## (11) EP 3 439 092 A1

(12)

## **EUROPEAN PATENT APPLICATION**

(43) Date of publication:

06.02.2019 Bulletin 2019/06

(51) Int Cl.:

H01M 8/18 (2006.01)

H01M 8/20 (2006.01)

(21) Application number: 17275121.6

(22) Date of filing: 04.08.2017

(84) Designated Contracting States:

AL AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HR HU IE IS IT LI LT LU LV MC MK MT NL NO PL PT RO RS SE SI SK SM TR

**Designated Extension States:** 

**BA ME** 

**Designated Validation States:** 

MA MD

(71) Applicants:

 Siemens Aktiengesellschaft 80333 München (DE)  The University of Newcastle Newcastle NE1 7RU (GB)

(72) Inventors:

 Friedl, Jochen Newcastle Upon Tyne, NE3 3NJ (GB)

 Hughes, Timothy Wantage, OX12 7AU (GB)

 Stimming, Ulrich Newcastle upon Tyne, NE15 9JT (GB)

 Wolfschmidt, Holger 91058 Erlangen (DE)

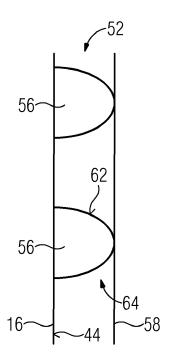
# (54) ELECTRODE STRUCTURE FOR A REDOX FLOW BATTERY AS WELL AS REDOX FLOW BATTERY

(57) The invention relates to an electrode structure (14) for a redox flow battery (10), the electrode structure (14) comprising at least one electrode (16),

the electrode (16) is configured as a surface element (42) having flat surfaces (44, 46) on opposite sides (48, 50) of the electrode (16), the surfaces (44, 46) extending in

respective parallel planes, wherein a plurality of three-dimensional protrusions (52) is applied on at least one of the surfaces (44, 46) and protrude, from the at least one surface (44), at least in a direction (54) being perpendicular to the respective planes.

FIG 3



EP 3 439 092 A1

## Description

[0001] The invention relates to an electrode structure for a redox flow battery according to the preamble of patent claim 1. The invention further relates to a redox flow battery according to the preamble of patent claim 14.

1

[0002] Such an electrode structure for a redox flow battery and such a redox flow battery are already known form US 2013/0022846 A1. The electrode structure comprises at least one electrode which is, for example, configured to provide electric current or electrical energy by means of which at least one electrically operable device can be operated.

[0003] Moreover, US 2010/018991 A1 shows a method of fabricating a material.

[0004] It is an object of the present invention to provide an electrode structure for a redox flow battery, and a redox flow battery so that the electrode structure and the redox flow battery can be manufactured in a particularly cost-effective way, wherein in a particularly advantageous operation of the redox flow battery can be realized. [0005] This object is solved by an electrode structure having the features of patent claim 1, as well as a redox flow battery having the features of patent claim 14. Advantageous embodiments with expedient developments of the invention indicated in the other patent claims.

[0006] A first aspect of the present invention relates to an electrode structure for a redox flow battery, the electrode structure comprising at least one electrode. For example, in a completely assembled state of the redox flow battery, the at least one electrode can provide electric current or electrical energy by means of which at least one electrically operable device can be operated. For example, the redox flow battery can be used to store electrical energy by means of which said device can be operated in a need-based manner. For example, the device is an electric machine which can be operated as an electric motor by means of the electric energy stored in the redox flow battery, and provided by means of said electrode.

[0007] In order to further develop the electrode structure in such a way that the electrode structure and thus the redox flow battery can be manufactured in a particularly cost-effective way, wherein a particularly advantageous operation of the electrode structure and thus the redox flow battery can be realized, according to the present invention, the electrode is configured as a surface element having flat surfaces on opposite sides of the electrode or the surface element respectively. In this regard the surfaces extend in respective parallel planes. [0008] Since, according to the present invention, the electrode is configured as said surface element the electrode is configured as an at least substantially two-dimensional (2D) electrode or structure. In this regard, the surface element or the electrode has a first extension extending in a first direction, and a second extension extending in a second direction being perpendicular to the first direction. The first and second extensions are sub-

stantially greater than a third extension of the surface element (electrode), the third extension extending in a third direction being perpendicular to the first and second directions. Thus, the surface element itself is flat and configured as, for example, a disc or plate. Since, in the electrode structure according to the present invention, said surface element is used as said electrode, the cost of the electrode structure can be kept particularly low. Moreover, in order to realize a particularly advantageous operation of the electrode structure and thus the redox flow battery, a plurality of three-dimensional (3D) protrusions is applied on at least one of the flat or even surfaces of the surface element (electrode), wherein said protrusions protrude, from the at least one surface, at least in a direction being perpendicular to the respective planes in which said surfaces extend or are arranged.

[0009] The respective plane is spanned by said first and second directions, so that said direction being perpendicular to the respective plane is said third direction running perpendicularly to the first and second directions and thus the respective plane. The protrusions are also referred to as three-dimensional structures or 3D structures which help realize a particularly advantageous, in particular efficient, operation of the redox flow battery. For example, the at least one surface on which the threedimensional structures are applied and the three-dimensional structures (protrusions) themselves form a total surface having a surface roughness which can be defined or adjusted by means of the protrusions in a need-based manner.

[0010] The idea behind the present invention is that, conventionally, redox flow battery systems normally use high-surface electrodes which are made from carbon fleece or carbon or graphite felt. Thus, said high surface electrodes are configured as carbon electrodes which are three-dimensionally formed and flattened and flown through by at least one electrolyte of the respective battery redox flow system. These high surface electrodes form high surface areas which are required by low reactive species in standard redox flow battery systems such as vanadium/vanadium.

[0011] The idea behind the invention is that high reactive species such as POMs (polyoxometalate redox flow batteries) as redox systems are able to work on more planar electrodes. Thus, said protrusions are used as three-dimensional structures on the at least substantially two-dimensional electrode for defining an advantageous surface roughness and high active areas. Thus, a particularly effective and efficient operation of the redox flow battery can be realized. Moreover, the costs of the redox flow battery can be kept particularly low. For example, by using said at least substantially two-dimensional surface element as said electrode of the electrode structure, flow resistance in a flow cell of the redox flow battery can be kept particularly low, which is particularly advantageous when using highly active redox species. Moreover the three-dimensional protrusions are used to achieve a defined surface roughness and active surface area.

40

45

25

30

40

45

Moreover, said three-dimensional protrusions can be used for a mechanical stabilization of at least one cell of the redox flow battery so that, for example, said protrusions can have a double function.

[0012] In an advantageous embodiment of the invention the protrusions are configured as spacers by means of which the electrode (surface element) is kept at a distance from at least one further component of the electrode structure. In this regard, the protrusions have a double function since the protrusions are used for both defining an advantageous surface roughness and keeping the electrode at a distance from said further component so as to realize a particularly high mechanical stability without having to use separate further components such as separate spaces. Thus, the number of parts of the electrode structure and thus the redox flow battery can be kept particularly low.

**[0013]** In the electrode structure according to the present invention, the electrode is a flat and cheap structure in a two-dimensional configuration equipped with three-dimensional add-ons formed by said protrusions. Thus, a scalable surface roughness of said total surface can be realized. For example, said surface roughness can be adjusted in a need-base manner by adjusting the protrusions, in particular with respect to their form and/or geometry and/or extension and/or number. Moreover, preferably, the protrusions can function as defined spacers thereby helping create a particularly high mechanical stability of the redox flow battery.

**[0014]** In a further advantageous embodiment of the invention the further component is configured as a second electrode or an ion exchange membrane for the redox flow battery, wherein the ion exchange membrane is also referred to as a membrane or a separator. Thus, by means of the protrusions, a particularly advantageous distance between the further component and the first electrode can be adjusted in a need-based manner.

**[0015]** In a particularly advantageous embodiment of the invention the further component is said membrane being arranged between the first electrode and said second electrode of the electrode structure. In this regard, preferably, the second electrode is configured as a second surface element having second flat or even surfaces on opposite sides of the second electrode. Embodiments and advantages of the first electrode and thus the first surface element are to be regarded as embodiments and advantages of the second electrode and thus the second surface element and vice versa.

**[0016]** The second surfaces extend in respective parallel second planes, wherein a plurality of three-dimensional second protrusions is applied on at least one of the second surfaces. The second protrusions protrude from the at least one second surface at least in a direction being perpendicular to the respective second planes. Preferably, said direction being perpendicular to the second planes is said third direction. Advantages and embodiments of the first protrusions are to be regarded as

embodiments and advantages of the second protrusions and vice versa. Thus, the electrodes can be kept at respective distances from the further component in a particularly advantageous way without having to use separate components such as separate spacers so that the number of parts and thus the costs and the weight of the redox flow battery can be kept low. Moreover, for example, the protrusions can be used as guiding elements which can help guide respective electrolytes flowing along the membrane on opposite sides and around the respective protrusions in a particularly advantageous way so that a particularly effective and efficient operation of the redox flow battery can be realized.

**[0017]** Preferably, the first protrusions contact the further component directly so that a particularly efficient operation of the redox flow battery can be realized in a particularly cost-effective way.

**[0018]** Moreover, preferably the second protrusions contact the further component directly so that a particularly advantageous operation of a redox flow battery can be realized.

**[0019]** In order to keep the cost of the redox flow battery particularly low, in a further embodiment of the present invention, the protrusions and the electrode are made from the same material. This means, for example, the first protrusions and the first electrode are made from the same material. Alternatively or additionally, for example, the second protrusions and the second electrode are made from the same material.

**[0020]** In a particularly advantageous embodiment of the invention the protrusions are made from an electrically non-conductive material so that, for example, additional separate isolators can be avoided.

**[0021]** For example, by using the protrusions as spacers a cell stack of the redox flow battery can be realized in a particularly cost-effective way, said cell stack comprising a plurality of battery cells which are also referred to as redox flow cells. Particularly, the battery cells can be realized as symmetrical redox flow cells in a particularly advantageous way by using said protrusions as spacers.

**[0022]** In a further embodiment of the invention the respective protrusion comprises a base body made from a first material, the base body being, at least partially, coated with a second material which is different from the first material. Thus, the electrode structure, in particular its characteristics, can be defended in a need-based manner and in a cost-effective way. For example, the electrode is made from the first material so that the respective base body and the respective electrode are made from the same material.

**[0023]** In a further advantageous embodiment of the invention, the second material is a metallic material or carbon or an electrically non-conductive material so as to, by means of the electrically non-conductive material, electrically isolate the respective protrusion at least partially, in particular in a major portion or completely. For example, said metallic material is at least one of nickel

15

20

40

45

50

55

(Ni), copper (Cu), gold (Au) and at least one metal alloy. For example, the surface element can be made from at least one of carbon, nickel, copper and at least one further material.

**[0024]** In a further advantageous embodiment of the invention, the protrusions are arranged in a flow channel for guiding an electrolyte of the redox flow battery. Thus, for example, the protrusions can help realize a particularly advantageous flow of said electrolyte so that a particularly efficient and effective operation of the redox flow battery can be realized.

[0025] For example, the respective protrusion can be configured as a hemisphere. Moreover, the respective protrusion can act as a flow body which is directed in a flow direction of the electrolyte and/or rotated or inclined in relation to the flow direction. Alternatively or additionally, the respective protrusion can be configured as a semi-tube, a tube or a rod. For example, the protrusion has a longitudinal extension which is parallel or inclined or perpendicular to a flow direction of the electrolyte. Moreover, the respective protrusion can be hollow or configured as a hollow structure which is, for example, a semi-tube, a tube or a rod. Moreover, the respective protrusion can have a form of a permit. The respective protrusion can be made from carbon, a metallic material, a hybrid material and/or coated material. Moreover, the protrusions can be configured as flow bodies which have wave structures. Moreover, the protrusions can have a scalable structure. Furthermore, the respective protrusion can have a head which is electrically isolated and/or mechanically smoothed and/or coated, in particular with

**[0026]** A second aspect of the invention relates to a redox flow battery comprising at least one electrode structure, in particular, at least one electrode structure according to the present invention. The electrode structure comprises at least one electrode.

[0027] In order to further develop the redox flow battery in such a way that the redox flow battery can be manufactured in a particularly cost-effective way, and that a particularly effective and efficient operation of the redox flow battery can be realized, according to the present invention, the electrode is configured as a surface element and thus an at least substantially two-dimensional structure having flat or even surfaces on opposite sides of the electrode, said surfaces extending respective parallel planes. Moreover, a plurality of three-dimensional protrusions is applied on at least one of the surfaces, and protrude from the at least one surface at least in a direction being perpendicular to the respective planes. Advantages and advantageous embodiments of the first aspect of the present invention are to be regarded as advantages and advantageous embodiments of the invention and vice versa.

**[0028]** Further advantages, features, and details of the invention derive from the following description of preferred embodiments as well as from the drawings. The features and feature combinations previously mentioned

in the description as well as the features and feature combinations mentioned in the following description of the figures and/or shown in the figures alone can be employed not only in the respective indicated combination but also in any other combination are taking alone without leaving the scope of the invention.

[0029] The drawings show in:

- FIG.1 a schematic view of a first embodiment of a redox flow battery according to the present invention;
- FIG.2 a schematic view of a method for manufacturing an electrode structure for the redox-flow battery according to the first embodiment;
- FIG.3 part of a schematic sectional view of the electrode structure according to the first embodiment;
- FIG.4 a schematic perspective view of an electrode of the electrode structure according to a second embodiment;
- <sup>25</sup> FIG.5 a schematic perspective view of the electrode according to a third embodiment;
  - FIG.6 a schematic perspective view of the electrode according to a fourth embodiment;
  - FIG.7 part of a schematic sectional view of the electrode structure according to a fifth embodiment; and
  - FIG.8 a schematic view of the redox flow battery according to a sixth embodiment.

**[0030]** In the figures the same elements or elements having the same functions are indicated by the same reference signs.

[0031] FIG. 1 shows in a schematic view a first embodiment of a redox flow battery 10 for storing electric current or electrical energy. In the first embodiment the redox flow battery 10 comprises at least one battery cell 12 which is also referred to as a cell, flow cell or redox flow cell. The redox flow battery 10, in particular said cell, comprises at least one electrode structure 14 having a first electrode 16 and at least two further components. A first one of said further components is an ion exchange membrane 18 which is also referred to as a membrane or separator. The second further component is a second electrode 20. As can be seen from FIG. 1, for example, the electrode 16 forms a positive pole, wherein the electrode 20 forms a negative pole of the cell. For example, the redox battery 10 can provide electrical energy or electric current by means of the electrodes 16 and 20, said electric current or electrical energy being stored in said cell. For example, at least one component 22 being, for

25

example, a load can be electrically connected with the electrodes 16 and 20 so that the component 22 can be supplied with the electrical energy stored in the cell. For example, the cell has a chamber 24 which is divided into two chamber areas 26 and 28 by means of said membrane. The chamber area 26 is supplied with a first electrolyte 30 stored in a first electrolyte tank 32. For example, the first electrolyte 30 is conveyed from the electrolyte tank 32 into and through the chamber area 26 by means of a first pump 34.

[0032] Furthermore, the chamber area 28 can be supplied with a second electrolyte 36 stored in a second electrolyte tank 38. For example, the second electrolyte 46 can be conveyed from the electrolyte tank 38 into and through the chamber area 28 by means of a second pump 40. Having flown through the respective chamber area 26 or 28 the respective electrolyte 30 or 36 can flow from the respective chamber area 26 or 28 back into the respective electrolyte tank 32 or 38. When flowing through the respective chamber area 26 or 28, the respective electrolyte 30 or 36 flows against and along the respective electrode 16 or 20. In this regard, for example, the chamber area 26 is at least partially bounded by the electrode 16 wherein the chamber area 28 is at least partially bounded by the electrode 20 so that the electrolyte 30 can contact the electrode 16 directly, and the electrolyte 36 can contact the electrode 20 directly when flowing through the chamber area 26 or 28 respectively. Thus, as can be seen from FIG.1, the electrolytes 30 and 36 circulate in two separate circuits between which an ion exchange occurs in the cell via said separator. During the ion exchange first substances solved in the electrolyte 30 and/or 36 are reduced and second substances solved in the electrolyte 30 and/or 36 are oxidized whereby electrical energy is set free. Said electrical energy can be provided by the redox flow battery 10 via the electrodes 16 and 20.

[0033] As can be seen from FIG. 2, for example, the electrode 16 is configured as a surface element 42 being an at least substantially two-dimensional structure. The surface element 42 has flat or even surfaces 44 and 46 on opposite sides 48 and 50 of the electrode 16 or the surface element 42. The surfaces 44 and 46 are arranged or extended in respective virtual parallel planes. Moreover, a plurality of three-dimensional protrusions 52 is applied on, for example, the surface 44. The protrusions 52 are three-dimensional structures or form such a threedimensional structure protruding, from the surface 44, at least in a direction being perpendicular to the respective planes. Said direction is illustrated by an arrow 54 in FIG. 2. The previous and following statements relating to the electrode 16 can be applied to the electrode 20 and vice versa. As can be seen from FIG. 2, in the first embodiment the protrusions 52 are configured as spacers 56 by means of which the electrode 16 is kept at a distance from at least one further component 58 of the electrode structure 14. In FIG.2, said distance or a direction along which said distance extends is illustrated by an arrow 60,

wherein the direction along which said distance extends is the distance in which the protrusions 52 protrude from the surface 44.

**[0034]** For example, the further component 58 is the membrane 18 or a third electrode of the electrode structure 14. As can be seen from FIG. 3, for example, the respective protrusion 52 has a curved or round outer circumferential surface 62 which, for example, contacts the further component 58 directly so that the respective protrusion 52 contacts the further component 58 directly.

[0035] FIG. 4 shows a second embodiment of the electrode 16 and thus the redox flow battery 10. Since, for example, the further component 58 is said separator, the protrusions 52 are arranged in a flow channel 64 through which the electrolyte 30 or 36 can flow. Thus, for example, the flow channel 64 is the chamber area 26 or 28. In FIG. 4, respective arrows 66 and 68 illustrate respective flow directions in or along which the respective electrolyte 30 or 36 can flow through the flow channel 64. Moreover, in the second embodiment, the respective protrusion 52 has a longitudinal direction. In this regard, for example, the respective protrusion 52 is configured as a tube having a round or circular cross-section. Moreover, the protrusion 52 can be configured as a semi-tube having a half-round or half-circular cross section. Moreover, the protrusion 52 can be configured as a rod having a rectangular cross-section. The longitudinal extension of the respective protrusion 52 is illustrated by an arrow 70.

[0036] If, for example, the flow direction of the electrolyte corresponds to the arrow 66, the longitudinal extension of the protrusion 52 is parallel or corresponds to the flow direction. Further, if the flow direction of the electrolyte 30 or 36 corresponds to the arrows 68, the longitudinal extension of the respective protrusion 52 is angular, in particular perpendicular, to the flow direction. Moreover, the respective protrusion 52 can be massive or hollow.

**[0037]** FIG.5 shows a third embodiment in which, for example, the respective protrusions 52 are configured as or have the forms of drops.

**[0038]** FIG. 6 shows a fourth embodiment in which the protrusions 52 are configured as or have the forms of cylinders.

[0039] FIG. 7 shows a fifth embodiment in which both electrodes 16 and 20 are configured as surface elements 42 having surfaces 44 facing towards each other, wherein respective protrusions 52 are applied on the respective surfaces 44. Thus, the electrodes 16 and 20 are kept at respective distances from the separator by means of the protrusions 52.

**[0040]** Furthermore, FIG.8 shows a sixth embodiment in which the redox flow battery 10 comprises a cell stack 72 having a plurality of battery cells. This means, the cell stack 72 comprises at least two battery cells. The cell stack 72 comprises a plurality of ion exchange membranes 18, a plurality of end electrodes 74 and a plurality of current collectors 76. For example, the respective electrode 74 can be configured as the electrode 16 or 20.

15

25

35

40

45

50

55

Moreover, the cell stack 72 comprises at least one bipolar electrode 78 having a first surface 80 and an opposite surface 82. In this regard, the electrolyte 30 can flow along the surface 80, wherein the electrolyte 36 can flow along the surface 82. For example, the respective surface 80 or 82 can be configured as said surface 44.

[0041] For example, the bipolar electrode 78 is configured in such a way that the bipolar electrode 78 comprises a first electrode element and a second electrode element. For example, the first electrode element and/or the second electrode element can be configured as the electrode 16 or 20. Moreover, the bipolar electrode 78 can comprise a separator arranged between the first and second electrode elements. For example, a respective surface such as the surface 44 of the respective electrode element can be equipped with protrusions 52 so that said electrode elements are kept at a distance from each other and/or from the separator arranged in between by said protrusions 52 applied on the respective surface 44 of the respective electrode element. Thus, for example, the protrusions 52 can be used as spacers between the electrode 16 or 20 and the separator or between two electrodes or the electrode elements. Thus, for example, said plurality of batteries cells can be realized as symmetrical redox flow cells in a particular cost-effective way. In other words, the first electrode element can be configured as the electrode 16 or 20 wherein the further component 58 can be said second electrode element which can also be configured as the electrode 16 or 20.

## Claims

1. An electrode structure (14) for a redox flow battery (10), the electrode structure (14) comprising at least one electrode (16),

### characterized in that

the electrode (16) is configured as a surface element (42) having flat surfaces (44, 46) on opposite sides (48, 50) of the electrode (16), the surfaces (44, 46) extending in respective parallel planes, wherein a plurality of three-dimensional protrusions (52) is applied on at least one of the surfaces (44, 46) and protrude, from the at least one surface (44), at least in a direction (54) being perpendicular to the respective planes.

2. The electrode structure (14) according to claim 1, characterized in that

the protrusions (52) are configured as spacers (56) by means of which the electrode (16) is kept at a distance (60) from at least one further component (58) of the electrode structure (14).

3. The electrode structure (14) according to claim 2, characterized in that

the further component (58) is configured as a second electrode (20) or an ion exchange membrane (18)

for the redox flow battery (10).

 The electrode structure (14) according to claim 2, characterized in that

the further component (58) is an ion exchange membrane (18) for the redox flow battery (10), the ion exchange membrane (18) being arranged between the electrode (16) and a second electrode (20) of the electrode structure (14).

**5.** The electrode structure (14) according to claim 4, characterized in that

the second electrode (20) is configured as a second surface element (42) having second flat surfaces (44, 46) on opposite sides (48, 50) of the second electrode (20), the second surfaces (44, 46) extending in respective parallel second planes, wherein a plurality of three-dimensional second protrusions (52) is applied on at least one of the second surfaces (44, 46) and protrude, from the at least one second surface (44), at least in a direction (54) being perpendicular to the respective second planes.

The electrode structure (14) according to any one of claims 2 to 4.

#### characterized in that

the protrusions (52) contact the further component (58) directly.

 The electrode structure (14) according to claim 5 or according to claims 5 and 6,

#### characterized in that

the second protrusions (52) contact the further component (58) directly.

8. The electrode structure (14) according to any one of the preceding claims,

#### characterized in that

the protrusions (52) and the electrode (16) are made from the same material.

**9.** The electrode structure (14) according to any one of the preceding claims,

## characterized in that

the protrusions (52) are made from an electrically non-conductive material.

**10.** The electrode structure (14) according to any one of the preceding claims,

#### characterized in that

the respective protrusion (52) comprises a bases body made from a first material, the base body being, at least partially, coated with a second material different from the first material.

**11.** The electrode structure (14) according to claim 10, **characterized in that** the electrode (16) is made from the first material.

15

**12.** The electrode structure (14) according to claim 10 or 11.

#### characterized in that

the second material is a metallic material or carbon or an electrically non-conductive material so as to electrically isolate the respective protrusion (52) at least partially.

**13.** The electrode structure (14) according to any one of the preceding claims,

## characterized in that

the protrusions (52) are arranged in a flow channel (64) for guiding an electrolyte (30) of the redox flow battery (10).

**14.** A redox flow battery (10) comprising at least one electrode structure (14) which comprises at least one electrode (16),

#### characterized in that

the electrode (16) is configured as a surface element (42) having flat surfaces (44, 46) on opposite sides (48, 50) of the electrode (16), the surfaces (44, 46) extending in respective parallel planes, wherein a plurality of three-dimensional protrusions (52) is applied on at least one of the surfaces (44, 46) and protrude, from the at least one surface (44), at least in a direction (54) being perpendicular to the respective planes.

30

35

40

45

50

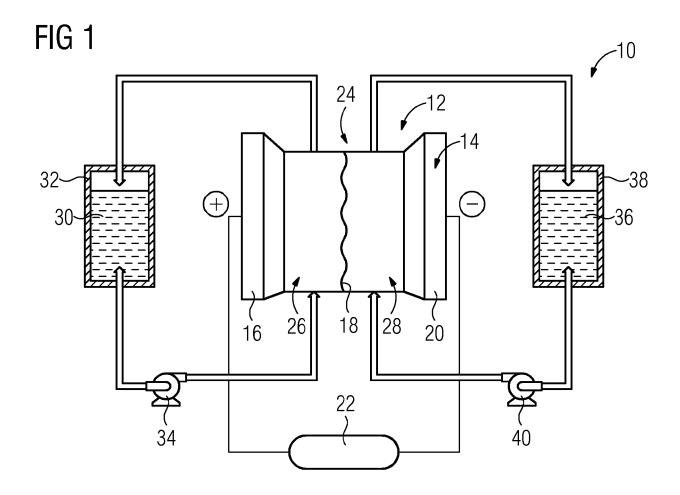


FIG 2

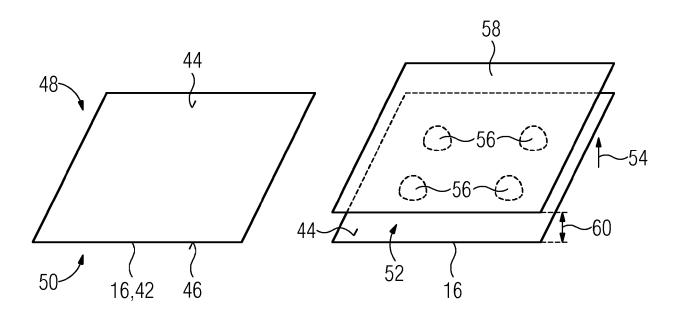
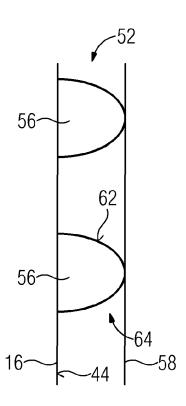


FIG 3



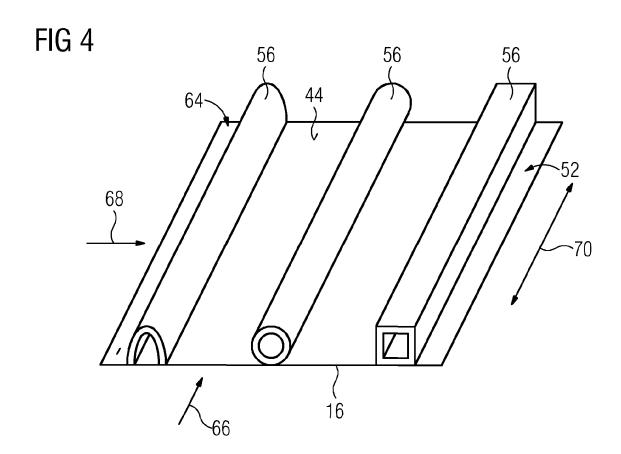


FIG 5

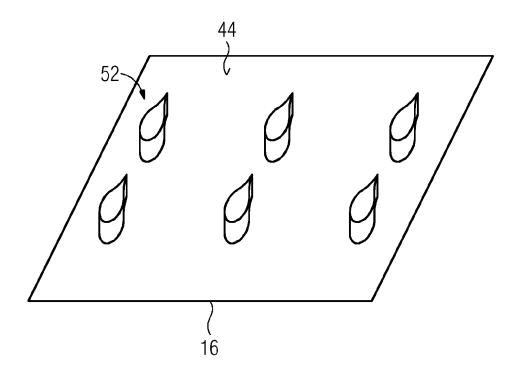
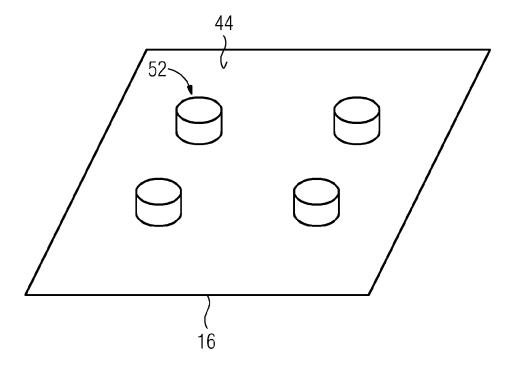
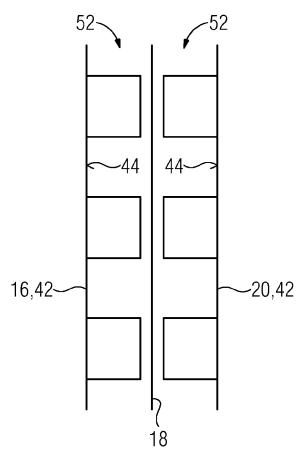
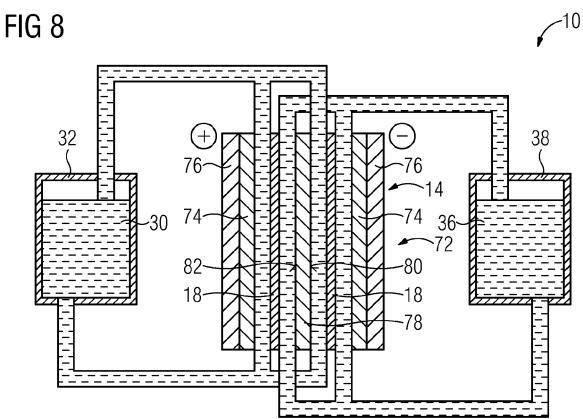


FIG 6











Category

Χ

Χ

Χ

χ

Α

#### **EUROPEAN SEARCH REPORT**

**DOCUMENTS CONSIDERED TO BE RELEVANT** 

US 2011/274988 A1 (FAN QINBAI [US] ET AL) 10 November 2011 (2011-11-10)

WO 2012/033692 A2 (PRIMUS POWER CORP [US]; 1-3,6,8, WINTER RICK [US]; HALL JONATHAN [US]; KELL 10-14

Citation of document with indication, where appropriate,

\* paragraph [0076] - paragraph [0113];

US 2014/065460 A1 (EVANS CRAIG [US] ET AL)

CN 204 651 407 U (ZHANGJIAGANG SMARTGRID

US 2015/200405 A1 (SCHARFENBERGER GUNTER

CO) 16 September 2015 (2015-09-16) \* the whole document \*

FANGHUA ELECTRICAL ENERGY STORAGE RES INST 14

of relevant passages

BRA) 15 March 2012 (2012-03-15)

\* the whole document \*

6 March 2014 (2014-03-06)

\* the whole document \*

figures 5a, 5b \*

X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category

technological background

O : non-written disclosure P : intermediate document

**Application Number** 

EP 17 27 5121

CLASSIFICATION OF THE APPLICATION (IPC)

INV. H01M8/18

H01M8/20

Relevant

to claim

1-4,6-8,

1-7,13,

1-14

& : member of the same patent family, corresponding

L: document cited for other reasons

document

10-14

1 - 14

1	0		

15

20

25

30

35

40

45

50

55

	A   03 2013/200403 A1	(SCHARLENDERGER GONTER	T T T			
	[DE] ET AL) 16 Jul	[DE] ET AL) 16 July 2015 (2015-07-16)  * the whole document *				
				SEARCHED (IPC) H01M		
-						
2	The present search report has					
_	Place of search	Date of completion of the search		Examiner		
04C01	Munich	22 September 2017	Кое	ssler, Jean-Luc		
1503 03.82 (P04C01)	CATEGORY OF CITED DOCUMENTS  X: particularly relevant if taken alone Y: particularly relevant if combined with ano document of the same category	E : earlier patent doou after the filing date ther D : document cited in	T: theory or principle underlying the invention E: earlier patent document, but published on, or after the filing date D: document cited in the application L: document cited for other reasons			

## EP 3 439 092 A1

## ANNEX TO THE EUROPEAN SEARCH REPORT ON EUROPEAN PATENT APPLICATION NO.

EP 17 27 5121

This annex lists the patent family members relating to the patent documents cited in the above-mentioned European search report. The members are as contained in the European Patent Office EDP file on The European Patent Office is in no way liable for these particulars which are merely given for the purpose of information.

22-09-2017

C	Patent document sited in search report		Publication date		Patent family member(s)		Publication date
U	S 2011274988	A1	10-11-2011	US WO	2011274988 2011142797		10-11-2011 17-11-2011
W	0 2012033692	A2	15-03-2012	AU CN EP JP JP KR TW	2011299439 103262336 2614552 6034788 2013541810 20140004627 201230458 2012033692	A A2 B2 A A A	07-03-2013 21-08-2013 17-07-2013 30-11-2016 14-11-2013 13-01-2014 16-07-2012 15-03-2012
U	S 2014065460	A1	06-03-2014	US WO	2014065460 2014039731		06-03-2014 13-03-2014
C	N 204651407	U	16-09-2015	NON	E		
U	S 2015200405	A1	16-07-2015	CN EP JP JP KR US WO	104471774 2875544 6177324 2015532761 20150038144 2015200405 2014012649	A1 B2 A A A1	25-03-2015 27-05-2015 09-08-2017 12-11-2015 08-04-2015 16-07-2015 23-01-2014
ORM P0459							

For more details about this annex : see Official Journal of the European Patent Office, No. 12/82

## EP 3 439 092 A1

#### REFERENCES CITED IN THE DESCRIPTION

This list of references cited by the applicant is for the reader's convenience only. It does not form part of the European patent document. Even though great care has been taken in compiling the references, errors or omissions cannot be excluded and the EPO disclaims all liability in this regard.

## Patent documents cited in the description

US 20130022846 A1 [0002]

• US 2010018991 A1 [0003]