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# (11) **EP 2 770 086 A3**

EUROPEAN PATENT APPLICATION

(88) Date of publication A3: (51) Int Cl.: C25C 3/28 (2006.01) C22B 34/12 (2006.01) 29.10.2014 Bulletin 2014/44 (43) Date of publication A2: 27.08.2014 Bulletin 2014/35 (21) Application number: 14163470.9 (22) Date of filing: 13.03.2003 (84) Designated Contracting States: (72) Inventors: AT BE BG CH CY CZ DE DK EE ES FI FR GB GR Stretzov, Les HU IE IT LI LU MC NL PT RO SE SI SK TR Adamstown, New South Wales 2289 (AU) Ratchev, Ivan (30) Priority: 13.03.2002 AU PS107102 Georgetown, New South Wales 2298 (AU) 19.06.2002 AU PS304902 Osborn, Steve Valentine, New South Wales 2280 (AU) (62) Document number(s) of the earlier application(s) in accordance with Art. 76 EPC: (74) Representative: Goodman, Simon John Nye 03743767.0 / 1 492 905 Reddie & Grose LLP **16 Theobalds Road** (71) Applicant: Metalysis Limited London Wath Upon Dearne WC1X 8PL (GB) Rotherham S63 5DB (GB)

#### (54) Reduction of metal oxides in an electrolytic cell

(57) A method of reducing a metal oxide, such as titania in a solid state in an electrolytic cell is disclosed. The electrolytic cell includes an anode, a cathode, and a molten electrolyte. The electrolyte includes cations of a metal that is capable of chemically reducing the metal oxide. The metal oxide in a solid state is immersed in the

electrolyte. The method includes a step of operating the cell at a potential that is above a potential at which cations of the metal that is capable of chemically reducing the metal oxide deposit as the metal on the cathode, whereby the metal chemically reduces the metal oxide.



Fig. 1 Schematic of the experimental set up

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