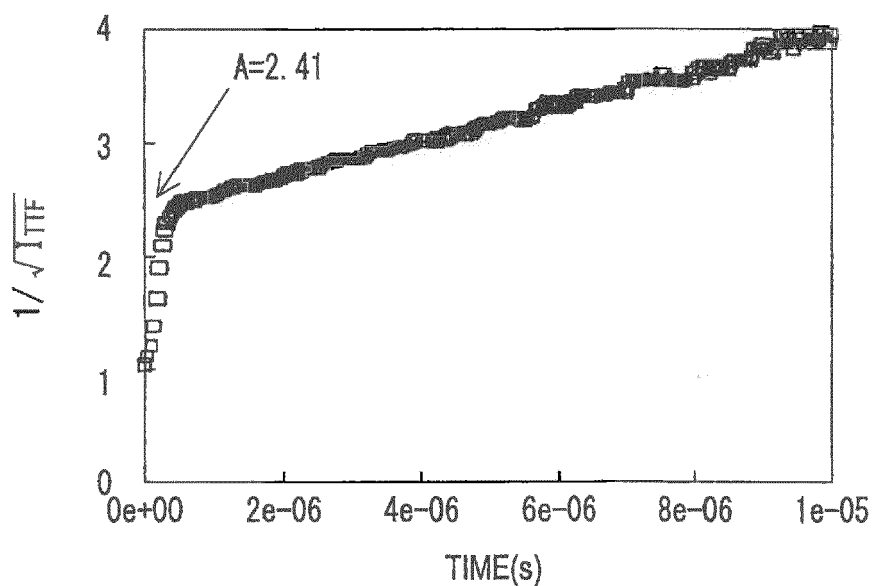


FIG. 14B



REFERENCES CITED IN THE DESCRIPTION

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