

(19)



(11)

**EP 3 415 590 A1**

(12)

**EUROPEAN PATENT APPLICATION**

(43) Date of publication:

**19.12.2018 Bulletin 2018/51**

(51) Int Cl.:

**C11D 1/37** (2006.01)

**C11D 1/29** (2006.01)

**C11D 3/37** (2006.01)

**C11D 3/22** (2006.01)

**C11D 17/00** (2006.01)

(21) Application number: **18166903.7**

(22) Date of filing: **11.04.2018**

(84) Designated Contracting States:

**AL AT BE BG CH CY CZ DE DK EE ES FI FR GB  
GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO  
PL PT RO RS SE SI SK SM TR**

Designated Extension States:

**BA ME**

Designated Validation States:

**KH MA MD TN**

• **RAVIDAT, Nans Elian**

**1853 Strombeek-Bever (BE)**

• **RICHMOND, Annalise Charlotte**

**1853 Strombeek-Bever (BE)**

• **KEULEERS, Robby Renilde François**

**1853 Strombeek-Bever (BE)**

(30) Priority: **13.06.2017 EP 17175827**

(71) Applicant: **The Procter & Gamble Company  
Cincinnati, OH 45202 (US)**

(72) Inventors:

• **BONNET, Guillaume**

**1853 Strombeek-Bever (BE)**

(74) Representative: **Pickford, James Lawrence et al  
Procter & Gamble**

**Technical Centres Limited**

**Whitley Road**

**Longbenton**

**Newcastle upon Tyne NE12 9TS (GB)**

(54) **CONSUMER PRODUCT**

(57) A consumer product composed of a laundry treatment composition housed in a container, wherein the container has a spray applicator and an internal chamber housing the laundry treatment composition and the laundry treatment composition is in contact with the

spray applicator and the spray applicator is capable of spraying the laundry treatment composition into the environment external to the container upon activation of the spray applicator.

**EP 3 415 590 A1**

**Description**

## FIELD OF THE INVENTION

5 **[0001]** The present invention relates to a consumer product comprising a container, spray applicator and a liquid laundry treatment composition.

## BACKGROUND OF THE INVENTION

10 **[0002]** Laundry wash operations and the use of laundry treatment compositions are well known.

**[0003]** However, for some wash operations, such as delicate garments, lightly soiled garments, or garments simply in need of a 'refresh' (maybe after being hung in a wardrobe for a period of time), current wash practices can be wasteful in terms of resource and environmentally unfriendly. Furthermore, consumers typically would prefer to wash these delicate items in isolation or with a very few items solely, this in order to further reduce the risk of damage due to inter-garment friction. In other words, the volume of wash composition, full washing loads and full wash cycles are not needed nor desired to achieve the desired benefit. In addition, with respect to the garments themselves, especially delicate garments, current wash process can be quite harsh on said garments.

**[0004]** For such applications it was found that low active or 'mild' spray formulations can overcome such issues. Items to be washed are sprayed with the spray formulation and then added into the wash process, for example into an automatic washing machine, without the need for any further detergent to be added. The spray application spreads the detergent composition over the fabric minimizing as such local product excess hence risk of potential local damage, and reduces the overall amount of treatment composition to be used and hence leads to reduced amount of water required in a rinse operation hence rinsing time requirement. As such short or delicate cycles or even just rinse cycles might already fulfill the need. The low active formulation provides a 'less harsh' treatment of the fabrics whilst still providing sufficient cleaning and/or refreshing of garments.

**[0005]** However, an issue encountered with such formulations is that as the fabric treatment composition is sprayed directly on delicate clothes, local accumulation of chemistry on the fabrics may occur. Furthermore, if the spraying pattern is too concentric, this may motivate the consumer to spray more than needed to get the entire fabric surface area covered with the composition. This would again unnecessarily increase the amount of chemistry through the wash, rendering the overall treatment 'more harsh' again, and might eventually require more rinsing water and rinsing time again.

**[0006]** Therefore, there is a need to provide a consumer product comprising a spray applicator and a low active laundry treatment composition wherein the composition can be sprayed as a broad fine dispersion of droplets onto fabrics, preferably without impacting performance of the composition.

It was surprisingly found that the careful balance of non-soap anionic surfactant to nonionic surfactant in the laundry treatment composition provided for such broader spray patterns whilst still providing a laundry treatment composition capable of being sprayed and providing laundry treatment benefits.

## SUMMARY OF THE INVENTION

40 **[0007]** A first aspect is a consumer product comprising a liquid laundry treatment composition housed in a container, wherein;

a. the container comprises a spray applicator and an internal chamber comprising the laundry treatment composition and the liquid laundry treatment composition is in contact with the spray applicator and the spray applicator is capable of spraying the liquid laundry treatment composition into the environment external to the container upon activation of the spray applicator;

b. the liquid laundry treatment composition comprises at least 50%, preferably at least 60%, more preferably at least 70%, most preferably at least 80% by weight of the liquid laundry detergent composition of water; and between 1% and 15% preferably between 4% and 12%, more preferably between 5% and 10% by weight of the liquid laundry treatment composition of non-soap surfactant,

wherein the non-soap surfactant comprises non-soap anionic surfactant and the non-soap surfactant comprises between 90% and 100%, preferably between 95% and 100%, most preferably about 100% by weight of the non-soap surfactant of the non-soap anionic surfactant.

55 **[0008]** A second aspect is a process of treating at least one garment comprising the steps of;

a. Using the consumer product according to the present invention to spray at least one garment with the laundry treatment composition;

- b. Optionally adding the at least one garment from step a to the drum of an automatic washing machine;  
 c. Optionally washing the at least one garment in the automatic washing machine wherein the drum comprises a wash liquor, wherein the wash liquor comprises water and the one or more garments and wherein the only laundry treatment composition added to the drum of the automatic washing machine is that used in step a.

## DETAILED DESCRIPTION OF THE INVENTION

### Consumer Product

**[0009]** The present invention discloses a consumer product comprising a liquid laundry treatment composition housed in a container.

**[0010]** By 'consumer product' we herein mean a product intended for end use by a consumer. The consumer product may be purchased as is, or may comprise at the point of purchase external packaging that is removed prior to use of the consumer product.

**[0011]** The container is described in more detail below.

**[0012]** The liquid laundry treatment composition is described in more detail below.

### The container

**[0013]** The container comprises a spray applicator and an internal chamber comprising the laundry treatment composition. The liquid laundry treatment composition is in contact with the spray applicator and the spray applicator is capable of spraying the liquid laundry treatment composition into the environment external to the container upon activation of the spray applicator.

**[0014]** The container may be of any suitable shape or size. Those skilled in the art will be aware of suitable shapes and sizes for the container. Preferably the container is of a size that an adult can comfortably hold said container in their hand. Alternatively the container may comprise a handle to allow the user to hold said container in their hand.

**[0015]** The container may be made from any suitable material. The container may be manufactured from a material selected from plastic, metal, glass or a mixture thereof. The container may be made from a plastic material, preferably a polyolefin material. The container may be made from polypropylene, polystyrene, polyethylene, polyethylene terephthalate, PVC or a mixture thereof or more durable engineering plastics like Acrylonitrile Butadiene Styrene (ABS), Polycarbonates, Polyamides and the like. The material used to make the container may comprise other ingredients, such as colorants, preservatives, plasticisers, UV stabilizers, oxygen, perfume and moisture barriers, recycled materials and the like.

**[0016]** The container may be made using any suitable process. Suitable processes include but are not limited to thermoforming, injection molding, injection stretch blow molding, extrusion blow molding, tube forming from a flat laminate with a welding step, extruded tube forming.

**[0017]** The container may be opaque, transparent or translucent. Preferably, the container is opaque.

**[0018]** The container may be a pressurised container or a non-pressurised container.

**[0019]** The spray applicator may be any suitable spray applicator. Those skilled in the art will be aware of suitable spray applicators.

**[0020]** Suitable spray dispensers include hand pump (sometimes referred to as "trigger") devices, pressurized can devices, electrostatic spray devices, etc. Preferably the spray dispenser is nonsolvent propellant pressurized and the spray means are of the trigger dispensing type. The spray dispenser can be a pre-compression sprayer or an aerosol spray with a pressure control valve, both commercially available in the art. Suitable pre-compression sprayers in which a buffer mechanism to control the maximum pressure can be added include the Flairosol® spray dispenser, commercially available, manufactured and sold by Afa Dispensing Group (The Netherlands).

**[0021]** Preferably, upon activation of the spray applicator the laundry treatment composition is sprayed into the environment external to the container and is present as droplets having an average particle size (Dv 50%) between 100 micron and 300 micron, preferably between 100 micron and 200 micron, most preferably between 100 and 150 micron.

### Liquid laundry treatment composition

**[0022]** The term 'liquid laundry treatment composition' refers to any laundry treatment composition that is a liquid and is capable of wetting and treating fabric e.g., cleaning clothing in a domestic washing machine. The liquid composition can include solids or gases in suitably subdivided form, but the liquid composition excludes forms which are non-fluid overall, such as tablets or granules.

**[0023]** The laundry treatment composition may be a laundry detergent composition, a laundry softening composition, a laundry care composition, a laundry scent refresher composition or a mixture thereof. Preferably the laundry treatment

composition has both cleaning, refreshing and care properties, the latter including fabric softness and fabric shape and texture retention properties.

**[0024]** The liquid laundry treatment composition comprises at least 50%, preferably at least 60%, more preferably at least 70%, most preferably at least 80% by weight of the liquid laundry detergent composition of water. Such water levels are preferred to enable a 'less harsh' composition and also allow for easy spray application.

**[0025]** The liquid laundry treatment composition comprises between 1% and 15% preferably between 4% and 12%, more preferably between 5% and 10% by weight of the liquid laundry treatment composition of non-soap surfactant.

**[0026]** The non-soap surfactant is described in more detail below.

**[0027]** Preferably, the liquid laundry treatment composition comprises between 0.1% and 5%, preferably from 0.2% to 1%, more preferably from 0.3% to 0.7% by weight of the liquid laundry treatment composition of a polymer selected from a cationic polymer, a polysaccharide polymer, or a mixture thereof, preferably a cationic polymer. The cationic polysaccharide technology will further provide fabric conditioning benefits including softness and multi-cycle shape retention benefits.

**[0028]** Preferably, the cationic polymer is a cationically modified polysaccharide, preferably selected from cationic guar gums, cationic cellulosic polymers, and mixtures thereof, most preferably cationic cellulosic polymers even more preferably cationically modified hydroxyethyl cellulose, most preferably, hydroxyethyl cellulose derivatised with trimethyl ammonium substituted epoxide.

**[0029]** By "hydrophobically modified" we herein mean that one or more hydrophobic groups are bound to the polymer. By "cationically modified" we herein mean that one or more cationically charged groups are bound to the polymer.

**[0030]** The cationically modified hydroxyethyl cellulose preferably is hydroxyethyl cellulose derivatised with trimethyl ammonium substituted epoxide.

**[0031]** The cationic polysaccharide polymer can be synthesized in, and are commercially available in, a number of different molecular weights. In order to achieve optimal softening and care performance from the product, it is desirable that the cationic polymer used in this invention be of an appropriate molecular weight. Without wishing to be bound by theory, it is believed that polymers that are too high in mass can entrap soils and prevent them from being removed, as well as will be providing physical stability challenges especially in low viscous liquors required for spraying. The use of cationic polymers with an average molecular weight of less than about 850,000 daltons, and especially those with an average molecular weight of less than 500,000 daltons can help to minimise this effect without significantly reducing the softening performance of properly formulated products while providing the desired optimum spray pattern. On the other hand, polymers with a molecular weight of about 10,000 daltons or less are believed to be too small to give an effective softening benefit. Therefore, the cationic polymer according to the invention preferably has a molecular weight of from about 10,000 daltons to about 850,000 daltons, preferably from about 50,000 daltons to about 750,000 daltons, more preferably from about 100,000 daltons to about 600,000 daltons, most preferably from about 200,000 daltons to about 500,000 daltons.

**[0032]** The cationic polymers according to the invention may also have a cationic charge density ranging from about 0.1meq/g to about 5meq/g, preferably from about 0.15meq/g to about 4 meq/g, more preferably from about 0.2meq/g to about 2.5 meq/g, even more preferably from about 0.25meq/g to about 1.5 meq/g, most preferably from about 0.25 meq/g to about 0.7 meq/g, at the pH of intended use of the laundry composition. As used herein the "charge density" of the cationic polymers is defined as the number of cationic sites per polymer gram atomic weight (molecular weight), and can be expressed in terms of meq/gram of cationic charge. In general, adjustments of the proportions of amine or quaternary ammonium moieties in the polymer in function of the pH of the liquid laundry formulation in the case of amines, will affect the charge density. Without intending to be bound by theory, cationic polymers with a too high charge density are thought to be too sensitive to precipitate out with anionic compounds in the formulation, while cationic polymers with a too low charge density are thought to have a too low affinity to fabrics, compromising softness accordingly. Any anionic counterions can be used in association with cationic polymers. Non-limiting examples of such counterions include halides (e.g. chlorine, fluorine, bromine, iodine), sulphate and methylsulfate, preferably halides, more preferably chlorine.

**[0033]** The cationic polymer according to the invention might be "hydrophobically modified". We herein mean that one or more hydrophobic groups are bound to the polymer. Without intending to be bound by theory we believe that hydrophobic modification can increase the affinity of the polymer towards the fabric. Without intending to be limiting, the one or more hydrophobic groups can be independently selected from C<sub>1</sub>-C<sub>32</sub> preferably C<sub>5</sub>-C<sub>32</sub> alkyl; C<sub>1</sub>-C<sub>32</sub> preferably C<sub>5</sub>-C<sub>32</sub> substituted alkyl, C<sub>5</sub>-C<sub>32</sub> alkylaryl, or C<sub>5</sub>-C<sub>32</sub> substituted alkylaryl, (poly)alkoxy C<sub>1</sub>-C<sub>32</sub> preferably C<sub>5</sub>-C<sub>32</sub> alkyl or (poly)alkoxy substituted C<sub>1</sub>-C<sub>32</sub> preferably C<sub>5</sub>-C<sub>32</sub> alkyl or mixtures thereof. Hydrophobic substitution on the polymer, preferably on the anhydroglucose rings of the cationic polymer may range from 0.01% to 5% per glucose unit, more preferably from 0.05% to 2% per glucose unit, of the polymeric material.

**[0034]** The cationic polysaccharide polymers according to the invention include those which are commercially available and further include materials which can be prepared by conventional chemical modification of commercially available materials. Commercially available cationic cellulose polymers according to the invention include those with the INCI name Polyquaternium 10, such as those sold under the trade names: Ucare Polymer JR 30M, JR 400, JR 125, LR 400

and LK 400 polymers; Polyquaternium 67 such as those sold under the trade name Softcat SK™, all of which are marketed by Amerchol Corporation, Edgewater NJ; and Polyquaternium 4 such as those sold under the trade name: Celquat H200 and Celquat L-200, available from National Starch and Chemical Company, Bridgewater, NJ. Other suitable polysaccharides include hydroxyethyl cellulose or hydroxypropylcellulose quaternized with glycidyl C<sub>12</sub>-C<sub>22</sub> alkyl dimethyl ammonium chloride. Examples of such polysaccharides include the polymers with the INCI names Polyquaternium 24 such as those sold under the trade name Quaternium LM 200 by Amerchol Corporation, Edgewater NJ.

**[0035]** Alternatively, synthetic derived cationic polymers can also be used within the scope of the application.

**[0036]** The liquid laundry treatment composition may comprise less than 10%, preferably less than 8%, more preferably less than 5%, even more preferably less than 3%, most preferably between 0.5% and 2% by weight of the laundry treatment composition of fatty acid, neutralised fatty acid soap or a mixture thereof. Without wishing to be bound by theory, a low level of fatty acid will enable faster wash suds to collapse during washing and upon rinsing, as such reducing risk of leaving detergent residues behind while maximizing active cleaning surfactant presence during the wash. A too high level of fatty acid though risks leaving calcium fatty acids deposits behind in presence of hard water and as such is not preferred.

**[0037]** It will be highly preferred to add cleaning performance boosting technologies into the consumer product of the invention, enabling the formulator to keep overall surfactant levels low, facilitating suds control hence residue risk accordingly. A range of possible performance boosting technologies are described below.

**[0038]** The laundry treatment composition may comprise less than 3%, preferably less than 2%, more preferably less than 1%, even more preferably between 0.01% and 0.5%, most preferably between 0.05% and 0.3% by weight of the laundry treatment composition of a chelant, preferably wherein the chelant is selected from amino carboxylates, amino phosphonates, polyfunctionally-substituted aromatic chelating agents and mixtures thereof, preferably selected from the group consisting of glutamic-N,N-diacetic acid (GLDA), methyl-glycine-diacetic acid (MGDA), Diethylenetriamine penta methylphosphonic acid (DTPMP), 1-hydroxyethane 1,1-diphosphonic acid (HEDP), ethylenediaminetetra-acetates (EDTA), N-hydroxyethylethylenediaminetriacetates, nitrilo-triacetates (NTA), ethylenediamine tetrapro-prionates, triethylenetetraaminehexacetates, diethylenetriaminepentaacetates, aspartic acid-N-monoacetic acid (ASMA), aspartic acid-N,N-diacetic acid (ASDA), aspartic acid-N-monopropionic acid (ASMP), iminodisuccinic acid (IDS), Imino diacetic acid (IDA), N- (2-sulfomethyl) aspartic acid (SMAS), N- (2-sulfoethyl) aspartic acid (SEAS), N- (2- sulfomethyl) glutamic acid (SMGL), N- (2- sulfoethyl) glutamic acid (SEGL), N- methyliminodiacetic acid (MIDA), alanine-N,N-diacetic acid (ALDA), serine-N,N-diacetic acid (SEDA), isoserine-N,N-diacetic acid (ISDA), phenylalanine-N,N-diacetic acid (PHDA), anthranilic acid- N ,N - diacetic acid (ANDA), sulfanilic acid-N, N-diacetic acid (SLDA), taurine-N, N-diacetic acid (TUDA) and sulfomethyl-N,N-diacetic acid (SMDA), ethylenediamine disuccinate ("EDDS"), Hydroxyethyleneiminodiacetic acid, Hydroxyiminodisuccinic acid, Hydroxyethylene diaminetriacetic acid, or a mixture thereof, more preferably the chelant is selected from the group consisting of glutamic-N,N-diacetic acid (GLDA), methyl-glycine-diacetic acid (MGDA) and derivatives thereof, and/or Diethylenetriamine penta methylphosphonic acid (DTPMP), 1-hydroxyethane 1,1-diphosphonic acid (HEDP), and derivatives thereof, and mixtures thereof, most preferably Diethylenetriamine penta methylphosphonic acid (DTPMP). Without wishing to be bound by theory, the lower chelant levels are preferred to provide a less harsh laundry treatment composition while still facilitating stain removal especially bleachable stain removal.

**[0039]** The laundry treatment composition preferably comprise a polycarboxylate, preferably selected from the group of consisting of malonic acid, (ethyl enedioxy) diacetic acid, maleic acid, diglycolic acid, tartaric acid, tartronic acid, fumaric acid, citric acid, more preferably citric acid, wherein the citric acid is preferably present at a level of 0.1% to 5%, preferably from 0.5% to 3% most preferably from 1% to 2% by weight of the liquid treatment composition. Without wishing to be bound by theory, the lower polycarboxylate levels are preferred to provide a less harsh laundry treatment composition while still facilitating stain removal especially bleachable stain removal. In addition, they may help to protect the surfactant system against water hardness by complexing calcium and magnesium ions present in the wash liquor.

**[0040]** The laundry treatment composition preferably comprises between 0.01% and 3%, preferably between 0.025% and 2%, more preferably between 0.05% and 1.5% by weight of the laundry treatment composition of a soil release polymer, a polyethyleneimine preferably an alkoxylated polyethyleneimine, a zwitterionic polyamine or a mixture thereof. Preferably, the soil release polymer is selected from the group of polyester terephthalates, polyethylene glycol containing soil release polymers and a mixture thereof, and the alkoxylated polyethyleneimine is preferably an ethoxylated polyethyleneimine.

**[0041]** The laundry treatment composition preferably comprises less than 3%, preferably less than 2%, more preferably less than 1.5%, even more preferably between 0.01% and 1%, most preferably between 0.05% and 0.5% by weight of the laundry treatment composition of a soil release polymer, preferably selected from the group of polyester terephthalates, polyethylene glycol containing soil release polymers and a mixture thereof. Without wishing to be bound by theory, the presence of a soil release polymer is preferred to provide improved cleaning benefit whilst still providing a less harsh treatment composition.

**[0042]** Preferably, the laundry treatment composition comprises a polyethyleneimine, preferably an alkoxylated polyethyleneimine, more preferably an ethoxylated polyethyleneimine and wherein preferably the laundry treatment compo-

sition comprises less than 3%, preferably less than 2%, more preferably less than 1.5%, even more preferably between 0.01% and 1%, most preferably between 0.05% and 0.5% by weight of the laundry treatment composition of the polyethyleneimine, preferably ethoxylated polyethyleneimine.

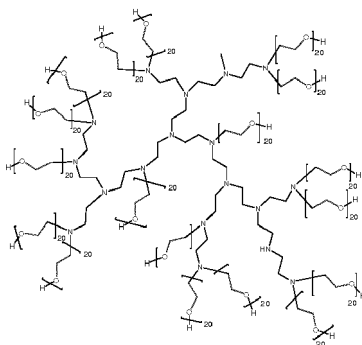
**[0043]** The ethoxylated polyethyleneimine may have a polyethyleneimine backbone of weight average molecular weight of between 100g/mol and 2000g/mol, preferably between 200g/mol and 1500g/mol, more preferably between 300g/mol and 1000g/mol, even more preferably between 400g/mol and 800g/mol, most preferably between 500g/mol and 700g/mol, preferably about 600.

**[0044]** The ethoxylation chains within the ethoxylated polyethyleneimine may be from 200g/mol to 2000g/mol weight average molecular weight, preferably from 400g/mol to 1500g/mol weight average molecular weight, more preferably from 600g/mol to 1000g/mol weight average molecular weight, most preferably about 880g/mol weight average molecular weight per ethoxylated chain.

**[0045]** The ethoxylation chains within the ethoxylated polyethyleneimine polymer of the present composition have on average 5 to 40, preferably 10 to 30, more preferably 15 to 25, even more preferably 18 to 22, most preferably about 20 ethoxy units per ethoxylation chain.

**[0046]** The ethoxylated polyethyleneimine may have a total weight average molecular weight of from 5000g/mol to 20000g/mol, preferably from 7500g/mol to 17500g/mol, more preferably from 10000g/mol to 15000g/mol, even more preferably from 12000g/mol to 13000g/mol, most preferably about 12700g/mol.

**[0047]** A preferred polyethyleneimine has the general structure of formula (I):



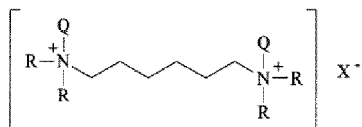
formula (I)

wherein the polyethyleneimine backbone has a weight average molecular weight of about 600g/mol, n of formula (I) has an average of about 20. Each polyethoxy chain is hydrogen capped. The degree of permanent quaternization of formula (I) is about 0% of the polyethyleneimine backbone nitrogen atoms. The molecular weight of this polyethyleneimine preferably is between 10000 and 15000g/mol, more preferably about 12700 g/mol.

**[0048]** The described ethoxylated polyethyleneimines can be made using techniques previously described in the art, and as such those skilled in the art would understand how to produce such compounds. These polyethyleneimines can be prepared, for example, by polymerizing ethyleneimine in the presence of a catalyst such as carbon dioxide, sodium bisulfite, sulfuric acid, hydrogen peroxide, hydrochloric acid, acetic acid, and the like, followed by an ethoxylation step.

**[0049]** Without wishing to be bound by theory, the presence of a polyethyleneimine is preferred to provide improved cleaning benefit whilst still providing a less harsh treatment composition.

**[0050]** Preferably, the laundry treatment composition comprises less than 3%, preferably less than 2%, more preferably less than 1.5%, even more preferably between 0.01% and 1%, most preferably between 0.05% and 0.5% of a zwitterionic polyamine. Particularly preferred zwitterionic polyamines are zwitterionic hexamethylene diamines according to the following formula:



**[0051]** R is an anionic or partially anionic unit-capped polyalkyleneoxy unit having the formula:  $-(R_2)_xR_3$  wherein  $R_2$  is C2-C4 linear or branched alkylene, and mixtures thereof, preferably C2 or branched C3 and mixtures thereof, even more preferably C2 (ethylene);  $R_3$  is hydrogen, an anionic unit, and mixtures thereof, in which not all  $R_3$  groups are hydrogen; x is from about 5 to about 50, preferably from about 10 to about 40, even more preferably from about 15 to

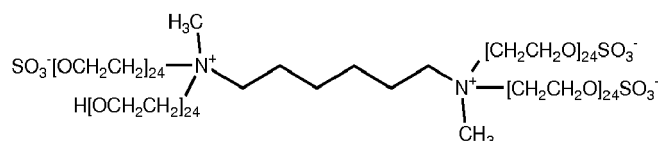
about 30, most preferably from about 20 to about 25. A preferred value for x is 24, especially when R comprises entirely ethyleneoxy units. Depending upon the method by which the formulator chooses to form the alkyleneoxy units, the wider or narrower the range of alkyleneoxy units present. The formulator will recognize that when ethoxylating a zwitterionic polyamine, only an average number or statistical distribution of alkyleneoxy units will be known. x values highlighted represent average values per polyalkoxy chain. Preferably the range of alkyleneoxy units within the zwitterionic polyamine is plus or minus two units, more preferably plus or minus one unit. Most preferably each R group comprises about the same average number of alkyleneoxy units. Non-limiting examples of R<sub>3</sub> anionic units include -(CH<sub>2</sub>)<sub>p</sub>CO<sub>2</sub>M; -(CH<sub>2</sub>)<sub>q</sub>SO<sub>3</sub>M; - (CH<sub>2</sub>)<sub>q</sub>OSO<sub>3</sub>M; -(CH<sub>2</sub>)<sub>q</sub>CH(SO<sub>2</sub>M)-CH<sub>2</sub>SO<sub>3</sub>M; -(CH<sub>2</sub>)<sub>q</sub>CH(OSO<sub>2</sub>M)CH<sub>2</sub>OSO<sub>3</sub>M; - (CH<sub>2</sub>)<sub>q</sub>CH(SO<sub>3</sub>M)CH<sub>2</sub>SO<sub>3</sub>M; -(CH<sub>2</sub>)<sub>p</sub>O<sub>3</sub>M; -P<sub>3</sub>M; -S<sub>3</sub>M and mixtures thereof; wherein M is hydrogen or a water soluble cation in sufficient amount to satisfy charge balance. Preferred anionic units are -(CH<sub>2</sub>)<sub>p</sub>CO<sub>2</sub>M; -S<sub>3</sub>M, more preferably -S<sub>3</sub>M (sulfonate group). The indices p and q are integers from 0 to 6, preferably 0 to 2, most preferably 0. For the purposes of the present invention, all M units, can either be a hydrogen atom or a cation depending upon the form isolated by the artisan or the relative pH of the system wherein the compound is used. Non-limiting examples of preferred cations include sodium, potassium, ammonium, and mixtures thereof.

**[0052]** Q is a quaternizing unit selected from the group consisting of C<sub>1</sub>-C<sub>30</sub> linear or branched alkyl, C<sub>6</sub>-C<sub>30</sub> cycloalkyl, C<sub>7</sub>-C<sub>30</sub> substituted or unsubstituted alkylenearyl, and mixtures thereof, preferably C<sub>1</sub>-C<sub>30</sub> linear or branched alkyl, even more preferably C<sub>1</sub>-C<sub>10</sub> or even C<sub>1</sub>-C<sub>5</sub> linear or branched alkyl, most preferably methyl; the degree of quaternization preferably is more than 50%, more preferably more than 70%, even more preferably more than 90%, most preferably about 100%.

**[0053]** X is an anion present in sufficient amount to provide electronic neutrality, preferably a water soluble anion selected from the group consisting of chlorine, bromine, iodine, methylsulfate, and mixtures thereof, more preferably chloride. To a great degree, the counter ion X will be derived from the unit which is used to perform the quaternization. For example, if methyl chloride is used as the quaternizing agent, chlorine (chloride ion) will be the counter ion X. Bromine (bromide ion) will be the dominant counter ion in the case where benzyl bromide is the quaternizing reagent.

**[0054]** Preferably from about 10% to about 100%, more preferably from about 20% to about 70%, even more preferably from 30% to about 50%, most preferably from about 35% to about 45% of the R<sub>3</sub> groups are an anionic unit, preferably a sulfonate unit, the remaining R<sub>3</sub> units being hydrogen.

**[0055]** Most preferred compound is the zwitterionic hexamethylene diamine represented by the following formula:



in which approximately 40% of the polyethoxy groups are sulfonated, the remaining polyethoxy groups being hydrogen capped. The degree of quaternization preferably is more than 90%, most preferably about 100%. Preferably the water soluble counter-anion is selected from the group consisting of chlorine, bromine, iodine, methylsulfate, and mixtures thereof, more preferably chloride.

**[0056]** Without wishing to be bound by theory, the presence of the zwitterionic polyamine is preferred to provide improved cleaning benefit whilst still providing a less harsh treatment composition.

**[0057]** The liquid laundry treatment composition may have a pH between 6 and 10, more preferably between 7 and 9, most preferably between 7.5 and 8.5.

**[0058]** Preferably, the laundry treatment composition may comprise but preferably is free of enzymes and bleaching compounds. This is preferred to allow for a less harsh treatment composition.

**[0059]** The laundry treatment composition may comprise adjunct materials selected from the group of dyes, opacifiers, pearlescent agents, preservatives, antioxidants, pH trimming agents, organic solvents, rheology control agents, dye transfer inhibitors, brighteners, perfume materials, silicone suds suppressors, or a mixture thereof.

**[0060]** The composition may be Newtonian or non-Newtonian. When Newtonian the composition preferably has a viscosity of from 1 mPa·s to 30 mPa·s, more preferably from 10 mPa·s to 25 mPa·s at 20°C, as measured using the method defined herein. Alternatively, the composition of the invention can be a shear thinning fluid. Especially suitable are shear thinning compositions having a high shear viscosity at 20°C of from 1 mPa·s to 40 mPa·s, preferably of from 10 mPa·s to 30 mPa·s at 1,000 s<sup>-1</sup>, and a low shear viscosity at 20°C of from 100 mPa·s to 1000 mPa·s, preferably of from 200 mPa·s to 500 mPa·s at 0.1 s<sup>-1</sup>, as measured using the method defined herein.

#### Test Method: Viscosity

**[0061]** The rheology profile is measured using a "TA instruments DHR1" rheometer, with a flat steel Peltier plate and

a 60mm, 2.026° cone plate geometry (TA instruments, serial number: SN960912). The flow curve procedure includes a conditioning step and a flow sweep step at 20°C. The conditioning step comprises a 10 seconds soaking step at 20°C, followed by a 10 seconds pre-shear step at 10 s<sup>-1</sup> at 20°C, followed by a 30 seconds zero shear equilibration step 20°C. The flow sweep step comprises a logarithmical shear rate increase from 0.01 s<sup>-1</sup> to 3,000 s<sup>-1</sup> at 20°C, with a 10 points per decade acquisition rate, a maximum equilibration time of 200 seconds, a sample period of 15 seconds and a tolerance of 3%.

**[0062]** When measuring shear thinning product compositions the high shear viscosity is defined at a shear rate of 1,000 s<sup>-1</sup>, and the low shear viscosity at a shear rate of 0.1 s<sup>-1</sup>. For Newtonian product compositions the shear rate at 1,000 s<sup>-1</sup> is recorded.

**[0063]** Preferably the laundry treatment composition comprises a polysaccharide rheology modifier or a mixture thereof, preferably a cationic polysaccharide, as described herein.

**[0064]** Preferably the laundry treatment composition comprises a polysaccharide rheology modifier or a mixture thereof, preferably a cationic polysaccharide, as described herein.

#### Non-soap surfactant

**[0065]** The non-soap surfactant comprises non-soap anionic surfactant. The non-soap surfactant comprises between 90% and 100%, preferably between 95% and 100%, most preferably about 100% by weight of the non-soap surfactant of the non-soap anionic surfactant.

**[0066]** The non-soap anionic surfactant may comprise a sulphate or a sulphonate anionic surfactant or a mixture thereof. Preferably the non-soap anionic surfactant comprises linear alkylbenzene sulphonate, alkyl sulphate, alkoxylated alkyl sulphate or a mixture thereof. More preferably, the non-anionic surfactant comprises a mixture of alkoxylated alkyl sulphate and linear alkylbenzene sulphonate. Preferably, the non-soap anionic surfactant comprises linear alkylbenzene sulphonate and ethoxylated alkyl sulphate, and preferably the weight ratio of ethoxylated alkyl sulphate to linear alkylbenzene sulphonate is between 100:0 and 50:50, preferably between 90:10 and 60:40, more preferably between 85:15 and 70:30. Without wishing to be bound by theory the ratio of non-soap anionic surfactants provides for optimal cleaning and rinsing whilst also providing less harsh treatment compositions.

**[0067]** Preferably, the alkoxylated alkyl sulphate is an ethoxylated alkyl sulphate with an average degree of ethoxylation of between 0.5 and 7, preferably between 1 and 5, more preferably between 2 and 4, most preferably about 3 and an average alkyl chain length of between 8 and 18. Preferably the alkoxylated alkyl sulphate has between 10 and 16, more preferably between 12 and 14 and the linear alkylbenzene sulphonate is a C<sub>10</sub>-C<sub>16</sub> linear alkylbenzene sulphonate or a C<sub>11</sub>-C<sub>14</sub> linear alkylbenzene sulphonate or a mixture thereof.

**[0068]** The liquid laundry treatment composition may comprise a non-ionic surfactant, preferably wherein the liquid laundry treatment composition comprises less than 10%, preferably less than 5%, by weight of the non-soap surfactant of the nonionic surfactant. Preferably the non-ionic surfactant is selected from an alcohol alkoxylate nonionic surfactant preferably selected from natural or olefin derived fatty alcohol alkoxylate, an oxo-synthesised fatty alcohol alkoxylate, Guerbet fatty alcohol alkoxylates, alkyl phenol alcohol alkoxylates or a mixture thereof. The fatty alcohol alkoxylate may have an average degree of alkoxylation of between 0.5 and 10, preferably between 1 and 9, more preferably between 3 and 8, more preferably a degree of ethoxylation of between 0.5 and 10, preferably between 1 and 9, more preferably between 3 and 8, most preferably between 5 and 8 or even from about 7 to about 8. The fatty alcohol alkoxylate may have an average alkyl chain length of between 8 and 18, preferably between 10 and 16, more preferably between 12 and 15.

**[0069]** More preferably the non-ionic surfactant comprises a mixture of alcohol ethoxylate surfactants, especially a mixture of a mid cut and a high cut alcohol ethoxylate with an average degree of ethoxylation of the mid cut and high cut alcohol ethoxylates of from 7 to 8. The mid cut alcohol ethoxylate is defined as having an average carbon chain length of from 12 to 14, the high cut alcohol ethoxylate is defined as having an average carbon chain length of from 14 to 15. The mid cut and high cut alcohol ethoxylate preferably are in a weight ratio of from 80:20 to 20:80, preferably 60:40 to 40:60.

**[0070]** The non-soap surfactant may comprise between 80% and 100% preferably between 90% and 100% or even between 95% and 100% preferably about 100% by weight of the non-soap surfactant of the anionic surfactant, or anionic surfactant non-ionic surfactant mixture.

Without wishing to be bound by theory the non-soap surfactant system according to the invention is believed to provide the correct balance of acceptable level of foam creation and easiness of rinsing in order to prevent detergent residues on fabrics, cleaning efficacy and minimizing harshness to fabrics.

#### Method of use

**[0071]** A further aspect of the present invention is a process of treating at least one garment comprising the steps of;



## EP 3 415 590 A1

- a. Using the consumer product according to the present invention to spray at least one garment with the laundry treatment composition;
- b. Optionally adding the at least one garment from step a to the drum of an automatic washing machine;
- c. Optionally washing the at least one garment in the automatic washing machine wherein the drum comprises a wash liquor, wherein the wash liquor comprises water and the one or more garments and wherein the only laundry treatment composition added to the drum of the automatic washing machine is that used in step a.

### Method of making

**[0072]** Any suitable method can be used to make the consumer product of the present invention. Those skilled in the art will be aware of techniques commonly known in the art to make the container and liquid treatment composition of the present invention.

**[0073]** The dimensions and values disclosed herein are not to be understood as being strictly limited to the exact numerical values recited. Instead, unless otherwise specified, each such dimension is intended to mean both the recited value and a functionally equivalent range surrounding that value. For example, a dimension disclosed as "40 mm" is intended to mean "about 40 mm."

### EXAMPLES

**[0074]** The impact of a full anionic surfactant system according to the invention (Example 1), compared to a nonionic based surfactant system outside the scope of the invention (Comparative example 1) has been assessed on its ability to generate a broadly dispersed spray pattern.

Formula compositions :

#### **[0075]**

Wt% (100% active)	Example 1 (anionic surfactant)	Comparative Example 1 (full nonionic surfactant)
Ethanol	0.81	0.81
1,2 Propylene glycol	0.37	0.37
Citric Acid	1.47	1.47
HLAS	1.25	-
C24 EO7 nonionic surfactant	-	3.1
C45 EO7 nonionic surfactant	-	3.1
Na-salt of Diethylene triamine pentamethylphosphonic acid (DTPMP) - chelant	0.11	0.11
Palm Kernel Fatty Acid	1.88	1.88
zwitterionic polyamine (Lutensit Z96 ex BASF)	0.14	0.14
C24 AE3S anionic surfactant	4.95	-
FWA36	0.001	0.001
Perfume	0.500	0.500
Water and minors (silicone suds suppressor - preservative - NaOH for pH-trimming)	Balance	Balance
pH	8.1	8.1

Spray bottles :

#### **[0076]**

- Flash Clean & Shine All Purpose Cleaning Spray (Lemon) ex Procter & Gamble Company, as commercially available in the UK on January 2017.

Test design :

**[0077]** A spray bottle is placed on a bench at a distance of 20cm between the front exit of the nozzle and the spraying target. The spraying target is a vertical positioned plastic slide used for laser printing (003R98203 - XEROX A3 TRANSPARENCY - PET material), collecting one single spray of detergent composition. An image of the spray pattern is achieved through scanning the sprayed-on plastic slide using a Xerox A3 scanner (Xerox® DocuMate® 4830). The image is consequently analyzed for number of liquid particles (at a lower size threshold for smaller particles of 0.01mm), using publicly available ImageJ Java-based image analysis software developed at the National Institutes of Health. 2 identical spray bottles were used to test the example 1 versus the comparative example formulation. The average values of two replicates are reported.

Test results :

**[0078]** The below table summarizes the analyzed spray patterns of the nonionic based surfactant system outside the scope of the invention (Comparative Example 1), compared to a full anionic surfactant system according to the invention (Example 1). It can clearly be seen that an anionic based surfactant system (Example 1) leads to a more broadly dispersed spray pattern compared to the full nonionic (Comparative Example 1) formulation, as supported by a higher number of sprayed particles despite equal sprayed volume.

# of sprayed particles	Example 1	Comparative Example 1
Spray bottle	1740	1056

## Claims

1. A consumer product comprising a liquid laundry treatment composition housed in a container, wherein;

a. the container comprises a spray applicator and an internal chamber comprising the laundry treatment composition and the liquid laundry treatment composition is in contact with the spray applicator and the spray applicator is capable of spraying the liquid laundry treatment composition into the environment external to the container upon activation of the spray applicator;

b. the liquid laundry treatment composition comprises at least 50%, preferably at least 60%, more preferably at least 70%, most preferably at least 80% by weight of the liquid laundry detergent composition of water; and between 1% and 15% preferably between 4% and 12%, more preferably between 5% and 10% by weight of the liquid laundry treatment composition of non-soap surfactant,

wherein the non-soap surfactant comprises non-soap anionic surfactant and the non-soap surfactant comprises between 90% and 100%, preferably between 95% and 100%, most preferably about 100% by weight of the non-soap surfactant of the non-soap anionic surfactant,

wherein the liquid laundry treatment composition comprises between 0.1% and 5%, preferably from 0.2% to 1%, more preferably from 0.3% to 0.7% by weight of the liquid laundry treatment composition of a polymer selected from a cationic polymer, a polysaccharide polymer, or a mixture thereof, preferably a cationic polymer.

2. The consumer product according to claim 1, wherein the cationic polymer is a cationically modified polysaccharide, preferably selected from cationic guar gums, cationic cellulosic polymers, and mixtures thereof, most preferably cationic cellulosic polymers even more preferably cationically modified hydroxyethyl cellulose, most preferably, hydroxyethyl cellulose derivatised with trimethyl ammonium substituted epoxide.

3. The consumer product according to any preceding claims wherein upon activation of the spray applicator the laundry treatment composition is sprayed into the environment external to the container and is present as droplets having an average particle size (Dv 50%) between 100 micron and 300 micron, preferably between 100 micron and 200 micron, more preferably between 100 micron and 150 micron.

4. The consumer product according to any preceding claims wherein the non-soap anionic surfactant comprises a

5 sulphate or a sulphonate anionic surfactant or a mixture thereof, preferably linear alkylbenzene sulphonate, alkyl sulphate, alkoxyated alkyl sulphate or a mixture thereof, more preferably a mixture of alkoxyated alkyl sulphate and linear alkylbenzene sulphonate, preferably wherein the non-soap anionic surfactant comprises linear alkylbenzene sulphonate and ethoxylated alkyl sulphate, and wherein the weight ratio of ethoxylated alkyl sulphate to linear alkylbenzene sulphonate is between 100:0 and 50:50, preferably between 90:10 and 60:40, more preferably between 85:15 and 70:30.

10 5. The consumer product according to claim 5 wherein the alkoxyated alkyl sulphate is an ethoxylated alkyl sulphate with an average degree of ethoxylation of between 0.5 and 7, preferably between 1 and 5, more preferably between 2 and 4, most preferably about 3 and an average alkyl chain length of between 8 and 18, preferably between 10 and 16, more preferably between 12 and 14, and the linear alkylbenzene sulphonate is a C<sub>10</sub>-C<sub>16</sub> linear alkylbenzene sulphonate or a C<sub>11</sub>-C<sub>14</sub> linear alkylbenzene sulphonate or a mixture thereof.

15 6. The consumer product according to any preceding claims wherein the liquid laundry treatment composition comprises a non-ionic surfactant, wherein the non-ionic surfactant is an alcohol alkoxylate surfactant, preferably selected from a natural or olefin derived fatty alcohol alkoxylate, an oxo-synthesised fatty alcohol alkoxylate, Guerbet fatty alcohol alkoxylates, alkyl phenol alcohol alkoxylates or a mixture thereof, preferably wherein the liquid laundry treatment composition comprises less than 10%, preferably less than 5%, by weight of the non-soap surfactant of the nonionic surfactant.

20 7. The consumer product according to claim 7, wherein the fatty alcohol alkoxylate has an average degree of alkoxylation of between 0.5 and 10, preferably between 1 and 9, more preferably between 3 and 8, more preferably a degree of ethoxylation of between 0.5 and 10, preferably between 1 and 9, more preferably between 3 and 8, most preferably between 5 and 8 or even from about 7 to about 8, and wherein the fatty alcohol alkoxylate preferably has an average alkyl chain length of between 8 and 18, preferably between 10 and 16, more preferably between 12 and 15.

25 8. The consumer product according to any preceding claims wherein the liquid laundry treatment composition comprises less than 10%, preferably less than 8%, more preferably less than 5%, even more preferably less than 3%, most preferably between 0.5% and 2% by weight of the laundry treatment composition of fatty acid, neutralised fatty acid soap or a mixture thereof.

30 9. The consumer product according to any preceding claims wherein the laundry treatment composition comprises less than 3%, preferably less than 2%, more preferably less than 1%, even more preferably between 0.01% and 0.5%, most preferably between 0.05% and 0.3% by weight of the laundry treatment composition of a chelant, preferably wherein the chelant is selected from amino carboxylates, amino phosphonates, polyfunctionally-substituted aromatic chelating agents and mixtures thereof, preferably selected from the group consisting of glutamic-N,N-diacetic acid (GLDA), methyl-glycine-diacetic acid (MGDA), Diethylenetriamine penta methylphosphonic acid (DTPMP), 1-hydroxyethane 1,1-diphosphonic acid (HEDP), ethylenediaminetetra-acetates (EDTA), N-hydroxyethylethylenediaminetriacetates, nitrilo-triacetates (NTA), ethylenediamine tetrapro-prionates, triethylenetetraamine-hexacetates, diethylenetriaminepentaacetates, aspartic acid-N-monoacetic acid (ASMA), aspartic acid-N,N-diacetic acid (ASDA), aspartic acid-N- monopropionic acid (ASMP), iminodisuccinic acid (IDS), Imino diacetic acid (IDA), N-(2-sulfomethyl) aspartic acid (SMAS), N- (2-sulfoethyl) aspartic acid (SEAS), N- (2- sulfomethyl) glutamic acid (SMGL), N- (2- sulfoethyl) glutamic acid (SEGL), N- methyliminodiacetic acid (MIDA), alanine-N,N-diacetic acid (ALDA), serine-N,N-diacetic acid (SEDA), isoserine-N,N-diacetic acid (ISDA), phenylalanine-N,N-diacetic acid (PH-DA), anthranilic acid- N ,N - diacetic acid (ANDA), sulfanilic acid-N, N-diacetic acid (SLDA), taurine-N, N-diacetic acid (TUDA) and sulfomethyl-N,N-diacetic acid (SMDA), ethylenediamine disuccinate ("EDDS"), Hydroxyethyleneiminodiacetic acid, Hydroxyiminodisuccinic acid, Hydroxyethylene diaminetriacetic acid, or a mixture thereof, more preferably the chelant is selected from the group consisting of glutamic-N,N- diacetic acid (GLDA), methyl-glycine-diacetic acid (MGDA) and derivatives thereof, and/or Diethylenetriamine penta methylphosphonic acid (DTPMP), 1-hydroxyethane 1,1-diphosphonic acid (HEDP), and derivatives thereof, and mixtures thereof, most preferably Diethylenetriamine penta methylphosphonic acid (DTPMP).

35 40 45 50 55 10. The consumer product according to any preceding claims wherein the laundry treatment composition comprises a polycarboxylate, preferably selected from the group of consisting of malonic acid, (ethyl enedioxy) diacetic acid, maleic acid, diglycolic acid, tartaric acid, tartronic acid, fumaric acid, citric acid, more preferably citric acid, wherein the citric acid is preferably present at a level of 0.1% to 5%, preferably from 0.5% to 3% most preferably from 1% to 2% by weight of the liquid treatment composition.

- 5
11. The consumer product according to any preceding claims wherein the laundry treatment composition comprises between 0.01% and 3%, preferably between 0.025% and 2%, more preferably between 0.05% and 1.5% by weight of the laundry treatment composition of a soil release polymer, a polyethyleneimine preferably an alkoxylated polyethyleneimine, a zwitterionic polyamine or a mixture thereof, preferably wherein the soil release polymer is selected from the group of polyester terephthalates, polyethylene glycol containing soil release polymers and a mixture thereof, and the alkoxylated polyethyleneimine is preferably an ethoxylated polyethyleneimine.
- 10
12. The consumer product according to any preceding claims wherein the liquid laundry treatment composition has a pH between 6 and 10, more preferably between 7 and 9, most preferably between 7.5 and 8.5.
13. The consumer product according to any preceding claims wherein the laundry treatment composition may comprise but preferably is free of enzymes and bleaching compounds.
- 15
14. A process of treating at least one garment comprising the steps of;
- 20
- a. Using the consumer product according to any preceding claims to spray at least one garment with the laundry treatment composition;
- b. Optionally adding the at least one garment from step a to the drum of an automatic washing machine;
- 25
- c. Optionally washing the at least one garment in the automatic washing machine wherein the drum comprises a wash liquor, wherein the wash liquor comprises water and the one or more garments and wherein the only laundry treatment composition added to the drum of the automatic washing machine is that used in step a.
- 30
- 35
- 40
- 45
- 50
- 55



## EUROPEAN SEARCH REPORT

Application Number  
EP 18 16 6903

5

10

15

20

25

30

35

40

45

50

55

2

EPO FORM 1503 03.82 (P04C01)

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (IPC)
X	DE 200 10 121 U1 (PROCTER & GAMBLE [US]) 28 December 2000 (2000-12-28) * claims 1, 9, 10, 13-15; examples II, IV, V *	1,3,4,8,9,11-13	INV. C11D1/37 C11D1/29 C11D3/37 C11D3/22 C11D17/00
X	WO 97/20913 A1 (PROCTER & GAMBLE [US]; PETRI MARCO [IT]) 12 June 1997 (1997-06-12) * examples *	1,4,8,9,13	
Y	GB 2 353 287 A (MCBRIDE ROBERT LTD [GB]) 21 February 2001 (2001-02-21) * test composition B on p.18; page 18 - page 19 * * page 12 *	1-14	
Y	WO 00/24856 A1 (PROCTER & GAMBLE [US]; BARNABAS MARY VIJAYARANI [US]; TRINH TOAN [US];) 4 May 2000 (2000-05-04) * claims 1, 46-53; examples Ig, Ij, Vif, VIj, XIVg, XIVj *	1-14	
T	WO 2015/078764 A1 (UNILEVER PLC [GB]; UNILEVER NV [NL]; CONOPCO INC DBA UNILEVER [US]) 4 June 2015 (2015-06-04) * comparative examples D, H in table 1, examples 2 and 4 in tables 1 and 2; tables 1, 2 *	1	
The present search report has been drawn up for all claims			TECHNICAL FIELDS SEARCHED (IPC) C11D
Place of search		Date of completion of the search	Examiner
The Hague		5 November 2018	Loiselet-Taisne, S
CATEGORY OF CITED DOCUMENTS		T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons & : member of the same patent family, corresponding document	
X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document			

**ANNEX TO THE EUROPEAN SEARCH REPORT  
ON EUROPEAN PATENT APPLICATION NO.**

EP 18 16 6903

5

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report.  
The members are as contained in the European Patent Office EDP file on  
The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

05-11-2018

10

15

20

25

30

35

40

45

50

55

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
DE 20010121	U1	28-12-2000	NONE
WO 9720913	A1	12-06-1997	AU 1329897 A 27-06-1997
		BR 9611905 A 02-03-1999	
		CA 2239992 A1 12-06-1997	
		CN 1207767 A 10-02-1999	
		CZ 9801742 A3 11-11-1998	
		EP 0783036 A1 09-07-1997	
		JP H11501976 A 16-02-1999	
		NZ 326023 A 28-02-2000	
		TR 199801026 T2 21-08-1998	
		WO 9720913 A1 12-06-1997	
		ZA 9610203 B 23-06-1997	
GB 2353287	A	21-02-2001	NONE
WO 0024856	A1	04-05-2000	AT 314451 T 15-01-2006
		AU 1133000 A 15-05-2000	
		AU 1223100 A 15-05-2000	
		AU 1321100 A 15-05-2000	
		BR 9914747 A 02-10-2001	
		BR 9915536 A 16-10-2001	
		CA 2346691 A1 04-05-2000	
		CA 2346771 A1 04-05-2000	
		CA 2346805 A1 04-05-2000	
		CN 1331739 A 16-01-2002	
		CN 1332787 A 23-01-2002	
		CZ 20011366 A3 12-06-2002	
		DE 69929233 T2 07-09-2006	
		EP 1123374 A1 16-08-2001	
		EP 1124924 A1 22-08-2001	
		EP 1144572 A2 17-10-2001	
		JP 4873781 B2 08-02-2012	
		JP 2002528652 A 03-09-2002	
		JP 2002528653 A 03-09-2002	
		JP 2002528654 A 03-09-2002	
		WO 0024851 A2 04-05-2000	
		WO 0024856 A1 04-05-2000	
		WO 0024858 A1 04-05-2000	
WO 2015078764	A1	04-06-2015	AR 098530 A1 01-06-2016
		CN 105874048 A 17-08-2016	
		EP 3074496 A1 05-10-2016	
		WO 2015078764 A1 04-06-2015	

EPO FORM P0459

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82