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(54) CURABLE RESIN COMPOSITION AND DISPLAY DEVICE

(57) A curable resin composition containing quantum dots (A), a resin (B), a photopolymerizable compound (C), a photopolymerization initiator (D), an antioxidant (E), and a solvent (F), wherein a water content based on a total mass of the curable resin composition is 50 ppm or more and less than 5000 ppm.

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Description

Technical Field

⁵ [0001] The present invention relates to a curable resin composition, a cured film thereof, and a display device including the cured film.

Background Art

10 [0002] Patent Literature 1 discloses a curable resin composition containing quantum dots.

Citation List

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Patent Literature

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[0003] Patent Literature 1: Japanese Patent Laid-Open No. 2016-065178

Summary of Invention

20 Technical Problem

[0004] The curable resin composition disclosed in Patent Literature 1 discloses that the transmittance and heat-resistant transparency of the cured film by containing an antioxidant. However, when an antioxidant is contained, the emission intensity of the obtained cured film may be decreased.

[0005] An object of the present invention is to provide a curable resin composition containing an antioxidant and capable of suppressing a decrease in the emission intensity of the cured film to be obtained.

Solution to Problem

- 30 [0006] The present invention provides a curable resin composition, a cured film, and a display device shown below.
 - [1] A curable resin composition comprising: quantum dots (A); a resin (B); a photopolymerizable compound (C); a photopolymerization initiator (D); an antioxidant (E); and a solvent (F),
 - wherein a water content based on a total mass of the curable resin composition is 50 ppm or more and less than 5000 ppm.
 - [2] The curable resin composition according to [1], wherein the antioxidant (E) contains a phosphorus-based antioxidant.
 - [3] The curable resin composition according to [1] or [2], wherein a water content, based on a mass of the antioxidant (E), of the curable resin composition is 60 ppm or less.
 - [4] The curable resin composition according to any one of [1] to [3], wherein the solvent (F) has a solubility in water of 0.01% by mass or more and 30% by mass or less at a temperature of 20°C.
 - [5] The curable resin composition according to any one of [1] to [4], further comprising an organic ligand.
 - [6] The curable resin composition according to [5], wherein the organic ligand has a gas phase acidity of -15 kJ/mol or more and 5 kJ/mol or less.
- [7] The curable resin composition according to [5] or [6], wherein the organic ligand is an organic compound having a polar group, and the polar group is at least one group selected from the group consisting of a thiol group, a carboxy group, and an amino group.
 - [8] The curable resin composition according to any one of [1] to [7], further comprising a light scattering agent (H).
 - [9] A cured film formed from the curable resin composition according to any one of [1] to [8].
 - [10] A display device comprising the cured film according to [9].

Advantageous Effects of Invention

[0007] According to the present invention, there is provided a curable resin composition containing an antioxidant and capable of suppressing a decrease in the emission intensity of the cured film to be obtained.

Description of Embodiments

<Curable resin composition>

[0008] A curable resin composition according to the present invention (hereinafter, also simply referred to as "curable resin composition") contains quantum dots (A), a resin (B), a photopolymerizable compound (C), a photopolymerization initiator (D), an antioxidant (E), and a solvent (F). The water content based on the total mass of the curable resin composition is less than 5000 ppm. Further, the water content based on the total mass of the curable resin composition is 50 ppm or more. The curable resin composition may further contain an organic ligand. The organic ligand may be coordinated to the quantum dots (A) in the curable resin composition, or may be free in the curable resin composition.

[1] Quantum dots (A)

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[0009] The quantum dots (A) are semiconductor particles having a particle size of 1 nm or more and 100 nm or less and absorb ultraviolet light or visible light and emit light by utilizing a band gap of the semi-conductor.

[0010] Examples of the quantum dots (A) include compounds of Group 12 element and Group 16 element such as CdS, CdSe, CdTe, ZnS, ZnSe, ZnTe, HgS, HgSe, HgTe, CdHgTe, CdSeS, CdSeTe, CdSTe, ZnSeS, ZnSeTe, ZnSTe, HgSeS, HgSeTe, HgSTe, CdZnSe, CdZnSe, CdZnSe, CdHgSe, CdHgSe, CdHgSe, CdHgSe, HgZnSe, HgZnSe, HgZnSe, HgZnSe, HgZnSe, HgZnSe, HgZnSe, CdZnSeTe, CdZnSES, CdHgSeTe, CdHgSeTe, CdHgSeTe, HgZnSeS, HgZnSeTe and HgZnSTe; compounds of Group 13 element and Group 15 element such as GaN, GaP, GaAs, AlN, AlP, AlAs, InN, InP, InAs, GaNP, GaNAs, GaPAs, AlNP, AlNAs, AlPAs, InNP, InNAs, InPAs, GaAlNP, GaAlNAs, GalnPAs, GalnNP, GalnNAs, GalnPAs, InAlNP, InAlNAs and InAlPAs; compounds of Group 14 element and Group 16 element such as PdS and PbSe.

[0011] Further, the quantum dots (A) may form a core-shell structure by combining the above compounds. Examples of such a combination include fine particles having a core of CdSe and a shell of ZnS, and fine particles having a core of InP and a shell of ZnSeS.

[0012] Since the energy state of the quantum dots (A) depends on its size, the emission wavelength can be freely selected by changing the particle size. For example, in the case of quantum dots composed of only CdSe, the peak wavelengths of the fluorescence spectrum when the particle sizes are 2.3 nm, 3.0 nm, 3.8 nm, and 4.6 nm are 528 nm, 570 nm, 592 nm, and 637 nm, respectively.

[0013] The light emitted from the quantum dots (A) has a narrow spectral width, and by combining the light having such a steep peak, it is possible to expand a displayable color gamut in a display device including a cured film formed from the curable resin composition. Further, the quantum dots (A) have high responsiveness, and can efficiently utilize light emitted from a light source.

[0014] The curable resin composition may contain only one kind of quantum dots that emit light having a specific wavelength in response to light emitted from a light source, or may contain two or more kinds of quantum dots that emit light having different wavelengths in combination. Examples of the light having a specific wavelength include red light, green light, and blue light.

[0015] The content ratio of the quantum dots (A) in the curable resin composition may be, for example, based on the total amount of the solid content of the curable resin composition, 1% by mass or more and 60% by mass or less, preferably 10% by mass or more and 50% by mass or less, more preferably 15% by mass or more and 50% by mass or less, still more preferably 20% by mass or more and 50% by mass or less, and yet still more preferably 20% by mass or more and 40% by mass or less.

[0016] As used herein, the total amount of solid content means the total of the components contained in the curable resin composition, excluding the solvent (F). The content ratio of the curable resin composition in the solid content can be measured by a known analytical means such as liquid chromatography or gas chromatography.

[2] Organic ligand

[0017] The semiconductor particles as the quantum dots (A) may be present in the curable resin composition in a state where the organic ligand is coordinated. Hereinafter, the semiconductor particles to which the organic ligand is coordinated are also referred to as ligand-containing semiconductor particles. The ligand to be coordinated to the semiconductor particles may be, for example, an organic compound having a polar group exhibiting a coordination ability based on the semiconductor particles. The organic ligand may be an organic ligand added for stabilization or due to synthetic restrictions on the ligand-containing semiconductor particles. For example, in National Publication of Japanese Patent Application No. 2015-529698, the ligand-containing semiconductor particles contain hexanoic acid as an organic ligand from the viewpoint of particle size control, and the organic ligand is replaced by DDSA (dodecenylsuccinic acid) for stabilization after synthesis.

[0018] The organic ligand can be coordinated, for example, to the surface of the semiconductor particle.

[0019] The organic ligand may be one kind of ligand or two or more kinds of ligands. When the organic ligand is an organic compound having a polar group, the organic ligand usually coordinates to the quantum dots (A) via the polar group. The coordination of the organic ligand is confirmed from the fact that the quantum dots (A) are uniformly dispersed in a dispersion medium suitable for the organic ligand.

[0020] The polar group is preferably at least one group selected from the group consisting of a thiol group (-SH), a carboxy group (-COOH), and an amino group (-NH₂). Organic ligands with a polar group selected from this group are preferred because it is easy to obtain an organic ligand having a gas phase acidity satisfying the numerical range described above. Further, the polar group selected from the group can be advantageous in increasing the coordination property to the semiconductor particles. The high coordination property can contribute to the improvement of the color unevenness of the cured film and/or the improvement of the patterning property of the curable resin composition. In particular, from the viewpoint of obtaining a cured film (wavelength conversion film or the like) having better light emission characteristics, the polar group is more preferably at least one group selected from the group consisting of a thiol group and a carboxy group. The organic ligand may have one or more polar groups.

[0021] The organic ligand may be, for example, an organic compound represented by the following formula (X):

 X^A-R^X (X)

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[0022] In the formula, X^A is the above-mentioned polar group, and R^X is a monovalent hydrocarbon group which may contain a heteroatom (N, O, S, halogen atom, or the like). The hydrocarbon group may have one or two or more unsaturated bonds such as carbon-carbon double bonds. The hydrocarbon group may have a linear, branched, or cyclic structure. The number of carbon atoms of the hydrocarbon group is, for example, 1 or more and 40 or less, and may be 1 or more and 30 or less. The methylene group contained in the hydrocarbon group is optionally replaced by -O-, -S-, -C(=O)-O-, -O-C(=O)-O-, -O-C(=O)-O-, -C(=O)-O-, -O-C(=O)-O-, or the like.

[0023] The group R^X may contain a polar group. With respect to specific examples of the polar group, the above description relating to the polar group X^A is referred to.

[0024] Specific examples of the organic ligand having a carboxy group as the polar group X^A include formic acid, acetic acid, propionic acid, and saturated or unsaturated fatty acids. Specific examples of saturated or unsaturated fatty acids include saturated fatty acids such as butyric acid, pentanoic acid, caproic acid, caprylic acid, capric acid, lauric acid, myristic acid, pentadecylic acid, palmitic acid, margaric acid, stearic acid, arachidic acid, behenic acid, and lignoceric acid; monounsaturated fatty acids such as myristoleic acid, palmitoleic acid, oleic acid, icosenoic acid, erucic acid, and nervonic acid; and polyunsaturated fatty acids such as linoleic acid, α -linolenic acid, γ -linolenic acid, stearic acid, dihomo- γ -linolenic acid, arachidonic acid, eicosatetraenoic acid, docosadienoic acid, and adrenic acid (docosatetraenoic acid). [0025] Specific examples of the organic ligand having a thiol group or an amino group as the polar group X^A include organic ligands in which the carboxy group of the organic ligands having a carboxy group as the polar group X^A exemplified above is replaced by a thiol group or an amino group.

[0026] Preferred examples of the organic ligand represented by the formula (X) include a compound (J-1) and a compound (J-2) described later.

[0027] The gas phase acidity of the organic ligand is not particularly limited, but is preferably -15 kJ/mol or more and 5 kJ/mol or less, and more preferably -10 kJ/mol or more and 0 kJ/mol or less. Within the numerical range, in ligand-containing semiconductor particles coordinated with the organic ligand, when the water content based on the total mass in the curable resin composition is 50 ppm or more, the organic ligand is dissociated by the presence of a trace amount of moisture, which is advantageous for improving the dispersibility of the organic ligand-containing semiconductor particles. High dispersibility can contribute to improvement in color unevenness and improvement in emission characteristics of the cured film.

[0028] The gas phase acidity of organic ligands can be calculated based on the most stable structure. The molecular orbital calculation used to obtain the most stable structure is an approximate solution of Schrodinger's wave equation for a target molecule, and predicts the energy and molecular properties caused by the electronic state from this solution. [0029] For this molecular orbital calculation in the structure optimization step, a semi-empirical molecular orbital method, a non-empirical molecular orbital method, or a density functional theory (DFT) may be used. Examples of the semi-empirical molecular orbital method include a PPP (Pariser-Parr-Pople) method, an INDO (Intermediate Neglect of Differential Overlap) method, an AM1 (Austin Model 1) method, and a PM3 (Parametric Method 3) method. Examples of the non-empirical molecular orbital method include an HF (Hertree-Fock) method, an MPn (Moller-Plesset) method (n = 2, 3, 4, ...), and a CI (Configuration interaction) method. Examples of the density functional theory include B3LYP and PBEPBE methods. Among them, the MP2 method is preferable for the non-empirical molecular orbital method, and the B3LYP method is preferable for the density functional theory. Further, a function including a polarization function is preferably used as the basis function, and examples thereof include 6-31+G** and aug-cc-pvDz. Further, programs such as Gaussian and GAMESS are used for this molecular orbital calculation.

[0030] Specifically, first, with respect to a group of compounds consisting of anions and their acids, an initial structure

of a molecule is input, and optimization calculation of the structure is performed. When the calculation has not converged, the optimization calculation of the structure is performed again. Next, a reference vibration analysis is performed based on the converged calculation results to verify that there is no negative frequency. When there is a negative frequency, it returns to the initial structure input again; when there is no negative frequency, the result is taken as the optimized structure. The optimized structure depends on the initial structure which is the input condition of the calculation. Accordingly, a plurality of optimized structures are obtained with several initial structures, and a structure having the most stable energy is selected and taken as the most stable structure.

[0031] First, reference vibration analysis is performed on each of the anions (A⁻) and their acids (HA) of the group of compounds having the most stable structure to calculate the free energies G(A⁻) and G(HA). In the reference vibration analysis, it is necessary to use the same calculation method as the calculation method performed in the structure optimization step. Usually, the calculation result of the free energy is obtained by the calculation of the reference vibration analysis for checking whether there is a negative frequency performed in the structure optimization step. The temperature and pressure at which the free energy is determined are 25°C and 1 atm.

[0032] Next, the free energy change ΔG of the acid dissociation reaction is calculated from the calculated $G(A^-)$ and G (HA) and the free energy $G(H^+)$ of the hydrogen ion (H+) using the following formula (1). The free energy of H⁺ is -26.3 kJ/mol.

$$HA \leftrightarrow H^+ + A^-$$

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$$\Delta G = [G(H^{+}) + G(A^{-})] - G(HA)$$
 (1)

[0033] The calculated ΔG is the Gibbs free energy change upon deprotonation of the acid (HA) and indicates the acidity of the acid (HA) in the gas phase. The smaller the value of ΔG , the more likely proton dissociation occurs. That is, it indicates that the acidity of the acid (HA) is high.

[0034] The gas phase acidity of the organic ligand as used herein can be regarded as ΔG calculated based on the formula (1). The B3LYP method as used herein can be used as the density function, 6-31+ G^{**} can be used as the basis function, and GAMESS can be used for the molecular orbital calculation for the optimization calculation of the structure used for the calculation of the formula (1).

[Compound (J-1)]

[0035] The compound (J-1) is a compound having a first functional group and a second functional group. The first functional group is a carboxy group (-COOH) and the second functional group is a carboxy group or a thiol group (-SH). The compound (J-1) has a carboxy group and/or a thiol group, and can thus be a ligand coordinated to the quantum dots (A).

[0036] The curable resin composition may contain only one compound (J-1) or two or more thereof.

[0037] When the compound (J-1) is contained in the curable resin composition, the development speed of the curable resin composition can be sufficiently increased, and the emission intensity (brightness) of a cured film formed from the curable resin composition can be increased. It is considered that this is because both the carboxy group and the thiol group of the compound (J-1) can impart high developability with an alkali developer to the curable resin composition, and can be well coordinated to the quantum dots (A) to increase the dispersibility of the quantum dots (A) in the curable resin composition. In particular, the carboxy group has a higher effect of increasing the developability of the alkali developer, and the thiol group has a higher effect of increasing the dispersibility of the quantum dots (A).

[0038] Increasing the development speed of the curable resin composition can also contribute to increasing the emission intensity (brightness) of the cured film. It is considered that this is because the permeation of water into the cured film during the developing step can be suppressed.

[0039] An example of the compound (J-1) is a compound represented by the following formula (J-1a). The compound (J-1) may be an acid anhydride of the compound represented by the formula (J-1a):

[Formula 1]

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$$HS - \left(-R^8\right)_p OH$$
 (J-1a)

[0040] [In the formula, R^B represents a divalent hydrocarbon group. A plurality of R^B , when present, are optionally the same or different. The hydrocarbon group may have one or more substituents. When there are a plurality of substituents, they may be the same or different, and they may be bonded to each other to form a ring together with the atoms to which they are bonded.- CH_2 - contained in the hydrocarbon group is optionally replaced by at least one of -O-, -S-, -SO₂-, -CO-, and-NH-.

p represents an integer of 1 to 10.]

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[0041] Examples of the divalent hydrocarbon group represented by R^B include a chain hydrocarbon group, an alicyclic hydrocarbon group, and an aromatic hydrocarbon group.

[0042] Examples of the chain hydrocarbon group include a linear or branched alkanediyl group, and the number of carbon atoms thereof is usually 1 to 50, preferably 1 to 20, and more preferably 1 to 10.

[0043] Examples of the linear or branched alkanediyl group include a propane-1,2-diyl group, a propane-1,3-diyl group, a butane-1,4-diyl group, and a pentane-1,5-diyl group, a hexane-1,6-diyl group, a heptane-1,7-diyl group, an octane-1,8-diyl group, a nonan-1,9-diyl group, a decane-1,10-diyl group, an undecane-1,11-diyl group, a dodecane-1,12-diyl group, a propane-1,2-diyl group, a 1-methylbutane-1,3-diyl group, a 2-methylpropane-1,3-diyl group, a pentane-1,4-diyl group, and a 2-methylbutane-1,4-diyl group.

[0044] Examples of the alicyclic hydrocarbon group include a monocyclic or polycyclic cycloalkanediyl group, and the number of carbon atoms thereof is usually 3 to 50, preferably 3 to 20, and more preferably 3 to 10.

[0045] Examples of the monocyclic or polycyclic cycloalkanediyl group include a cyclobutane-1,3-diyl group, a cyclopentane-1,3-diyl group, a cyclohexane-1,4-diyl group, a norbornane-1,4-diyl group, a norbornane-2,3-diyl group, a 5-norbornene-2,3-diyl group, an adamantane-1,5-diyl group, and an adamantan-2,6-diyl.

[0046] Examples of the aromatic hydrocarbon group include an arenediyl group, and the number of carbon atoms thereof is usually 6 to 20. Examples of the monocyclic or polycyclic arenediyl group include a benzenediyl group, a naphthalenediyl group, an anthracenediyl group, a phenanthrenidyl group, a pyrenediyl group, a pyridinediyl group, a pyrazinediyl group, and a pyrazolediyl group.

[0047] Examples of the substituent that the hydrocarbon group may have include an alkyl group having 1 to 50 carbon atoms, a cycloalkyl group having 3 to 50 carbon atoms, an aryl group having 6 to 20 carbon atoms, a carboxy group, an amino group, and a halogen atom.

[0048] Examples of the alkyl group having 1 to 50 carbon atoms include saturated or unsaturated linear or branched alkyl groups. Specific examples thereof include a methyl group, an ethyl group, a propyl group, a butyl group, a pentyl group, a heavyl group, a heavyl group, an octyl group, an isopropyl group, an isobutyl group, a sec-butyl group, a tertbutyl group, an isopentyl group, a neopentyl group, and a 2-ethylhexyl group.

[0049] Examples of the cycloalkyl group having 3 to 50 carbon atoms include a cyclopropyl group, a cyclopentyl group, a cyclohexyl group, and a cyclooctyl group.

[0050] Examples of the aryl group having 6 to 20 carbon atoms include a phenyl group, a biphenyl group, a tolyl group, a xylyl group, an ethylphenyl group and a naphthyl group.

[0051] Examples of the halogen atom include a fluorine atom, a chlorine atom, a bromine atom, and an iodine atom. [0052] The substituent that the hydrocarbon group may have is preferably a carboxy group, an amino group, or a halogen atom.

[0053] When $-CH_2$ - contained in the hydrocarbon group is replaced by at least one of -O-, -CO-, and -NH-, $-CH_2$ - is preferably replaced by at least one of -CO- and -NH-, and more preferably -NH-. p is preferably 1 or 2.

[0054] Examples of the compound represented by the formula (J-1a) include compounds represented by the following formulas (1-1) to (1-9). The gas phase acidity of the compounds represented by the following formulas (1-1) to (1-9) all satisfy a numerical range of -15 kJ/mol or more and 5 kJ/mol or less.

[Formula 2]

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[0055] Specific examples of the compound represented by the formula (J-1a) include mercaptoacetic acid, 2-mercaptopropionic acid, 3-mercaptobutanoic acid, 4-mercaptobutanoic acid, mercaptosuccinic acid, mercaptostearic acid, mercaptooctanoic acid, 4-mercaptobenzoic acid, 2,3,5,6-tetrafluoro-4-mercaptobenzoic acid, L-cysteine, N-acetyl-L-cysteine, 3-methoxybutyl 3-mercaptopropionate, and 3-mercapto-2-methylpropionic acid. All of these gas phase acidities satisfy a numerical range of -15 kJ/mol or more and 5 kJ/mol or less.

[0056] Of these, 3-mercaptopropionic acid and mercaptosuccinic acid are preferable.

[0057] Another example of the compound (J-1) is a polycarboxylic acid compound, preferably a compound (J-1b) in which -SH in the formula (J-1a) is replaced by a carboxy group (-COOH) in the compound represented by the formula (J-1a).

[0058] Examples of the compound (J-1b) include the following compounds:

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Succinic acid, glutaric acid, adipic acid, octafluoroadipic acid, azelaic acid, dodecanedioic acid, tetradecanedioic acid, hexadecanedioic acid, heptadecanedioic acid, octadecanedioic acid, nonadecandioic acid, dodecafluorosuberic acid, 3-ethyl-3-methylglutaric acid, hexafluoroglutaric acid, trans-3-hexenedioic acid, sebacic acid, hexadecafluorosebacic acid, acetylenedicarboxylic acid, trans-aconitic acid, 1,3-adamantandicarboxylic acid, bicyclo[2.2.2]octane-1,4-dicarboxylic acid, cis-4-cyclohexene-1,2-dicarboxylic acid, 1,1-cyclopropanedicarboxylic acid, 1,1-cyclobutanedicarboxylic acid, cis- or trans-1,3-cyclohexanedicarboxylic acid, cis- or trans-1,4-cyclohexanedicarboxylic acid, 1,1-cyclopentanediacetic acid, 1,2,3,4-cyclopentanetetracarboxylic acid, decahydro-1,4-naphthalenedicarboxylic acid, 2,3-norbornanedicarboxylic acid, 5-norbornene-2,3-dicarboxylic acid, phthalic acid, 3-fluorophthalic acid, isophthalic acid, tetrafluoroisophthalic acid, terephthalic acid, tetrafluoroterephthalic acid, 2,5-dimethylterephthalic acid, 2,6-naphthalenedicarboxylic acid, 2,3naphthalenedicarboxylic acid, 1,4-naphthalenedicarboxylic acid, 1,1'-ferrocenedicarboxylic acid, 2,2'-biphenyldicarboxylic acid, 4,4'-biphenyldicarboxylic acid, 2,5-furandicarboxylic acid, benzophenone-2,4'-dicarboxylic acid monohydrate, benzophenone-4,4'-dicarboxylic acid, 2,3-pyrazinedicarboxylic acid, 2,3-pyridinedicarboxylic acid, 2,4-pyridinedicarboxylic acid, 3,5-pyridinedicarboxylic acid, 2,5-pyridinedicarboxylic acid, 2,6-pyridinedicarboxylic acid, 3,4-pyridinedicarboxylic acid, pyrazole-3,5-dicarboxylic acid monohydrate, 4,4'-stilbenedicarboxylic acid, anthraquinone-2,3-dicarboxylic acid, 4-(carboxymethyl)benzoic acid, chelidonic acid monohydrate, azobenzene-4,4'-dicarboxylic acid, azobenzene-3,3'-dicarboxylic acid, chlorendic acid, 1H-imidazole-4,5-dicarboxylic acid, 2,2-bis(4-carboxyphenyl)hexafluoropropane, 1,10-bis(4-carboxyphenoxy)decane, dipropylmalonic acid, dithiodiglycolic acid, 3,3'-dithiodipropionic acid, 4,4'-dithiodibutanoic acid, 4,4'-dicarboxydiphenyl ether, 4,4'-dicarboxydiphenyl sulfone, ethylene glycol bis(4-carboxyphenyl)ether, 3,4-ethylenedioxythiophene-2,5-dicarboxylic acid, 4,4'-isopropylidenediphenoxyacetic acid, 1,3-acetonedicarboxylic acid, methylenedisalicylic acid, 5.5'-thiodisalicylic acid, tris(2-carboxyethyl)isocyanurate, tetrafluorosuccinic acid, $\alpha.\alpha.\alpha'.\alpha'$ tetramethyl-1,3-benzenedipropionic acid, and 1,3,5-benzenetricarboxylic acid.

[0059] The molecular weight of the compound (J-1) is preferably 3000 or less, more preferably 2000 or less, still more preferably 1000 or less, yet still more preferably 800 or less, and particularly preferably 500 or less, from the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film. The molecular weight of the compound (J-1) is usually 100 or more.

[0060] The molecular weight may be a number-average molecular weight or a weight-average molecular weight. In this case, the number-average molecular weight and the weight-average molecular weight are a number-average molecular weight and a weight-average molecular weight in terms of standard polystyrene measured by gel permeation chromatography (GPC), respectively.

[0061] In the curable resin composition, at least a part of the molecules of the compound (J-1) is preferably coordinated to the quantum dots (A), and all or almost all the molecules may be coordinated to the quantum dots (A). That is, the curable resin composition preferably contains a compound (J-1) coordinated to the quantum dots (A), but may also contain a compound (J-1) not coordinated to the quantum dots (A) and a compound (J-1) coordinated to the quantum dots (A).

[0062] Including the compound (J-1) coordinated to the quantum dots (A) can be advantageous from the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film. Usually, the compound (J-1) can be coordinated to the quantum dots (A) via the first functional group and/or the second functional group. The compound (J-1) can be coordinated to the surface of the quantum dots (A), for example.

[0063] The ratio of contents of the compound (J-1) to the quantum dots (A) in the curable resin composition is preferably 0.001 or more and 1 or less, more preferably 0.01 or more and 0.5 or less, and still more preferably 0.02 or more and 0.1 or less in terms of mass ratio. When the ratio of contents is in this range, it may be advantageous from the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film.

[0064] The content ratio of the compound (J-1) in the curable resin composition is preferably 0.1% by mass or more and 20% by mass or less, more preferably 0.1% by mass or more and 10% by mass or less, still more preferably 0.2% by mass or more and 8% by mass or less, yet still more preferably 0.2% by mass or more and 5% by mass or less, and particularly preferably 0.5% by mass or more and 4% by mass or less, based on the total amount of the solid content of the curable resin composition, from the viewpoint of increasing the development speed of the curable resin composition and increasing the emission intensity (brightness) of the cured film.

[Compound (J-2)]

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⁵ **[0065]** The compound (J-2) is a compound (J-2) different from the compound (J-1), and is a compound having a polyalkylene glycol structure and having a polar group at the molecular end. The molecular end is preferably the end of the longest carbon chain in the compound (J-2) (the carbon atom in the carbon chain is optionally replaced by another atom such as an oxygen atom).

[0066] The curable resin composition may contain only one compound (J-2) or two or more thereof.

[0067] Compounds having a polyalkylene glycol structure and having the first functional group and the second functional group are considered to belong to the compound (J-1).

[0068] From the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film, the curable resin composition may contain the compound (J-1), or may contain the compound (J-1) and the compound (J-2).

[0069] The polyalkylene glycol structure refers to a structure represented by the following formula:

[Formula 3]

RcO

wherein n is an integer of 2 or more. In the formula, R^C is an alkylene group, and examples thereof include an ethylene group and a propylene group.

[0070] Specific examples of the compound (J-2) include a polyalkylene glycol-based compound represented by the following formula (J-2a).

[Formula 4]

[0071] In the formula (J-2a), X is a polar group, Y is a monovalent group, and Z^C is a divalent or trivalent group. n is

an integer of 2 or more, m is 1 or 2. R^C is an alkylene group.

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[0072] In the curable resin composition, at least a part of the molecules of the compound (J-2) is preferably coordinated to the quantum dots (A), and all or almost all the molecules may be coordinated to the quantum dots (A). That is, the curable resin composition preferably contains a compound (J-2) coordinated to the quantum dots (A), but may also contain a compound (J-2) not coordinated to the quantum dots (A) and a compound (J-2) coordinated to the quantum dots (A).

[0073] Including the compound (J-2) coordinated to the quantum dots (A) can be advantageous from the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film. Usually, the compound (J-2) can be coordinated to the quantum dots (A) via the polar group X. When the group Y contains a polar group, the compound (J-2a) may be coordinated to the quantum dots(A) via the polar group of the group Y, or via the polar group X and the polar group of the group Y. The compound (J-2) can be coordinated to the surface of the quantum dots (A), for example.

[0074] The polar group X is preferably at least one group selected from the group consisting of a thiol group (-SH), a carboxy group (-COOH), and an amino group (-NH2). The polar group selected from the group can be advantageous in increasing the coordination property to the quantum dots (A). In particular, from the viewpoint of increasing the emission intensity (brightness) of the cured film, the polar group X is more preferably at least one group selected from the group consisting of a thiol group and a carboxy group.

[0075] The group Y is a monovalent group. The group Y is not particularly limited, and examples thereof include a monovalent hydrocarbon group which may have a substituent (N, O, S, halogen atom, or the like). $-CH_2$ - contained in the hydrocarbon group is optionally replaced by -O-, -S-, -C(=O)-, -C(=O)-, -O--C(=O)-, -C(=O)-, -C(=O)-NH-, -NH-, or the like. **[0076]** The number of carbon atoms of the hydrocarbon group is, for example, 1 or more and 12 or less. The hydrocarbon group may have an unsaturated bond.

[0077] Examples of the group Y include an alkyl group having a linear, branched, or cyclic structure having 1 or more and 12 or less carbon atoms; and an alkoxy group having a linear, branched, or cyclic structure having 1 or more and 12 or less carbon atoms. The number of carbon atoms of the alkyl group and the alkoxy group is preferably 1 or more and 8 or less, more preferably 1 or more and 6 or less, and still more preferably 1 or more and 4 or less. -CH₂- contained in the alkyl group and the alkoxy group is optionally replaced by -O-, -S-,-C(=O)-, -C(=O)-O-, -O-C(=O)-, -C(=O)-NH-, -NH-, or the like. In particular, the group Y is preferably a linear or branched alkoxy group having 1 or more and 4 or less carbon atoms.

[0078] The group Y may contain a polar group. Examples of the polar group include at least one group selected from the group consisting of a thiol group (-SH), a carboxy group (-COOH), and an amino group (-NH $_2$). However, as described above, compounds having a polyalkylene glycol structure and having the first functional group and the second functional group are considered to belong to the compound (J-1). The polar group is preferably located at the end of the group Y. [0079] The group Z^C is a divalent or trivalent group. The group Z^C is not particularly limited, and examples thereof include a divalent or trivalent hydrocarbon group which may contain a heteroatom (N, O, S, halogen atom, or the like). The number of carbon atoms of the hydrocarbon group is, for example, 1 or more and 24 or less. The hydrocarbon group may have an unsaturated bond.

[0080] Examples of the group Z^C which is a divalent group include an alkylene group having a linear, branched, or cyclic structure having 1 or more and 24 or less carbon atoms; and an alkenylene group having a linear, branched, or cyclic structure having 1 or more and 24 or less carbon atoms. The number of carbon atoms of the alkyl group and the alkenylene group is preferably 1 or more and 12 or less, more preferably 1 or more and 8 or less, and still more preferably 1 or more and 4 or less. $-CH_2$ -contained in the alkyl group and the alkenylene group is optionally replaced by -O-, -S-, -C(=O)-, -C-C(=O)-, -C-C(O)-, -C

[0081] The group Z^C may have a branched structure. The group Z^C having a branched structure may have a polyalkylene glycol structure different from the polyalkylene glycol structure represented by the formula (J-2a) in a branched chain different from the branched chain including the polyalkylene glycol structure represented by the formula (J-2a).

[0082] In particular, the group Z^C is preferably a linear or branched alkylene group having 1 or more and 6 or less carbon atoms, and more preferably a linear alkylene group having 1 or more and 4 or less carbon atoms.

[0083] R^C is an alkylene group, and is preferably a linear or branched alkylene group having 1 or more and 6 or less carbon atoms, and more preferably a linear alkylene group having 1 or more and 4 or less carbon atoms. n in the formula (J-2a) is an integer of 2 or more, preferably 2 or more and 540 or less, more preferably 2 or more and 120 or less, and still more preferably 2 or more and 60 or less.

[0084] The molecular weight of the compound (J-2) may be, for example, about 150 or more and 10000 or less, but is preferably 150 or more and 5000 or less and more preferably 150 or more and 4000 or less from the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film.

[0085] The molecular weight may be a number-average molecular weight or a weight-average molecular weight. In

this case, the number-average molecular weight and the weight-average molecular weight are a number-average molecular weight and a weight-average molecular weight in terms of standard polystyrene measured by GPC, respectively. **[0086]** The ratio of contents of the compound (J-2) to the quantum dots (A) in the curable resin composition is preferably 0.001 or more and 2 or less, more preferably 0.01 or more and 1.5 or less, and still more preferably 0.1 or more and 1 or less in terms of mass ratio. When the ratio of contents is in this range, it may be advantageous from the viewpoint of increasing the development speed of the curable resin composition and from the viewpoint of increasing the emission intensity (brightness) of the cured film.

[3] Resin (B)

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[0087] Examples of the resin (B) include the following resins [K1] to [K4]:

resin [K1]: copolymer of at least one (a) (hereinafter also referred to as "(a)") selected from the group consisting of an unsaturated carboxylic acid and an unsaturated carboxylic acid anhydride and a monomer (c) (hereinafter also referred to as "(c)") copolymerizable with (a) (but different from (a));

resin [K2]: resin obtained by reacting a copolymer of (a) and (c) with a monomer (b) (hereinafter also referred to as "(b)") having a cyclic ether structure having 2 to 4 carbon atoms and an ethylenically unsaturated bond;

resin [K3]: resin obtained by reacting a copolymer of (b) and (c) with (a); and

resin [K4]: resin obtained by reacting a copolymer of (b) and (c) with (a) and further reacting with a carboxylic acid anhydride.

[0088] Examples of (a) include unsaturated monocarboxylic acids such as (meth)acrylic acid, crotonic acid, and o-, m-, and p-vinylbenzoic acid;

unsaturated dicarboxylic acids such as maleic acid, fumaric acid, citraconic acid, mesaconic acid, itaconic acid, 3-vinylphthalic acid, 4-vinylphthalic acid, 3,4,5,6-tetrahydrophthalic acid, 1,2,3,6-tetrahydrophthalic acid, dimethyltetrahydrophthalic acid, and 1,4-cyclohexenedicarboxylic acid;

bicyclo unsaturated compounds containing a carboxy group, such as methyl-5-norbornene-2,3-dicarboxylic acid, 5-carboxybicyclo[2.2.1]hept-2-ene, 5,6-dicarboxybicyclo[2.2.1]hept-2-ene, 5-carboxy-5-methylbicyclo[2.2.1]hept-2-ene, 5-carboxy-6-methylbicyclo[2.2.1]hept-2-ene, and 5-carboxy-6-ethylbicyclo[2.2.1]hept-2-ene;

unsaturated dicarboxylic acid anhydrides such as maleic anhydride, citraconic anhydride, itaconic anhydride, 3-vinylphthalic anhydride, 4-vinylphthalic anhydride, 3,4,5,6-tetrahydrophthalic anhydride, 1,2,3,6-tetrahydrophthalic anhydride, dimethyltetrahydrophthalic anhydride, 5,6-dicarboxybicyclo[2.2.1]hept-2-ene anhydride;

unsaturated mono[(meth)acryloyloxyalkyl]esters of di- or higher polycarboxylic acids such as mono[2-(meth)acryloyloxyethyl]succinate and mono[2-(meth)acryloyloxyethyl]phthalate; and

unsaturated (meth)acrylates containing a hydroxy group and a carboxy group in the same molecule, such as α -(hydroxymethyl) (meth)acrylic acid.

[0089] Of these, (meth)acrylic acid, maleic anhydride and the like are preferable from the viewpoint of copolymerization reactivity and solubility of the obtained resin in an alkaline aqueous solution.

[0090] As used herein, the term (meth)acrylic acid means acrylic acid and/or methacrylic acid. The same applies to "(meth)acryloyl", "(meth)acrylote" and the like.

(b) is, for example, a monomer having a cyclic ether structure having 2 to 4 carbon atoms (for example, at least one selected from the group consisting of an oxirane ring, an oxetane ring, and a tetrahydrofuran ring) and an ethylenically unsaturated bond. (b) is preferably a monomer having a cyclic ether structure having 2 to 4 carbon atoms and a (meth)acryloyloxy group.

[0091] Examples of (b) include monomers having an oxirane ring and an ethylenically unsaturated bond, such as glycidyl (meth)acrylate, β -methylglycidyl (meth)acrylate, β -ethylglycidyl (meth)acrylate, glycidyl vinyl ether, o-vinylbenzyl glycidyl ether, m-vinylbenzyl glycidyl ether, p-vinylbenzyl glycidyl ether, α -methyl-o-vinylbenzyl glycidyl ether, α -methyl-benzyl glycidyl ether, 2,3-bis(glycidyloxymethyl)styrene, 2,4-bis(glycidyloxymethyl)styrene, 2,5-bis(glycidyloxymethyl) styrene, 2,6-bis(glycidyloxymethyl)styrene, 2,3,5-tris(glycidyloxymethyl)styrene, 2,3,5-tris(glycidyloxymethyl)styrene, 2,3,6-tris(glycidyloxymethyl) styrene, 3,4,5-tris(glycidyloxymethyl)styrene, and 2,4,6-tris(glycidyloxymethyl) styrene;

monomers having an oxetane ring and an ethylenically unsaturated bond, such as 3-methyl-3-methacryloyloxymethyloxetane, 3-methyl-3-acryloyloxymethyloxetane, 3-ethyl-3-methacryloyloxymethyloxetane, 3-ethyl-3-acryloyloxymethyloxetane, 3-methyl-3-methacryloyloxyethyloxetane, 3-methyl-3-methyl-3-methyl-3-methyl-3-methyloxetane, 3-methyl-3

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acryloyloxyethyloxetane, and 3-ethyl-3-acryloyloxyethyloxetane; and

monomers having a tetrahydrofuran ring and an ethylenically unsaturated bond, such as tetrahydrofurfuryl acrylate (for example, Viscoat V#150, manufactured by OSAKA ORGANIC CHEMICAL INDUSTRY LTD.) and tetrahydrofurfuryl methacrylate.

(b) is preferably a monomer having an oxirane ring and an ethylenically unsaturated bond, because the reactivity during the production of the resins [K2] to [K4] is high and unreacted (b) hardly remains.

[0092] Examples of (c) include (meth)acrylic esters such as methyl (meth)acrylate, ethyl (meth)acrylate, n-butyl (meth)acrylate, sec-butyl (meth)acrylate, tert-butyl (meth)acrylate, 2-ethylhexyl (meth)acrylate, dodecyl (meth)acrylate, lauryl (meth)acrylate, stearyl (meth)acrylate, cyclopentyl (meth)acrylate, cyclopexyl (meth)acrylate, 2-methylcyclohexyl (meth)acrylate, tricyclo[5.2.1.0^{2,6}]decan-8-yl (meth)acrylate (referred to as "dicyclopentanyl (meth)acrylate" (common name) in the art or sometimes referred to as "tricyclodecyl (meth)acrylate"), tricyclo[5.2.1.0^{2,6}]decen-8-yl (meth)acrylate (which is referred to as "dicyclopentenyl (meth)acrylate" (common name) in the art), dicyclopentanyloxyethyl (meth)acrylate, isobornyl (meth)acrylate, adamantyl (meth)acrylate, allyl (meth)acrylate, propargyl (meth)acrylate, phenyl (meth)acrylate, naphthyl (meth)acrylate, and benzyl (meth)acrylate;

hydroxy group-containing (meth)acrylic esters such as 2-hydroxyethyl (meth)acrylate and 2-hydroxypropyl (meth)acrylate;

dicarboxylic diesters such as diethyl maleate, diethyl fumarate, and diethyl itaconate;

bicyclo unsaturated compounds such as bicyclo[2.2.1]hept-2-ene, 5-methylbicyclo[2.2.1]hept-2-ene, 5-ethylbicyclo[2.2.1]hept-2-ene, 5-hydroxybicyclo[2.2.1]hept-2-ene, 5-hydroxymethylbicyclo[2.2.1]hept-2-ene, 5-(2'-hydroxyethyl)bicyclo[2.2.1]hept-2-ene, 5-methoxybicyclo[2.2.1]hept-2-ene, 5-ethoxybicyclo[2.2.1]hept-2-ene, 5,6-dihydroxybicyclo[2.2.1]hept-2-ene, 5,6-di(2'-hydroxyethyl)bicyclo[2.2.1]hept-2-ene, 5,6-di(2'-hydroxyethyl)bicyclo[2.2.1]hept-2-ene, 5,6-dimethoxybicyclo[2.2.1]hept-2-ene, 5,6-diethoxybicyclo[2.2.1]hept-2-ene, 5-hydroxy-5-methylbicyclo[2.2.1]hept-2-ene, 5-hydroxy-5-ethylbicyclo[2.2.1]hept-2-ene, 5-hydroxymethyl-5-methylbicyclo[2.2.1]hept-2-ene, 5-tert-butoxycarbonylbicyclo[2.2.1]hept-2-ene, 5,6-bis(tert-butoxycarbonyl)bicyclo[2.2.1]hept-2-ene, 5,6-bis(tert-butoxycarbonyl)bicyclo[2.2.1]hept-2-ene, 5,6-bis(cyclohexyloxycarbonyl)bicyclo[2.2.1]hept-2-ene;

dicarbonylimide derivatives such as N-phenylmaleimide, N-cyclohexylmaleimide, N-benzylmaleimide, N-succinimidyl-3-maleimidobenzoate, N-succinimidyl-4-maleimidobutyrate, N-succinimidyl-6-maleimide caproate, N-succinimidyl-3-maleimide propionate, and N-(9-acridinyl)maleimide; and

styrene, α -methylstyrene, m-methylstyrene, p-methylstyrene, vinyltoluene, p-methoxystyrene, acrylonitrile, methacrylonitrile, vinyl chloride, vinylidene chloride, acrylamide, methacrylamide, vinyl acetate, 1,3-butadiene, isoprene, and 2,3-dimethyl-1,3-butadiene.

[0093] Among these, from the viewpoint of copolymerization reactivity and heat resistance of the resin (B), styrene, vinyltoluene, N-phenylmaleimide, N-cyclohexylmaleimide, N-benzylmaleimide, bicyclo[2.2.1]hept-2-ene and the like are preferable

[0094] In the resin [K1], the ratio of the structural unit derived from each of (a) and (c) in the total structural units constituting the resin [K1] is preferably the following:

the structural unit derived from (a): 2 mol% or more and 60 mol% or less; and the structural unit derived from (c): 40 mol% or more and 98 mol% or less,

and more preferably the following:

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the structural unit derived from (a): 10 mol% or more and 50 mol% or less; and the structural unit derived from (c): 50 mol% or more and 90 mol% or less.

[0095] When the ratio of the structural units of the resin [K1] is within the above range, the curable resin composition tends to be excellent in storage stability, developability when forming a cured pattern, and solvent resistance of the obtained cured pattern.

[0096] As used herein, the cured pattern refers to a cured film formed in a pattern.

[0097] The resin [K1] can be produced with reference to the method disclosed in for example, a document "Experimental Method for Polymer Synthesis" (edited by Takayuki Otsu, published by Kagaku-Dojin Publishing Company, INC, First Edition, First Printed on Mar. 1, 1972) and cited documents described in the above-mentioned document.

[0098] Specific examples thereof include the following method: predetermined amounts of (a) and (c), a polymerization initiator, a solvent and the like are placed in a reaction vessel; for example, a deoxidization atmosphere is formed by

replacing oxygen with nitrogen; and these are heated or kept warm during stirring.

[0099] The polymerization initiator, the solvent and the like which are used here are not particularly limited, and those commonly used in the art can be used. Examples of the polymerization initiator include azo compounds (2,2'-azobi-sisobutyronitrile, 2,2'-azobis(2,4-dimethylvaleronitrile) and the like) and organic peroxides (benzoyl peroxide and the like), and the solvent may be a solvent capable of dissolving each monomer, and examples of the solvent (F) contained in the curable resin composition of the present invention include solvents to be described later.

[0100] A solution after a reaction, of the resultant copolymer may be used as it is; a concentrated or diluted solution of the copolymer may be used; or a solid (powder) taken out from the copolymer by a method such as reprecipitation may be used. When the solvent (F) described later is used as the solvent for the polymerization, the solution after the reaction can be used as it is for the preparation of the curable resin composition, whereby the producing process of the curable resin composition can be simplified.

[0101] The resin [K2] can be produced by adding a cyclic ether having 2 to 4 carbon atoms of (b) to the copolymer of (a) and (c), that is, to a carboxylic acid and/or a carboxylic acid anhydride of (a).

[0102] The copolymer of (a) and (c) is first produced in the same manner as in the method described as the method for producing the resin [K1]. In this case, the ratio of the structural unit derived from each of (a) and (c) is preferably the same ratio as that described in the resin [K1].

[0103] Next, a cyclic ether having 2 to 4 carbon atoms of (b) is reacted with a part of the carboxylic acid and/or the carboxylic acid anhydride derived from (a) in the copolymer.

[0104] Subsequent to the production of the copolymer of (a) and (c), the resin [K2] can be produced by replacing a nitrogen atmosphere in a flask with air, and reacting (b) in the presence of a reaction catalyst for a carboxylic acid or a carboxylic acid anhydride and a cyclic ether (for example, an organic phosphorus compound, a metallic complex, or an amine compound), and a polymerization inhibitor (for example, hydroquinone and the like), for example, at 60°C or more and 130°C or less for 1 to 10 hours.

[0105] The amount of (b) used is preferably 5 mol or more and 80 mol or less, and more preferably 10 mol or more and 75 mol or less, based on 100 mol of (a). Within this range, the storage stability of the curable resin composition, the developability when forming the cured pattern, and the solvent resistance, heat resistance, mechanical strength and sensitivity of the cured pattern tend to be well balanced.

[0106] Examples of the organic phosphorus compound as a reaction catalyst include triphenylphosphine. As the amine compound as the reaction catalyst, for example, an aliphatic tertiary amine compound or an aliphatic quaternary ammonium salt compound can be used, and specific examples thereof include tris(dimethylaminomethyl)phenol, triethylamine, tetrabutylammonium bromide, and tetrabutylammonium chloride. From the viewpoint of developability and emission intensity of the cured film, the reaction catalyst is preferably an organic phosphorus compound.

[0107] The amount of the reaction catalyst used is preferably 0.001 part by mass or more and 5 parts by mass or less based on 100 parts by mass of the total amount of (a), (b), and (c).

[0108] The amount of the polymerization inhibitor used is preferably 0.001 part by mass or more and 5 parts by mass or less based on 100 parts by mass of the total amount of (a), (b), and (c).

[0109] The reaction conditions such as the charging method, the reaction temperature and the time can be appropriately adjusted in consideration of the production equipment, the amount of heat generated by the polymerization, and the like. In the same manner as the polymerization conditions, the charging method and the reaction temperature can be appropriately adjusted in consideration of the production equipment, the amount of heat generated by the polymerization, and the like.

[0110] The resin [K3] is produced by producing a copolymer of (b) and (c) in the same manner as in the abovementioned method for producing the resin [K1] as a first step. In the same manner as in the above, a solution after a reaction, of the resultant copolymer may be used as it is; a concentrated or diluted solution of the copolymer may be used; or a solid (powder) taken out from the copolymer by a method such as reprecipitation may be used.

[0111] The ratio of the structural unit derived from each of (b) and (c) based on the total number of moles of the total structural units constituting the copolymer is preferably the following:

the structural unit derived from (b): 5 mol% or more and 95 mol% or less; and the structural unit derived from (c): 5 mol% or more and 95 mol% or less,

and more preferably the following:

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the structural unit derived from (b): 10 mol% or more and 90 mol% or less; and the structural unit derived from (c): 10 mol% or more and 90 mol% or less.

[0112] The resin [K3] can be produced by reacting a carboxylic acid or a carboxylic acid anhydride of (a) with the cyclic ether derived from (b) contained in the copolymer of (b) and (c) under the same conditions as those of the method for

producing the resin [K2].

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[0113] The amount of (a) used which is reacted with the copolymer is preferably 5 mol or more and 80 mol or less based on 100 mol of (b).

[0114] The resin [K4] is a resin obtained by further reacting the resin [K3] with a carboxylic acid anhydride. A carboxylic acid anhydride is reacted with a hydroxy group generated by a reaction between a cyclic ether and a carboxylic acid or a carboxylic anhydride.

[0115] Examples of the carboxylic acid anhydride include maleic anhydride, citraconic anhydride, itaconic anhydride, 3-vinylphthalic anhydride, 4-vinylphthalic anhydride, 3,4,5,6-tetrahydrophthalic anhydride, 1,2,3,6-tetrahydrophthalic anhydride, dimethyltetrahydrophthalic anhydride, 5,6-dicarboxybicyclo[2.2.1]hept-2-ene anhydride.

[0116] The amount of the carboxylic acid anhydride used is preferably 0.5 to 1 mol based on 1 mol of the amount used in (a).

[0117] Specific examples of the resin (K1), the resin (K2), the resin (K3), and the resin (K4) include a resin [K1] such as a benzyl (meth)acrylate/(meth)acrylic acid copolymer and a styrene/(meth)acrylic acid copolymer;

a resin [K2] such as a resin produced by adding glycidyl (meth)acrylate to a benzyl (meth)acrylate/(meth)acrylic acid copolymer, a resin produced by adding glycidyl (meth)acrylate to a tricyclodecyl (meth)acrylate/styrene/(meth)acrylic acid copolymer, or a resin produced by adding glycidyl (meth)acrylate to a tricyclodecyl (meth)acrylate/benzyl (meth)acrylate/glycidyl (meth)acrylic acid copolymer; a resin [K3] such as a resin produced by reacting a tricyclodecyl (meth)acrylate/glycidyl (meth)acrylate copolymer with (meth)acrylic acid or a resin produced by reacting a tricyclodecyl (meth)acrylate/styrene/glycidyl (meth)acrylate copolymer with (meth)acrylic acid; and a resin [K4] such as a resin produced by reacting a tricyclodecyl (meth)acrylate/glycidyl (meth)acrylate copolymer with (meth)acrylic acid to produce a resin and then reacting this resin with tetrahydrophthalic anhydride.

[0118] In particular, the resin (B) preferably contains at least one selected from the group consisting of the resin [K2], the resin [K3], and the resin [K4].

[0119] As a further example of the resin (B), the alkalisoluble resin disclosed in Japanese Patent Laid-Open No. 2018-123274 can be mentioned. Examples of the alkalisoluble resin include a polymer (hereinafter, also referred to as "resin (Ba)") which has a double bond in a side chain, includes a structural unit (α) represented by the following general formula (I) and a structural unit (β) represented by the following general formula (II) in a main chain, and further includes an acid group.

[0120] The acid group may be introduced into the resin when, for example, the resin (Ba) contains a structural unit (γ) derived from an acid group-containing monomer (for example, (meth)acrylic acid). The resin (Ba) preferably contains the structural units (α), (β), and (γ) in the main chain skeleton.

[Formula 5]

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[0121] [In the formula, RA and RB are the same or different and each represent a hydrogen atom or a hydrocarbon group having 1 to 25 carbon atoms, n represents the average number of repeating units of the structural units represented by the general formula (I), and is a number of 1 or more.]

[Formula 6]

[0122] [In the formula, each R^C is the same or different and represents a hydrogen atom or a methyl group. Each R^D is the same or different and represents a linear or branched chain hydrocarbon group having 4 to 20 carbon atoms. m represents the average number of repeating units of the structural units represented by the general formula (II), and is

a number of 1 or more.]

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[0123] The content proportion of the structural unit (α) in the resin (Ba) is, for example, from the viewpoint of heat resistance and storage stability of the resin (Ba), 0.5% by mass or more and 50% by mass or less, preferably 1% by mass or more and 40% by mass or less, and more preferably 5% by mass or more and 30% by mass or less based on 100% by mass of the total amount of all monomer units that give the main chain skeleton of the resin (Ba) In the general formula (I), n represents the average number of repeating units of the structural units (α) in the resin (Ba), and n can be set so that the content proportion of the structural units (α) falls within the above range.

[0124] The content proportion of the structural unit (β) is, for example, from the viewpoint of the solvent resistance of the cured film, 10% by mass or more and 90% by mass or less, preferably 20% by mass or more and 80% by mass or less, and more preferably 30% by mass or more and 75% by mass or less based on 100% by mass of the total amount of all monomer units that give the main chain skeleton of the resin (Ba). In the general formula (II), m represents the average number of repeating units of the structural units (β) in the resin (Ba), and m can be set so that the content proportion of the structural units (β) falls within the above range.

[0125] The content proportion of the structural unit (γ) is, for example, from the viewpoint of the solubility of the resin (Ba) in the alkaline substance and the solubility of the resin (Ba) in the solvent, 0.5% by mass or more and 50% by mass or less, preferably 2% by mass or more and 50% by mass or less, and more preferably 5% by mass or more and 45% by mass or less based on 100% by mass of the total amount of all monomer units that give the main chain skeleton of the resin (Ba).

[0126] The resin (B) can include one or more selected from the group consisting of the resin [K1], resin [K2], resin [K3], resin [K4], and resin (Ba) described above.

[0127] The resin (B) preferably has a weight-average molecular weight of 9000 or less in terms of polystyrene measured by gel permeation chromatography (GPC). Since the resin (B) has the above weight-average molecular weight, a cured film having a high development speed and a high emission intensity can be obtained even in a thick film.

[0128] The weight-average molecular weight of the resin (B) in terms of polystyrene is, for example, 1000 or more and 9000 or less, and is preferably 2000 or more and 8500 or less and more preferably 3000 or more and 8500 or less, from the viewpoints of development speed and emission intensity.

[0129] The weight-average molecular weight of the resin (B) in terms of polystyrene is measured according to the measurement method described in the section of Examples below.

[0130] The weight-average molecular weight of the resin (B) can be adjusted within the above range by appropriately combining the selection of raw materials to be used, a charging method, and reaction conditions such as reaction temperature and time.

[0131] The molecular weight distribution [weight-average molecular weight (Mw) / number-average molecular weight (Mn)] of the resin (B) is, for example, 1.0 or more and 6.0 or less, and preferably 1.2 or more and 4.0 or less from the viewpoint of developability of a cured film.

[0132] The acid value of the resin (B) is preferably 90 mg KOH/g or more and 150 mg KOH/g or less based on the solid content. When the acid value is less than 90 mg KOH/g, the solubility of the cured film in an alkali developer may be lowered and a residue may be left on the substrate, and when the acid value is more than 150 mg KOH/g, peeling of the cured pattern obtained by development is more likely to occur.

[0133] The acid value of the resin (B) is preferably 95 mg KOH/g or more and 140 mg KOH/g or less, more preferably 100 mg KOH/g or more and 130 mg KOH/g or less, from the viewpoint of the developability of the cured film.

[0134] The acid value is a value measured as the amount (mg) of potassium hydroxide necessary for neutralizing 1 g of the resin (B), and can be obtained by titrating with, for example, an aqueous potassium hydroxide solution. The acid value of the resin (B) is measured according to the measurement method described in the section of Examples below. **[0135]** The resin (B) may contain a resin having a double bond equivalent of, for example, 300 g/eq or more and 2000

g/eq or less, preferably 500 or more and 1500 g/eq or less. Since the resin (B) contains a resin having a double bond equivalent of 300 g/eq or more and 2000 g/eq or less, the phenomenon of being quenched during the step of producing the cured pattern tends to be easily prevented. When the resin (B) contains a resin having a double bond equivalent exceeding 2000 g/eq, the ability of the resin (B) to effectively protect the quantum dots (A) tends to decrease. When the resin (B) contains a resin having a double bond equivalent of less than 300 g/eq, the cured film tends to be easily peeled off without being dissolved during development.

[0136] Examples of the resin having a double bond equivalent of 300 g/eq or more and 2000 g/eq or less include (meth)acrylic resins. The resin (B) is preferably composed of a (meth)acrylic resin.

[0137] The content ratio of the resin (B) in the curable resin composition is, for example, based on the total amount of the solid content of the curable resin composition, 5% by mass or more and 80% by mass or less, preferably 10% by mass or more and 70% by mass or less, more preferably 13% by mass or more and 60% by mass or less, and still more preferably 17% by mass or more and 55% by mass or less. When the content ratio of the resin (B) is within the above range, the quantum dots (A) tend to be easily dispersed and the emission efficiency tends to be easily maintained at a high level during the step of producing a cured pattern.

[0138] In the curable resin composition, the mass ratio (solid content ratio) of the resin (B) to the photopolymerizable compound (C) described later is, for example, 1 or more, and is preferably 1.5 or more, more preferably 2 or more, and still more preferably 2.5 or more from the viewpoint of developability and emission intensity of the cured film.

⁵ [4] Photopolymerizable compound (C)

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- **[0139]** The photopolymerizable compound (C) is a compound that can be polymerized by an active radical, an acid, or the like generated from the photopolymerization initiator (D) described later, and examples thereof include a compound having an ethylenically unsaturated bond, and preferable is a (meth)acrylic ester compound.
- **[0140]** In particular, the photopolymerizable compound (C) is preferably a polymerizable compound having three or more ethylenically unsaturated bonds. The weight-average molecular weight of the photopolymerizable compound (C) is preferably 150 or more and 2900 or less, and more preferably 250 or more and 1500 or less.
 - **[0141]** Examples of the polymerizable compound having three or more ethylenically unsaturated bonds include a photopolymerizable compound (Ca) having three or more ethylenically unsaturated bonds and having a carboxy group in its molecule, and a photopolymerizable compound (Cb) having three or more ethylenically unsaturated bonds and having no carboxy group in its molecule. The photopolymerizable compound (C) preferably contains at least one photopolymerizable compound (Ca) and a photopolymerizable compound (Cb), and may contain two or more photopolymerizable compounds (Ca), two or more photopolymerizable compounds (Cb), or at least one photopolymerizable compound (Ca) and at least one photopolymerizable compound (Cb).
- **[0142]** When the photopolymerizable compound (C) contains the photopolymerizable compound (Ca), the dispersibility of the quantum dots (A) in the curable resin composition can be improved, whereby the emission intensity of the cured film can be improved. Further, when the photopolymerizable compound (C) contains the photopolymerizable compound (Ca), the curability and heat resistance of the curable resin composition can be improved.
 - **[0143]** The ethylenically unsaturated bond of the photopolymerizable compound (Ca) is preferably a (meth)acryloyloxy group. The number of ethylenically unsaturated bonds contained in one molecule of the photopolymerizable compound (Ca) is preferably 3 to 5, and more preferably 3. The number of carboxy groups contained in one molecule of the photopolymerizable compound (Ca) is preferably 1.
 - **[0144]** Examples of the photopolymerizable compound (Ca) include compounds obtained by modifying a compound having three or more (meth)acryloyloxy groups and a hydroxy group, such as pentaerythritol tri(meth)acrylate or dipentaerythritol penta(meth)acrylate, with a polybasic acid. Examples of the compound include a compound obtained by monoesterifying pentaerythritol tri(meth)acrylate and succinic acid, a compound obtained by monoesterifying pentaerythritol tri(meth)acrylate and maleic acid, and a compound obtained by monoesterifying dipentaerythritol penta(meth)acrylate and maleic acid. Of these, a compound obtained by monoesterifying pentaerythritol tri(meth)acrylate and succinic acid is preferable.
- ³⁵ **[0145]** The ethylenically unsaturated bond of the photopolymerizable compound (Cb) is preferably a (meth)acryloyloxy group. The number of ethylenically unsaturated bonds contained in one molecule of the photopolymerizable compound (Cb) is preferably 3 to 6.
 - [0146] Examples of the photopolymerizable compound (Cb) include trimethylolpropane tri(meth)acrylate, pentaerythritol tri(meth)acrylate, pentaerythritol tetra(meth)acrylate, dipentaerythritol penta(meth)acrylate, dipentaerythritol penta(meth)acrylate, dipentaerythritol penta(meth)acrylate, tripentaerythritol hepta(meth)acrylate, tetrapentaerythritol deca(meth)acrylate, tris(2-(meth)acryloloxyethyl)isocyanurate, ethylene glycol-modified pentaerythritol tetra(meth)acrylate, ethylene glycol-modified dipentaerythritol hexa(meth)acrylate, propylene glycol-modified pentaerythritol tetra(meth)acrylate, propylene glycol-modified dipentaerythritol hexa(meth)acrylate, caprolactone-modified pentaerythritol tetra(meth)acrylate, caprolactone-modified dipentaerythritol hexa(meth)acrylate. Of these, dipentaerythritol penta(meth)acrylate and dipentaerythritol hexa(meth)acrylate are preferable.
 - [0147] The content ratio of the photopolymerizable compound (C) in the curable resin composition is preferably, based on the total amount of the solid content of the curable resin composition, 7% by mass or more and 60% by mass or less, more preferably 10% by mass or more and 45% by mass or less, and still more preferably 13% by mass or more and 30% by mass or less. When the content ratio of the photopolymerizable compound (C) is within the above range, the residual film ratio of the cured pattern and the chemical resistance of the cured pattern tend to be further improved.
 - **[0148]** From the viewpoint of improving the development speed of the curable resin composition, the content ratio of the photopolymerizable compound (C) is preferably 13% by mass or more, more preferably 15% by mass or more, and still more preferably 20% by mass or more, based on the total amount of the solid content of the curable resin composition.
 - **[0149]** From the viewpoint of improving the development speed of the curable resin composition, the photopolymerizable compound (C) preferably contains 50% by mass or more of the photopolymerizable compound (Cb). From the viewpoint of improving the solvent resistance (residual film ratio) of the cured film, the photopolymerizable compound (C) preferably contains a photopolymerizable compound (Ca) and a photopolymerizable compound (Cb).

[5] Photopolymerization initiator (D)

[0150] The photopolymerization initiator (D) is a compound capable of initiating polymerization by generating an active radical, an acid, or the like by the action of light or heat, and can contain at least one selected from the group consisting of an oxime compound, a biimidazole compound, a triazine compound, and an acylphosphine compound. Above all, the photopolymerization initiator (D) preferably contains an oxime compound. When these photopolymerization initiators are used, the residual film ratio of the cured pattern can be further increased.

[0151] From the viewpoint of further increasing the polymerization sensitivity during the production of the cured film, the oxime compound, the biimidazole compound, the triazine compound, and the acylphosphine compound are preferably compounds having at least two aromatic rings in the molecule.

[0152] Examples of the aromatic ring include 5-membered rings such as a furan ring, a pyrrole ring, an imidazole ring, a thiophene ring, and a thiazole ring; 6-membered rings such as a benzene ring, a pyridine ring, a pyrimidine ring, and a triazine ring; and condensed rings thereof.

[0153] The oxime compound preferably contains an oxime compound having a first molecular structure represented by the following formula (1). Hereinafter, the oxime compound is also referred to as "oxime compound (1)".

[Formula 7]

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$$*$$
 R^1
 (1)

[0154] In the formula (1), R¹ represents R¹¹, OR¹¹, COR¹¹, SR¹¹, CONR¹²R¹³, or CN.

[0155] R¹¹, R¹², and R¹³ each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an aralkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0156] The hydrogen atoms of the group represented by R^{11} , R^{12} , or R^{13} are optionally replaced by OR^{21} , COR^{21} , SR^{21} , $NR^{22}Ra^{23}$, $CONR^{22}R^{23}$, $-NR^{22}-OR^{23}$, $-N(COR^{22})-OCOR^{23}$, $-C(=N-OR^{21})-R^{22}$, $-C(=N-OCOR^{21})-R^{22}$, COR^{21}

[0157] R²⁴ represents a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an aralkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms. **[0158]** When the group represented by R¹¹, R¹², R¹³, R²¹, R²², or R²³ has an alkyl moiety, the alkyl moiety is optionally branched or cyclic, and R¹² and R¹³, and R²² and R²³ are each optionally joined together to form a ring.

* represents a bond with a second molecular structure which is a molecular structure other than the first molecular structure of the oxime compound.

[0159] When the photopolymerization initiator (D) contains the oxime compound (1), it is possible to form a cured film having good emission intensity while suppressing film thickness unevenness due to curing shrinkage even in a thick film. **[0160]** Examples of alkyl groups having 1 to 20 carbon atoms represented by R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1) include a methyl group, an ethyl group, a propyl group, an isopropyl group, a butyl group, an isobutyl group, a sec-butyl group, a tert-butyl group, a pentyl group, an isopentyl group, a tert-pentyl group, a hexyl group, a hexyl group, a decyl group, an isodecyl group, an isooctyl group, a 2-ethylhexyl group, a tert-octyl group, a nonyl group, an isononyl group, a decyl group, an isodecyl group, an undecyl group, a dodecyl group, a tetradecyl group, and a cyclohexylethyl group, an icosyl group, a cyclopentyl group, a cyclohexyl group, a cyclohexylmethyl group, and a cyclohexylethyl group. **[0161]** Examples of aryl groups having 6 to 30 carbon atoms represented by R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1) include a phenyl group, a tolyl group, a xylyl group, an ethylphenyl group, an anthryl group, an anthryl group, and an anthryl group, and a phenyl group, a biphenylyl group, a naphthyl group, and an anthryl group substituted

with one or more of the alkyl groups.

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[0162] Examples of aralkyl groups having 7 to 30 carbon atoms represented by R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1) include a benzyl group, an α -methylbenzyl group, an α , α -dimethylbenzyl group, and a phenylethyl group. **[0163]** Examples of the heterocyclic groups having 2 to 20 carbon atoms represented by R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1) include a pyridyl group, a pyrimidyl group, a furyl group, a thienyl group, a tetrahydrofuryl group a diavelegyl group a hopegyggel 2 yl group a tetrahydrofuryl group a diavelegyl group a hopegyggel 2 yl group a tetrahydrofuryl group a pyrrolidyl group, an imidszelidyl group

and R²⁴ in the formula (1) include a pyridyl group, a pyrimidyl group, a furyl group, a thienyl group, a tetrahydrofuryl group, a dioxolanyl group, a benzoxazol-2-yl group, a tetrahydropyranyl group, a pyrrolidyl group, an imidazolidyl group, a pyrazolidyl group, a thiazolidyl group, an isothiazolidyl group, an oxazolidyl group, an isoxazolidyl group, a piperazyl group, and a morpholinyl group, and preferable is a 5- to 7-membered heterocyclic group.

[0164] R^{12} and R^{13} and R^{22} and R^{23} in the formula (1) are each optionally joined together to form a ring, meaning that R^{12} and R^{13} and R^{22} and R^{23} are each optionally joined together to form a ring together with a nitrogen atom, a carbon atom, or an oxygen atom to be connected.

[0165] Examples of the ring that can be formed by joining R^{12} and R^{13} and R^{22} and R^{23} in the formula (1) together include a cyclopentane ring, a cyclopentene ring, a benzene ring, a piperidine ring, a morpholine ring, a lactone ring, and a lactam ring, and preferable is a 5- to 7-membered ring.

[0166] Examples of the halogen atom that R¹¹, R¹², R¹³, R²¹, R²², and R²³ in the formula (1) may have as a substituent include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom.

[0167] R¹ in the formula (1) is preferably R¹¹, more preferably an alkyl group having 1 to 20 carbon atoms, still more preferably an alkyl group having 1 to 10 carbon atoms, and yet still more preferably an alkyl group having 1 to 6.

[0168] An example of the second molecular structure linked to the first molecular structure represented by the formula (1) is a structure represented by the following formula (2). The second molecular structure refers to a molecular structure portion of the oxime compound (1) other than the first molecular structure.

[0169] The bond represented by "*" in the formula (2) is directly bonded to the bond represented by "*" in the formula (1). That is, when the second molecular structure is a structure represented by the formula (2), the benzene ring having "-*" in the formula (2) and the carbonyl group having "-*" in the formula (1) are directly bonded to each other.

[Formula 8]

$$\begin{array}{c}
(R^3)t \\
R^4 \\
(R^2)s
\end{array}$$
(2)

In the formula (2), R^2 and R^3 each independently represent R^{11} , OR^{11} , SR^{11} , COR^{11} , $CONR^{12}R^{13}$, $NR^{12}COR^{11}$, $OCOR^{11}$, $COOR^{11}$, $SCOR^{11}$, $OCSR^{11}$, $COSR^{11}$, $CSOR^{11}$,

[0170] A plurality of R², when present, are optionally the same or different,

a plurality of R³, when present, are optionally the same or different, and

R¹¹, R¹², and R¹³ have the same meaning as described above.

s and t each independently represent an integer of 0 to 4.

L represents a sulfur atom, CR³¹R³², CO, or NR³³.

R³¹, R³², and R³³ each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, or an aralkyl group having 7 to 30 carbon atoms.

[0171] When the group represented by R^{31} , R^{32} , or R^{33} has an alkyl moiety, the alkyl moiety is optionally branched or cyclic, and R^{31} , R^{32} , and R^{33} are each independently and optionally joined together to form a ring with any of adjacent benzene rings.

[0172] R⁴ is a hydroxy group, a carboxy group, or the following formula (2-1): [Formula 9]

$$(R^{4a})_{v}-L^{2}-L^{1}-$$
 (2-1)

⁵⁵ (In the formula (2-1), L¹ represents -O-, -S-, -NR²²-, -NR²²CO-, -SO₂-, -CS-, -OCO-, or -COO-.

[0173] R²² has the same meaning as described above.

[0174] L² represents a group obtained by removing v hydrogen atoms from an alkyl group having 1 to 20 carbon atoms, a group obtained by removing v hydrogen atoms from an aryl group having 6 to 30 carbon atoms, a group obtained by

removing v hydrogen atoms from an aralkyl group having 7 to 30 carbon atoms, or a group obtained by removing v hydrogen atoms from a heterocyclic group having 2 to 20 carbon atoms.

[0175] When the group represented by L² has an alkylene moiety, the alkylene moiety is optionally interrupted one to five times by -O-, -S-, -COO-, -OCO-, -NR²²-, -NR²²COO-, -OCONR²²-, -SCO-, -COS-, -OCS-, or -CSO-, and the alkylene moiety is optionally branched or cyclic.

[0176] R^{4a} is each independently OR⁴¹, SR⁴¹, CONR⁴²R⁴³, NR⁴²COR⁴³, OCOR⁴¹, COOR⁴¹, SCOR⁴¹, OCSR⁴¹, CSOR⁴¹, CSOR⁴¹, CN, or a halogen atom.

[0177] A plurality of R^{4a} , when present, are optionally the same or different.

[0178] R⁴¹, R⁴², and R⁴³ each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, or an aralkyl group having 7 to 30 carbon atoms, and when the group represented by R⁴¹, R⁴², and R⁴³ has an alkyl moiety, the alkyl moiety is optionally branched or cyclic, and R⁴² and R⁴³ are optionally joined together to form a ring.

[0179] v represents an integer of 1 to 3.)

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[0180] Examples of alkyl groups having 1 to 20 carbon atoms, aryl groups having 6 to 30 carbon atoms, and aralkyl groups having 7 to 30 carbon atoms represented by R¹¹, R¹², R¹³, R²¹, R²², R²³, R²⁴, R³¹, R³², and R³³ in the formula (2) and R²², R⁴¹, R⁴², and R⁴³ in the formula (2-1) are the same as examples for R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1).

[0181] Examples of heterocyclic groups having 2 to 20 carbon atoms represented by R^{11} , R^{12} , R^{13} , R^{21} , R^{22} , and R^{24} in the formula (2) and R^{22} in the formula (2-1) are the same as examples for R^{11} , R^{12} , R^{13} , R^{21} , R^{22} , R^{23} , and R^{24} in the formula (1).

[0182] R^{31} , R^{32} , and R^{33} in the formula (2) are each independently and optionally joined together to form a ring with any of adjacent benzene rings, meaning that R^{31} , R^{32} , and R^{33} are each independently and optionally joined together to form a ring with any of adjacent benzene rings together with a nitrogen atom to be connected.

[0183] Examples of the ring which can be formed by joining R^{31} , R^{32} , and R^{33} in the formula (2) together with any of adjacent benzene rings are the same as those of the ring which can be formed by joining R^{31} and R^{31} and R^{32} and R^{32} in the formula (1) together.

[0184] L² in the above formula (2-1) represents groups obtained by removing v hydrogen atoms from an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an aralkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0185] Examples of the group obtained by removing v hydrogen atoms from an alkyl group having 1 to 20 carbon atoms when v is 1 include alkylene groups such as a methylene group, an ethylene group, a propylene group, a methylene group, a 1-methylpropylene group, a 2-methylpropylene group, a 1,2-dimethylpropylene group, a 3-methylbutylene group, a 3-methylbutylene group, a 4-methylbutylene group, a 2,4-dimethylbutylene group, a 1,3-dimethylbutylene group, a hexylene group, a heptylene group, an octylene group, a nonylene group, a decylene group, a dodecylene group, a tridecylene group, a tetradecylene group, a pentadecylene group, an ethane-1,1-diyl group, and a propane-2,2-diyl group.

[0186] Examples of the group obtained by removing v hydrogen atoms from an aryl group having 6 to 30 carbon atoms when v is 1 include arylene groups such as a 1,2-phenylene group, 1,3-phenylene group, and a 1,4-phenylene group, a 2,6-naphthylene group, a 1,4-naphthylene group, a 2,5-dimethyl-1,4-phenylene group, a diphenylmethane-4,4'-diyl group, a 2,2-diphenylpropane-4,4'-diyl group, a diphenylsulfide-4,4'-diyl group, and a diphenylsulfon-4,4'-diyl group.

[0187] Examples of the group obtained by removing v hydrogen atoms from an aralkyl group having 7 to 30 carbon atoms when v is 1 include a group represented by the following formula (a) and a group represented by the following formula (b).

[Formula 10]

$$-L^{3}$$
 (a) $-L^{5}$ (b)

[0188] [In the formulas (a) and (b), L^3 and L^5 represent an alkylene group having 1 to 10 carbon atoms, and L^4 and L^6 represent a single bond or an alkylene group having 1 to 10 carbon atoms.]

[0189] Examples of the alkylene group having 1 to 10 carbon atoms include alkylene groups such as a methylene group, an ethylene group, a propylene group, a methylethylene group, a butylene group, a 1-methylpropylene group, a 2-methylpropylene group, a 1,3-dimethylpropylene group, a 1-methylbutylene group, a 2-methylbutylene group, a 3-methylbutylene group, a 4-methylbutylene group, a 2,4-dimethylbutylene group, a 1,3-dimethylbutylene group, a pentylene group, a hexylene group, a heptylene group, an octylene group, a nonylene group,

and a decylene group.

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[0190] Examples of the group obtained by removing v hydrogen atoms from a heterocyclic group having 2 to 20 carbon atoms when v is 1 include divalent heterocyclic groups such as a 2,5-pyridinediyl group, a 2,6-pyridinediyl group, a 2,5-pyridinediyl group, a 2,5-thiophenediyl group, a 3,4-tetrahydrofurandiyl group, a 2,5-tetrahydrofurandiyl group, a 2,5-tetrahydrofurandiyl group, a 2,5-benzothiophenediyl group, an N-methylindole-2,5-diyl group, a 2,5-benzothiazolediyl group, and a 2,5-benzoxazolediyl group.

[0191] Examples of the halogen atom represented by R^2 and R^3 in the formula (2) and R^{4a} in the formula (2-1) include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom.

[0192] From the viewpoint of solubility in the solvent (F) and/or the development speed of the cured film, a preferable example of the structure represented by the formula (2) is the structure represented by the following formula (2a):

[Formula 11]

[0193] [In the formula (2a), L' represents a sulfur atom or NR 50 , R 50 represents a linear, branched, or cyclic alkyl group having 1 to 20 carbon atoms, and R 2 , R 3 , R 4 , s, and t have the same meaning as described above.]

[0194] From the same viewpoint as above, another preferable example of the structure represented by the formula (2) is the structure represented by the following formula (2b):

[Formula 12]

[0195] [In the formula (2b), R⁴⁴ represents a hydroxy group, a carboxy group, or a group represented by the following formula (2-2):

[Formula 13]

$$R^{44a}-L^{12}-L^{11}$$
 (2-2)

(In the formula (2-2), L¹¹ represents -O- or *-OCO-,

* represents a bond to L¹², L¹² represents an alkylene group having 1 to 20 carbon atoms, the alkylene group may be interrupted by one to three -O-, R^{44a} represents OR⁵⁵ or COOR⁵⁵, R⁵⁵ represents a hydrogen atom or an alkyl group having 1 to 6 carbon atoms.).]

[0196] R^{44} is preferably a group represented by the formula (2-2). In this case, it is advantageous in terms of the solubility of the oxime compound (1) in the solvent (F) and the development speed of the cured film.

[0197] The number of carbon atoms of the alkylene group represented by L^{12} is preferably 1 to 10, and more preferably 1 to 4.

[0198] R^{44a} is preferably a hydroxy group or a carboxy group, and more preferably a hydroxy group.

[0199] The method for producing the oxime compound (1) having the second molecular structure represented by the formula (2) is not particularly limited, but the oxime compound (1) can be produced by the method disclosed in Japanese Patent Laid-Open No. 2011-132215.

[0200] Another example of the second molecular structure linked to the first molecular structure represented by the formula (1) is a structure represented by the following formula (3).

[0201] The bond represented by "*" in the formula (3) is directly bonded to the bond represented by "*" in the formula (1). That is, when the second molecular structure is a structure represented by the formula (3), the benzene ring having "-*" in the formula (3) and the carbonyl group having "-*" in the formula (1) are directly bonded to each other.

[Formula 14]

 R^7 R^8 R^9 R^5 R^5 R^8 R^9 R^5

[0202] In the formula (3), R⁵ represents a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylalkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0203] When the group represented by R⁵ has an alkyl moiety, the alkyl moiety is optionally branched or cyclic.

[0204] The hydrogen atoms of the group represented by R^5 are optionally replaced by R^{21} , OR^{21} , COR^{21} , SR^{21} , $NR^{22}R^{23}$, $CONR^{22}R^{23}$, $-NR^{22}-OR^{23}$, $-N(COR^{22})-OCOR^{23}$, $NR^{22}COR^{21}$, $OCOR^{21}$, COR^{21} , -C (=N-OCOR²¹)-R²², $SCOR^{21}$, $OCSR^{21}$, $COSR^{21}$, $CSOR^{21}$, a hydroxyl group, a nitro group, CN, a halogen atom, or $COOR^{21}$. **[0205]** R^{21} , R^{22} , and R^{23} have the same meaning as described above.

[0206] The hydrogen atoms of the group represented by R²¹, R²², or R²³ are optionally replaced by CN, a halogen atom, a hydroxy group, or a carboxy group.

[0207] When the group represented by R^{21} , R^{22} , and R^{23} has an alkylene moiety, the alkylene moiety is optionally interrupted one to five times by -O-, -S-, -COO-, -OCO-, -NR²⁴-, -NR²⁴CO-, -NR²⁴CO-, -OCONR²⁴-, -SCO-, -COS-, -OCS-, or -CSO-.

[0208] R²⁴ has the same meaning as described above.

[0209] When the group represented by R^{21} , R^{22} , and R^{23} has an alkyl moiety, the alkyl moiety is optionally branched or cyclic, and R^{22} and R^{23} are optionally joined together to form a ring.

[0210] R^6 , R^7 , R^8 , and R^9 are each independently R^{61} , OR^{61} , SR^{61} , COR^{62} , $CONR^{63}R^{64}$, $NR^{65}COR^{61}$, $OCOR^{61}$, $COOR^{62}$, $SCOR^{61}$, $OCSR^{61}$, $OCSR^{62}$, $CSOR^{61}$, a hydroxy group, a nitro group, CN, or a halogen atom.

[0211] R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylalkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0212] The hydrogen atoms of the group represented by R^{61} , R^{62} , R^{63} , R^{64} , or R^{65} are optionally replaced by OR^{21} , COR^{21} , SR^{21} , $NR^{22}Ra^{23}$, $CONR^{22}R^{23}$, $-NR^{22}-OR^{23}$, $-N(COR^{22})-OCOR^{23}$, -C (= NOR^{21}) $-R^{22}$, -C(= NOR^{21}) -C(=N

[0213] R⁶ and R⁷, R⁷ and R⁸, and R⁸ and R⁹ are each optionally joined together to form a ring.

* represents a bond with the first molecular structure of the oxime compound (1).

[0214] Examples of alkyl groups having 1 to 20 carbon atoms, aryl groups having 6 to 30 carbon atoms, aralkyl groups having 7 to 30 carbon atoms, and heterocyclic groups having 2 to 20 carbon atoms represented by R⁵, R²¹, R²², R²³, R²⁴, R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ in the formula (3) are the same as examples for R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1).

[0215] R²² and R²³ in the formula (3) are each optionally joined together to form a ring, meaning that R²³ are each optionally joined together to form a ring together with a nitrogen atom, a carbon atom, or an oxygen atom to be connected.

[0216] Examples of the ring which can be formed by joining R^{22} and R^{23} , and R^{23} in the formula (3) together are the same as those of the ring which can be formed by R^{12} and R^{13} , and R^{22} and R^{23} in the formula (1).

[0217] Examples of the halogen atom represented by R^6 , R^7 , R^8 , and R^9 , and the halogen atoms which may replace the hydrogen atoms of R^5 , R^{21} , R^{22} , R^{23} , R^{61} , R^{62} , R^{63} , R^{64} , and R^{65} in the formula (3) include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom.

[0218] From the viewpoint of solubility in the solvent (F) and/or the development speed of the cured film, in one preferable embodiment, R⁵ is a group represented by the following formula (3-1):

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[Formula 15]

Z N C—R²² (3-1)

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[0219] [In the formula (3-1), Z represents a group obtained by removing one hydrogen atom from an alkyl group having 1 to 20 carbon atoms, a group obtained by removing one hydrogen atom from an aryl group having 6 to 30 carbon atoms, a group obtained by removing one hydrogen atom from an aralkyl group having 7 to 30 carbon atoms, or a group obtained by removing one hydrogen atom from a heterocyclic group having 2 to 20 carbon atoms.

[0220] When the group represented by Z has an alkylene moiety, the alkylene moiety is optionally interrupted one to five times by -O-, -S-, -COO-, -OCO-, -NR²⁴-, -NR²⁴COO-, -OCONR²⁴-, -SCO-, -COS-, -OCS-, or -CSO-, and the alkylene moiety is optionally branched or cyclic.

[0221] R^{21} , R^{22} , and R^{24} have the same meaning as described above.]

[0222] From the same viewpoint as above, Z in the formula (3-1) is preferably a methylene group, an ethylene or phenylene group.

[0223] From the same viewpoint as above, R^{21} and R^{22} in the formula (3-1) are preferably an alkyl group having 1 to 20 carbon atoms or an aryl group having 6 to 30 carbon atoms, and more preferably a methyl group, an ethyl group, or a phenyl group.

[0224] From the same viewpoint as above, in another preferred embodiment, R⁷ is a nitro group.

[0225] The method for producing the oxime compound (1) having the second molecular structure represented by the formula (3) is not particularly limited, but the oxime compound (1) can be produced by the method disclosed in Japanese Patent Laid-Open No. 2000-80068 and Japanese Patent Laid-Open No. 2011-178776.

[0226] Yet another example of the second molecular structure linked to the first molecular structure represented by the formula (1) is a structure represented by the following formula (4).

[0227] The bond represented by "*" in the formula (4) is directly bonded to the bond represented by "*" in the formula (1). That is, when the second molecular structure is a structure represented by the formula (4), the benzene ring having "-*" in the formula (4) and the carbonyl group having "-*" in the formula (1) are directly bonded to each other.

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[Formula 16]

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$$R^{73}$$
 R^{74}
 R^{74}
 R^{74}
 R^{74}
 R^{74}

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[0228] In the formula (4), R⁷¹ represents a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an aralkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms. When the group represented by R⁷¹ has an alkyl moiety, the alkyl moiety is optionally branched or cyclic. [0229] The hydrogen atoms of the group represented by R⁷¹ are optionally replaced by R²¹, OR²¹, COR²¹, SR²¹, NR²²R²³, CONR²²R²³, -NR²²-OR²³, -N(COR²²)-OCOR²³, NR²²COR²¹, OCOR²¹, COOR²¹, -C(=N-OR²¹)-R²², -C(=N-OCOR²¹)-R²², SCOR²¹, OCSR²¹, COSR²¹, CSOR²¹, a hydroxyl group, a nitro group, CN, a halogen atom, or COOR²¹. [0230] R²¹, R²², and R²³ have the same meaning as described above.

[0231] The hydrogen atoms of the group represented by R²¹, R²², or R²³ are optionally replaced by CN, a halogen atom, a hydroxy group, or a carboxy group.

[0232] When the group represented by R^{21} , R^{22} , and R^{23} has an alkylene moiety, the alkylene moiety is optionally interrupted one to five times by -O-, -S-, -COO-, -OCO-, -NR²⁴-, -NR²⁴CO-, -NR²⁴CO-, -OCONR²⁴-, -SCO-, -COS-, -OCS-, or -CSO-.

[0233] R²⁴ has the same meaning as described above.

[0234] When the group represented by R^{21} , R^{22} , and R^{23} has an alkyl moiety, the alkyl moiety is optionally branched or cyclic, and R^{22} and R^{23} are optionally joined together to form a ring.

[0235] R⁷², R⁷³, and three R⁷⁴ each independently represent R⁶¹, OR⁶¹, SR⁶¹, COR⁶², CONR⁶³R⁶⁴, NR⁶⁵COR⁶¹, OCOR⁶¹, COOR⁶², SCOR⁶¹, OCSR⁶¹, COSR⁶², CSOR⁶¹, a hydroxy group, a nitro group, CN, or a halogen atom.

[0236] R^{61} , R^{62} , R^{63} , R^{64} , and R^{65} each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylalkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0237] The hydrogen atoms of the group represented by R^{61} , R^{62} , R^{63} , R^{64} , or R^{65} are optionally replaced by OR^{21} , COR^{21} , SR^{21} , $NR^{22}Ra^{23}$, $COR^{22}R^{23}$, $-N(COR^{22})-OCOR^{23}$, $-C(=NOR^{21})-R^{22}$, $-C(=N-OCOR^{21})-R^{22}$, -C(=N-O

[0238] R^{72} and R^{73} and two R^{74} are each optionally joined together to form a ring.

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* represents a bond with the first molecular structure of the oxime compound (1).

[0239] Examples of alkyl groups having 1 to 20 carbon atoms, aryl groups having 6 to 30 carbon atoms, aralkyl groups having 7 to 30 carbon atoms, and heterocyclic groups having 2 to 20 carbon atoms represented by R^{71} , R^{21} , R^{22} , R^{23} , R^{24} , R^{61} , R^{62} , R^{63} , R^{64} , and R^{65} in the formula (4) are the same as examples for R^{11} , R^{12} , R^{13} , R^{21} , R^{22} , R^{23} , and R^{24} in the formula (1).

[0240] R^{22} and R^{23} in the formula (4) are each optionally joined together to form a ring, meaning that R^{22} and R^{23} are each optionally joined together to form a ring together with a nitrogen atom, a carbon atom, or an oxygen atom to be connected

[0241] Examples of the ring which can be formed by joining R²² and R²³ in the formula (4) together are the same as those of the ring which can be formed by joining Ra¹² and Ra¹³, and Ra²³ and Ra²³ in the formula (1) together.

[0242] Examples of the halogen atom represented by R^{72} , R^{73} , and R^{74} , and the halogen atoms which may replace the hydrogen atoms of R^{71} , R^{21} , R^{22} , R^{23} , R^{61} , R^{62} , R^{63} , R^{64} , and R^{65} in the formula (4) include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom.

[0243] The method for producing the oxime compound (1) having the second molecular structure represented by the formula (4) is not particularly limited, but the oxime compound (1) can be produced by the method disclosed in International Publication No. 2017/051680 and International Publication No. 2020/004601.

[0244] Yet another example of the second molecular structure linked to the first molecular structure represented by the formula (1) is a structure represented by the following formula (5).

[0245] The bond represented by "*" in the formula (5) is directly bonded to the bond represented by "*" in the formula (1). That is, when the second molecular structure is a structure represented by the formula (5), the pyrrole ring having "-*" in the formula (5) and the carbonyl group having "-*" in the formula (1) are directly bonded to each other.

[Formula 17]

 R^{85} R^{86} R^{81} R^{85} R^{84} R^{83} R^{82} R^{82} R^{83}

[0246] In the formula (5), R⁸¹ represents a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an aralkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0247] When the group represented by R81 has an alkyl moiety, the alkyl moiety is optionally branched or cyclic.

[0248] The hydrogen atoms of the group represented by R⁸¹ are optionally replaced by R²¹, OR²¹, COR²¹, SR²¹, NR²²R²³, CONR²²R²³, -NR²²-OR²³, -N(COR²²)-OCOR²³, NR²²COR²¹, OCOR²¹, COOR²¹, -C(=N-OR²¹)-R²², SCOR²¹, OCSR²¹, COSR²¹, CSOR²¹, a hydroxyl group, a nitro group, CN, a halogen atom, or COOR²¹. [0249] R²¹, R²², and R²³ have the same meaning as described above.

[0250] The hydrogen atoms of the group represented by R²¹, R²², or R²³ are optionally replaced by CN, a halogen atom, a hydroxy group, or a carboxy group.

[0251] When the group represented by R²¹, R²², and R²³ has an alkylene moiety, the alkylene moiety is optionally interrupted one to five times by -O-, -S-, -COO-, -OCO-, -NR²⁴-, -NR²⁴CO-, -NR²⁴CO-, -OCONR²⁴-, -SCO-, -COS-, -OCS-, or -CSO-.

[0252] R²⁴ has the same meaning as described above.

[0253] When the group represented by R^{21} , R^{22} , and R^{23} has an alkyl moiety, the alkyl moiety is optionally branched or cyclic, and R^{22} and R^{23} are optionally joined together to form a ring.

[0254] R⁸², R⁸³, R⁸⁴, R⁸⁵, and R⁸⁶ are each independently R⁶¹, OR⁶¹, SR⁶¹, COR⁶², CONR⁶³R⁶⁴, NR⁶⁵COR⁶¹, OCOR⁶¹, COOR⁶², SCOR⁶¹, OCSR⁶¹, COSR⁶², CSOR⁶¹, a hydroxy group, a nitro group, CN, or a halogen atom.

[0255] R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylalkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0256] The hydrogen atoms of the group represented by R^{61} , R^{62} , R^{63} , R^{64} , or R^{65} are optionally replaced by OR^{21} , COR^{21} , SR^{21} , $NR^{22}Ra^{23}$, $CONR^{22}R^{23}$, $-NR^{22}-OR^{23}$, $-N(COR^{22})-OCOR^{23}$, -C (= NOR^{21}) $-R^{22}$, -C(= NOR^{21})- R^{22} , -C(= NOR^{21})- $-R^{22}$, -C(= NOR^{21})--C(= NOR^{21})--C

[0257] R^{83} and R^{84} , R^{84} and R^{85} , and R^{85} and R^{86} are each optionally joined together to form a ring.

* represents a bond with the first molecular structure of the oxime compound (1).

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[0258] Examples of alkyl groups having 1 to 20 carbon atoms, aryl groups having 6 to 30 carbon atoms, aralkyl groups having 7 to 30 carbon atoms, and heterocyclic groups having 2 to 20 carbon atoms represented by R⁸¹, R²¹, R²², R²³, R²⁴, R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ in the formula (5) are the same as examples for R¹¹, R¹², R¹³, R²¹, R²², R²³, and R²⁴ in the formula (1).

[0259] R²² and R²³ in the formula (5) are each optionally joined together to form a ring, meaning that R²² and R²³ are each optionally joined together to form a ring together with a nitrogen atom, a carbon atom, or an oxygen atom to be connected

[0260] Examples of the ring which can be formed by joining R^{22} and R^{23} in the formula (5) together are the same as those of the ring which can be formed by joining R^{12} and R^{13} , and R^{23} in the formula (1) together.

[0261] Examples of the halogen atom represented by R⁸², R⁸³, R⁸⁴, R⁸⁵, and R⁸⁶, and the halogen atoms which may replace the hydrogen atoms of R⁸¹, R²¹, R²², R²³, R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ in the formula (5) include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom.

[0262] The method for producing the oxime compound (1) having the second molecular structure represented by the formula (5) is not particularly limited, but the oxime compound (1) can be produced by the method disclosed in International Publication No. 2017/051680 and International Publication No. 2020/004601.

[0263] Yet another example of the second molecular structure linked to the first molecular structure represented by the formula (1) is a structure represented by the following formula (6).

[0264] The bond represented by "*" in the formula (6) is directly bonded to the bond represented by "*" in the formula (1). That is, when the second molecular structure is a structure represented by the formula (6), the benzene ring having "-*" in the formula (6) and the carbonyl group having "-*" in the formula (1) are directly bonded to each other.

[Formula 18]

$$R^{96}$$
 R^{97}
 R^{92}
 R^{93}
 R^{93}

[0265] In the formula (6), four R⁹¹, R⁹², R⁹³, R⁹⁴, R⁹⁵, R⁹⁶, and R⁹⁷ are each independently R⁶¹, OR⁶¹, SR⁶¹, COR⁶², CONR⁶³R⁶⁴, NR⁶⁵COR⁶¹, OCOR⁶¹, COOR⁶², SCOR⁶¹, OCSR⁶¹, COSR⁶², CSOR⁶¹, a hydroxy group, a nitro group, CN, or a halogen atom.

[0266] R^{61} , R^{62} , R^{63} , R^{64} , and R^{65} each independently represent a hydrogen atom, an alkyl group having 1 to 20 carbon atoms, an aryl group having 6 to 30 carbon atoms, an arylalkyl group having 7 to 30 carbon atoms, or a heterocyclic group having 2 to 20 carbon atoms.

[0267] The hydrogen atoms of the group represented by R⁶¹, R⁶², R⁶³, R⁶⁴, or R⁶⁵ are optionally replaced by OR²¹, COR²¹, SR²¹, NR²²Ra²³, CONR²²R²³, -NR²²-OR²³, -N(COR²²)-OCOR²³, -C (=NOR²¹) -R²², -C(=N-OCOR²¹)-R²², CN, a halogen atom, or COOR²¹.

[0268] R²¹, R²², and R²³ have the same meaning as described above.

[0269] R⁹² and R⁹³, R⁹⁴ and R⁹⁵, R⁹⁵ and R⁹⁶, and R⁹⁶ and R⁹⁷ are each optionally joined together to form a ring.

* represents a bond with the first molecular structure of the oxime compound (1).

[0270] Examples of alkyl groups having 1 to 20 carbon atoms, aryl groups having 6 to 30 carbon atoms, aralkyl groups having 7 to 30 carbon atoms, and heterocyclic groups having 2 to 20 carbon atoms represented by R²¹, R²², R²³, R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ in the formula (6) are the same as examples for R¹¹, R¹², R¹³, R²¹, R²², and R²³ in the formula (1). **[0271]** R²² and R²³ in the formula (6) are each optionally joined together to form a ring, meaning that R²² and R²³ are each optionally joined together to form a ring together with a nitrogen atom, a carbon atom, or an oxygen atom to be connected.

[0272] Examples of the ring which can be formed by joining R^{22} and R^{23} in the formula (6) together are the same as those of the ring which can be formed by joining R^{32} , and R^{33} and R^{32} and R^{33} in the formula (1) together.

[0273] Examples of the halogen atom represented by R⁹¹, R⁹², R⁹³, R⁹⁴, R⁹⁵, R⁹⁶, and R⁹⁷, and the halogen atoms which may replace the hydrogen atoms of R²¹, R²², R²³, R⁶¹, R⁶², R⁶³, R⁶⁴, and R⁶⁵ in the formula (6) include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom.

[0274] The method for producing the oxime compound (1) having the second molecular structure represented by the formula (6) is not particularly limited, but the oxime compound (1) can be produced by the method disclosed in International Publication No. 2017/051680 and International Publication No. 2020/004601.

[0275] The photopolymerization initiator (D) may contain an oxime compound other than the oxime compound (1).

[0276] Examples of the oxime compound other than the oxime compound (1) include an oxime compound having a partial structure represented by the following formula (d1). * represents a bond.

[Formula 19]

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[0277] Examples of the oxime compound having a partial structure represented by the formula (d1) include N-benzoyloxy-1-(4-phenylsulfanylphenyl)butan-1-one-2-imine, N-benzoyloxy-1-(4-phenylsulfanylphenyl)octan-1-one-2-imine, N-benzoyloxy-1-(4-phenylsulfanylphenyl)-3-cyclopentylpropan-1-one-2-imine, N-acetoxy-1-[9-ethyl-6-(2-methylbenzoyl)-9H-carbazole-3-yl]ethane-1-imine, N-acetoxy-1-[9-ethyl-6-(2-methyl-4-(3,3-dimethyl-2,4-dioxacyclopentanylmethyloxy)benzoyl}-9H-carbazole-3-yl]ethane-1-imine, N-acetoxy-1-[9-ethyl-6-(2-methylbenzoyl)-9H-carbazole-3-yl]-3-cyclopentylpropane-1-imine, and N-benzoyloxy-1-[9-ethyl-6-(2-methylbenzoyl)-9H-carbazole-3-yl]-3-cyclopentylpropan-1-one-2-imine, and compounds disclosed in Japanese Patent Laid-Open No. 2011-132215, International Publication No. 2008/78678, International Publication No. 2008/78686, and International Publication No. 2012/132558. Commercially available products such as Irgacure OXE01 and Irgacure OXE02 (all manufactured by BASF Japan Ltd.), and N-1919 (manufactured by ADEKA Corporation) may also be used.

[0278] Of these, the oxime compound having a partial structure represented by the formula (d1) is preferably at least one selected from the group consisting of N-benzoyloxy-1-(4-phenylsulfanylphenyl)butan-1-one-2-imine, N-benzoyloxy-1-(4-phenylsulfanylphenyl)-3-cyclopentylpropan-1-one-2-imine, and M-benzoyloxy-1-(4-phenylsulfanylphenyl)octan-1-one-2-imine.

[0279] Examples of the biimidazole compound include a compound represented by the formula (d5):

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[In the formula (d5), RE to RJ represent an aryl group having 6 to 10 carbon atoms which may have a substituent.]

[0280] Examples of the aryl group having 6 to 10 carbon atoms include a phenyl group, a toluyl group, a xylyl group, an ethylphenyl group, and a naphthyl group, and preferable is a phenyl group.

[0281] Examples of the substituent include a halogen atom and an alkoxy group having 1 to 4 carbon atoms. Examples of the halogen atom include a fluorine atom, a chlorine atom, a bromine atom and an iodine atom, and preferable is a chlorine atom. Examples of the alkoxy group having 1 to 4 carbon atoms include a methoxy group, an ethoxy group, a propoxy group, and a butoxy group, and preferable is a methoxy group.

[0282] Examples of the biimidazole compound include 2,2'-bis(2-chlorophenyl)-4,4',5,5'-tetraphenylbiimidazole, 2,2'-bis(2,3-dichlorophenyl)-4,4',5,5'-tetraphenylbiimidazole (for example, see Japanese Patent Laid-Open No. 06-75372 and Japanese Patent Laid-Open No. 06-75373), 2,2'-bis(2-chlorophenyl)-4,4',5,5'-tetraphenyl biimidazole, 2,2'-bis(2-chlorophenyl)-4,4',5,5'-tetra(dialkoxyphenyl)biimidazole, 2,2'-bis(2-chlorophenyl)-4,4',5,5'-tetra(dialkoxyphenyl)biimidazole, 2,2'-bis(2-chlorophenyl)-4,4',5,5'-tetra(dialkoxyphenyl)biimidazole (for example, see Japanese Patent No. 48-38403 and Japanese Patent Laid-Open No. 62-174204), and an imidazole compound in which phenyl groups at the 4,4',5,5' positions are each substituted with a carboalkoxy group (for example, see Japanese Patent Laid-Open No. 7-10913). Of these, a compound represented by the following formula or a mixture thereof is preferable.

[Formula 21]

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[0283] Examples of the triazine compound include 2,4-bis(trichloromethyl)-6-(4-methoxyphenyl)-1,3,5-triazine, 2,4-bis(trichloromethyl)-6-(4-methoxynaphthyl)-1,3,5-triazine, 2,4-bis(trichloromethyl)-6-piperonyl-1,3,5-triazine, 2,4-bis(trichloromethyl)-6-[2-(5-methylfuran-2-yl)ethenyl]-1,3,5-triazine, 2,4-bis(trichloromethyl)-6-[2-(4-diethylamino-2-methylphenyl)-6-[2-(4-diethylamino-2-methylphenyl)-1,3,5-triazine, 2,4-bis(trichloromethyl)-6-[2-(3,4-dimethoxyphenyl)ethenyl]-1,3,5-triazine. Of these, 2,4-bis(trichloromethyl)-6-piperonyl-1,3,5-triazine is preferable.

[0284] Examples of the acylphosphine compound include bis(2,4,6-trimethylbenzoyl)phenylphosphine oxide and (2,4,6-trimethylbenzoyl)diphenylphosphine oxide.

[0285] Only one kind of the photopolymerization initiator described above may be used alone, or two or more kinds of photopolymerization initiators may be used in combination. When two or more photopolymerization initiators are used in combination, may be used in combination with other photopolymerization initiators other than the oxime compound, biimidazole compound, triazine compound, and acylphosphine compound described above may be used in combination. **[0286]** Examples of the other photopolymerization initiator include benzoin compounds such as benzoin, benzoin methyl ether, benzoin ethyl ether, benzoin isopropyl ether, and benzoin isobutyl ether; benzophenone compounds such as benzophenone, methyl o-benzoylbenzoate, 4-phenylbenzophenone, 4-benzoyl-4'-methyldiphenyl sulfide, 3,3',4,4'-tetra(tert-butylperoxycarbonyl)benzophenone, 2,4,6-trimethylbenzophenone, and 4,4'-bis(diethylamino)benzophenone; quinone compounds such as 9,10-phenanthrene quinone, 2-ethylanthraquinone, and camphorquinone; and 10-butyl-2-chloroacridone, benzyl, methyl phenylglyoxylate, and a titanocene compound.

[0287] Example of the combination of two or more kinds of photopolymerization initiators include a combination of an oxime compound and a biimidazole compound, an oxime compound and a triazine compound, an oxime compound and an acylphosphine compound, a biimidazole compound and a triazine compound, a combination of a biimidazole compound and an acylphosphine compound, and a combination of a triazine compound and an acylphosphine compound. [0288] The content of the photopolymerization initiator (D) in the curable resin composition is preferably 0.1 parts by mass or more and 300 parts by mass or less, and more preferably 0.1 parts by mass or more and 200 parts by mass or less, based on 100 parts by mass of the photopolymerizable compound (C). Further, the content of the photopolymerization initiator (D) in the curable resin composition is preferably 0.1 parts by mass or more and 30 parts by mass or less, and more preferably 1 part by mass or more and 20 parts by mass or less, based on 100 parts by mass of the total amount of the resin (B) and the photopolymerizable compound (C). When the content of the photopolymerization initiator (D) is within the above range, the curable resin composition tends to have high sensitivity and the exposure time tends to be shortened, so that the productivity of the cured film tends to improve.

[6] Photopolymerization initiation aid (D1)

[0289] The curable resin composition may further contain a photopolymerization initiation aid (D1) in combination with a photopolymerization initiator (D). The photopolymerization initiation aid (D1) is a compound or a sensitizer used to promote the polymerization of the photopolymerizable compound (C) initiated by the photopolymerization initiator (D). Examples of the photopolymerization initiation aid (D1) include amine compounds, alkoxyanthracene compounds, thioxanthone compounds, and carboxylic acid compounds.

[0290] Examples of the amine compound include triethanolamine, methyldiethanolamine, triisopropanolamine, methyl 4-dimethylaminobenzoate, ethyl 4-dimethylaminobenzoate, isoamyl 4-dimethylaminobenzoate, 2-dimethylaminobenzoate, 2-ethylhexyl 4-dimethylaminobenzoate, N,N-dimethylparatoluidine, 4,4'-bis(dimethylamino)benzophenone (common name: Michler's ketone), 4,4'-bis(diethylamino)benzophenone, and 4,4'-bis(ethylamino)benzophenone, and of these, 4,4'-bis(diethylamino)benzophenone is preferable. Commercially available products such as EAB-F (manufactured by Hodogaya Chemical Co., Ltd.), may be used.

[0291] Examples of the alkoxy anthracene compound include 9,10-dimethoxyanthracene, 2-ethyl-9,10-dimethoxyanthracene, 9,10-diethoxyanthracene, 2-ethyl-9,10-diethoxyanthracene, 9,10-dibutoxyanthracene, and 2-ethyl-9,10-dibutoxyanthracene.

[0292] Examples of the thioxanthone compound include 2-isopropylthioxanthone, 4-isopropylthioxanthone, 2,4-diethylthioxanthone, 2,4-dichlorothioxanthone, and 1-chloro-4-propoxythioxanthone.

[0293] Examples of the carboxylic acid compound include phenylsulfanylacetic acid, methylphenylsulfanylacetic acid, ethylphenylsulfanylacetic acid, methylphenylsulfanylacetic acid, dimethylphenylsulfanylacetic acid, methoxyphenysulfanylacetic acid, dimethoxyphenylsulfanylacetic acid, chlorophenylsulfanylacetic acid, dichlorophenylsulfanylacetic acid, N-phenylglycine, phenoxyacetic acid, naphthylthioacetic acid, N-naphthylglycine, and naphthoxyacetic acid. [0294] When the curable resin composition contains a photopolymerization initiation aid (D1), the content of the photopolymerization initiation aid (D1) in the curable resin composition is preferably 0.1 parts by mass or more and 300 parts by mass or less, and more preferably 0.1 parts by mass or more and 200 parts by mass or less, based on 100 parts by mass or less, composition is preferably 0.1 parts by mass or more and 30 parts by mass or less, and more preferably 1 part by mass or more and 20 parts by mass or less, based on 100 parts by mass of the total amount of the resin (B) and the photopolymerizable compound (C). When the content of the photopolymerization initiation aid (D1) is within the above range, the sensitivity of the curable resin composition can be further increased.

[7] Antioxidant (E)

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[0295] The antioxidant (E) is not particularly limited as long as it is an antioxidant that is generally used industrially, and a phosphorus-based antioxidant, a phenol-based antioxidant, a phosphorus/phenol composite antioxidant, a sulfur-based antioxidant, and the like can be used. Two or more kinds of the antioxidants (E) may be used in combination. The phosphorus/phenol composite antioxidant may be a compound having one or more phosphorus atoms and one or more phenol structures in its molecule. When the curable resin composition contains the antioxidant (E), the emission intensity, transmittance, and heat-resistant transparency of the obtained cured film can be improved. The antioxidant (E) preferably contains a phosphorus-based antioxidant from the viewpoint of further improving the emission intensity of the obtained cured film and from the viewpoint of being excellent in the effect of suppressing a change in emission intensity.

[0296] Examples of the phosphorus-based antioxidant include Irgafos (R) 168 (Irgafos 168: tris(2,4-di-tert-butylphe-nyl)phosphite, manufactured by BASF Japan Ltd.), Irgafos 12: tris[2-[[2,4,8,10-tetra-tert-butyldibenzo[d,f] [1,3,2]-dioxaphosphine-6-yl]oxy]ethyl]amine, manufactured by BASF Japan Ltd.), Irgafos 38 (Irgafos 38: bis(2,4-bis(1,1-dimethylethyl)-6-methylphenyl)ethyl ester phosphite, BASF Japan Ltd.), ADK STAB (R) 329K, ADK STAB PEP36, and ADK STAB PEP-8 (all manufactured by ADEKA Corporation), Sandstab P-EPQ (manufactured by Clariant AG), Weston (R) 618 and Weston 619G (manufactured by GE), and Ultranox 626 (manufactured by GE).

[0297] Examples of the phenol-based antioxidant include Irganox (R) 1010 (Irganox 1010: pentaerythritol tetrakis[3-(3,5-di-tert-butyl-4-hydroxyphenyl)propionate], manufactured by BASF Japan Ltd.), Irganox 1076 (Irganox 1076: Octadecyl-3-(3,5-di-tert-butyl-4-hydroxyphenyl)propionate, manufactured by BASF Japan Ltd.), Irganox 1330 (Irganox 1330: 3,3',3",5,5',5"-hexa-tert-butyl-a,a',a''-(mesitylene-2,4,6-triyl)tri-p-cresol, BASF Japan Ltd.), Irganox 3114 (Irganox 3114: 1,3,5-tris(3,5-di-tert-butyl-4-hydroxybenzyl)-1,3,5-triazine-2,4,6(1H,3H,5H)-trione, BASF Japan Ltd.), Irganox 3790 (Irganox 3790: 1,3,5-tris((4-tert-butyl-3-hydroxy-2,6-xylyl)methyl)-1,3,5-triazine-2,4,6(1H,3H,5H)-trione, BASF Japan Ltd.), Irganox 1035 (Irganox 1035: thiodiethylenebis[3-(3,5-di-tert-butyl-4 hydroxyphenyl)propionate], manufactured by BASF Japan Ltd.), Irganox 1135: 3,5-bis(1,1-dimethylethyl)-4-hydroxy-C7-C9 side-chain alkyl ester of benzene propanoic acid, manufactured by BASF Japan Ltd.), Irganox 3125 (Irganox 3125 (Irganox 3125, BASF Japan Ltd.), Irganox 565 (Irganox 565: 2,4-bis(n-octylthio)-6-(4-hydroxy-3',5'-di-tert-butylanilino)-1,3,5-triazine, manufactured by BASF Japan Ltd.), ADK STAB

(R) AO-80 (ADK STAB AO-80: 3,9-bis(2-(3-(3-tert-butyl-4-hydroxy-5-methylphenyl)propionyloxy)-1,1-dimethylethyl)-2,4,8,10-tetraoxaspiro(5,5)undecane, manufactured by ADEKA Corporation), SUMILIZER (R) BHT, SUMILIZER GA-80, SUMILIZER GS (manufactured by Sumitomo Chemical Co., Ltd.), Cyanox (R) 1790 (Cyanox 1790, manufactured by Cytec Industries Inc.), and vitamin E (manufactured by Eisai Co., Ltd.).

[0298] Examples of the phosphorus/phenol composite antioxidant include SUMILIZER (R) GP (6-[3-(3-tert-butyl-4-hydroxy-5-methylphenyl)propoxy]-2,4,8,10-tetra-tert-butyldibenz[d,f][1.3.2]dioxaphosphepine) (manufactured by Sumitomo Chemical Co., Ltd.).

[0299] Examples of the sulfur-based antioxidant include dialkyl thiodipropionate compounds such as dilauryl thiodipropionate, dimyristyl thiodipropionate, and distearate thiodipropionate; and β -alkylmercaptopropionic acid ester compounds of polyols such as tetrakis[methylene(3-dodecylthio)propionate]methane.

[0300] The content of the antioxidant (E) in the curable resin composition is, for example, 1 part by mass or more and 50 parts by mass or less based on 100 parts by mass of the resin (B), and is preferably 5 parts by mass or more and 40 parts by mass or less, and more preferably 7 parts by mass or more and 30 parts by mass or less from the viewpoint of the light emission amount and heat resistance of the cured film.

[8] Solvent (F)

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[0301] The solvent (F) is not particularly limited as long as it dissolves the resin (B), the photopolymerizable compound (C), and the photopolymerization initiator (D), and any solvent which has been used conventionally in the art can be used. The solvent (F) preferably has a solubility in water of 0.01% by mass or more and 30% by mass or less at a temperature of 20°C. Within this range, it is easy to prepare a curable resin composition having a water content based on the total mass of 50 ppm or more and 5000 ppm or less.

[0302] Examples of the solvent (F) include an ester solvent (a solvent which contains -COO- but does not contain -O- in its molecule), an ether solvent (a solvent which contains -O- but does not contain-COO- in its molecule), an ether ester solvent (a solvent which contains -COO- and-O- in its molecule), a ketone solvent (a solvent which contains -CO- but does not contain -COO- in its molecule), an alcohol solvent (a solvent which contains OH but does not contain -O-, -CO- nor -COO- in its molecule), an aromatic hydrocarbon solvent, an amide solvent, and dimethyl sulfoxide.

[0303] Examples of the ester solvent include methyl lactate, ethyl lactate, n-butyllactate, methyl 2-hydroxy isobutanoate, ethyl acetate, n-butylacetate, isobutylacetate, n-pentyl formate, isopentyl acetate, n-butylpropionate, isopropyl butyrate, ethyl butyrate, n-butylbutyrate, methyl pyruvate, ethyl pyruvate, propyl pyruvate, methyl acetoacetate, ethyl acetoacetate, cyclohexanol acetate, and γ -butyrolactone.

[0304] Examples of the ether solvent include ethylene glycol monomethyl ether, ethylene glycol monoethyl ether, ethylene glycol monopropyl ether, ethylene glycol monobutyl ether, diethylene glycol monomethyl ether, diethylene glycol monomethyl ether, propylene glycol monomethyl ether, propylene glycol monomethyl ether, propylene glycol monopropyl ether, propylene glycol monobutyl ether, 3-methoxy-1-butanol, 3-methoxy-3-methylbutanol, tetrahydrofuran, tetrahydropyran, 1,4-dioxane, diethylene glycol dimethyl ether, diethylene glycol diethyl ether, diethylene glycol methyl ether, diethylene glycol dipropyl ether, diethylene glycol dibutyl ether, anisole, phenetol, and methyl anisole.

[0305] Examples of the ether ester solvent include methyl methoxyacetate, ethyl methoxyacetate, butyl methoxyacetate, methyl ethoxyacetate, ethyl ethoxyacetate, methyl 3-methoxypropionate, ethyl 3-methoxypropionate, methyl 3-methoxypropionate, ethyl 3-ethoxypropionate, methyl 2-methoxypropionate, ethyl 2-methoxypropionate, propyl 2-methoxypropionate, methyl 2-ethoxypropionate, ethyl 2-methoxy-2-methylpropionate, ethyl 2-ethoxy-2-methylpropionate, ethyl 2-ethoxy-2-methylpropionate, 3-butyl methoxyacetate, 3-methyl-3-butyl methoxyacetate, propylene glycol monomethyl ether acetate, propylene glycol monomethyl ether acetate, ethylene glycol monomethyl ether acetate, diethylene glycol monoethyl ether acetate, and diethylene glycol monobutyl ether acetate.

[0306] Examples of the ketone solvent include 4-hydroxy-4-methyl-2-pentanone, acetone, 2-butanone, 2-heptanone, 3-heptanone, 4-heptanone, 4-methyl-2-pentanone, cyclopentanone, and isophorone.

[0307] Examples of the alcohol solvent include methanol, ethanol, propanol, butanol, hexanol, cyclohexanol, ethylene glycol, propylene glycol, and glycerin.

[0308] Examples of the aromatic hydrocarbon solvent include benzene, toluene, xylene, and mesitylene.

[0309] Examples of the amide solvent include N,N-dimethylformamide, N,N-dimethylacetamide, and N-methylpyrrolidone.

[0310] The solvent (F) is preferably propylene glycol monomethyl ether acetate, ethyl lactate, propylene glycol monomethyl ether, ethyl 3-ethoxypropionate, ethylene glycol monomethyl ether, diethylene glycol monomethyl ether, diethylene glycol monomethyl ether, 4-hydroxy-4-methyl-2-pentanone, or toluene, or a mixture of two or more of these is preferred.

[0311] The solvent (F) is a component other than the solid content, and for example, the solvent contained in the

quantum dots (A), the resin (B), or the like is also included in the solvent (F).

[0312] The content ratio of the solvent (F) in the curable resin composition is the ratio of the total mass of all the solvents contained in the composition to the total amount of the composition, and is, for example, based on the total amount of the curable resin composition, 40% by mass or more and 95% by mass or less, preferably 55% by mass or more and 90% by mass or less. In other words, the solid content of the curable resin composition is preferably 5% by mass or more and 60% by mass or less, and more preferably 10% by mass or more and 45% by mass or less. When the content ratio of the solvent (F) is within the above range, there is a tendency that the flatness of the composition layer at the time of coating becomes better and a cured film having an appropriate film thickness is easily formed.

[9] Leveling agent (G)

[0313] The curable resin composition may further contain a leveling agent (G).

[0314] Examples of the leveling agent (G) include a silicone-based surfactant, a fluorine-based surfactant, and a silicone-based surfactant having a fluorine atom. These may have a polymerizable group at its side chain. The leveling agent (G) is preferably a fluorine-based surfactant from the viewpoint of developability and emission intensity of the cured film.

[0315] Examples of the silicone-based surfactant include a surfactant having a siloxane bond in its molecule. Specific examples thereof include Toray Silicone DC3PA, Toray Silicone SH7PA, Toray Silicone DC11PA, Toray Silicone SH21PA, Toray Silicone SH28PA, Toray Silicone SH29PA, Toray Silicone SH30PA, and Toray Silicone SH8400 (manufactured by Dow Corning Toray Co., Ltd.); KP321, KP322, KP323, KP324, KP326, KP340, and KP341 (manufactured by Shin-Etsu Chemical Co., Ltd.); and TSF400, TSF401, TSF410, TSF4300, TSF4440, TSF4445, TSF4446, TSF4452, and TSF4460 (manufactured by Momentive Performance Materials Inc.).

[0316] Examples of the fluorine-based surfactant include a surfactant having a fluorocarbon chain in its molecule. Specific examples thereof include Fluorad (R) FC430 and Fluorad FC431 (manufactured by Sumitomo 3M Limited); MEGAFACE (R) F142D, MEGAFACE F171, MEGAFACE F172, MEGAFACE F173, MEGAFACE F177, MEGAFACE F183, MEGAFACE F554, MEGAFACE F575, MEGAFACE R30, and MEGAFACE RS-718-K (manufactured by DIC Corporation); EFTOP (R) EF301, EFTOP EF303, EFTOP EF351, and EFTOP EF352 (manufactured by Mitsubishi Materials Electronic Chemicals Co., Ltd.); Surflon (R) S381, Surflon S382, Surflon SC101, and Surflon SC105 (manufactured by Asahi Glass Co., Ltd.); and E5844 (manufactured by Daikin Fine Chemical Laboratory).

[0317] Examples of the silicone-based surfactant having a fluorine atom include a surfactant having a siloxane bond and a fluorocarbon chain in its molecule. Specific examples thereof include MEGAFACE (R) R08, MEGAFACE BL20, MEGAFACE F475, MEGAFACE F477, and MEGAFACE F443 (manufactured by DIC Corporation).

[0318] When the curable resin composition contains a leveling agent (G), the content ratio of the leveling agent (G) in the curable resin composition is, for example, based on the total amount of the curable resin composition, 0.001% by mass or more and 1.0% by mass or less, preferably 0.005% by mass or more and 0.75% by mass or less, more preferably 0.01% by mass or more and 0.5% by mass or less, and still more preferably 0.05% by mass or more and 0.5% by mass or less. When the content ratio of the leveling agent (G) is within the above range, the flatness of the cured film can be further improved.

40 [10] Light scattering agent (H)

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[0319] The curable resin composition may further contain a light scattering agent (H). Examples of the light scattering agent (H) include particles of metal or metal oxide, and glass particles.

[0320] Examples of the metal oxide include TiO_2 , SiO_2 , $BaTiO_3$, and ZnO. The particle size of the light scattering agent (H) is, for example, 0.03 μ m or more and 20 μ m or less, preferably 0.05 μ m or more and 1 μ m or less, and more preferably 0.05 μ m or more and 0.5 μ m or less.

[0321] As the light scattering agent (H), a light scattering agent previously dispersed in a part or the whole of the solvent (F) using a dispersant may be used. Commercially available products may also be used as the dispersant.

[0322] Examples of the commercially available products include:

50 Examples of such commercial products include:

DISPERBYK-101, 102, 103, 106, 107, 108, 109, 110, 111, 116, 118, 130, 140, 154, 161, 162, 163, 164, 165, 166, 170, 171, 174, 180, 181, 182, 183, 184, 185, 190, 192, 2000, 2001, 2020, 2025, 2050, 2070, 2095, 2150, 2155; ANTI-TERRA-U, U100, 203, 204, 250, ; BYK-P104, P104S, P105, 220S, 6919; BYK-LPN6919, and 21116; LAC-TIMON and LACTIMON-WS; Bykumen; and the like, manufactured by BYK Japan KK; SOLSPERSE-3000, 9000, 13000, 13240, 13650, 13940, 16000, 17000, 18000, 20000, 21000, 24000, 26000, 27000, 28000, 31845, 32000, 32500, 32550, 33500, 32600, 34750, 35100, 36600, 38500, 41000, 41090, 53095, 55000, 76500, and the like, manufactured by Lubrizol Japan Limited;

EFKA-46, 47, 48, 452, 4008, 4009, 4010, 4015, 4020, 4047, 4050, 4055, 4060, 4080, 4400, 4401, 4402, 4403, 4406, 4408, 4300, 4310, 4320, 4330, 4340, 450, 451, 453, 4540, 4550, 4560, 4800, 5010, 5065, 5066, 5070, 7500, 7554, 1101, 120, 150, 1501, 1502, 1503, and the like, manufactured by BASF Japan Ltd.; and AJISPER PA111, PB711, PB821, PB822, and PB824, manufactured by Ajinomoto Fine-Techno Co., Inc.

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[0323] The content ratio of the light scattering agent (H) in the curable resin composition is, for example, based on the total solid content of the curable resin composition, 0.001% by mass or more and 50% by mass or less, and from the viewpoint of developability and emission intensity of the cured film, preferably 1% by mass or more and 30% by mass or less, and more preferably 2% by mass or more and 10% by mass or less.

[0324] If necessary, the curable resin composition may further contain an additive known in the art, such as a polymerization inhibitor, a filler, other polymeric compound, an adhesion promoter, a light stabilizer, or a chain transfer agent.

[11] Water content

[0325] The water content based on the total mass of the curable resin composition is less than 5000 ppm, preferably 3000 ppm or less, and more preferably 2500 ppm or less. In the present invention, when the water content based on the total mass of the curable resin composition is set to less than 5000 ppm, a decrease in the emission intensity of the obtained cured film can be suppressed. Further, in the present invention, when the water content based on the total mass of the curable resin composition is set to less than 5000 ppm, a cured film having high emission intensity can be formed. The water content of the curable resin composition is a value measured according to the method described in Examples below.

[0326] The present inventors have found that the antioxidant (E) may be hydrolyzed. When the antioxidant (E) is hydrolyzed, foreign substances are precipitated, and the emission intensity of the obtained cured film may be decreased. According to the present invention, it is presumed that when the water content based on the total mass of the curable resin composition is set to less than 5000 ppm, the hydrolysis of the antioxidant (E) can be suppressed, and thus a decrease in the emission intensity of the obtained cured film can be suppressed. The presence or absence and degree of hydrolysis of the antioxidant (E) can be measured by, for example, liquid chromatography mass spectrometry (LC/MS). [0327] The water content based on the total mass of the curable resin composition is 50 ppm or more. Within this range, when organic ligand-containing semiconductor particles are used as the quantum dots (A), the dispersibility of the quantum dots (A) can be improved, and this can contribute to improvement in color unevenness and improvement in emission characteristics of the cured film.

[0328] In the curable resin composition, the water content based on the mass of the antioxidant (E) is preferably 60 ppm or less, more preferably 50 ppm or less, and still more preferably 30 ppm or less. Within this range, the hydrolysis of the antioxidant (E) can be suppressed. Further, in the curable resin composition, the water content based on the mass of the antioxidant (E) is preferably 1 ppm or more. The water content based on the mass of the antioxidant (E) can be calculated by calculation based on the measured value of the water content based on the total mass of the curable resin composition.

[0329] The origin of the moisture measured as the water content of the curable resin composition is not limited, but includes moisture previously contained in the quantum dots (A), the resin (B), the photopolymerizable compound (C), the photopolymerization initiator (D), the antioxidant (E), the solvent (F), or the like, moisture added in advance, moisture included in the preparation or after the preparation of the curable resin composition, moisture added, and the like. The water content of the curable resin composition can be adjusted by adding moisture to the components before mixing or the curable resin composition and/or by reducing the moisture by dehydration using a dehydrating agent or the like. The dehydrating agent is not limited as long as it has a dehydrating action in each component or in the curable resin composition, and examples thereof include molecular sieves and magnesium sulfate.

<Method for producing curable resin composition>

[0330] The method for producing the curable resin composition may include a step of mixing the quantum dots (A), the resin (B), the photopolymerizable compound (C), the photopolymerization initiator (D), the antioxidant (E), the solvent (F), and other components used as necessary. The method for producing the curable resin composition may further include a step of preparing the resin (B).

<Cured film>

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[0331] The cured film according to the present invention (hereinafter, also simply referred to as "cured film") is obtained by applying the curable resin composition according to the present invention to a substrate and further curing by the action of light and, if necessary, further by the action of heat.

[0332] The cured film contains the quantum dots (A) and the resin (B) contained in the curable resin composition. Further, the cured film contains a cured reaction product of the photopolymerizable compound (C) and the photopolymerization initiator (D) contained in the curable resin composition. The cured reaction product is a substance containing a structure derived from the structure of the photopolymerizable compound (C) and the photopolymerization initiator (D). The structure derived from the structure of the photopolymerizable compound (C) or the photopolymerization initiator (D) is, for example, a skeleton structure or a portion thereof other than the curing reaction site of the photopolymerizable compound (C) or the photopolymerization initiator (D).

[0333] The cured film may be formed on the entire surface of the substrate or may be formed on a part of the substrate (that is, a cured pattern). The cured pattern is one embodiment of the cured film, and refers to a cured film formed in a pattern.

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[0334] Examples of the method for forming a cured film on a part of the substrate include a photolithography method, an inkjet method, and a printing method. Of these, the photolithography method is preferable. The photolithography method is a method in which a curable resin composition is applied to a substrate, dried as necessary to form a composition layer, and the composition layer is exposed to light through a photomask and developed.

[0335] Examples of a method for forming a cured film on the entire surface of a substrate include a method in which a curable resin composition is applied to a substrate, dried as necessary to form a composition layer, and the composition layer is heated and/or the entire surface of the composition layer is exposed to light.

[0336] Examples of the substrate include a glass plate made from, for example, quartz glass, borosilicate glass, alumina silicate glass, soda lime glass of which the surface is coated with silica, or the like; a resin plate made from, for example, polycarbonate, polymethyl methacrylate, polyethylene terephthalate or the like; a substrate made from silicon; and a substrate produced by forming a thin film made from aluminum, silver or a silver/copper/palladium alloy or the like on a substrate.

[0337] The formation of a cured pattern using a photolithography method can be carried out using a known or conventional device or under known or conventional conditions. For example, it can be produced as follows.

[0338] First, the curable resin composition is applied onto a substrate, and then dried by heat-drying (prebaking) and/or drying under reduced pressure to remove volatile components including a solvent from the composition, thereby producing a composition layer. Examples of the coating method include a spin coating method, a slit coating method, and a slit and spin coating method.

[0339] The temperature to be employed in the case where heat-drying is carried out is preferably 30°C or more and 120°C or less, more preferably 50°C or more and 110°C.

The heating time is preferably 10 seconds or more and 60 minutes or less, and more preferably 30 seconds or more and 30 minutes or less.

[0340] In the case where drying under reduced pressure is carried out, it is preferred to carry out the drying procedure under a pressure of 50 Pa or more and 150 Pa or less and at a temperature of 20°C or more and 25°C or less.

[0341] The film thickness of the composition layer is not particularly limited, and may be selected appropriately depending on the desired film thickness of the cured pattern to be produced, and is, for example, 1 μ m or more and 20 μ m or less, preferably 1.5 μ m or more and 18 μ m or less, more preferably 2 μ m or more and 14 μ m or less, and still more preferably 2 μ m or more and 12 μ m or less.

[0342] Next, the composition layer is exposed to light through a photomask for forming a desired cured pattern. The pattern on the photomask is not particularly limited.

[0343] The light sources used for exposure are preferably a light source that emits light having a wavelength of 250 nm or more to 450 nm or less. For example, light in the vicinity of 436 nm, 408 nm, or 365 nm may be selectively extracted from the light having the wavelength according to the absorption wavelength of the photopolymerization initiator (D) by a band-pass filter. Specific examples of the light source include a mercury lamp, a light-emitting diode, a metal halide lamp, and a halogen lamp.

[0344] A light-exposing device such as a mask aligner and a stepper is preferably used because the device is capable of emitting a parallel light beam uniformly over the whole area of the exposed surface or aligning the photomask accurately to the substrate which has the composition layer formed thereon. The exposed composition layer is cured by polymerizing the photopolymerizable compound (C) and the like contained in the composition layer.

[0345] By bringing the exposed composition layer into contact with a developing solution for development, the unexposed portion of the composition layer is dissolved and removed in the developer to obtain a cured pattern. Examples of the developer include an aqueous solution of an alkaline compound such as potassium hydroxide, sodium hydrogen carbonate, sodium carbonate and tetramethylammonium hydroxide; and organic solvents. The concentration of the alkaline compound in the aqueous solution is preferably 0.01% by mass or more and 10% by mass or less, and more preferably 0.03% by mass or more and 5% by mass or less. Examples of the organic solvent include the same as the solvent (F). The developer may contain a surfactant.

[0346] The developing method may be any of a paddle method, a dipping method, a spray method and the like. Further, the substrate may be inclined at any degree during development.

[0347] The cured pattern obtained by development is preferably further subjected to heating (post-baking). The heating temperature is carried out is preferably 150°C or more and 250°C or less, more preferably 160°C or more and 235°C. The heating time is preferably 1 minute or more and 120 minutes or less, and more preferably 10 minutes or more and 60 minutes or less. By performing heating after development, polymerization of the unreacted photopolymerizable compound (C) and the like contained in the cured pattern can be advanced, and therefore, a cured pattern having more excellent chemical resistance can be obtained. Even in the case where development is not performed, it is preferable to further perform heating (post-baking) on the exposed composition layer.

[0348] By irradiating the cured film with ultraviolet light or visible light, the cured film can emit light having a wavelength different from that of the irradiation light. The wavelength of light to be emitted can be selected by selecting the components and the particle size of the quantum dots (A) in the curable resin composition used for forming the cured film.

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color filter can be used.

[0349] Since the cured film has a function of converting the wavelength of irradiation light as described above, it can be used as a color conversion layer of a display device.

[0350] The film thickness of the cured film is 1 μ m or more and 20 μ m or less, preferably 1.5 μ m or more and 18 μ m or less, more preferably 2 μ m or more and 14 μ m or less, and still more preferably 2 μ m or more and 12 μ m or less.

[0351] In one preferred embodiment, the cured film or cured pattern is formed from the curable resin composition containing the quantum dots (A), the resin (B), the polymerizable compound (C), the photopolymerization initiator (D), and an antioxidant (E), and other components used as necessary, and has a brightness of 200 cd/m² or greater, and even after storing the curable resin composition for one month, the brightness change rate of the resulting cured film or cured pattern is preferably within \pm 7.0%, and more preferably within \pm 5.0%.

[0352] The curable resin composition according to the present invention contains an antioxidant and can suppress a decrease in brightness in the state of the curable resin composition or in the state after the cured film is formed, and thus is useful as a color conversion layer of a display device, in particular, a liquid crystal display device, an organic EL display device, or an inorganic EL display device. Examples of such a display device include display devices disclosed in Japanese Patent Laid-Open No. 2006-309219, Japanese Patent Laid-Open No. 2006-310303, Japanese Patent Laid-Open No. 2013-15812, Japanese Patent Laid-Open No. 2009-251129, and Japanese Patent Laid-Open No. 2014-2363. [0353] The display device includes at least a light source and the cured film. The display device is not particularly limited, and may further include layers such as a light absorption layer, a light reflection member (such as a reflection film), a diffusion film, a brightness enhancement unit, a prism sheet, a light guide plate, and a medium material layer

between elements.

[0354] The light absorption layer is a layer having wavelength selectivity that transmits light in a specific wavelength range and absorbs light in other wavelength ranges. The light absorption layer is usually a layer containing a colorant such as a dye or a pigment, and can be arranged on the cured film. As the light absorption layer, a conventionally known

[0355] The light reflection member is a member for reflecting the light of the light source toward the cured film, and may be a reflecting mirror, a film of reflective particles, a reflective metal film, a reflective body, or the like. The diffusion film is a film for diffusing the light of a light source or the light emitted from a cured film, and may be an amplified diffusing film or the like. The brightness enhancement unit is a member for reflecting a part of light back in the direction in which the light is transmitted.

[0356] The prism sheet typically has a base material section and a prism section. The base material section may be omitted, depending on the adjacent members. The prism sheet can be pasted together with the adjacent members via any appropriate adhesive layer (for example, an adhesive layer or a pressure sensitive adhesive layer). The prism sheet is configured to have a plurality of unit prisms that are convex in parallel on the side opposite to the visible side (rear side). By arranging the convex sections of the prism sheet toward the rear side, the light that passes through the prism sheet is more likely to be focused. Also, when the convex sections of the prism sheet are arranged toward the rear side, there is less light reflected without incident on the prism sheet compared to the case where the convex sections are arranged toward the visible side, and a display device with high brightness can be obtained.

[0357] Any appropriate light guide plate may be used as the light guide plate. For example, a light guide plate having a lens pattern formed on the rear side such that light from the lateral direction can be deflected in the thickness direction or a light guide plate having a prism shape or the like formed on the rear side and/or the visible side can be used.

[0358] The display device may include a layer composed of one or more medium materials on the optical path between adjacent elements (layers). Examples of the one or more medium materials include, but not limited to, vacuum, air, gas, optical materials, adhesives, optical adhesives, glass, polymers, solids, liquids, gels, cured materials, optical bonding materials, refractive index matching or refractive index mismatching materials, refractive index gradient materials, cladding or anti-cladding materials, spacers, silica gel, brightness enhancement materials, scattering or diffusion materials, reflective or anti-reflective materials, wavelength selective materials, or other suitable media known in the art.

[0359] Specific examples of the display device include those provided with a wavelength conversion material for an EL display or a liquid crystal display. Specific examples thereof include a display device in which the cured film as a

wavelength conversion layer is disposed between a blue light source (A) and a light guide plate along an end surface (side surface) of the light guide plate to form a backlight (on-edge type backlight) that emits white light, and a light absorption layer is arranged on the light guide plate side; a display device in which the cured film as a wavelength conversion layer is disposed on a light guide plate to form a backlight (surface mounting type backlight) that emits light emitted from a blue light source disposed on an end surface (side surface) of the light guide plate to the wavelength conversion layer through the light guide plate as white light, and a light absorption layer is disposed on the wavelength conversion layer; and a display device in which the cured film is disposed in the vicinity of a light-emitting portion of a blue light source to form a wavelength conversion layer, a backlight (on-chip backlight) that emits irradiated light as white light, and a light absorption layer is disposed on the wavelength conversion layer.

Examples

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[0360] Hereinafter, the present invention will be described in more detail by way of examples. Unless otherwise specified, "%" and "part" in the example are % by mass and parts by mass.

[Measurement of emission intensity]

[0361] Using curable resin compositions immediately after production or immediately after adjustment of the water content by the dehydrating agent (hereinafter also referred to as "composition before storage") and curable resin compositions after storing the curable resin composition immediately after production or immediately after adjustment of the water content by the dehydrating agent after in a brown glass container at 25°C and a humidity of 30 to 40% RH for one month (hereinafter also referred to as "composition after storage"), cured films were prepared and emission intensity was measured by the following method.

[0362] Onto a 5-cm square glass substrate (Eagle 2000; manufactured by Corning Incorporated), a curable resin composition (composition before storage or composition after storage) was applied by a spin coat method so that the film thickness set to 10 μ m, and then prebaked at 100°C for 3 minutes, to form a curable composition layer. This curable composition layer was subjected to light irradiation in an exposure amount (basis: 365 nm) of 80 mJ/cm² under an air atmosphere using an exposure device (TME-150RSK; manufactured by TOPCON CORPORATION) and after development, post-baking was performed 180°C for 60 minutes, to obtain a substrate having a cured film.

[0363] A narrow directional angle type sensing backlight illumination (OPF series; manufactured by OPTEX FA CO., LTD.) equipped with an LED lamp having an emission wavelength of 444 nm and a scratch resistant cover was prepared as a backlight. A backlight was placed with the scratch resistant cover facing upward, and an optical fiber for detecting light emission connected to the following spectrometer was installed at a position 4 cm high from the surface of the scratch resistant cover. A glass substrate (Eagle 2000; manufactured by Corning Incorporated) as a reference was placed on the surface of the scratch resistant cover of the backlight. In this state, the backlight was turned on, and the light amount of the backlight was adjusted so that the total radiant flux (μ W) of the backlight was 5000 μ W through a reference glass substrate (Eagle 2000; manufactured by Corning Incorporated).

[0364] Next, the substrate having the cured film prepared above was disposed on the surface of the glass substrate. In this state, the backlight was turned on, and the emission intensity (brightness, unit: μ W) of the light emitted from the cured film was measured as an integrated radiant flux in a range of wavelengths of 485 nm or more and 780 nm or less. The emission intensity was measured using a spectrum analyzer (Spectrum meter, manufactured by Ocean Optics, Inc). [0365] The emission intensity L0 of the cured film produced using the composition before storage and the emission intensity L of the cured film produced using the rate storage were measured, and the change rate R [%] of the emission intensity was calculated by the following formula.

 $R (\%) = \{(L - L0) / L0\} \times 100$

[Measurement of water content]

[0366] Two grams of the composition before storage was weighed and the water content was measured based on the total mass using a moisture measuring device (manufactured by Mitsubishi Chemical Analytech Co., Ltd., CA-200) under the following conditions. The water content based on the mass of the antioxidant can be calculated by calculation based on the measured value of the water content based on the total mass and the amount of antioxidant added.

Measurement method: vaporization heating coulometry

Anode solution: AQUAMICRON AX Cathode solution: AQUAMICRON CXU

Measurement temperature: 130°C

[Weight-average molecular weight]

⁵ [0367] The weight-average molecular weight (Mw) of the resin was measured by the GPC method under the following conditions.

Equipment: K2479 (manufactured by Shimadzu Corporation)

Column: SHIMADZU Shim-pack GPC-80M

Column temperature: 40°C Solvent: tetrahydrofuran Flow rate: 1.0 mL/min

Detector: RI

Standard for calibration; TSK STANDARD POLYSTYRENE F-40, F-4, F-288, A-2500, A-500 (manufactured by

15 Tosoh Corporation)

[Solid content]

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[0368] About 1 g of the resin (B-1) solution was weighed in an aluminum cup, dried at 180°C for 1 hour, and then the mass was measured. The solid content (% by mass) of the resin (B-1) solution was calculated from the mass reduction amount.

(Production Example 1: Preparation of dispersion of quantum dots (A-1))

[0369] A toluene dispersion of InP/ZnSeS quantum dots coordinated with oleic acid as a ligand was prepared.

[0370] The dispersion was distilled under reduced pressure to remove toluene. To 30 parts of the solid content, 70 parts of cyclohexyl acetate was added to obtain a dispersion (solid content: 30%) of quantum dots (a-1).

(Production Example 2: Preparation of resin (B-1) solution)

[0371] After 110 parts of propylene glycol monomethyl ether acetate (hereinafter referred to as "PGMEA") were put into a flask equipped with a stirrer, a reflux cooling tube with a thermometer, a dropping funnel, and a nitrogen introduction tube, the mixture was stirred while replacing nitrogen, and the temperature was raised to 80°C. A solution obtained by dissolving 25 parts of dicyclopentanyl methacrylate, 26 parts of methyl methacrylate, 16 parts of methacrylic acid, and 11 parts of 2,2'-azobis(2,4-dimethylvaleronitrile) in 110 parts by mass of PGMEA was dropped from a dropping funnel into a flask, and stirred at 80°C for 3 hours.

[0372] Next, 16 parts of glycidyl methacrylate, 0.4 parts of 2,2'-methylenebis(4-methyl-6 tert-butylphenol), and 0.8 parts of triphenylphosphine were put into a flask, and the temperature was raised to 110°C and stirred for 8 hours to react the carboxylic acid in the polymer with the epoxy group to introduce a polymerizable unsaturated bond. Then, 17 parts of 1,2,3,6-tetrahydrophthalic acid anhydride was added and the reaction was continued for 3 hours to introduce carboxylic acid groups into the side chains. The reaction solution was cooled to room temperature to obtain a resin (B-1) solution.

[0373] The resin (B-1) had a weight-average molecular weight of 5200 and an acid value of 100 mg KOH/g, and the solid content in the resin (B-1) solution was 40% by mass.

(Production Example 3: Preparation of dispersion of light scattering agent (H-1))

[0374] DISPERBYK21116 (manufactured by BYK Japan KK) was added to 70 parts of titanium oxide nanoparticles in a solid content of 3 parts and propylene glycol monomethyl ether acetate (hereinafter referred to as "PGMEA") in a total amount of 100 parts, and the mixture was stirred with a paint shaker until it was sufficiently dispersed to obtain a dispersion (solid content: 73%) of a light scattering agent (H-1).

<Examples 1 to 5, Comparative Examples 1 and 2>

[0375] A curable resin composition was prepared by mixing quantum dots (A-1) dispersion obtained in Production Example 1, a resin (B-1) solution obtained in Production Example 2, and other components shown in Table 1 in a predetermined amount. The content of each component in the curable resin composition obtained from the addition amount is as shown in Table 1. In Table 1, the contents of components other than the solvent (F) are in terms of solid

content (unit: parts by mass). The unit of the content of the solvent (F) is parts by mass. For example, the quantum dots (A-1) are blended as a dispersion of the quantum dots (A-1) in the preparation of the curable resin composition, and the content shown in Table 1 is the amount of the quantum dots (A-1) itself contained in the solution.

[0376] The solvent (F) in Table 1 contains a solvent contained in the dispersion or solution used for preparing the curable resin composition.

[0377] In Examples 1, 2, 4, 5, and Comparative Example 1, the water content of the obtained curable resin composition was adjusted by adding water to PGMEA (propylene glycol monomethyl ether acetate), which is the solvent (F-1), and in Example 3 and Comparative Example 2, the water content of the curable resin composition was adjusted by adding a dehydrating agent (molecular sieve) to the obtained curable resin composition. The water content of the curable resin compositions thus adjusted was measured as described above, and the values are listed in Table 1.

[0378] The content of oleic acid (the organic ligand of the quantum dot) in the curable resin composition shown in Table 1 was calculated based on the measurement of the concentration of oleic acid in the solid content contained in the dispersion of the quantum dots (A-1) obtained in Production Example 1 according to the method [a] below.

15 [a] Measurement of oleic acid concentration

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[0379] After the solvent was removed by vacuum-drying the dispersion of the quantum dots (A-1) at 150°C, the weight change of the remaining solid content was measured at a temperature increasing rate of 5°C/min from 50°C to 550°C using a thermogravimetric analyzer "TGDTA 6200". The weight changed from 50°C to 500°C was taken as the weight of oleic acid. From this, the concentration of oleic acid in the solid content was calculated to be 28% by mass.

[0380] The obtained curable resin composition was measured and evaluated for emission intensities L0 and L as described above, and the change rate R [%] of the emission intensities was calculated based on the emission intensities L0 and L. The results are also shown in Table 1.

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5		Comparative Example 2	25.2	8.6		38.7	8.9	11.9	1.8	3.5		0.1	186	22	0.3	1158	-13.3
10 15		Comparative Example 1	25.2	8.6		38.7	8.9	11.9	1.8	3.5		0.1	186	2000	61.2	1166	-7.1
20		Example 5 C	21.6	8.4	5.0	38.7	8.9	11.9	1.8	3.5		0.1	186	1000	12.3	1487	0.1
25		Example 4	25.2	9.8		38.7	8.9	11.9	1.8		3.5	0.1	186	1000	12.3	996	-0.5
30	[Table 1]	Example 3	25.2	9.8		38.7	8.9	11.9	1.8	3.5		0.1	186	100	1.2	1190	-0.1
35		Example 2	25.2	9.8		38.7	8.9	11.9	1.8	3.5		0.1	186	1000	12.3	1182	0.1
40		Example 1	25.2	8.6		38.7	8.9	11.9	1.8	3.5		0.1	186	2000	24.5	1174	-3.7
45			A-1		H-1	B-1	C-1	C-2	D-1	E-1	E-2	G-1	F-1	l mass) [ppm]	ss of antioxidant)		
50			ots (A)	and	Light scattering agent (H)		Polymerizable compound		Polymerizatio n initiator (D)	(E)	(E)	lent (G)		Water content (based on total mass) [ppm]	Water content (based on mass of antioxidant) [ppm]	Emission intensity (L0) $[\mu \text{W}]$	e (R) [%]
55			Quantum dots (A)	Organic ligand	Light scatte	Resin (B)	Polymerizal	(C)	Polymerizat	tachiyoita A	Alilloxidalit (E)	Leveling agent (G)	Solvent (F)	Water conte	Water conte [ppm]	Emission in	Change rate (R) [%]

[0381] Photopolymerizable compound (C-1): M-510 (polybasic modified acrylate, manufactured by Toagosei Co., Ltd., solid content: 100%)

[0382] Photopolymerizable compound (C-2): A-9550 (dipentaerythritol polyacrylate, manufactured by SHIN-NAKA-MURA CHEMICAL Co., Ltd., solid content: 100%)

[0383] Photopolymerization Initiator (D-1): compound represented by the following formula. Production was performed by the method disclosed in Japanese Patent Laid-Open No. 2011-132215.

[Formula 22]

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Antioxidant (E-1): SUMILIZER-GP (phosphorus-based antioxidant, manufactured by Sumitomo Chemical Co., Ltd., solid content: 100%)

Antioxidant (E-2): AO-60 (phenol-based antioxidant, manufactured by ADEKA Corporation, solid content: 100%) Solvent (F-1): PGMEA (propylene glycol monomethyl ether acetate, solubility in water is 18.5% by mass at a temperature of 20°C)

Leveling agent (G-1): F-554 (fluorine-based leveling agent, manufactured by DIC Corporation, solid content: 100%)

25 <Hydrolysis of antioxidant (E-1)>

> [0384] The curable resin compositions of Examples 1 to 3 and Comparative Example 1 prepared by adding the antioxidant (E-1) were separated into a supernatant and a precipitate by a centrifuge, and the supernatant was diluted 10 times to prepare and analyzed by liquid chromatography mass spectrometry (LC/MS) under the following conditions.

Equipment: model 1200, manufactured by Agilent Technologies, Inc.

Column: L column ODS (3.0 mm $\phi \times$ 150 μ m, 3.0 μ m)

Liquid A: 10 mM AAA/H₂O Liquid B: 10 mM AAA/MTA

Liquid B%: 40% (50 minutes), 1000 (30 minutes), post time 15 minutes

Flow rate: 0.3 mL/min Temperature: 40°C

Detection wavelength: UV 280 nm

Injection volume: 3.0 µL

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[0385] As a result of the LC/MS analysis, in addition to the compound (E-1) which was the main component of the antioxidant (E-1) added as the antioxidant (E), the following compounds (e1), (e2), and (e3) were detected as the decomposition component (e) after hydrolysis of compound (E-1). In the decomposition component (e), the main component was the compound (e1), and the compound (e2) and the compound (e3) were trace components. In Examples 1 to 3, the proportion of the decomposition component (e) was lower than that in Comparative Example 1.

[Formula 23]

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55 (E-1) [Formula 24]

+(=) ОН +(=) ОР ОР

(e1)

[Formula 25]

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[Formula 26]

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(e3)

Claims

- 1. A curable resin composition comprising: quantum dots (A); a resin (B); a photopolymerizable compound (C); a photopolymerization initiator (D); an antioxidant (E); and a solvent (F), wherein a water content based on a total mass of the curable resin composition is 50 ppm or more and less than 5000 ppm.
- 2. The curable resin composition according to claim 1, wherein the antioxidant (E) contains a phosphorus-based antioxidant.
 - 3. The curable resin composition according to claim 1 or 2, wherein a water content, based on a mass of the antioxidant (E), of the curable resin composition is 60 ppm or less.
- **45 4.** The curable resin composition according to any one of claims 1 to 3, wherein the solvent (F) has a solubility in water of 0.01% by mass or more and 30% by mass or less at a temperature of 20°C.
 - 5. The curable resin composition according to any one of claims 1 to 4, further comprising an organic ligand.
- 50 **6.** The curable resin composition according to claim 5, wherein the organic ligand has a gas phase acidity of -15 kJ/mol or more and 5 kJ/mol or less.
 - 7. The curable resin composition according to claim 5 or 6, wherein the organic ligand is an organic compound having a polar group, and the polar group is at least one group selected from the group consisting of a thiol group, a carboxy group, and an amino group.
 - 8. The curable resin composition according to any one of claims 1 to 7, further comprising a light scattering agent (H).

	9.	A cured film formed from the curable resin composition according to any one of claims 1 to 8.
	10.	A display device comprising the cured film according to claim 9.
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INTERNATIONAL SEARCH REPORT International application No. PCT/JP2021/011418 A. CLASSIFICATION OF SUBJECT MATTER 5 Int. C1. C08F290/12(2006.01)i, C09K11/08(2006.01)i, C09K11/70(2006.01)i, B82Y20/00(2011.01)i, C08F2/44(2006.01)i, G02B5/20(2006.01)i FI: C08F2/44 Z, C08F290/12, C09K11/08 G, G02B5/20, B82Y20/00, C09K11/70 ZNM According to International Patent Classification (IPC) or to both national classification and IPC FIELDS SEARCHED 10 Minimum documentation searched (classification system followed by classification symbols) Int. C1. C08F290/12, C09K11/08, C09K11/70, B82Y20/00, C08F2/44, G02B5/20 Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched Published examined utility model applications of Japan Published unexamined utility model applications of Japan Registered utility model specifications of Japan Published registered utility model applications of Japan 15 1922-1996 1994-2021 Electronic data base consulted during the international search (name of data base and, where practicable, search terms used) 20 DOCUMENTS CONSIDERED TO BE RELEVANT Category* Citation of document, with indication, where appropriate, of the relevant passages Relevant to claim No. WO 2018/105545 A1 (DIC CORP.) 14 June 2018, entire 1 - 10Α 25 JP 2017-78120 A (FUJIFILM CORP.) 27 April 2017, Α 1-10 entire text JP 2010-9995 A (SEIKO EPSON CORP.) 14 January 30 Α 1 - 102010, entire text JP 2018-124413 A (DAINIPPON PRINTING CO., LTD.) 09 1-10 Α August 2018, entire text 35 See patent family annex. Further documents are listed in the continuation of Box C. 40 later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention Special categories of cited documents: document defining the general state of the art which is not considered to be of particular relevance "E" earlier application or patent but published on or after the international document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone filing date document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other "L" 45 document of particular relevance; the claimed invention cannot be special reason (as specified) considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art document referring to an oral disclosure, use, exhibition or other means document published prior to the international filing date but later than the priority date claimed document member of the same patent family Date of the actual completion of the international search Date of mailing of the international search report 50 27.04.2021 18.05.2021 Name and mailing address of the ISA/ Authorized officer Japan Patent Office 3-4-3, Kasumigaseki, Chiyoda-ku, Tokyo 100-8915, Japan Telephone No. 55

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